

27/4/2022

# ME8491 - Engineering Metallurgy

## Assignment - I

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2020 PECME 171

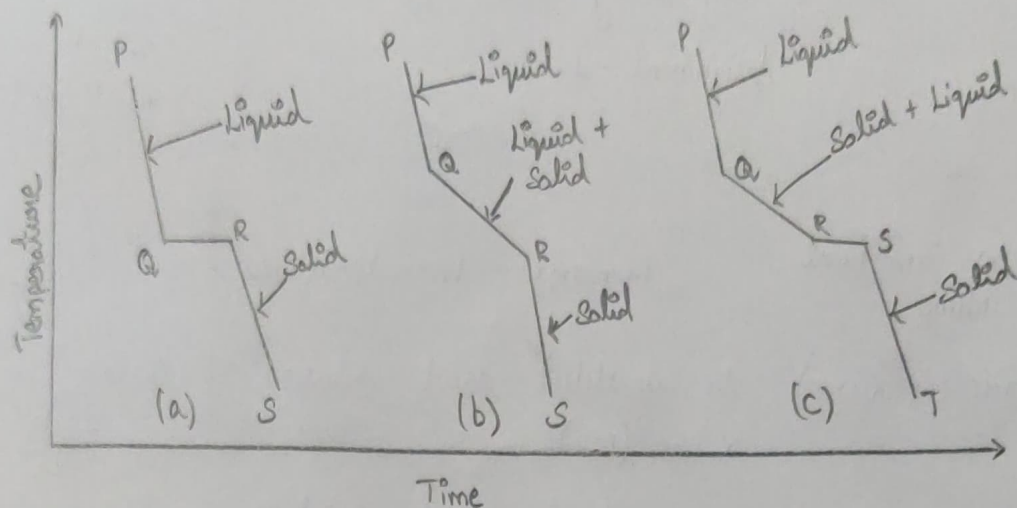
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11) a)

S.No	Micro-Constituent Name	General Characteristics
1.	Ferrite (or $\alpha$ iron)	An interstitial solid solution of Carbon in $\alpha$ iron (BCC).
2.	Austenite (or $\gamma$ iron)	An interstitial solid solution of Carbon in $\gamma$ iron (FCC).
3.	Cementite	A Compound of iron and Carbon ( $\text{Fe}_3\text{C}$ ).
4.	Pearlite	Eutectoid of ferrite and Cementite with a lamellar microstructure of alternate $\alpha$ iron and Cementite plates.
5.	Ledeburite	Eutectic mixture of austenite ( $\gamma$ iron) and Cementite ( $\text{Fe}_3\text{C}$ ).
6.	Martensite	An interstitial solid solution of Carbon in ferrite ( $\alpha$ iron).
7.	Troostite	A mixture of radial lamellae of ferrite and Cementite.
8.	Sorbite	A mixture of ferrite and finely divided Cementite.
9.	Bainite	Eutectoid of ferrite and Cementite. The ferrite has either a feathery appearance or occurs as plates. Carbide particles lie between the ferrite regions.

12(a)



a) Cooling Curve for pure metal or Compound:

- \* From P to Q, the curve proceeds at a uniform rate and at point Q, the first crystals began to form.

- \* As solidification proceeds, the latent heat of fusion is liberated in such amount that the temperature remains constant from Q to R until whole mass has entirely solidified. The period QR is known as the horizontal thermal arrest.

- \* Further cooling from point R will cause the temperature to drop along curve RS. The slopes of PQ and RS curves depends upon the specific heats of liquid and solid metals respectively.

- \* In the region PQ, only liquid phase is present while in the region RS only solid phase is present.

(b) Cooling Curves of a binary solid solution:

- \* Curve portion PQ is similar in character to that for pure metal.

- \* The slope of the cooling curves will change due to the evolution of latent heat of crystallisation.

- \* Beyond point R, there will be only solid phase and the temperature falls along line RS.

(c) Cooling Curve of a binary eutectic system:

- \* A binary eutectic system has two components that are completely soluble in the liquid state but entirely insoluble in the solid state.



\* The system is liquid along PQ up to point Q.

\* At point Q, the temperature drops along QR and crystallisation of the one component starts.

\* At point R, the liquid composition reaches such a level that two components crystallise simultaneously from the solution and the temperature remains constant until all the liquid solidifies.

\* Thus eutectic reaction can be defined as the transformation, during cooling, of a liquid phase isothermally and reversibly into two solid phases.

12) b) Peritectoid Reaction:

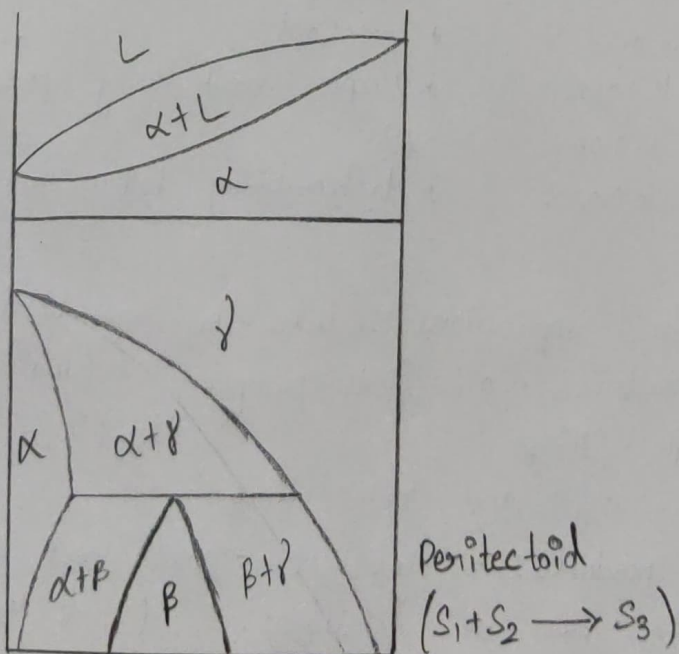
\* Peritectoid reaction is an isothermal reversible reaction in which two solid phases transform into a third solid phase, upon cooling.

$$\text{Solid 1} + \text{Solid 2} \xrightarrow{\text{cooling}} \text{Solid 3}$$

\* Solid 1 + Solid 2  $\xrightleftharpoons[\text{Heating}]{\text{Cooling}}$  Solid 3

\* The peritectoid is an 'upside-down' eutectoid.

\* Peritectoid systems are found in Ni-Zn, Fe-Nb, Cu-Sn, Ni-Mo, and many other systems.



## 11) b) (a) Malleable Cast Iron:

Composition:

Components	Percentage
Carbon	2.0 - 2.65 %
Silicon	0.9 - 1.4 %
Manganese	0.25 - 0.55 %
Phosphorus	< 0.18
Sulphur	0.05

Properties:

- \* Similar to ductile iron
- \* Good shock resistance
- \* Good ductility
- \* Good machinability.

Application:

- \* Malleable iron is better for thinner castings.
- \* Vehicle Components [Good Machinability]
  - Power trains, frames, suspensions and wheels.
  - Steering components, connecting rods.
- \* Railway components [Good tensile strength]
- \* Pipe fitting [High ductility].

## (b) Spheroidal Cast Iron:

Composition:

Iron Carbon	3.3 to 3.4 %
Silicon	2.2 to 2.8 %
Manganese	0.1 to 0.5 %
Magnesium	0.03 to 0.05 %
Phosphorus	0.005 to 0.04 %
Sulphur	0.005 to 0.02 %

Properties:

- \* Strength higher than grey cast iron
- \* Ductility up to 6% as cast or 20% annealed
- \* Low Cost
- \* Simple manufacturing process makes complex shapes.
- \* Machinability better than steel.

Applications:

- \* Major industrial applications include - highway diesel trucks, agricultural tractors, - oil well pumps. (High strength)
- \* Pipe and pipe fittings used for water and sewer lines.
- \* Machinery products: - Crankshafts, - front wheel spindle supports, - steering knuckles, - disc brake callipers. (Good Machinability).