

# Assignment for SSC Examinees 2022

Subject: Chemistry		1 <sup>st</sup> Paper		Subject Code: 176		Level: HSC		
Assignment Number	Assignment	Learning Outcomes	Guidelines (cues/steps or stages)	Assessment Criterion /Rubric				Com'ts
03  Chapter 2  Qualitative Chemistry	Determining the application of ionic product and solubility product of a solution	Students will be able to:  - explain solubility, solubility rules and solubility product of ionic compound.	Students will:  A. explain the affect of different factors on solubility; B. derive solubility product equation for various less soluble salts; C. calculate the solubility product of a saturated solution of $A_2B$ salt in water at $25^\circ C$ temperature in which the concentration of $B^{2-}$ ion is $6.4 \times 10^{-5} M$ . D. $2mL$ $0.25 M$ $NH_4OH$ solution is added in $10mL$ $0.30M$ $AB_3$ solution. If the solubility product of $A(OH)_3$ at $25^\circ C$ is $3.98 \times 10^{-38}$ then justify the probability of getting precipitation in the solution.	Assessment Criterion /Rubric				Com'ts
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Marks Obtained	Comments
14-16	Excellent
11-13	Very good
8-10	Good
7 or below	Needs Improvement