

EXPERIMENT - 7DETERMINATION OF PLANCK'S CONSTANT

Aim: Determination of Planck's Constant

APPARATUS: 0-1 V power supply, a one-way key, a rheostat, a digital milliammeter, a digital voltmeter, a 1K resistor and different known wavelength LED's (Light Emitting Diode).

THEORY: The significance of Planck's constant is that 'quanta' (small packets of energy) can be determined by frequency of radiation and Planck's constant. It describes the behaviour of particle waves at atomic level as well as the particle nature of light.

The light energy emitted during forward biasing is given as,

$$E = \frac{hc}{\lambda} \quad \text{--- (1)}$$

where

c = velocity of light

h = Planck's constant

λ = wavelength of light

If V is the forward voltage applied across the LED where it begins to emit light (the knee voltage) the energy given to electrons crossing the junction is

$$E = eV \quad (2)$$

equating (1) and (2) we get

$$eV = \frac{hc}{\lambda} \quad (3)$$

Knee voltage V can be measured for LED's with different values of λ (wavelength of light)

$$V = \frac{hc}{e} \left(\frac{1}{\lambda} \right) \quad (4)$$

Now from eq (4), slope of graph of V on the vertical axis vs $1/\lambda$ on the horizontal axis is

$$S = \frac{hc}{e}$$

To determine Planck's constant h ,

$$h = \frac{e}{c} S \quad (\text{True value } h = \frac{e}{c} = 5.33 \times 10^{-28} \text{ Js/m})$$

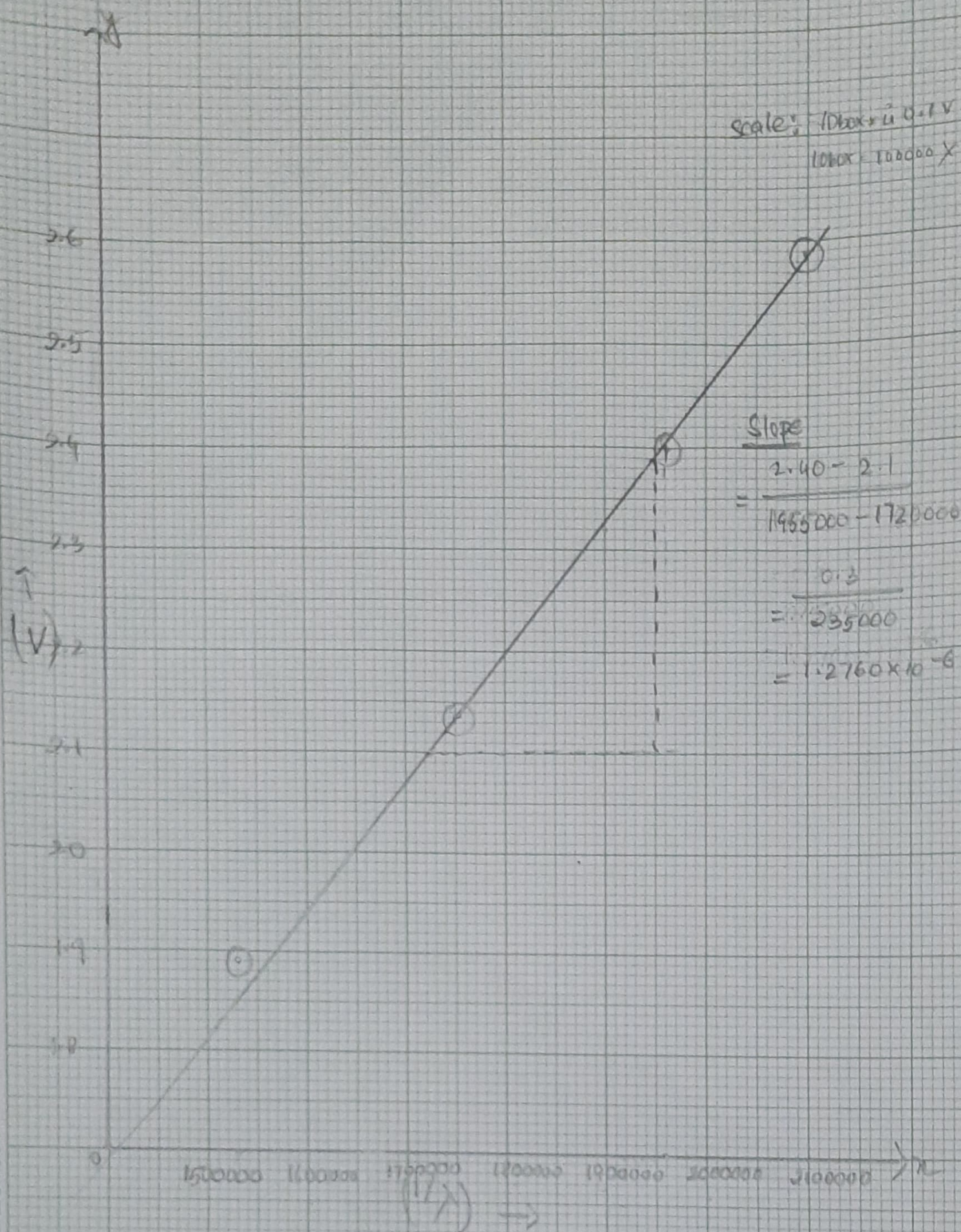
Alternatively, eq (3) can be written as

$$h = \frac{e}{c} \lambda V$$

OBSERVATIONS :

COLOUR OF LED	WAVELENGTH (nm $\times 10^{-9}$) (m)	VOLTAGE (V)	λV (Vnm)	$h = \frac{e \times V}{f}$ (Js) PLANCK'S CONSTANT
Red	650×10^{-9}	1.89	1228.5×10^{-9}	6.54×10^{-34}
Green	510×10^{-9}	2.40	1224×10^{-9}	6.52×10^{-34}
Yellow	570×10^{-9}	2.14	1219.8×10^{-9}	6.50×10^{-34}
Blue	475×10^{-9}	2.59	1230.2×10^{-9}	6.55×10^{-34}

$$\text{Mean} = 6.5275 \times 10^{-34} \text{ Js}$$



$\frac{1}{\lambda}$	V
1538461.638	1.89
1960784.314	2.40
1754385.965	2.14
2105263.158	2.59

CALCULATION :

$$\begin{aligned} \text{Slope} &= \frac{2.40 - 2.1}{195000 - 172000} \\ &= \frac{0.3}{23000} \\ &= 1.2760 \times 10^{-6} \end{aligned}$$

$$\begin{aligned} h &= \frac{e}{e} \times \text{slope} \\ &= 5.33 \times 10^{-28} \times 1.2760 \times 10^{-6} \\ &= 6.8010 \times 10^{-34} \text{ Js} \end{aligned}$$

\therefore Graphical Planck's Constant = $6.8010 \times 10^{-34} \text{ Js}$.

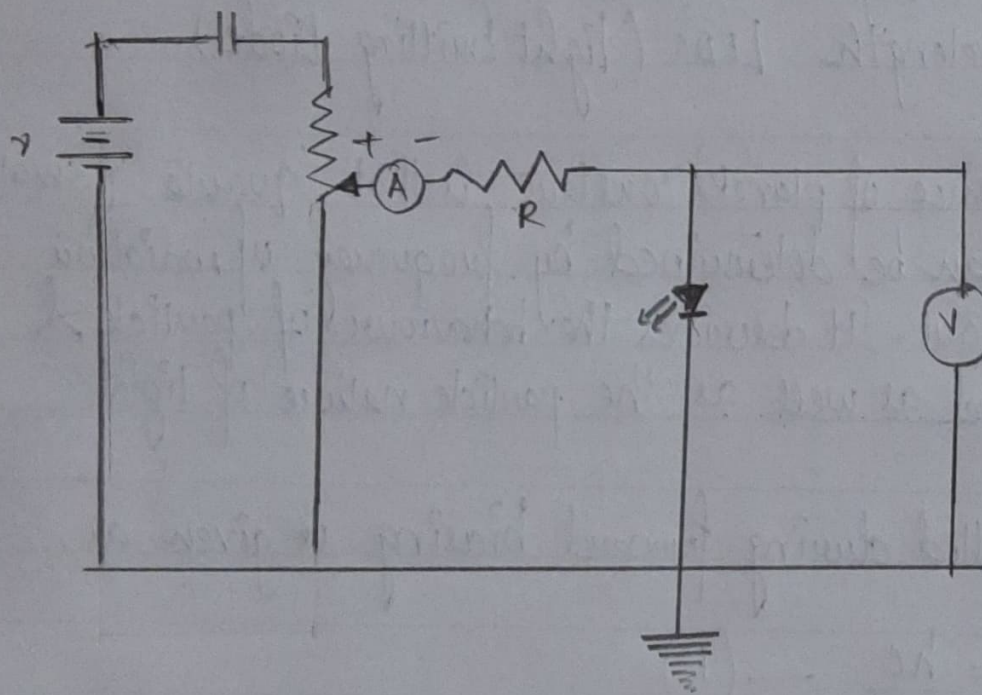
$$\begin{aligned} \text{Error} &= \frac{O_1 - O_2}{100} = \frac{(\text{Observed planck's constant} - \text{graphical planck's constant})}{100} \\ &= \frac{6.8010 \times 10^{-34} - 6.5275 \times 10^{-34}}{100} = 2.735 \times 10^{-37} \end{aligned}$$

CONCLUSION :-

a) Observed Planck's Constant = $6.5275 \times 10^{-34} \text{ Js}$

b) Graphical Planck's Constant = $6.8010 \times 10^{-34} \text{ Js}$

—X—X—



CIRCUIT DIAGRAM