NEGATIVE POLYATOMIC IONS		chlorate	ClO ₃	hydrogen oxalate	HC ₂ O ₄	nitrate	NO ₃	pyrophosphate	P ₂ O ₇ ⁴⁻
acetate	CH₃COO ⁻	chlorite	ClO ₂	hydrogen sulfate	HSO ₄	nitrite	NO ₂	sulfate	SO ₄ ²⁻
asenate	AsO ₄ ³⁻	chromate	CrO ₄ ²⁻	hydrogen sulfide	HS ⁻	orthosilicate	SiO ₄ ⁴⁻	sulfite	SO ₃ ²⁻
arsenite	ASO ₃ ³⁻	cyanate	CNO	hydrogen sulfite	HSO ₃	oxalate	C ₂ O ₄ ²⁻	thiocyanate	SCN ⁻
benzoate	C ₆ H ₅ COO¯	cyanide	CN ⁻	hydroxide	OH.	perchlorate	ClO ₄	thiosulfate	S ₂ O ₃ ²⁻
borate	BO ₃ ³⁻	dichromate	Cr ₂ O ₇ ²⁻	hypochlorite	ClO	periodate	IO ₄	POSTITIVE POLYATOMIC IONS	
bromate	BrO ₃	dihydrogen phosphate	H ₂ PO ₄	iodate	IO ₃	permanganate	MnO ₄	ammonium	NH ₄ ⁺
carbonate	CO ₃ ²⁻	hydrogen carbonate (also bicarbonate)	HCO ₃	monohydrogen phosphate	HPO ₄ 2-	peroxide	O ₂ ²⁻	hydronium	H₃O⁺
						phosphate	PO ₄ ³⁻	mercury l	Hg ₂ ²⁺

