

# **Jashore University of Science and Technology**

**Bachelor of Science in Electrical and Electronic Engineering**

**1st semester of 3rd year, Academic session: 2024–2025**

**Course no.: PHY 3101**

**Course title: Electrical Engineering Materials**

**Class test no.: 02**

**Date: 21 October 2025**

**Roll:**

- 1.** What is atomic packing fraction? Calculate the atomic packing fraction of a BCC lattice. [4]
  
- 2.** Draw (010), (101), (111) and (112) plane of a simple cubic lattice. [6]
  
- 3.** Show that the reciprocal lattice of a BCC lattice is an FCC lattice. [5]
  
- 4.** Na is a monovalent metal (BCC) with a density of  $0.9712 \text{ g cm}^{-3}$ . Its atomic mass is  $22.99 \text{ g mol}^{-1}$ . The drift mobility of electrons in Na is  $53 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ . Calculate the electrical conductivity of Na and compare this with the experimental value of  $2.1 \times 10^7 \text{ } \Omega^{-1} \text{ m}^{-1}$  and comment on the difference. [5]