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## **Chemistry Merit Badge Worksheet**

## **Requirement 1**

Explain the meaning of chemistry. Discuss the branches of chemistry. Discuss how chemistry is different from other fields of natural science.

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|   |          |        |
| Requirement 2   |          |        |
| ☐ Show that a candle flame uses up oxygen from the air and produces carbon dioxide. |          |        |
| ☐ Demonstrate that heating sawdust or wood chips produces a gas that burns.         |          |        |
| Sketch the carbon dioxide – oxygen cycle.   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
|   |          |        |
| Requirement 3   |          |        |
| Explain how the following gases are produced industrially.                          |          |        |
| • Oxygen  |          |        |
|   |          |        |
|   |          |        |
| Hydrogen  |          |        |
|   |          |        |
| • Chlorine  |          |        |
|   |          |        |
|   |          |        |
| • Ammonia   |          |        |
|   |          |        |
|   |          |        |

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| rement 4   |        |      |
| tiate the following.   |        |      |
|  |        |      |
| Carbonates   |        |      |
|  |        |      |
|  |        |      |
| Sulfides   |        |      |
|  |        |      |
|  |        |      |
| Chlorides  |        |      |
|  |        |      |
|  |        |      |
| Acids  |        |      |
|  |        |      |
|  |        |      |
| Gases  |        |      |
|  |        |      |
|  |        |      |
|  |        |      |
|  |        |      |
| rement 5   |        |      |
|  |        |      |
| e formulas for two compounds that make water hard.   |        |      |
| e formulas for two compounds that make water hard.  Provide the Name of Compound # 1:  |        |      |
| e formulas for two compounds that make water hard.  Provide the Name of Compound # 1:  |        |      |
| e formulas for two compounds that make water hard.  Provide the Name of Compound # 1:  |        |      |
| e formulas for two compounds that make water hard.  Provide the Name of Compound # 1:  |        |      |
| e formulas for two compounds that make water hard.  Provide the Name of Compound # 1:  Formula for Compound # 1  |        |      |
| Provide the Name of Compound # 1:  Formula for Compound # 1  Provide the Name of Compound # 2:   |        |      |
| e formulas for two compounds that make water hard.  Provide the Name of Compound # 1:  Formula for Compound # 1  |        |      |
| Provide the Name of Compound # 1:  Formula for Compound # 1  Provide the Name of Compound # 2:   |        |      |
| Provide the Name of Compound # 1:  Formula for Compound # 1  Provide the Name of Compound # 2:   |        |      |
| rement 5  e formulas for two compounds that make water hard.  Provide the Name of Compound # 1:  Formula for Compound # 1  Provide the Name of Compound # 2:  Formula for Compound # 2  equation describing how a home water softener works. |        |      |

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|         |  |        |      |
| Requ    | irement 6  |        |      |
|         |  |        |      |
| Explair | n what oxygen does in the animal body.                               |        |      |
| 7       |  |        |      |
|         |  |        |      |
| Tell ho | w the following are carried in the body.                             |        |      |
|         |  |        |      |
| •       | Oxygen   |        |      |
|         |  |        |      |
|         |  |        |      |
| •       | Carbon Dioxide   |        |      |
|         |  |        |      |
|         |  |        |      |
| •       | Carbon Monoxide  |        |      |
|         |  |        |      |
|         |  |        |      |
| Describ | be the chemical reaction that take place when the following happens. |        |      |
| •       | Vegetables Cook  |        |      |
|         |  |        |      |
|         |  |        |      |
| •       | Meats Cook   |        |      |
|         |  |        |      |
|         |  |        |      |
| •       | Bread Dough Rises  |        |      |
|         |  |        |      |
|         |  |        |      |
|         | Bread Bakes  |        |      |
| •       | Dieau Dakes  |        |      |
|         |  |        |      |
|         |  |        |      |
| •       | Bread is Chewed  |        |      |
|         |  |        |      |
|         |  |        |      |

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| Requirement 7  |               |      |
| ☐ Carry out an experiment to show the different ways of protecting iron or steel f | from rusting. |      |
| ☐ Give examples using Scouting utensils.   | _             |      |
| Why aluminum does not rust?  |               |      |
| wity attributed does not rust:   |               |      |
|  |               |      |
| Why iron does rust?  |               |      |
|  |               |      |
|  |               |      |
| Requirement 8  |               |      |
| Do any two of the following:   |               |      |
| $\square$ Visit a plant that makes chemical products, and uses chemical processes. |               |      |
| What plant did you visit?  |               |      |
| Describe the chemical processes used.  |               |      |
|  |               |      |
|  |               |      |
| ☐ Visit a laboratory or place of business that uses chemicals.                     |               |      |
| What laboratory did you visit?   |               |      |
| Write down how and why the chemicals are stored.                                   |               |      |
|  |               |      |
| ☐ Learn how chemistry is meeting farm problems of soil fertility and crop pests.   |               |      |
| What did you learn about soil fertility?   |               |      |
|  |               |      |
|  |               |      |
| What did you learn about crop pests?   |               |      |
|  |               |      |
|  |               |      |
| ☐ Explain the differences in university courses for training the following:        |               |      |
| Chemical Technicians   |               |      |

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|   |                   |        |      |
|   |                   |        |      |
|   |                   |        |      |
|   |                   |        |      |
| Chemists  |                   |        |      |
|   |                   |        |      |
|   |                   |        |      |
|   |                   |        |      |
| Chemical Engineers  |                   |        |      |
|   |                   |        |      |
|   |                   |        |      |
|   |                   |        |      |
|   |                   |        |      |
| Describe two (2) different kinds of                       | f work done by:   |        |      |
| Describe two (2) different kinds of Chemical Technicians: | •                 |        |      |
|   | (1)               |        |      |
|   | •                 |        |      |
|   | (1)               |        |      |
|   | (1)               |        |      |
| Chemical Technicians:                                     | (1)<br>(2)<br>(1) |        |      |
| Chemical Technicians:                                     | (1)(2)            |        |      |
| Chemical Technicians:                                     | (1)<br>(2)<br>(1) |        |      |
| Chemical Technicians:                                     | (1)<br>(2)<br>(1) |        |      |
| Chemical Technicians:  Chemists:                          | (1)               |        |      |