+4 +2 = 2 +5 Oksidasi = Mn0 -> Mn04

+2 +7 =5 × 2

(VISION)

5Pb02+2Mn0+8H+ -> 5Pb2++2Mn04-+4H20 Suasana Basa 1. Cl2 + 103 -> Cl + 104 0 +6 -6 -1 +7 -8 Redulesi = Cl2 -> 2cl Rehulesi 0 -1 =1 x 2 = 2 oksidasi Oksidasi = 103 -> 104 +5 +7 = 2x1 = 21 C12+105+20H -> 201+101+H20 2 Al + No3-6 -> Al (0H)4- + NH3+7 Redulesi = No; + 3H2O -> NH3 + 60H + 30H x 3 Oksidasi = A1+40H -> A1(0H)4 OKSIDASI = 8 AI + 320FT -> 8 AI (0H) 4-Redulesi = 3 NO3 + 18 H20-> 3 NH3 + 370H 8AI+ 3NO3 + 50H + 18H20 -> 8AI (0H)4 + 3NH2 3. Mn09 + C2042 --> Mn02 + CO2 +7 -2 +3 -2 Reducisi oksid asi Redulcsi = Mn 04 -> Mn 02 +7 +4 = 3 x+ oksidasi = C2042 -> 2001 +3 +4 = +x2 2 Mn 04 + 3C2042 + 4H20 -> 2 MnOz+ 6CO2 + 80H (VISION)