A B C D E F

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3						
	Atomic	Element Name	Symbol	Valency	Radical	Radical
4	Number				Formed	Туре
5	1	Hydrogen	Н	1	H^+	Basic
6	2	Helium	He	0	No radical	
7	3	Lithium	Li	1	Li ⁺	Basic
8	4	Beryllium	Be	2	Be^{2+}	Basic
9	5	Boron	В	3	B^{3+}	Basic
10	6	Carbon	С	4	Shares electrons	
11	7	Nitrogen	N	3	N ³⁻	Acidic
12	8	Oxygen	0	2	0^{2-}	Acidic
13	9	Fluorine	F	1	F ⁻	Acidic
14	10	Neon	Ne	0	No radical	
15	11	Sodium (Natrium)	Na	1	Na ⁺	Basic
16	12	Magnesium	Mg	2	Mg^{2+}	Basic
17						
	13	Aluminium	Al	3	Al^{3+}	Basic
18		Aluminium Silicon	Al Si	3	Al ³⁺ Shares electrons	Basic
	14		_			Basic Acidic
18	14 15	Silicon	Si	4	Shares electrons	
18 19	14 15 16	Silicon Phosphorus	Si P	4	Shares electrons P ³⁻	Acidic
18 19 20	14 15 16 17	Silicon Phosphorus Sulphur	Si P S	4 3 2	Shares electrons P^{3-} S^{2-} Cl^{-}	Acidic Acidic
18 19 20 21	14 15 16 17 18	Silicon Phosphorus Sulphur Chlorine	Si P S Cl	4 3 2 1	Shares electrons P^{3-} S^{2-} Cl^{-}	Acidic Acidic
18 19 20 21 22	14 15 16 17 18 19	Silicon Phosphorus Sulphur Chlorine Argon	Si P S Cl Ar	4 3 2 1 0	Shares electrons P^{3-} S^{2-} Cl^{-} No radical	Acidic Acidic Acidic