


	A	B	C	D	E	F
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4	<b>Atomic Number</b>	<b>Element Name</b>	<b>Symbol</b>	<b>Valency</b>	<b>Radical Formed</b>	<b>Radical Type</b>
5	1	Hydrogen	H	1	$H^+$	Basic
6	2	Helium	He	0	No radical	
7	3	Lithium	Li	1	$Li^+$	Basic
8	4	Beryllium	Be	2	$Be^{2+}$	Basic
9	5	Boron	B	3	$B^{3+}$	Basic
10	6	Carbon	C	4	Shares electrons	
11	7	Nitrogen	N	3	$N^{3-}$	Acidic
12	8	Oxygen	O	2	$O^{2-}$	Acidic
13	9	Fluorine	F	1	$F^-$	Acidic
14	10	Neon	Ne	0	No radical	
15	11	Sodium (Natrium)	Na	1	$Na^+$	Basic
16	12	Magnesium	Mg	2	$Mg^{2+}$	Basic
17	13	Aluminium	Al	3	$Al^{3+}$	Basic
18	14	Silicon	Si	4	Shares electrons	
19	15	Phosphorus	P	3	$P^{3-}$	Acidic
20	16	Sulphur	S	2	$S^{2-}$	Acidic
21	17	Chlorine	Cl	1	$Cl^-$	Acidic
22	18	Argon	Ar	0	No radical	
23	19	Potassium (Kalium)	K	1	$K^+$	Basic
24	20	Calcium	Ca	2	$Ca^{2+}$	Basic
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