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Are Comets the source of our water?

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Summary

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1 Introduction

Scientists aren't sure where the water on Earth and other terrestrial planets originated from. There are multiple explanations of where the water came from. But there is currently no perfect explanation how the terrestrial planets gained water. It is believed that the answer can be found in studying the D/H ratio in comets and asteroids. Because this ratio is directly linked to the place in the solar system the objects formed.

The most believed explanation is that the water was imported from comets and/ or asteroids that are rich in water. The proto-Earth would have been quite hot because of the collisions with other protoplanets. In this case the proto-Earth accreted dry. However, it is also possible that the Earth accreted wet, because there was water in the early solar system. In this case there should be a way for the proto-Earth to hold on to this water.

In this report the main research question is "Are comets the source of water?". To answer this question firstly the formation of comets and asteroids in the early solar system is being researched in chapter 2. In this chapter the measured data from the comet 67P/Churyumov-Gerasimenko is being researched. This is the only comet where a lander did direct measurements. Because there is the only comet, it can't be ruled out that other comets have the same D/H ratio. This is why also measured data from other comets are used. These measurements have higher tolerances because there hasn't been any direct measurements.

In chapter 3 the possibility of the proto-Earth that accreted wet is being researched. The most important question is if the proto-planets could hold on to this water. It is believed that that because of the huge collisions the terrestial planets with other protoplanets there was a lot of heat and there were even magma pools. But there are theories that the protoplanets could hold on to those volatile elements inside the planet. In these theories the magma pools were even necessary to get the volatile elements out of the proto-Earth and into the atmosphere.

In the last chapter 4 the options that comets and asteroids brought the water to proto-Earth during the late bombardment. During this time there weren't many large collisions with proto-planets. The proto-Earth could cool down and there were only small impacts. Because of this the atmosphere wouldn't be completely destroyed and could absorb the water vapor. In chapter 5 the conclusion of this research is stated and are recommendations given to get more information about the source of water on Earth.

2 The formation of comets and asteroids in the early solar system

2.1 Creation of asteroids and comets and their H/D Ratio

2.1.1 Creation of asteroids

In the Solar System, there are two main sources for asteroids and comets: the asteroid belt and the Kuiper Belt. The former is located between the terrestrial planets and the gas planets in a region that is often referred to as the snow line [1], while the latter is located further out than any of the known planets. In the early states of the solar system, asteroids in the asteroid belt formed within the solar nebula. This process would not differ from the formation of the other planets were it not for the presence of Jupiter, the closest gas giant to the snow line. The gravitational forces and disturbances prevent the small planetesimals to form a planet. In addition to this, these disturbances could cause the asteroids to leave the asteroid belt and collide with other bodies in the solar system. [1]

2.1.2 Creation of comets

In the present solar system the comets encountered in the inner solar system usually originate from either the Kuiper Belt or the Oort cloud. However, it is likely that the comets were not originally formed there. For the formation of comets there are two plausible locations: either in the neighborhood of Uranus and Neptune or further out where the temperature was lower and the gravitational forces not as disturbing as in the vicinity of Uranus and Neptune. The low-density environment of the Oort cloud would not provide enough particles for the comets to form, so it is suggested that comets found in this region were formed somewhere else and then ejected into the Oort cloud. If comets were formed near Uranus and Neptune the comets would probably be absorbed by the planets or ejected onto hyperbolic trajectories rather than just being ejected into larger-distance orbits within the Oort cloud [2]. This leaves the most likely point of formation outside the orbits of Neptune and Jupiter [3].

2.2 Water distribution in the early states of the solar system and contribution to the formation of asteroids and comets

The early states of the Solar System with its hot inner core near the proto sun have mainly contributed to the fact that the only water-rich asteroids originally came from the outer asteroid belt and the Kuiper Belt. Up to 10 percent of their mass is water [4].

The possible areas where water particles can aggregate and therefore be found in bodies such as asteroids and comets as well as planets and planetary satellites are beyond the snow line that marks the area where H_2O only occurs as water ice and not as water particles or water vapor. Figure 1 shows the expected lifetime of pure and dirty ice grains depending on the grain size [5].

2.3 H/D ratio in comets and asteroids

The D/H ratio in comets and asteroids is linked to their formation region and therefore differs for various asteroids as well as comets from the Oort cloud and the Kuiper Belt and can be related to the D/H ratio in the solar nebula and the different planets. An overview of measured and estimated data is given in Figure 2

However, more recent observations of comet 103P/Hartley~2 have shown that it's H/D ratio is $(161 \pm 24) \times 10^{-6}$ and therefore significantly lower than estimated from previous measurements of other comets. This could be the case because the origin of 103P/Hartley~2 is not known for a certainty but it is believed that it could have originated in the Kuiper Belt, contrarily to a lot of other comets such as 1P/Halley who are widely believed to have originated in the Oort cloud. [7]

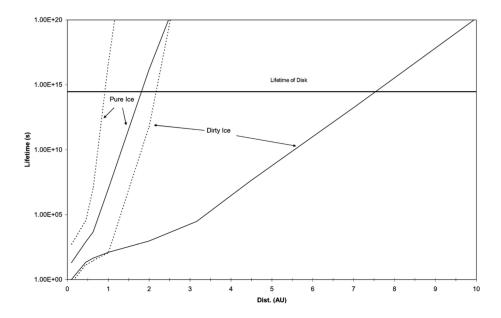


Figure 1: Comparison of lifetimes for pure and dirty ice grains. The dashed curves are for 0.1 μ m grains and the solid curves are for 10 μ m grains. The solid horizontal line shows a conservative upper limit to the age of the gas disk. [5]

For comets it has been suggested that the deuterium enrichment factor depends on the creation epoch of the comet which is different for Kuiper Belt comets and comets from the Oort cloud, due to different temperature scales and depends on the evolution of the solar nebula over time. Using the enrichment factor f for water

$$f = \frac{1/2}{1/2} \frac{HDO/H_2O}{HD/H_2} \tag{1}$$

which quantifies the D/H ratio for a certain deuteriated species, in this case water. As mentioned above, the deuterium enrichment factor f depends on the temperature distribution of the isotopic exchange as well as the pressure distribution which is also temperature-dependent. The differential equation is given in Equation 2 [8]:

$$\partial_t f = \underbrace{k(T)P(A(T) - f)}_{\text{diffusion term}} + \underbrace{\frac{1}{\sum R} \partial_R(\kappa R \Sigma \partial_R f)}_{\text{exchange term}}$$
(2)

A(T) is the fractation at equilibrium, P is the total pressure R is the heliocentric distance, k(T) is the isotopic exchange, Σ is the gas surface density and κ describes the turbulent diffusivity which depends on the nature of the turbulence. Equation 2 can be split up in an exchange and a diffusion term, the former describes the dependency of the heliocentric distance while the latter describes the dependency on the temperature distribution.

This connection is displayed and compared with measurement data from various comets in Figure 3. Meteorites with high D/H ratio are only those of the following two types: LL3 and carbonaceous chondrites. In both types, the insoluble organic matter shows elevated values of deuterium enrichment as well as clay minerals. The values obtained from the measurements of clay minerals can then be used to determine the D/H ratio of water because of the negligible isotopic fractionation between clay minerals and liquid water. [6]

$D/H(10^{-6})$	Species	Location	f	References
Galactic ⁽⁶⁾				
20 to 50	Н	Big-Bang (Theoretical)	-	Schramm, 1998
34 ± 2.5	Н	Other Galaxy	-	Burles and Tytler, 1998
16±1	Н	Local Interstellar Medium	≡1	Linsky et al., 1993
110^{+150}_{-65}	H_2O	Hot Cores (HC)	7	Gensheimer et al., 1996
$1 \times 10^4 (10)$	H_2O	Cold Interstellar Clouds (10 K)	625	Brown and Millar, 1989
$1^{+3}_{-0.2} \times 10^3$ $2^{+0.7}_{-1.8} \times 10^4$	$OM^{(1)}$	HC in Interstellar Clouds (IC)	63	Pers. compilation ⁽³⁾
$2^{+0.7} \times 10^4$	$OM^{(1)}$	Cold Interst. Clouds (<20 K)	1250	Pers. compilation ⁽³⁾
950 ± 550	H ₂ O (Ice)	Cold Interst. Clouds (10 K)	60 ± 34	Texeira et al., 1999
Proto Sun	- '	, ,		
21±5	H_2	Nebula 4.5 Gyrs ago	≡1	Geiss and Gloeckler, 199
Gaseous Pla	nets ⁽⁷⁾			
21±8	H_2	Jupiter	≡1	Lellouch et al., 1996
26±7	H_2	Jupiter	≡1	Mahaffy et al., 1998
15 to 35	H_2	Saturn	≡1	Griffin et al., 1996
65+2.5	H_2	Neptune	2.6	Feuchtgruber et al., 1999
55^{+35}_{-15}	H_2	Uranus	2.2	Feuchtgruber et al., 1999
Comets ⁽⁷⁾	112	Ciaius	2.2	redefitigraber et at., 199
310±30	H_2O	Comet P/Halley	12.4	Eberhardt et al., 1995
290±100	H ₂ O	Comet Hyakutake	11.6	Bockelée et al., 1998
330±80	H ₂ O	Comet Hale-Bopp	13.2	Meier <i>et al.</i> , 1998a
2300±400	$OM^{(1)}$	HCN in Comet Hale-Bopp	92	Meier <i>et al.</i> , 1998b
Interplaneta			/2	Wicier et at., 17700
91 –4 018	$(?)^{(11)}$	IDP and AMM ⁽¹²⁾	3 6-161	Pers. compilation ⁽³⁾
Telluric Plar		IDI and AMM	3.0 101	reis. compilation
149±3	H ₂ O	Bulk Earth	6	Lécuyer et al., 1998
16000 ± 200	H ₂ O	Venus (Atmosph. in situ)	640	Donahue et al, 1982
780±80	H ₂ O	Mars (Atmosph. in situ)	31	Owen et al., 1988
200-780	-OH	SNC (Mars mantle)	8–31	Watson et al., 1994
L L3.0 and L			0-31	watson et at., 1774
	compounds			
730 ± 120		in Clays and Chondrules	29	Deloule and Robert, 199
800-1100	$OM^{(1)}$	in Kerogen - like in Matrix	32–44	Pers. compilation ⁽³⁾
Protosolar		in rerogen like in Maurix	32 11	reis. compilation
88±11	-OH	in Chondrules	3.5	Deloule et al., 1998
Carbonaceo			0.0	2 210 010 01 011, 1770
140±10	-OH	Mean Statistical Value	5.6	Pers. compilation ⁽³⁾
	<170 (66%) ⁽		2.0	r ets. compilation
380–620	$OM^{(1)}$	Kerogen - like in Matrix	15-25	Pers. compilation ⁽³⁾
(CM,CV,C			20 20	- III. Compilation
370 ± 6	$OM^{(1)}$	Kerogen - like in Matrix	15-25	Halbout et al., 1990
(Orgueil C			10 20	
315–545	$OM^{(1)}$	Amino Acids	12.5-22	Pizzarello et al., 1991
185–310	$OM^{(1)}$	Hydrocarbons and		Pizzarello et al., 1991
100 010	ON	Carboxylic acids	1.5-12.5	1 122mono et ut., 1991
		Carboxyric acids		

Figure 2: D/H ratios of various objects in the solar system. [6]

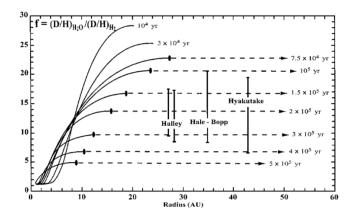


Figure 3: Calculated deuterium enrichment factor f in H2O as a function of the heliocentric distance R, at various epochs in years of the evolution of the nebula. Dots indicate the heliocentric distance where H2O condenses. Dashed lines correspond to the value of f in ices. D/H enrichments obtained in comets Halley, Hyakutake, and HaleBopp are shown for comparison. The location of comets is arbitrary. [9]

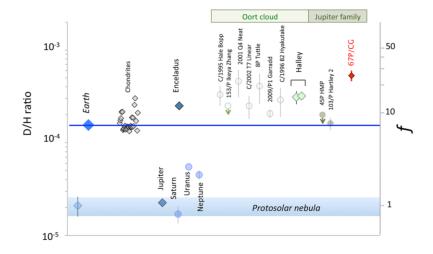


Figure 4: D/H ratios in different objects of the solar system. Diamonds represent data obtained by in-situ mass spectrometry measurements, and circles refer to data obtained by astronomical methods. [10]

2.4 Measurement data from 67P/Churyumov-Gerasimenko

67P/Churyumov-Gerasimenko is a Jupiter-family comet that probably originated in the Kuiper Belt. During the Rosetta Mission detailed measurements of the D/H ratio were conducted, resulting in a D/H ratio of $(530 \pm 70) \times 10^{-6}$, 3 times higher than the values found on Earth. Although 67P/Churyumov-Gerasimenko is expected to come from the same region of the Solar System as 103P/Hartley 2 (discussed in the previous section), they show significantly different values that suggest that the variety of Jupiter-family comets is greater than expected.

3 Water content of terrestrial planets after their creation

There is currently no perfect explanation of how the terrestrial planets gained water. The most common hypothesis is that water arrived later on those planets because the temperatures were too high. Therefore the planets accreted dry. This hypothesis is researched in this chapter together with the less famous hypothesis that the planets were accreted wet. For the latter one, also the possibility of holding on to this water is researched.

3.1 Early Accretion Model

The hypothesis that states that Earth had contained water while formation was occurring is known as the Accretion Model. This theory also has other names such as Wet Earth and Cool Earth, for consistency and clarity. This hypothesis will further be addressed as Early Accretion Model, to distinguish itself from the previous hypothesis. This model has been proposed in opposition to the Late Heavy Bombardment. This hypothesis was started because of the isotopic differences between Earth and extraplanetary objects. The hypothesis is based on the accretion of volatile elements and water in grains of dust and being able to retain them during the development of the planet. The developments in the hypothesis comes from showing that water can be absorbed into different materials at high temperatures of the accretion disc. There has been a study showing that olivine can be hydrated in high temperatures [11] [12]. A new article [13] discusses evidence that there was significant water before the collision that formed the moon, which is before the time period of the Late Heavy Bombardment (LHB). This hypothesis evaluated the claims of the ability of Earth to retain early water. This hypothesis has not been as deeply researched as the Late Heavy Bombardment has been.

3.2 Formation of terrestrial planets in the early solar system

The terrestrial planets formed from the protoplanetary disk, a large, cold and slowly rotating cloud of gas and dust. The gas and dust sometimes collided due to gravitational forces. If the collisions where gentle enough the gas and dust would grow into a bigger object and eventually into a protoplanet. The gas consisted mostly of hydrogen, helium and oxygen, some of the hydrogen and oxygen combined to make water vapor [14, p. 523].

The amount of water vapor within 3 AU in the protoplanetary disk has been estimated at three earth masses [15] in 1994 and two earth masses [16] [17] in 2003. The mass of Earth is $5,9722 \pm 0.0006 \cdot 10^{27}$ g and the earth mass of all Earths oceans is $\pm 1, 4 \cdot 10^{24}$ g. The amount of water inside the Earth is not exactly known, most estimations are around 10 Earth oceans with a extreme maximum of 50 Earth oceans [14, p. 523]. The water storage of Earth in minirals is about 5 - 6 Earth oceans [18].

If all four terrestrial planets accreted with 50 Earth oceans of water, then that would still only be 4,7% of the available water. There was probably enough water vapor in the early solar system to account for Earths oceans and the water on the other terrestrial bodies. The main problem is if the terrestrial bodies could hold on to this water during the formation of the protoplanets due to heat.

3.3 Water storage in terrestrial planets during formation

There are two main possibilities of water being stored, hydrated minerals and absorption onto grains. The water can be depleted by high temperatures and collisions. In this paragraph the storage is studied.

There are hydrated minerals observed in the mid-asteroid belt. Some were apparently heated to several hundred degrees Celsius. This is a confirmation that water was present at the early solar system and can be stored. The hotter an asteroid would have been, the less of these minerals are found. The chances of the terrestrial bodies holding on to water in hydrated minerals are pretty slim because the temperature of those planets were much higher than those of the asteroids [19].

There is also the possibility of water being absorbed by grains. Grains are small objects with a rough surface. The two forms of absorption that could have happened are physisorption and chemisorption [14, p. 523]. Physisorption is caused by the 'van der Waals force' which is $10 - 100 \text{ meV} \approx 1, 6 - 16 \cdot 10^{-21} \text{ J}$. Chemisorption involves a chemical reaction between the surface of the grain and in this case water. This force is stronger with $\approx 0, 5 \text{ eV} \approx 8 \cdot 10^{-20} \text{ J}$.

The effects of physisorption has been simulated by creating an Earth out of grains with a 100 times greater surface area than a spherical grain with the same volume. The finding were that with a temperature of 1000 K a quarter of Earths ocean could be absorbed. If the temperature would be 700 K than one Earth ocean would be absorbed and with 500 K three Earth oceans [20] [21].

If chemisorption is taken into account the force that keeps the water onto the grains will increase if there are multiple water molecules. How much it exactly contributes is unknown because tests are only performed at temperatures below 30 K. But it can be said that the force that keeps the water onto the grains will increase even at higher temperatures due to chemisorption [22].

As the grains collide with each other, some or all of the water will be lost depending on the impact. In the physisorption simulation they accounted for this, if two grains collided with a force greater than two times the total bond energy. In that case all water would be gone. If chemisorption was also taken into consideration the total amount of water that could be stored in the proto-Earth would increase [20] [21].

3.4 Water loss in terrestrial protoplanets

The early terrestrial planets probably melted one or multiple times while being accreted, probably from multiple major impacts, of which one created the moon. This means that there where magma oceans at some point in their lifetimes, before reaching their final solid state. [23].

These magma oceans could have played a key role in delivering volatile elements into the growing atmosphere through degassing. For this to happen the magma ocean should have been over saturated by these elements and there was no boundary layer for these elements [24, p. 128-129].

As the interior gained more heat the volatile elements broke down and formed bubbles that were being transported to the exterior. The melted material would become drier because most volatile elements would migrate to the atmosphere. Most of the big impacts would take away some of the Earths atmosphere but it could also bring other materials to Earth. The impacts would create more heat in the interior and the process of volatile elements out gassing continued until the interior was completely dry [24, p. 130-131].

Due to the heat there could not be any oceans on Earth, but the Earth was massive enough to keep most water vapor in the atmosphere. After the Earth cooled down this vapor could condense and become Earths first oceans. Through plate tectonics and volcanic activity more water could be transported from inside the Earth to the oceans [24, p. 130-131].

3.5 Collisions between protoplanets

The formation of protoplanets can be explained by figure 5. Small amounts of matter form planetsimals because of mostly gravitational forces between small objects and collisions. This happens in approximately 10⁵ years. These planetsimals clam together because of the same forces and become bigger until they become embryos. When these objects combine to a round object and clear out their orbit a planet forms [24, p. 118-120] [25].

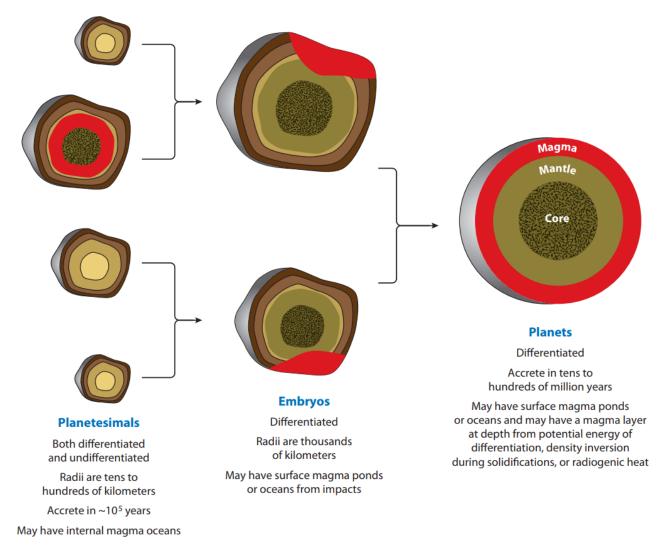


Figure 5: Formation of protoplanets with magma oceans [24, p. 120].

The planetesimals might have a internal magma ocean. But the volatile elements will be trapped because there mantle will be solid. This is because of the energy released from nuclear reactions. The amount of energy from the collisions will be minimal, so no magma pool will be formed after a collision. The embryos are larger and a collisions can create a magma pool. The volatile elements in this area can form bubbles and rise to the surface. to form an atmosphere. This atmosphere will probably escape after the next collision as the gravitational forces are minimal [24, p. 118-123].

When embryos collide and make planets there will be magma pools. Again, the volatile elements can rise out of the planet into the atmosphere. If the planet is large enough it can hold on to its atmosphere. After another collision it can loose some of its atmosphere. But it can also get more volatile elements from the impact [24, p. 118-123].

3.6 The current D/H ratio of terrestrial planets

The distribution of the D/H ratio through the solar system can be seen in figure 6. It's clear that how further away from the sun, the higher the D/H ratio. Mars has a D/H ratio which is 7 times larger and Venus has a D/H ratio which is 150 times larger. The latter one is a discrepancy because Venus is closer to the sun than Earth. It's possible that Venus got more water from comets from far way in the solar system. This confirms that the origin of water is from comets. Another explanation is that the ultraviolet radiation from the sun breaks water molecules into H and OH, the lighter gass will escape into space. It is possible that Venus lost an oceans worth of water but Earth did not because Earth was too far from the Sun for the instability to develop [26].

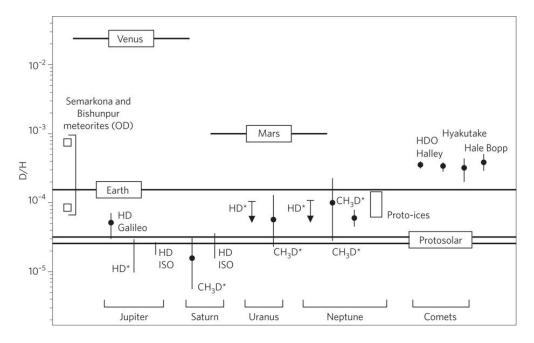


Figure 6: D/H ratio in the solar system [26].

4 The bombardment of Earth

There are differing trains of thought on the timing, how and where the Earth received its water. The two possibilities where water could come from are either through internal reserves acquired during the accretion process or from extraplanetary sources after Earths development. These extraplanetary sources of water would have come from asteroids, comets and protoplanet collisions. The hypothesis that concerns these objects delivering water to a developed and dry Earth is referred to as the late heavy bombardment. The hypothesis that Earths water comes from extraplanetary origins is not flawless and has controversy. There is another hypothesis where the water was accumulated during the Earths accretion development. This hypothesis is referred to as the Accretion Model. These are the two base theories about the origin of Earths water. These hypotheses have evolved over time to account for previously unknown scientific data.

4.1 Late Heavy Bombardment

The Late Heavy Bombardment has been addressed by different names such as Late Veneer, Late Bombardment Hypothesis, Late Accretion, and Late Heavy Bombardment: for consistency within this report, this hypothesis will be referred as Late Heavy Bombardment or LHB. The late heavy bombardment has variations that cover the various objects that could deliver water to the Earth as well as the influences of different planetary orbits. The various objects included are comets, asteroids and planetesimals. The time frame of this bombardment happened between 4.4 and 3.5 Ga [27]. This hypothesis was first noted by Hopp[28] but not named as such in an article from Kimura et al. concerning the discrepancy in gold distribution [29]. However, another article from Maruyama [30] report that another article written by E. Anders [31] first came up with the theory. The start and progression of this theory concerns the siderophile elements and their distribution within the Earth. The elements include tungsten, ruthenium, gold, osmium, iridium, and more elements [32] [33] [34]. These reports on the different elements shows the gathering of data for this hypothesis. The late heavy bombardment has competing views on the origin of Earths water. The most popular to least popular opinions are asteroids, comets and then planetesimals. There are reports and articles [35] [36] [37] that highlight the differences between the sources as well as likelihood and the effects coming from the different sources. The most commonly discussed of these sources are the carbonaceous chondrite asteroids. The asteroids are the viewed as the most likely source of the Earths Water in the late heavy bombardment.

4.1.1 Comets

Comets have showered Earth in dust and ice in their wake and the occasionally impact on Earth. There are two classes of comets: short period and long period comets. The distinction is whether their orbital period is less or greater than 200 years. The long period comets are from the Kuiper Belt and the Oort Cloud while the short period comets are from the Jupiter and Saturn Belt. With the discovery of comets comprising significantly of water, comets were the first extraplanetary objects to be considered as the source of Earth's water. The impact of comets could deliver significant amount of water and would effect Earth's climate and atmosphere.

4.1.1.1 Probability and Numbers

Comets are abundant in the solar system. Estimates of the number of comets in the solar system are in the magnitudes of $10^{11} \sim 10^{12}$ objects [38]. The current amount of comets is approximately one-fifth of the original amount of comets [39]. The amount of comet material could amount to almost 2% [39] the solar system mass. However scientists have only been able to observe under 6000 [40] comets and explored and studied only a handful of comets.

The mass of comet 67P is $9.98 \times 10^{12} \mathrm{kg}$ [41] with water ice accounting for 75% [42] of the comet which corresponds to $7.48 \times 10^{12} kg$. Comet Halley is $2.2 \times 10^{14} kg$ [43] and contains 40% water ice [44] by weight. The comet Halley should have $8.8 \times 10^{13} kg$ of water ice. Assuming average spherical size and density of short period comets are 4.4 kilometers [39] and $0.6 \, g/cc$ [45] respectively meaning an average mass of $2.14 \times 10^{14} kg$. Assuming similar hydration levels as come Halley with 40% water ice by weight, an average comet would hold about $8.56 \times 10^{13} kg$ kg of ice.

Under ideal conditions of perfect transfer of all material from the comet to the Earth and no loss of previously accumulated water, it would take almost 180 million comets to provide Earth its $1.54 \times 10^{22} kg$ of water. A total of $3.85 \times 10^{22} kg$ of comets would be needed for this set of ideal conditions.

Assuming non-ideal conditions, such as, an non perfect transfer on water and non-negligible of previous water due to the impact. The amount of water accumulated will not reach the desired value of $1.54 \times 10^{22} kg$, this is due to the loss of previous water being equal to the amount of water transferred to Earth as the assumed average comet is not large enough to transfer more water than the Earth would have lost to space.

4.1.1.2 Controversy and Disagreements

The hypothesis of comets being the source of water on Earth is not without contention. The main points that cause scientist to disagree with this assertion are the deuterium to hydrogen ratios and the time frame of the bombardment from Jupiter's comets. The comet deuterium hydrogen ratios are much higher than Earth's, especially with comet 67P being 3 times the value of Earth's. The second point that causes disagreements which includes Jupiter's development into the equation of scattering comets and asteroids towards Earth. Comets were most likely scattered within a million years after Jupiter was fully developed, and at that point of time in the solar system, the Earth was a much smaller target [46]. These arguments against tend to discount comets being a main source of water for Earth.

4.1.1.3 Conclusion

Comets are bountiful objects of ice and dust that people throughout the ages can admire or fear. They did deliver ice and other volatile elements to Earth through collisions. There is enough cometary mass in the solar system to provide Earth the required amount of water. However due to the issue with the size of the planet at the time of bombardment and the high deuterium to hydrogen ratio, scientists have stated that this is improbable source. The high deuterium to hydrogen ratio found in comets, some scientists speculate that at most 10% of the Earth's water comes from comets [46]. Comets have contributed to Earth's water but in a minor role rather than the primary contributor.

4.1.2 Asteroids

Asteroids are becoming the popular origin of water on Earth. Asteroids are different from comets, both in origin location and composition. The asteroid belt can be separated by the Kirkward gaps at 2.5 and 2.82 AU[47], into the inner, middle and the outer asteroid belt. These inner belt has differing properties compared to the middle and outer belts with the most notable in this report being hydration values.

4.1.2.1 Calculations and Associated Probability

The inner asteroid belt is much dryer with an hydration of < 0.1% by weight while the asteroid belt beyond 2.5 AU contains more water with up to 10% by weight [46]. The sizes of asteroids range from meters to hundreds of kilometers large with the asteroid belt being composed of asteroids smaller and larger than 1 kilometer in size [48]. The average density of an asteroid depends on its classification, the densities varies from $1.38\,g/cc$ for carbonaceous, $2.71\,g/cc$ for silicate, and $5.32\,g/cc$ for metallic asteroids[49]. With an estimate of 1 kilometer sized asteroid with an averaged density of $3.1\,g/cc$ and assuming the hydration to be 0.1% for inner belt asteroids and 10% for middle and outer belt asteroids. In Ideal transfer of perfect mass capture and negligible previously retained water escaping due to the collision, the amount of inner belt asteroids to collide with Earth would be just under 1 trillion asteroids to give Earth its $1.54 \times 10^{22} kg$ of water. This amount of asteroids is problematic as it is more than twice the mass of planet Earth at $1.5 \times 10^{25} kg$. It would take 10 billion middle and outer belt asteroids with a combined mass of $1.5 \times 10^{23} kg$ or 2.5% of Earth's mass. The current asteroid belt has $3.58 \times 10^{21} kg$ [50]. It is hypothesized that the current asteroid belt is a remnant compared to the primordial asteroid belt at 1% its former size [51]. If using the former size, there is sufficient material if all the asteroids were well hydrated.

4.1.2.2 Controversy and Disagreements

There is not a consensus on if asteroids is the origin of water on Earth. One of the issues that cause contention is Jupiter's influence on the asteroid belt. Jupiter's influence on the scattering of the primordial asteroid belt would have been too early. Morbidelli concluded that the large majority of asteroids would collide with the Earth before the Earth was 60% formed [46]. There is also a contention on the primordial asteroid belt size, with a new hypothesis stating that it was empty and was filled instead of losing nearly all of its objects [52]. This would prevent asteroids being the source of water as there would have not been any to deliver water to the Earth. The primordial belt if two magnitudes larger than today's value would have a similar variation of hydration and would not deliver the needed amount of water on Earth.

4.1.2.3 Conclusion

Asteroids are considered the leading explanation for water on Earth. The deuterium to hydrogen ratio of asteroids are much closer to Earth's values than comet's. However this scenario still leaves much to desire as the primordial asteroid belt might not have significant enough amount of water to hydrate the Earth.

4.1.3 ABEL Model

The new hypothesis in Late Heavy Bombardment is "Advent of Bio-ELements" (ABEL) model [53]. ABEL attempts to reconcile evidence of water within the Earth with LHB. ABEL model comprises of a two step process: Earth was formed dry with no atmosphere or oceans then gained an atmosphere and oceans.

4.1.3.1 Differences between ABEL and other LHB

ABEL model considers multiple geological characteristics compared to the previously described Late Heavy Bombardment scenarios. The late bombardment that occurred in the middle of Hadean era is renamed to the ABEL bombardment. ABEL model considers the origin of plate tectonics and the presence of water within the Earth which is not considered within the previous models. The areas that ABEL differs from that of the cometary, asteroidal or planetesimal scenarios are having no atmosphere, a rigid lithosphere, and a completely reductive planet [53].

ABEL model states that the Earth originally formed without an atmosphere. The atmosphere was developed during the ABEL bombardment with the presence of volatiles and water. The bombardment provided the oceanic and atmospheric elements to the dry Earth [54].

A large concentration of ABEL is the plate tectonics. The model proposes that there was a large solid lithosphere [54]. The ABEL bombardment with massive asteroids broke and subducted the primordial continents into the mantle as well as introducing volatiles which is needed for plate tectonics [55].

4.1.4 Effects of Late Heavy Bombardment

The impacts from extraplanetary object have significant effects on the planet Earth. The consistent impacts consistent with LHB would produce large lava oceans, up to 10% of Earth's surface and would have increased the temperature significantly and possibly evaporate all the oceans [56]. The impacts would have caused massive amounts of rock dust and water vapor to enter the high atmosphere. The sub-micrometer dust cloud can severely block incoming light from the sun and cool the planet, and the water vapor may be photolyzed and be lost to space [57]. These are very harsh environments for life to form from.

4.2 Conclusion

The main two hypotheses are the Late Heavy Bombardment and the early accretion model. The Late Heavy Bombardment says that Earth was formed dry and received water and volatiles from extraplanetary sources. This model dismisses all claims from the early accretion model concerning the possibility of the Earth forming wet. The early accretion model does not preclude the Late Heavy Bombardment, however does severely limit the amount of water accumulated from extraplanetary sources. There is agreement on our knowledge of detail of terrestrial planet formation and hydration is currently insufficient [58].

5 Conclusion & Recommendations

Conclusion: We are still not sure.

What can we do/measure to get more information about the origin of water.

recommendation: Create simulations of water being absorbed by grains that takes into account both physisorption and chemisorption at reasonable temperatures (paragraph 3.2).

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