

# PC3242 Part II Lectures 3

Topic	Text Book (Zhen Cui '05)	Lectures
<a href="#">Optical lithography</a>	Chapter 2	1, 2 & 3
<a href="#">Electron Beam Lithography</a>	Chapter 3	4 & 5
<u>Focused Ion Beam Technology</u>	Chapter 4	
Low Energetic Ions (keV)		6, 7 & 8
SIMS	Extra material provided	
FIB in Lithography	Chapter 4	
High Energetic Ions (MeV)	Extra material provided	9
RBS	Extra material provided	10
Light ions in lithography	Extra material provided	11
<u>Nano Imprint Lithography</u>	Chapter 7	12
<u>3DP Three Dimensional Printing</u>	Extra material provided	13
<u>Etching</u>	Chapter 6	14

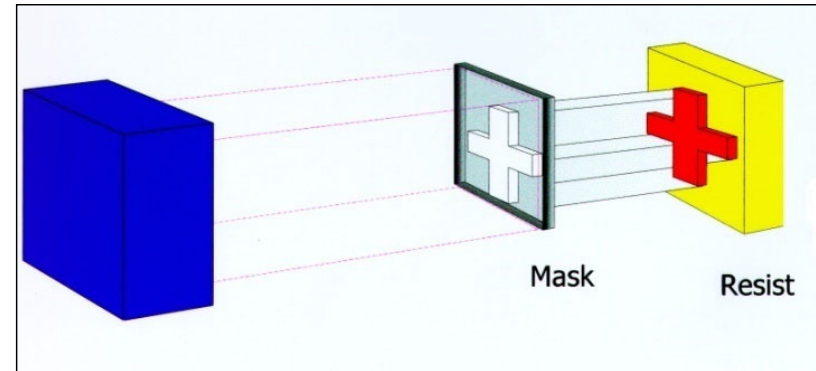
# Optical Lithography

## Masked processes (electromagnetic)

- Light (Coherent)
- X-rays, UV

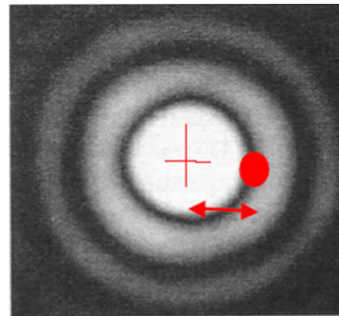
## Contact

$$w = k\sqrt{\lambda z}$$



## Projection printing

$$R = l_m = k_1 \frac{\lambda}{NA}$$

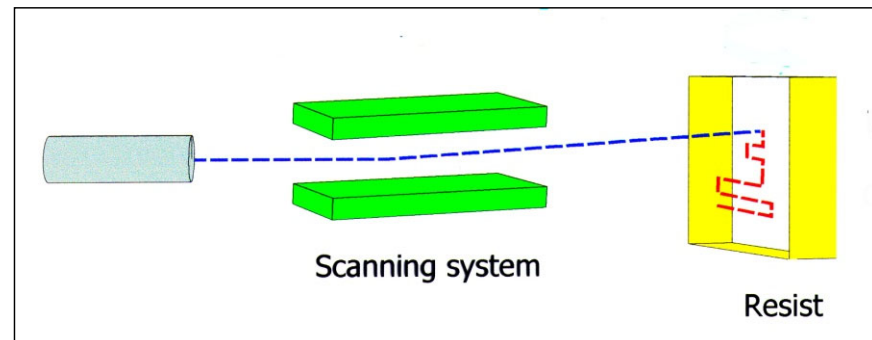


$$DOF = k_2 \frac{\lambda}{(NA)^2}$$

Coherence  $\sigma$  is a critical parameter here to achieve ultimate resolution

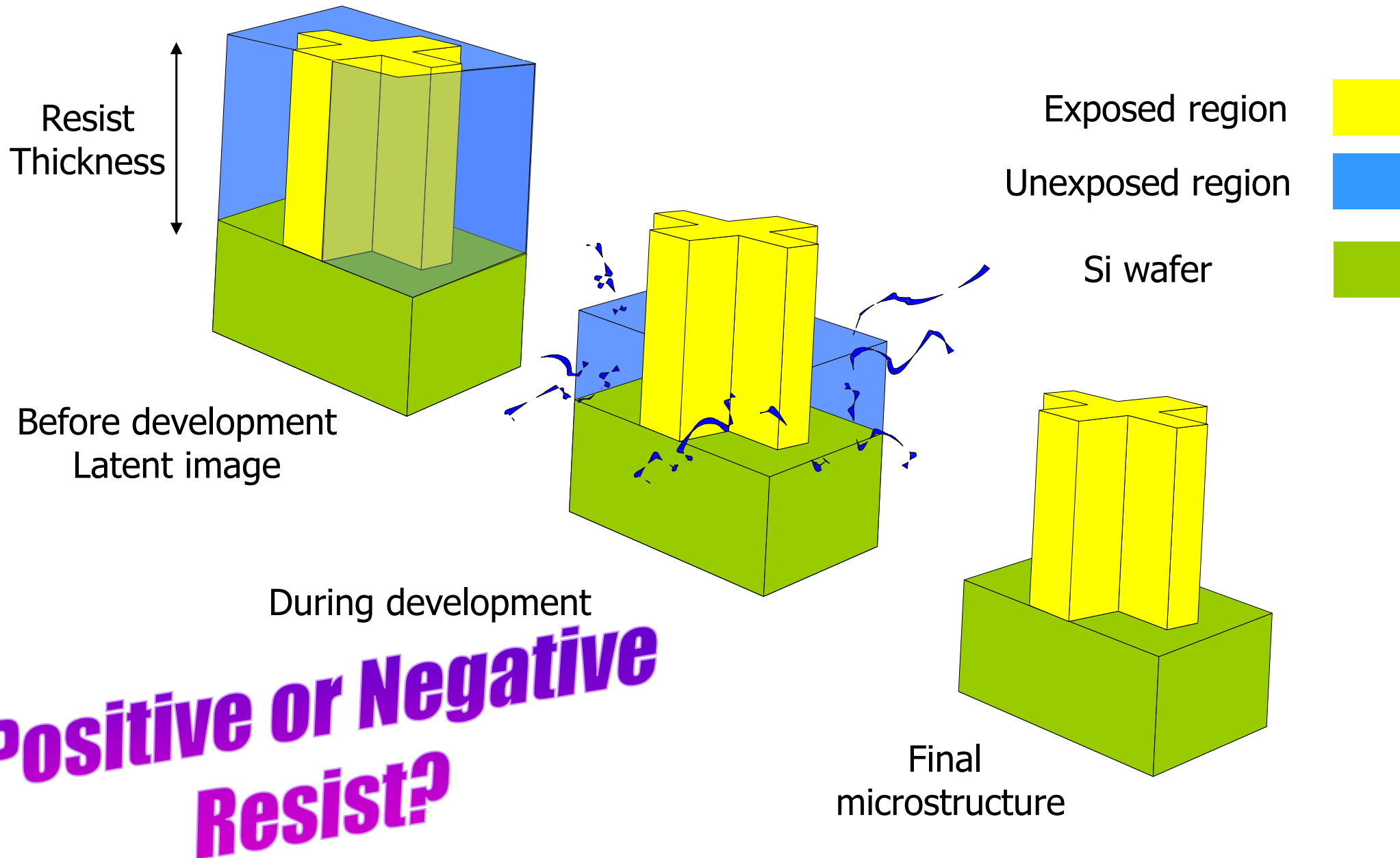
## Direct write processes

- Charged particles



# Optical Lithography

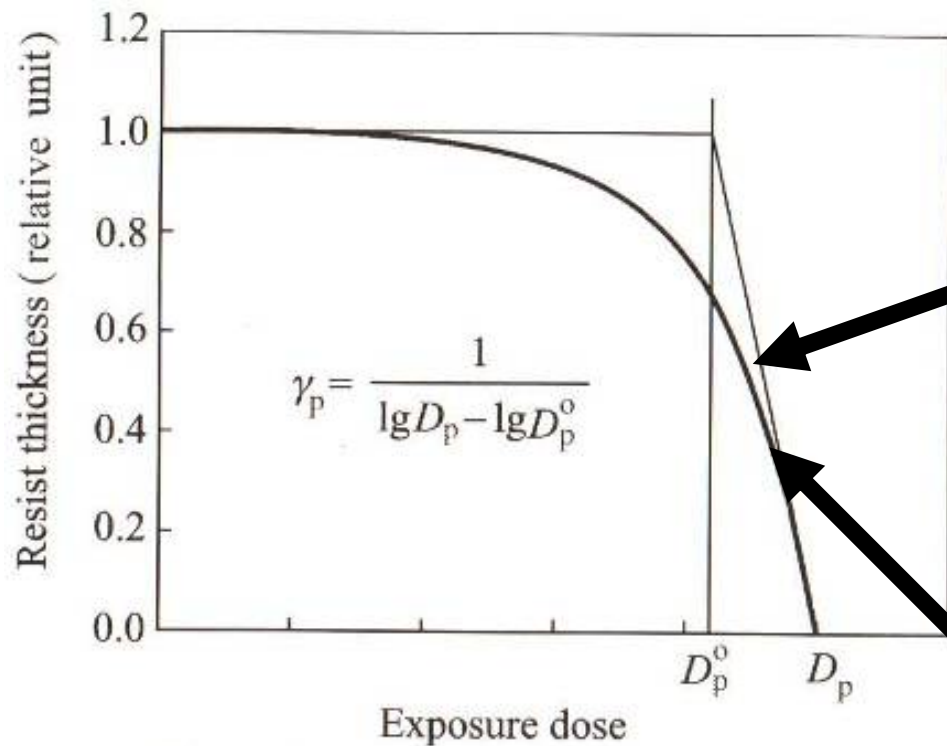
## Chemical development



# Optical Lithography

## Grey-scale

Optical lithography can also produce curved resist profiles; **How?**



$D_p^0$  threshold dose  $D_p$  sensitivity  $\gamma_p$  contrast

(a) Sensitivity and contrast for positive photoresists

→ Using grey scale lithography

In electron beam you can vary the energy deposition per unit area in combination with a **low contrast** resist this give nice control over resist thickness.

In optical lithography we need grey scale masks + **low contrast** resist. Mask requirements:

Opaque pixels in transparent area or transparent pixels in opaque areas  
Here the pixel size < resolution

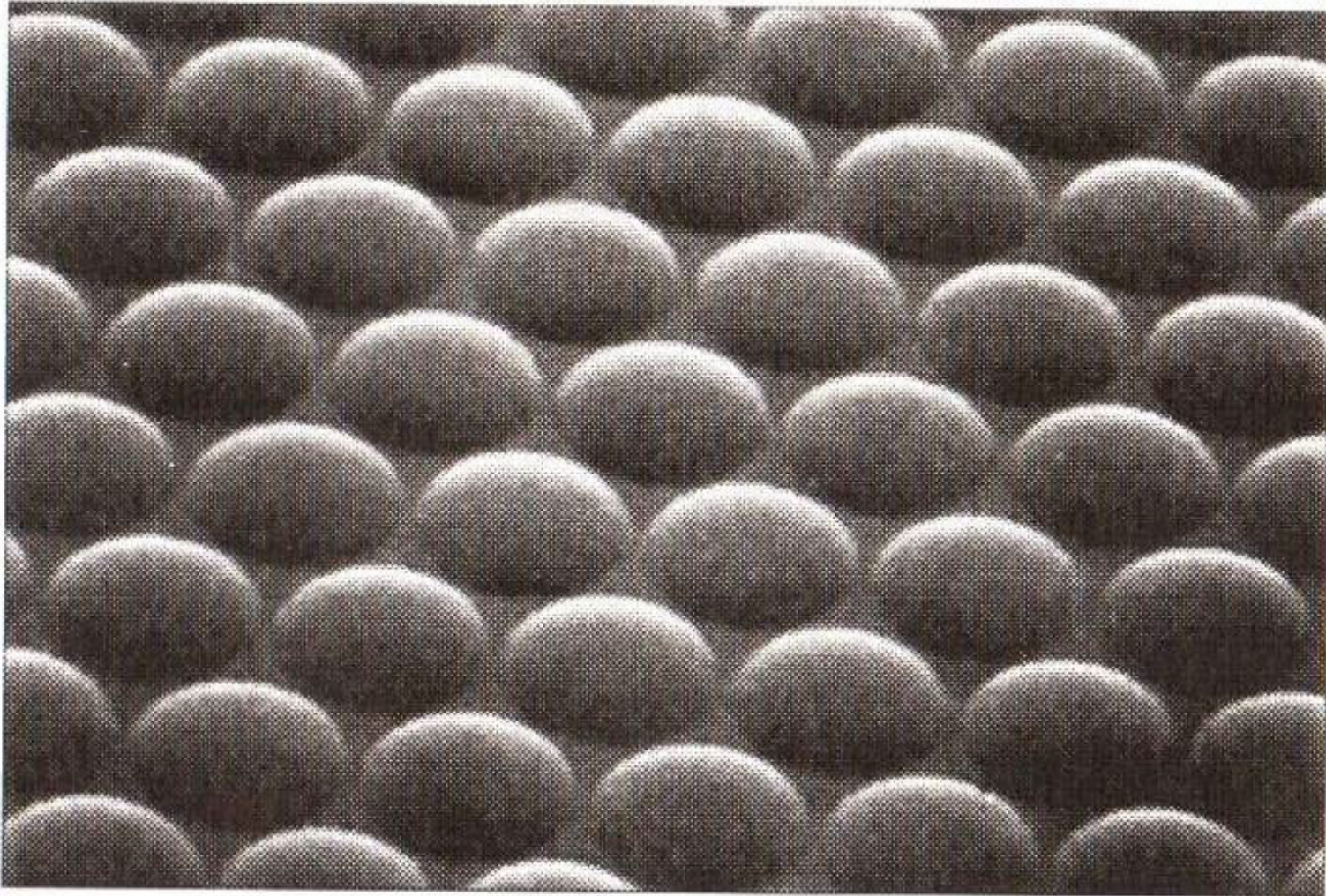
→ Transparency modulation



# Optical Lithography

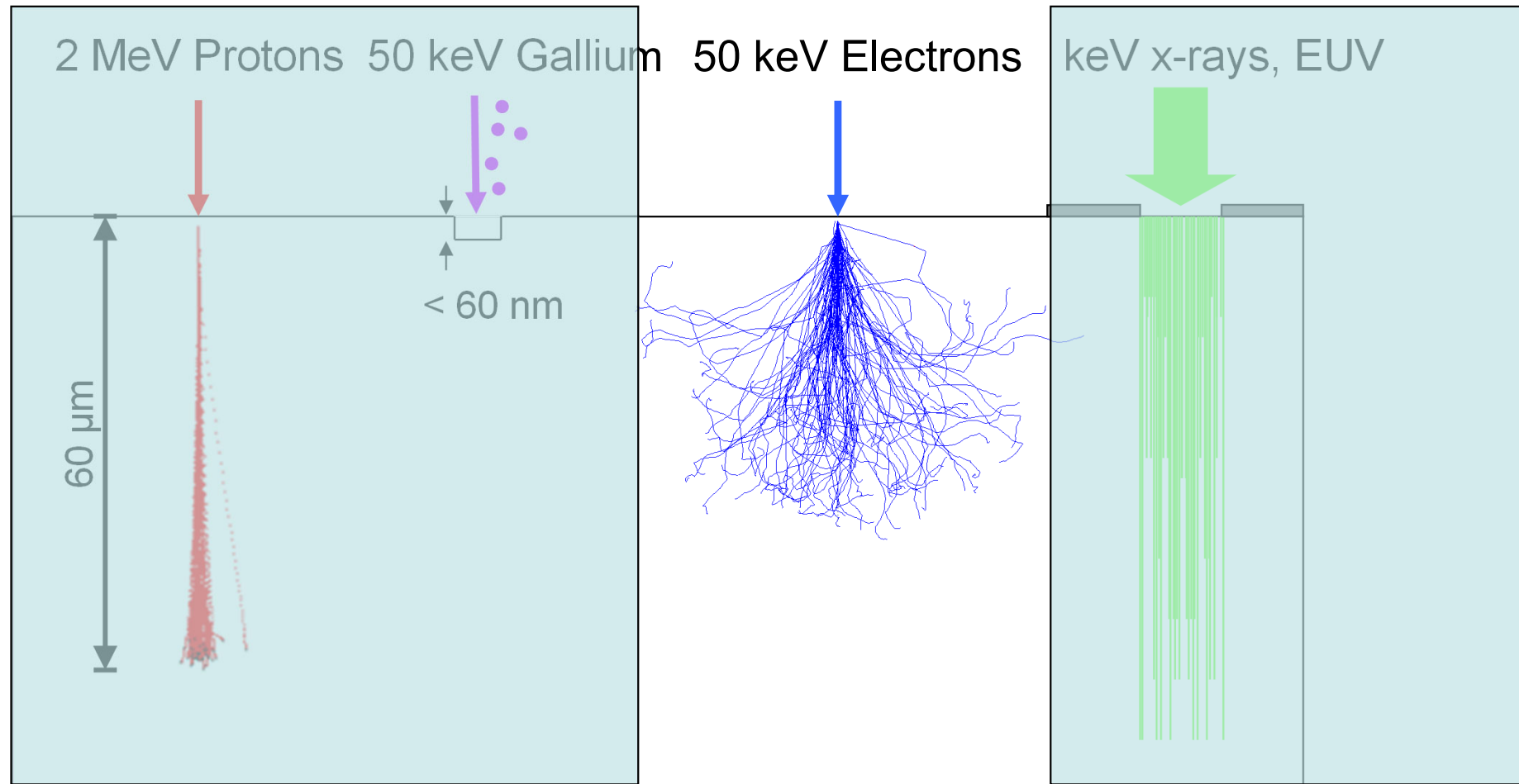
## *Grey-scale*

Application of grey scale lithography can be found in the area of integrated optics:



Micro lens array made by photo lithography through a grey scale mask

# Particle/wave interaction with material



2 MeV protons: Well defined path + Dose homogeneity

(Focused ion beam FIB): Surface atoms removed by sputtering

50 keV electrons: beam broadening below the surface

X-Rays need mask, well defined path: dose exp decay with depth

- **Nanolithography** using particles or waves can be planned if we understand the physics of the interactions.  
E.g. like in Optical lithography, Electron Beam Writing, FIB and Proton Beam Writing.

# ***Electron Beam Lithography / Microscopy***

Other books on electron microscopy:

Scanning Electron Microscopy and X-ray Microanalysis

*Joseph Goldstein, Dale E Newbury, David C Joy, Charles E Lyman,  
Patrick Echlin, Eric Lifshin, LC Sawyer, JR Michael*

Transmission Electron Microscopy: A Textbook for Materials Science

*DB Williams and C Barry Carter*

# E-Beam

## Introduction

**Electron** means we use electrons to form our image. Electrons behave as waves just like light, but have a much shorter wavelength.

**Microscopy** means we are looking at small things

**What is electron microscopy?**

$$\lambda_{light} = \frac{hc}{E} \geq 157nm$$

**Why use electrons not light?**

Electrons have a much shorter wavelength than light. You cannot see anything smaller than half the wavelength of the radiation you are using

$$\begin{aligned} \lambda_e &= \frac{1.226}{\sqrt{V}} (nm) \\ &= h/p = \frac{h}{m_e v} = \frac{h}{\sqrt{2em_e U}} = \frac{6.63 \times 10^{-34}}{\sqrt{2 \times 9.1 \times 10^{-31} \times 10^3 \times 1.6 \times 10^{-19}}} = \frac{6.63 \times 10^{-34}}{1.71 \times 10^{-23}} = 3.86 \times 10^{-11} m \end{aligned}$$



# E-Beam

## *Techniques and acronyms*

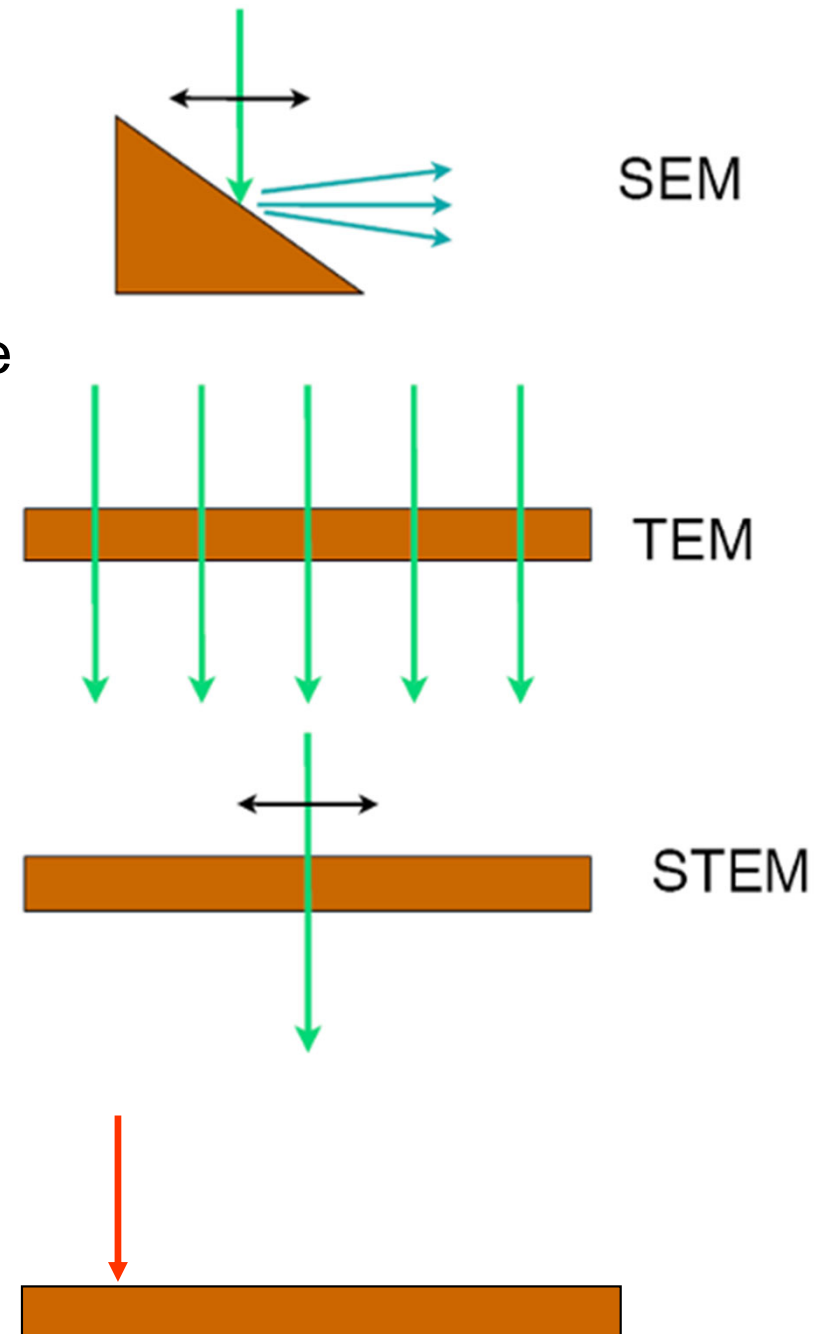
**EM:** Electron microscopy. Covers TEM, SEM, STEM, etc

**SEM:** Scanning electron microscopy. Collect the secondary electrons emitted from the surface.

**TEM:** Transmission electron microscopy

**STEM:** Scanning transmission electron microscopy. Like TEM, but scan a finely focused beam of electrons across the specimen rather than image using a broad beam

**E-beam (e-beam) lithography**



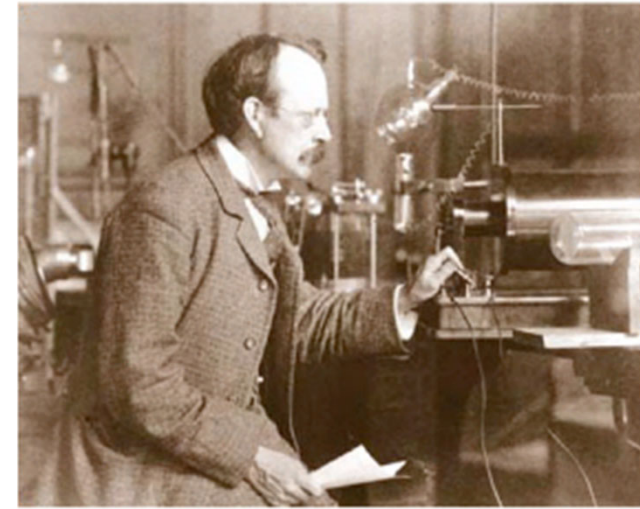
# E-Beam History

**Optical microscopy:** Resolution limited by wavelength of light to  $\sim 300\text{nm}$   
Other radiation (X-rays,  $\gamma$ -rays) cannot be focused.

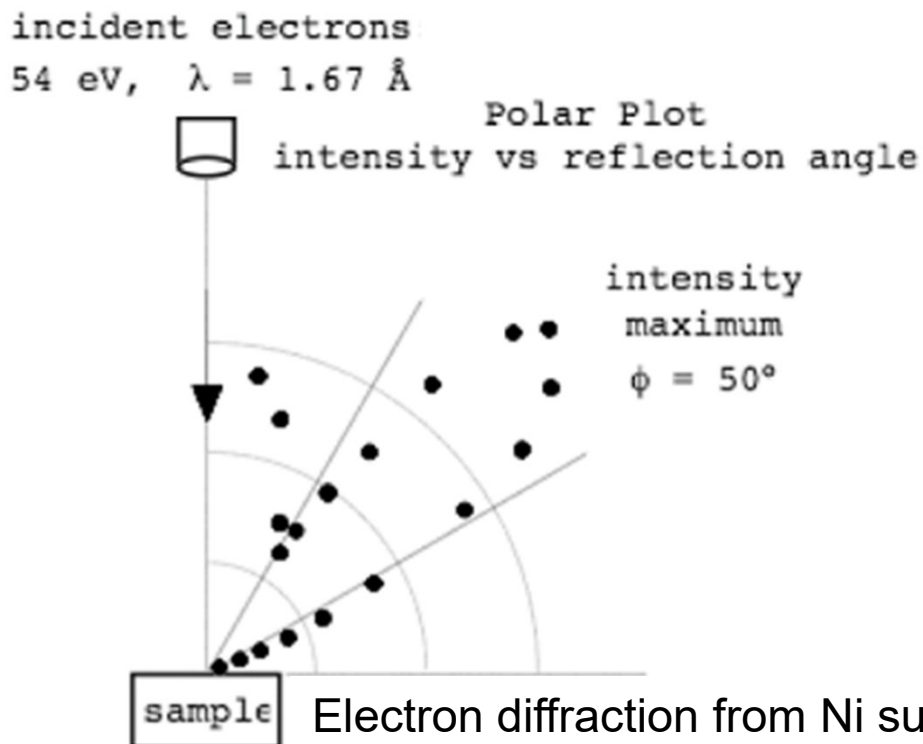
**1897:** JJ Thompson discovers the electron

**1925:** de Broglie proposes electrons are waves with small wavelength

**1927:** Electron diffraction demonstrated by CH Davisson and Lh Germer (reflection) and GP Thompson and A Reid (transmission)



JJ Thompson, Cavendish Labs



Electron diffraction from Ni surface Davisson and Germer



Davisson and Germer

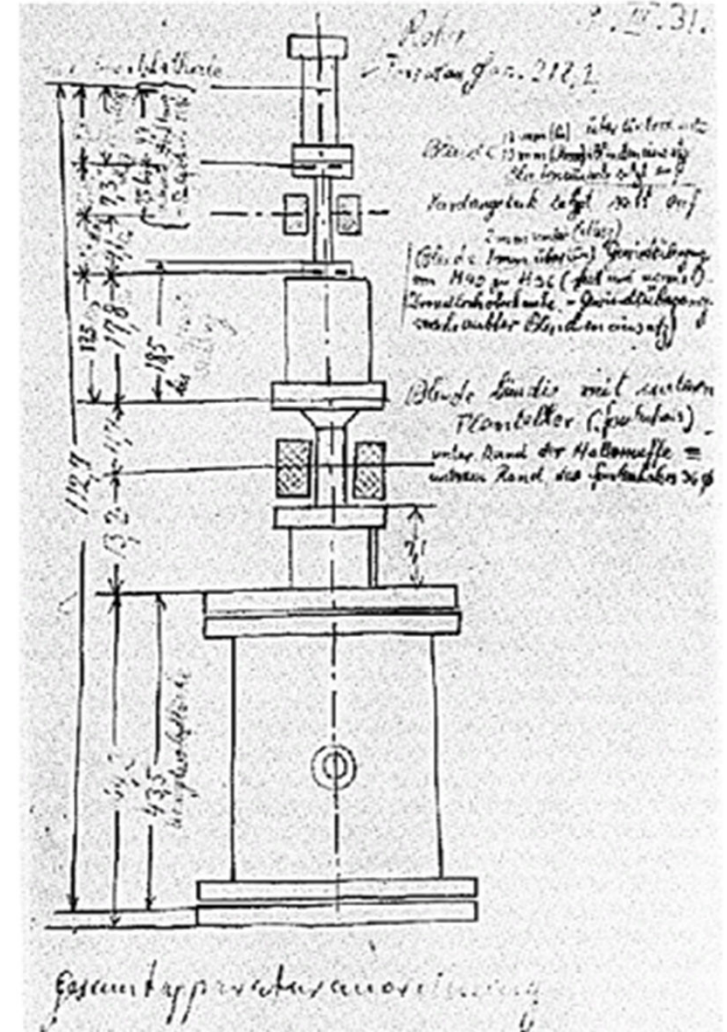
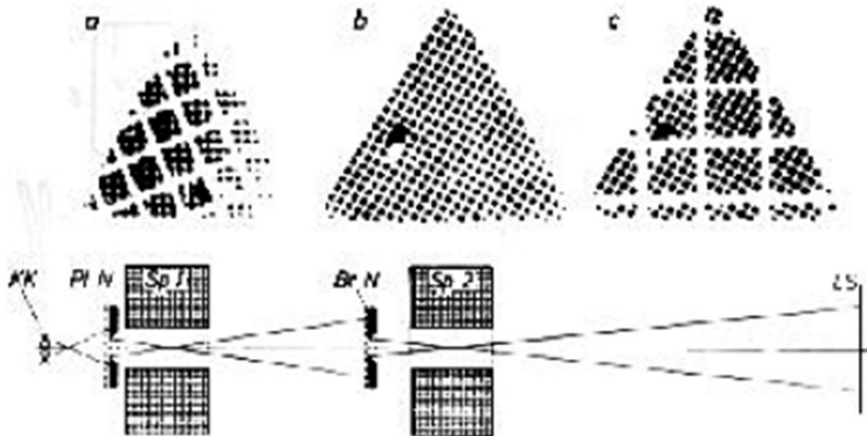
# E-Beam History

**1931:** M Knoll and E Ruska build first electron microscope

M Knoll and E Ruska, Das Elektronenmikroskop. Z. Physik **78** (1932) 318–339



Ruska &  
Knoll,  
1931



Ruska's sketch of first TEM

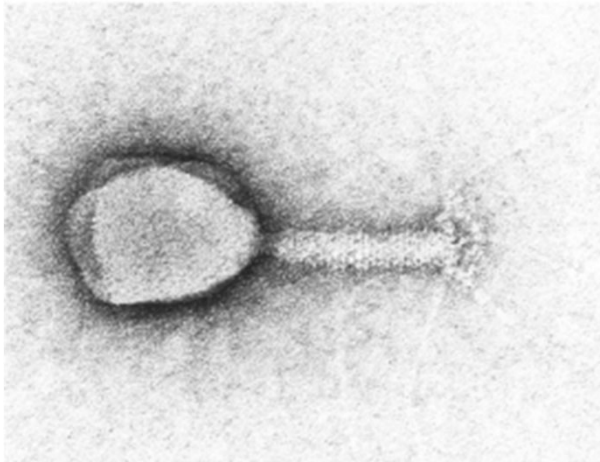
First TEM image, magnification  $17.4\times$ , 50kV

[http://ernst.ruska.de/daten\\_e/library/documents/999.nobellecture/lecture.html](http://ernst.ruska.de/daten_e/library/documents/999.nobellecture/lecture.html)



# E-Beam History

- 1934:** Resolution of electron microscope better than light microscope – Driest & Muller
- 1936:** First commercial TEM – Metropolitan-Vickers AEI EM1
- 1938:** First practical commercial TEM – von Borries & Ruska, Siemens. 10 nm resolution. M von Ardenne builds first STEM
- 1940:** RCA TEM, 2.4 nm resolution
- 1941:** First electron micrographs of viruses
- 1942:** First SEM built by Zworykin et al
- 1945:** Resolution 1 nm



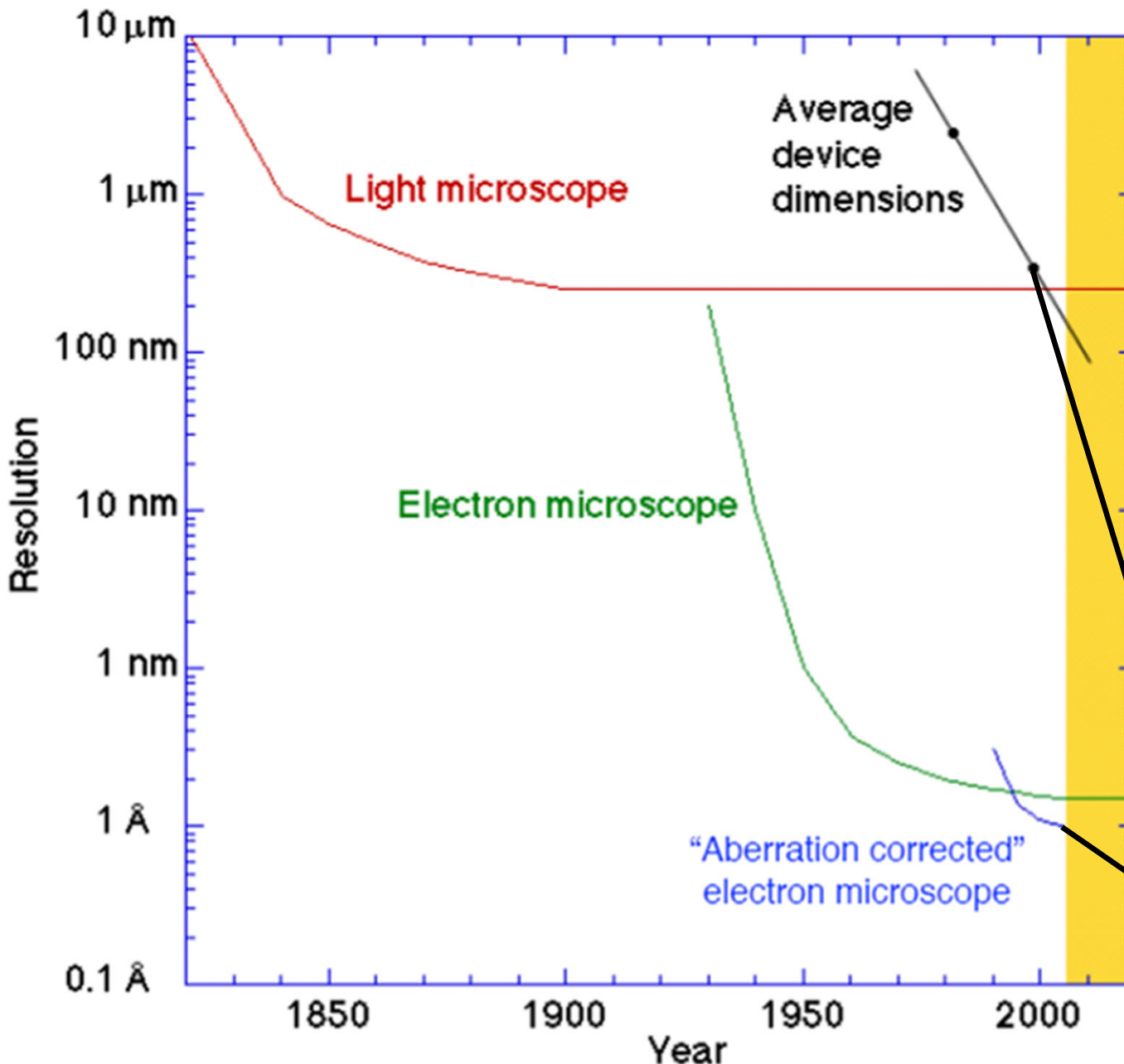
Luria and Anderson, first TEM image of a bacteriophage, 1942



Siemens TEM

# E-Beam History

## Electron microscope resolution



**Light microscope:**  
resolution limit  $\sim 300\text{nm}$

**Electron microscope:**  
resolution limit should be  
 $\sim 0.001\text{nm}$

**TEM:** resolution limit  
 $\sim 0.05\text{nm}$  "Aberration  
corrected" EM

Why such a huge  
difference?

Atomic spacing



# E-Beam

## Ultimate resolution

### Secondary electron scattering

Types:

- ❖ Forward & Backward by primary beam  
→ Proximity effects

### **1983 STEM (2-3 nm beam spot)**

was used to write in 10 nm thick resist  
with 300 keV electron beam

This gave **10 nm features**

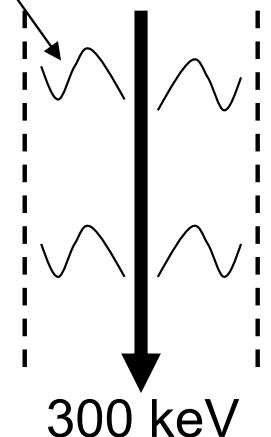
Low energy electrons are needed for lithography,  
they break the bonds in the resist chains!

Low energy secondary  
electrons 80% <200 eV

Radius: 5 nm  
Diameter: 10 nm

### **Solution Resist Processing**

4 nm line width has been achieved through ultrasonic resist development. Use energy close to minimum exposure dose.



# E-Beam

## Electron scattering and proximity effect

Besides the system resolution and resist performance the **scattering of the electrons** in the resist is of crucial importance for the final resolution

Through what mechanism will the incoming electron beam lose its energy?

Electronic and Nuclear scattering

Which one is more important for lithography?

Electronic because of large cross section (ie high likely hood)

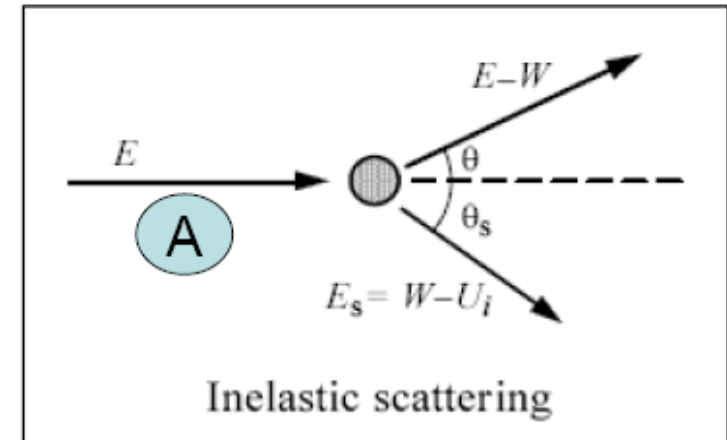
# E-Beam

## How do electrons interact with matter?

- I Inelastic scattering on atomic orbital electrons →

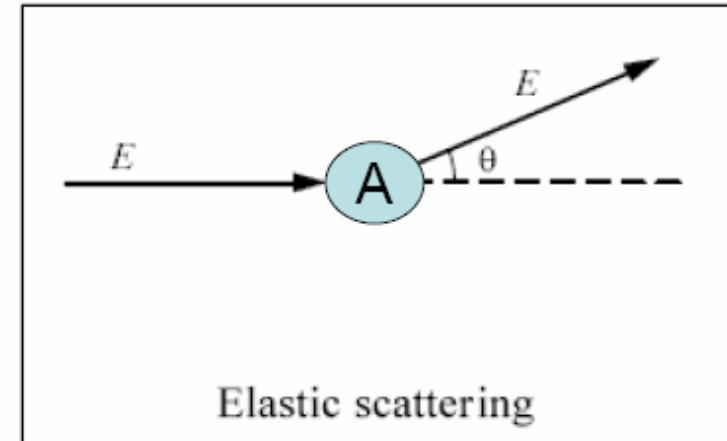
Excitations + Ionization of Atoms

Collision Stopping Power



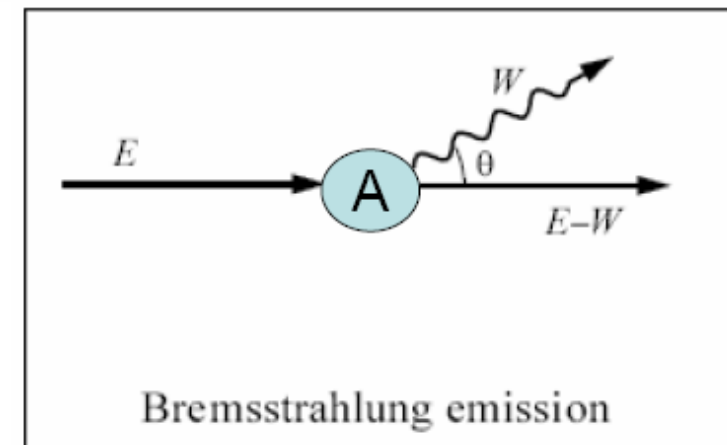
- II Elastic scattering on atoms without significant energy exchange

*Larger atoms (with a greater atomic number, Z) have a higher probability of producing an elastic collision because of their greater cross-sectional area*



- II Inelastic nuclear scattering. This results in radiation which is known as Bremsstrahlung

Radiative Stopping Power

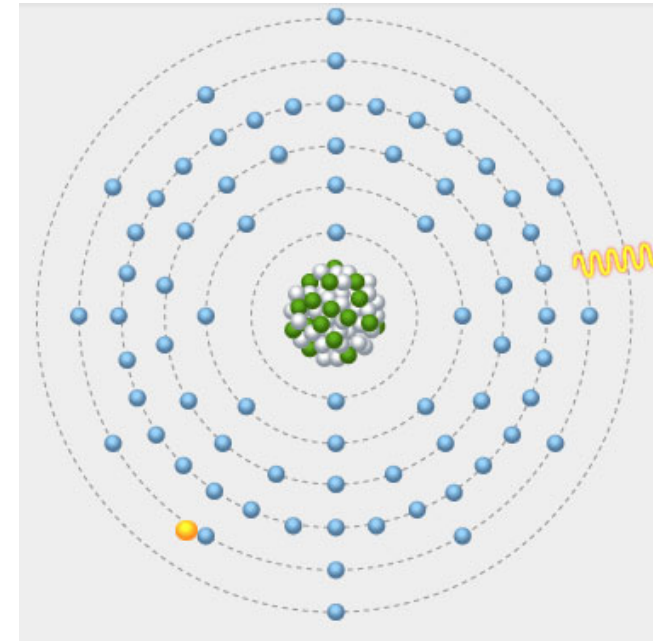


# E-Beam / X-rays

## X-ray spectroscopy

When an electron hits a material X-rays are formed by 2 processes:

- 1) **Bremsstrahlung**
- 2) **Characteristic X-rays/photons**



### Bremsstrahlung process

Caused by electrons being decelerated. Contains all energies from 0 to beam energy  
Intensity given by Kramers law

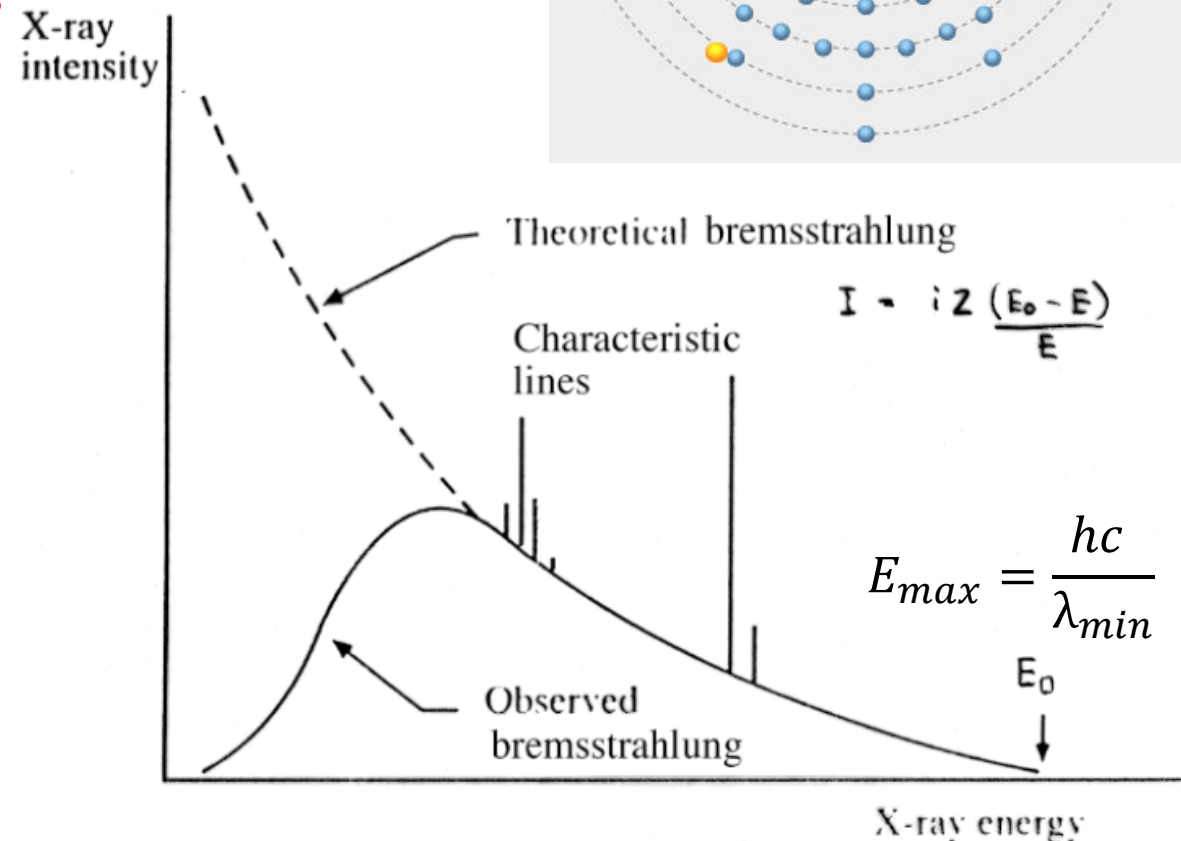
$$I = \frac{iZ(E_0 - E)}{E}$$

where

$i$  = beam current

$Z$  = average atomic number

$E_0$  = incident electron beam energy



*Why does the observed X-ray intensity drop?*

# E-Beam / X-rays

## X-ray spectroscopy (Characteristic X-rays)

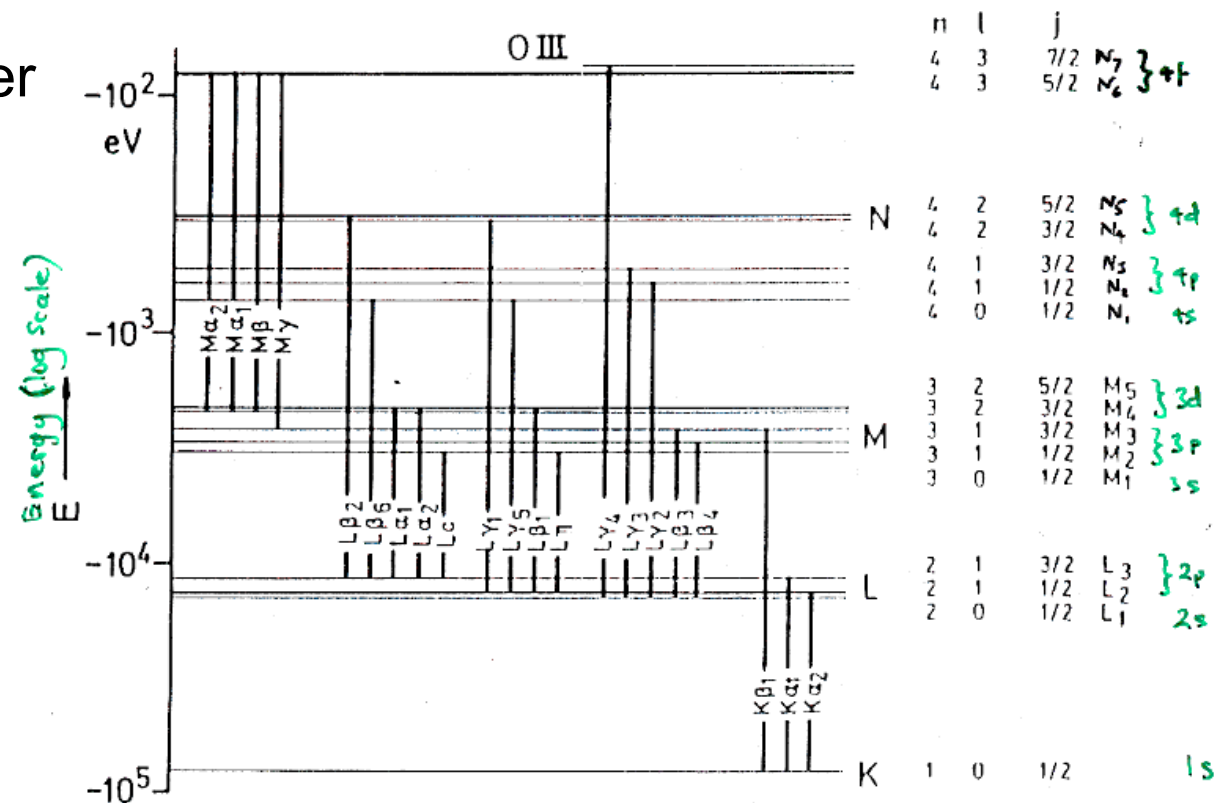
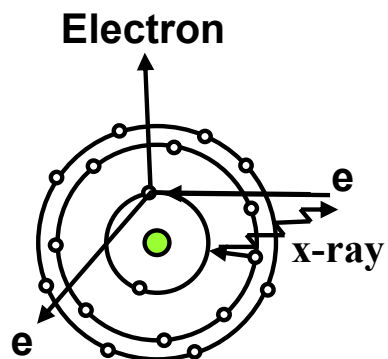
Primary electron removes electron from inner shell of target atom putting an ion in an excited state

Ion loses energy by outer shell electron falling into vacancy

Excess energy emitted as either an X-ray or an Auger Electron

Probability of X-ray emission given by fluorescence yield,  $\omega_k, \omega_l, \omega_m$

$\omega$  small for low Z



K, L and M characteristic lines for Au



# E-Beam / X-rays

## Energy dispersive X-ray spectroscopy (EDX)

Typical limits of detection 0.1%

