Monoatomic ideal gas

$$\mathcal{H} = \sum_{i=1}^{n} \frac{1}{2n} \overrightarrow{p_i}^2 + \sum_{i < j} V(|\overrightarrow{r_i} - \overrightarrow{r_j}|)$$

for 
$$V=0$$

$$\Omega(\mathcal{E}_{1}V,N;\delta\mathcal{E}) = \left(\frac{m}{2}\right)^{1/2} C_{3N}(2m)^{3\nu/2-1} V^{N} \mathcal{E}^{3/2N-1} \delta\mathcal{E}$$

$$\frac{1}{N}\ln\left(\Omega(E,V,N;\delta E)\right) = \left(\frac{3}{2} - \frac{1}{N}\right)\ln(u) + \ln(v) + \left(\frac{3}{2} - \frac{1}{N}\right)\ln(2m)$$

$$+\frac{1}{N}\ln(C_{3N})+\ln(N)+\left(\frac{3}{2}-\frac{1}{N}\right)\ln(N)+\frac{1}{2N}\ln(\frac{m}{2})$$

$$\lim_{N\to\infty}\frac{1}{N}\ln(\delta\epsilon)=0$$

$$S(U,V) = \lim_{N \to \infty} \frac{1}{N} K_B \ln(\Omega(E,V,N;\delta E))$$

$$= \frac{3}{2} K_B | n(u) + K_B | n(v) + S_0$$

$$S_0 = \frac{3}{2} \ln(2m) + \frac{1}{N} \ln(C_{3N}) + \ln(N) + \frac{3}{2} \ln(N)$$

$$\frac{1}{T} = \frac{3S}{\delta u} = \frac{3K_B}{2u} \longrightarrow u = \frac{3}{2}K_BT$$

$$\frac{p}{T} = \frac{\partial S}{\partial V} = \frac{K_B}{V} \longrightarrow pV = K_BT$$

$$\frac{2\delta}{M} = \frac{3s}{T}$$

$$S(E,V,N;\delta E) = K_B \ln(\Omega(E,V,N;\delta E))$$

$$= \frac{3}{3} K_B N \ln(E) + K_B N \ln(V) + f(N,\delta E)$$

$$\frac{1}{T} = \frac{8S}{8E} = \frac{3}{2} \frac{NK_8}{E} \longrightarrow E = \frac{3}{2} K_8 TN$$

$$\frac{p}{T} = \frac{35}{3V} = \frac{KBN}{V} \longrightarrow pV = KBTN$$

state equation

As 
$$C_{3N} = \frac{2\pi}{2\pi} \sim N^{-3/2N} b^{N}$$

$$\frac{1}{N} \ln(C_{3N}) = \frac{3}{2} \ln(N) + \ln(b)$$

what about if

$$C_{3N} \longrightarrow C_{3N} \sim N^{\frac{-3/2N}{N}} \stackrel{N}{b} = N^{\frac{5}{2N}} \stackrel{N}{O}^{N}$$

then,

finally