Computational Physics 2023

Sommersemester, 3th April, 2022 – 14th Juli, 2022

- 1)Introduction
- 2) Numbers and errors
- 3) Differentiation and integration
- 4) Ordinary differential equations
- 5) Molecular dynamics simulations
- 6) Partial differential equations
- 7) Iteration processes
- 8) Matrixdiagonalisation & Eigenvalue problems
- 9)Minimization
- 10) Random numbers
- 11) Monte Carlo (MC) Simulations
- 12)Perkolation
- 13)Stochastic Dynamics



Atomistic force-fields

$$U(R) = \sum_{bonds} k_r (r - r_{eq})^2 \qquad bond$$

$$+ \sum_{angles} k_{\theta} (\theta - \theta_{eq})^2 \qquad angle$$

$$+ \sum_{dihedrals} k_{\phi} (1 + \cos[n\phi - \gamma]) \qquad dihedral$$

$$+ \sum_{impropers} k_{\omega} (\omega - \omega_{eq})^2 \qquad improper$$

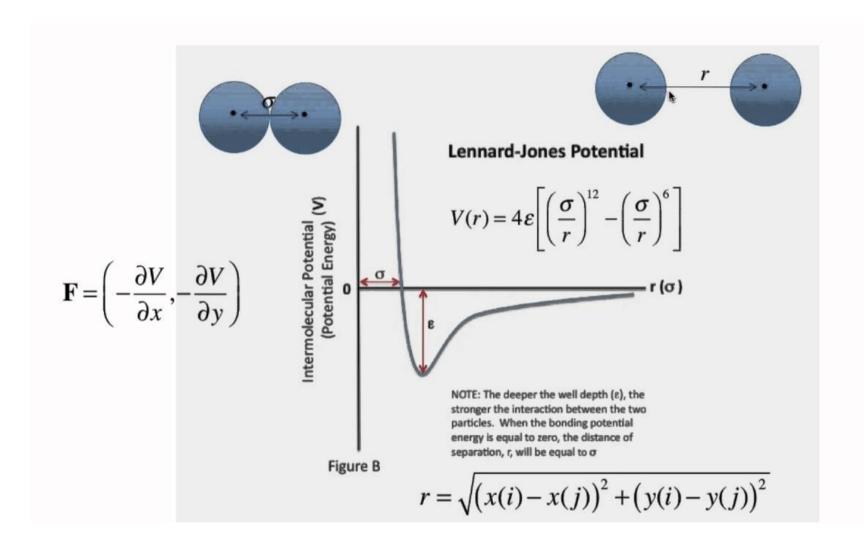
$$+ \sum_{i < j} k_{\omega} (\omega - \omega_{eq})^2 \qquad van \ der \ Waals$$

$$+ \sum_{i < j} \epsilon_{ij} \left[\left(\frac{r_m}{r_{ij}} \right)^{12} - 2 \left(\frac{r_m}{r_{ij}} \right)^6 \right] \qquad van \ der \ Waals$$

$$+ \sum_{i < j} \frac{q_i q_j}{4\pi\epsilon_0 r_{ij}} \qquad electrostatic$$



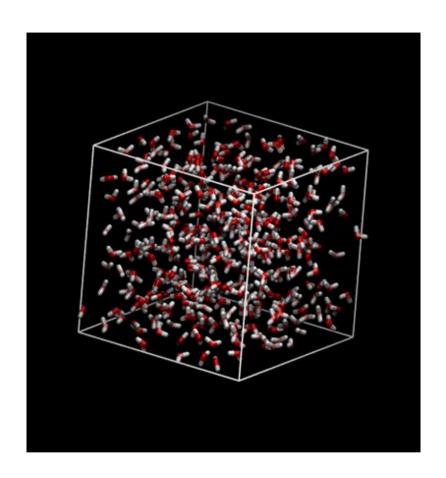
Lennard-Jones





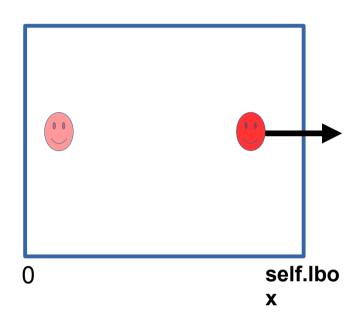


The simulation box: Periodic boundaries



- Fluid-air (Fluid-vacuum) interfaces are characterized by a substantial surface tension. The water-air surface tension is about 70 mN/m or about 18 k_BT/nm².
- Periodic boundaries form an excellent and widely used method to obtain bulk properties of fluids (no surface tension) within small, finite systems.

Periodic boundary Conditions (PBC)



In the "Argon code":

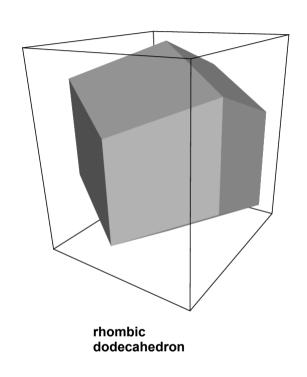
```
# Update current positions (applying PBC)
      if newX < 0:
        self.atoms[atom].x = newX + self.lbox
      elif newX > self.lbox:
        self.atoms[atom].x = newX - self.lbox
      else:
        self.atoms[atom].x = newX
      if newY < 0:
        self.atoms[atom].y = newY + self.lbox
      elif newY > self.lbox:
        self.atoms[atom].y = newY - self.lbox
      else:
        self.atoms[atom].y = newY
      if newZ < 0:
        self.atoms[atom].z = newZ + self.lbox
      elif newZ > self.lbox:
        self.atoms[atom].z = newZ - self.lbox
      else:
        self.atoms[atom].z = newZ
```

Different simulation boxes

All space-filling periodic boxes can be described by a 3x3 matrix:

Table 6 Overview over different box type	Table 6	Overview	over	different	box	type
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	image	box	box vectors box vector angle			ales		
box type	distance			b	С	∠ bc	∠ ac	∠ ab
cubic	d	d^3	d	0	0	90°	90°	90°
			0	d	0			
			0	0	d			
rhombic	d	$\frac{1}{2}\sqrt{2} d^3$	d	0	$\frac{1}{2}$ d	60°	60°	60°
dodcahdron		$0.707\ d^3$	0	d	$\frac{1}{2}$ d			
(xy-square)			0	0	$\frac{1}{2}\sqrt{2} d$	-		
rhombic	d	$rac{1}{2}\sqrt{2}\ d^3$	d	$\frac{1}{2}$ d	$\frac{1}{2}$ d	60°	60°	60°
dodcahdron		$0.707d^3$	0	$\frac{1}{2}\sqrt{3} d$	$\frac{1}{6}\sqrt{3} d$	-		
(xy- hexagon)			0	0	$\frac{1}{3}\sqrt{6} d$	-		
truncated	d	$\frac{4}{9}\sqrt{3} d^3$	d	$\frac{1}{3}d$	$-\frac{1}{3} d$	71.53°	109.47°	71.53°
octahedron		$0.770 d^3$	0	$\frac{2}{3}\sqrt{2} d$	$\frac{1}{3}\sqrt{2} d$	_		
		$0.770 \ d^3$	0	0	$\frac{1}{3}\sqrt{6} d$	-		



- The matrix is asymmetric with 0,0,0 below the diagonal. This assures that the 3 box vectors are linear independent!
- The volume of the rhombic dodcahdron is 71% of the volume of a cube having the same image distance. This typically shaves of 29% of solvent and thus speeds up the simulation by a factor of 1.3.

Atomistic force-fields: The main computational cost

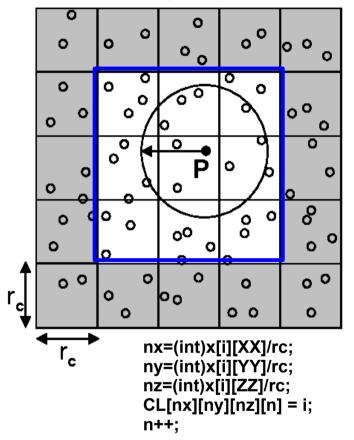
$$\begin{split} U(R) &= \sum_{bonds} k_r \big(r - r_{eq}\big)^2 & bond \\ &+ \sum_{angles} k_{\theta} \big(\theta - \theta_{eq}\big)^2 & angle \\ &+ \sum_{dihedrals} k_{\phi} \big(1 + \cos[n\phi - \gamma]\big) & dihedral \\ &+ \sum_{l \neq j} k_{\omega} \big(\omega - \omega_{eq}\big)^2 & improper \\ &+ \sum_{i \neq j} \epsilon_{ij} \left[\left(\frac{r_m}{r_{ij}}\right)^{12} - 2 \left(\frac{r_m}{r_{ij}}\right)^{6} \right] & van \ der \ Waals \\ &+ \sum_{i \neq j} \frac{q_i q_j}{4\pi \varepsilon_0 r_{ij}} & electrostatic \\ \end{split}$$

 Nonbonded pair-wise interactions are not cheap since there exists 1/2N²-N unique pairs in total. This quadratic scaling is killing the performance!

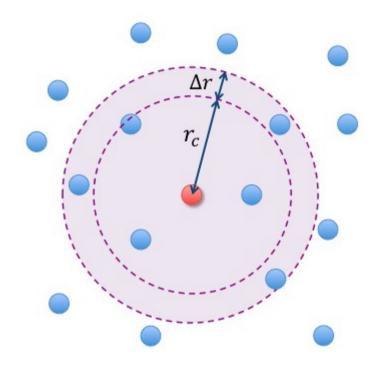


Simulation cutoff

Cell list: Finding neighbors efficiently



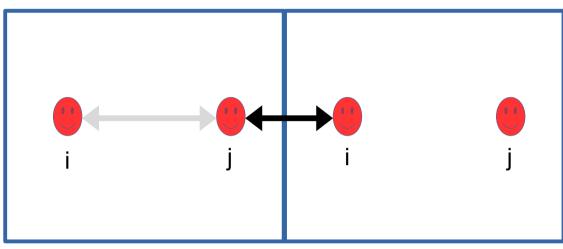
Verlet list:



- The cell list can be used to efficiently generate a Verlet list.
 A substantial volume mismatch between the cutoff sphere and the corresponding subcells exists. Updating of the cell list itself, however, scales with N_{global}.
- The Verlet list (scales with N_{local}^2) conserves all neighbors until a particle moves beyond "its skin", $r_c + \Delta r$. This check is inexpensive since these distances are anyway calculated to assign the forces.
- Infrequent updating makes the simulation faster but is also a source of noise.

Periodic boundary Conditions (PBC)

$$R_c < \frac{1}{2}\min(\|\mathbf{a}\|, \|\mathbf{b}\|, \|\mathbf{c}\|),$$



In the "Argon code":

def calculateForce(self, atom1, atom2):

"""Calculates the force between two atoms using LJ 12-6 potential"""
Calculate distance between two atoms

dx = self.atoms[atom1].x - self.atoms[atom2].x

dy = self.atoms[atom1].y - self.atoms[atom2].y

dz = self.atoms[atom1].z - self.atoms[atom2].z

Minimum Image Convention

dx -= self.lbox*round(dx/self.lbox)

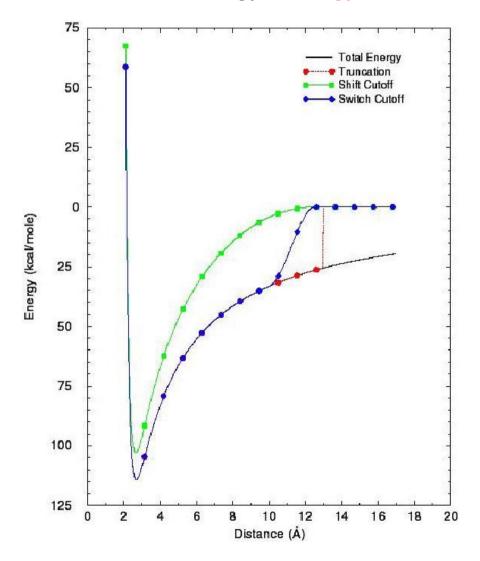
dy -= self.lbox*round(dy/self.lbox)

dz -= self.lbox*round(dz/self.lbox)

 The minimum image convention implies that the cut-off radius used to truncate non-bonded interactions may not exceed half the shortest box vector.

Simulation cutoff: Lennard-Jones

- The smaller the cutoff the more efficient the speed up. In atomistic simulations the typical 'cutoff for Lennard-Jones interactions is about 1.0-1.2 nm.
- Truncation of energy! Energy derivatives must go to zero at the cutoff!



$$\mathbf{F}(\mathbf{x_i}) = -\nabla V(\mathbf{x_i})$$

"Shift" and "Switch" Cutoffs are achieved via addition of the function S(r), if r1 = 0 -> "Shift" Cutoff else "Switch" Cutoff:

$$F_s(r) = F_{\alpha}(r)$$
 $r < r_1$
 $F_s(r) = F_{\alpha}(r) + S(r)$ $r_1 \le r < r_c$
 $F_s(r) = 0$ $r_c \le r$

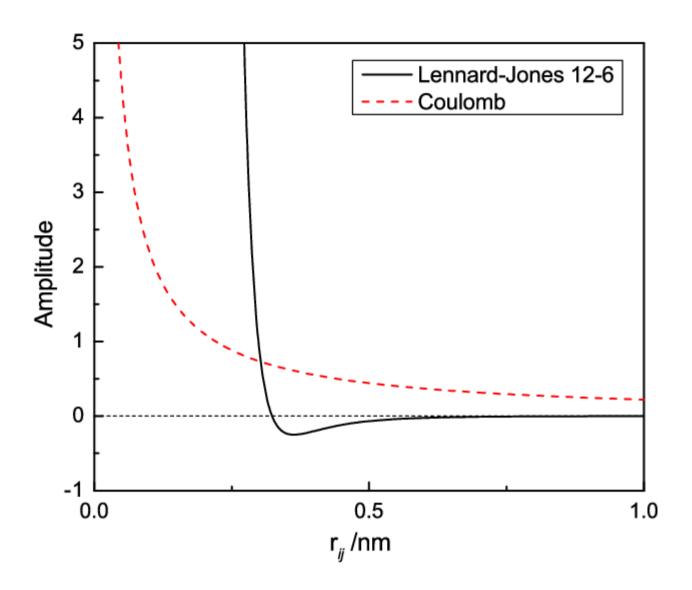
$$S_V(r; r_1, r_c) = \frac{1 - 10(r - r_1)^3 (r_c - r_1)^2 + 15(r - r_1)^4 (r_c - r_1) - 6(r - r_1)}{(r_c - r_1)^5}$$

Atomistic force-fields

How expensive are chemical bonds?

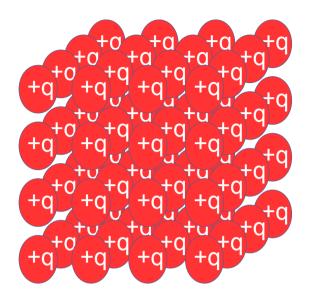
$$\begin{split} U(R) &= \sum_{bonds} k_r \left(r - r_{eq}\right)^2 & bond \\ &+ \sum_{angles} k_{\theta} \left(\theta - \theta_{eq}\right)^2 & angle \\ &+ \sum_{dihedrals} k_{\theta} \left(1 + \cos[n\phi - \gamma]\right) & dihedral \\ &+ \sum_{i < j} k_{\omega} \left(\omega - \omega_{eq}\right)^2 & improper \\ &+ \sum_{i < j} \varepsilon_{ij} \left[\left(\frac{r_m}{r_s}\right)^{12} - 2\left(\frac{r_m}{r_y}\right)^{6} \right] & van \ der \ Waals \\ &+ \sum_{i < j} \frac{q_i q_j}{4\pi\varepsilon_0 r_{ij}} & electrostatic \\ \end{split}$$

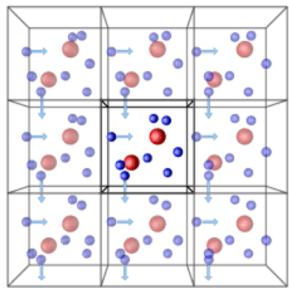
• Electrostatic interactions only scale with 1/r: The long-range interactions are significant! Shifting or switching neglects these contributions.





Electrostatic interactions: Monopoles





periodic boundary conditions (PBC)

- Coulomb explosion: Potential energy builds up quicker (r³) than the decay (1/r)!
- Accurate description of systems with a net charge is generally challenging in periodic boundary conditions.



Electrostatic interactions: Multipoles

$$V_{total} = V_{mono} + V_{dipole} + V_{quad} + V_{octo} + \dots$$

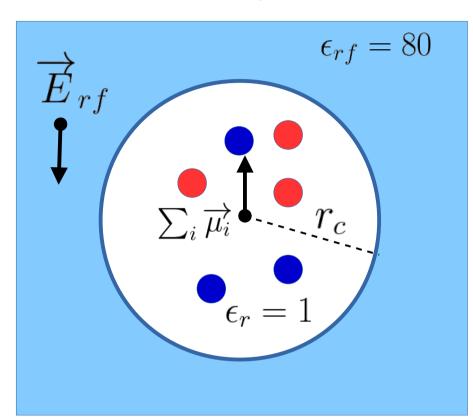
monopole dipole $\frac{1}{r}$ $\frac{1}{r}$ $\frac{1}{r^2}$ $\frac{1}{r^2}$ $\frac{1}{r}$ $\frac{1}{r}$ $\frac{1}{r}$ $\frac{1}{r}$

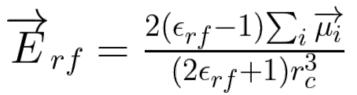
• In neutral systems the contribution of monopoles to the potential vanishes and the dipole terms become dominant!

 $\sum q = 0 \longrightarrow V_{total} \approx V_{dipole}$

Electrostatics: Onsager reaction field

- 1) Direct interactions with other particles in R.
- 2) Treat material outside sphere as continuum dielectric.







Lars Onsager

$$V_{crf} = f \frac{q_i q_j}{\varepsilon_r r_{ij}} \left[1 + \frac{\varepsilon_{rf} - \varepsilon_r}{2\varepsilon_{rf} + \varepsilon_r} \frac{r_{ij}^3}{r_c^3} \right] - f \frac{q_i q_j}{\varepsilon_r r_c} \frac{3\varepsilon_{rf}}{2\varepsilon_{rf} + \varepsilon_r}$$

- Instantaneous response of the dielectric to the dipole in the sphere. Some variants of this method introduce a relaxation time.
- The method works well in isotropic systems with large electrostatic screening due to the presence of many ion pairs (high ionic strength → small Debye length).

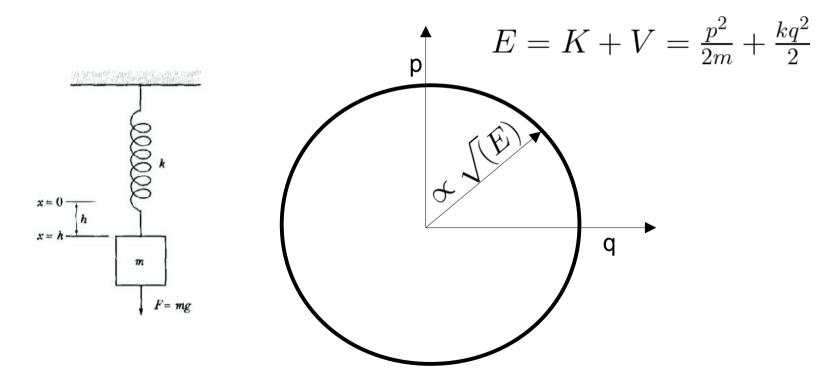


Fluctations in the canonical ensemble

$$\dot{q} = \frac{\partial H}{\partial p}$$

$$\dot{p} = -\frac{\partial H}{\partial q}$$

Hamiltons equations of motion are energy conserving (NVE)



Fluctations in the canonical ensemble

The general proof:

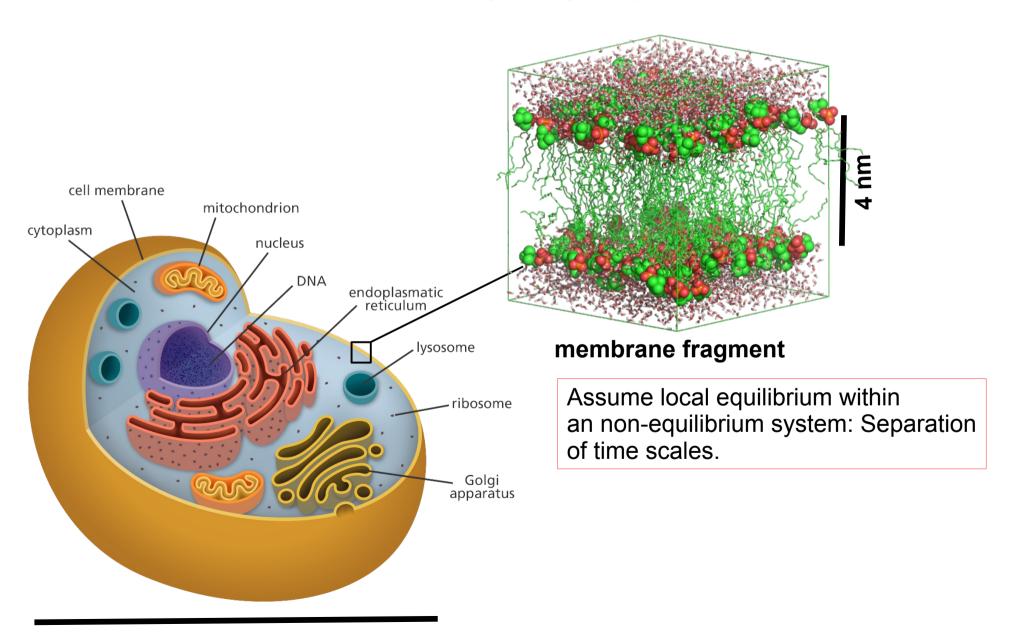
$$dH(p,r) = \frac{\partial H}{\partial p} dp + \frac{\partial H}{\partial r} dr$$

$$\frac{\partial H}{\partial p} = \dot{r} \qquad \frac{\partial H}{\partial r} = -\dot{p}$$

$$\frac{dH(p,r)}{dt} = \frac{\partial H}{\partial p} \dot{p} + \frac{\partial H}{\partial r} \dot{r} = \dot{r}\dot{p} - \dot{p}\dot{r} = 0$$

Fluctations in the canonical ensemble

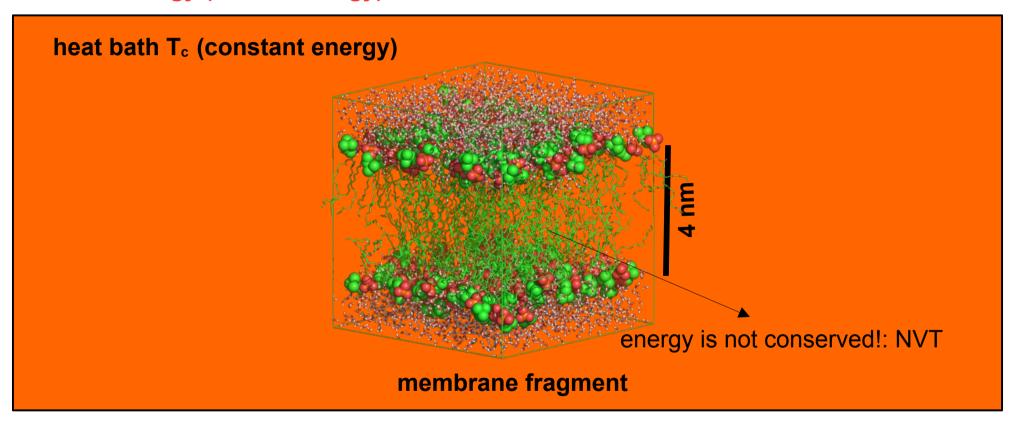
Simulation of "living" biological systems?

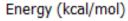


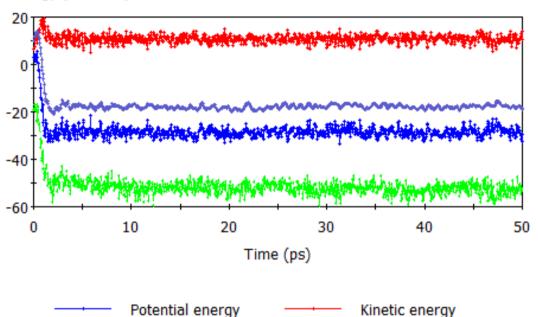
 $100000 \text{ nm} = 100 \mu \text{m}$

Constant temperature in finite systems

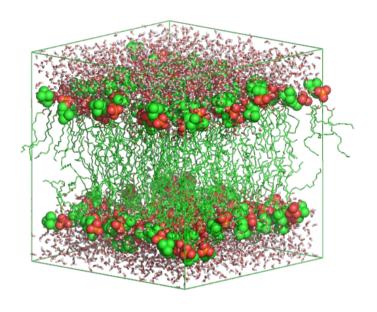
The simulation system is coupled to a heath bath with a constant temperature and energy (infinite energy)!



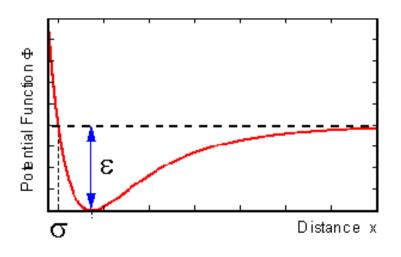




Non-bond energy



Note that the potential energy is typically negative!



Total energy

$$\phi(r_{ij}) = 4\varepsilon \left[\left(\frac{\sigma}{r_{ij}} \right)^{12} - \left(\frac{\sigma}{r_{ij}} \right)^{6} \right]$$

Question: How large are energy fluctuations in the finite simulation system at equilibrium?

The average energy E and the average of E² are Boltzmann distributed:

$$\langle E \rangle = \frac{1}{Z} \sum_{j} E_{j} e^{\frac{-E_{j}}{kT}}$$

$$\langle E^{2} \rangle = \frac{1}{Z} \sum_{j} E_{j}^{2} e^{\frac{-E_{j}}{kT}},$$

$$\langle E \rangle = \frac{1}{Z} \sum_{j} E_{j} e^{\frac{-E_{j}}{kT}}$$

$$\langle E^{2} \rangle = \frac{1}{Z} \sum_{j} E_{j}^{2} e^{\frac{-E_{j}}{kT}},$$

Take the derivative with respect to the temperature T:

$$\frac{\partial \langle E \rangle}{\partial T} = \frac{1}{Z} \sum_{j} \frac{E_{j}^{2}}{kT^{2}} e^{\frac{-E_{j}}{kT}} + \left(\sum_{j} E_{j} e^{\frac{-E_{j}}{kT}}\right) \left(\frac{-1}{Z^{2}} \frac{\partial Z}{\partial T}\right)$$

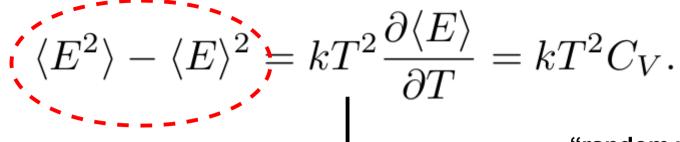
$$= \frac{1}{kT^{2}} \langle E^{2} \rangle - \langle E \rangle \frac{\partial \ln Z}{\partial T} = \frac{1}{kT^{2}} \left(\langle E^{2} \rangle - \langle E \rangle^{2}\right),$$

$$\stackrel{\langle E \rangle/kT^{2}}{}$$

$$\frac{\partial \langle E \rangle}{\partial T} = \frac{1}{Z} \sum_{j} \frac{E_{j}^{2}}{kT^{2}} e^{\frac{-E_{j}}{kT}} + \left(\sum_{j} E_{j} e^{\frac{-E_{j}}{kT}}\right) \left(\frac{-1}{Z^{2}} \frac{\partial Z}{\partial T}\right)$$

$$= \frac{1}{kT^{2}} \langle E^{2} \rangle - \langle E \rangle \frac{\partial \ln Z}{\partial T} = \frac{1}{kT^{2}} \left(\langle E^{2} \rangle - \langle E \rangle^{2}\right), \qquad (E^{2}) - \langle E \rangle^{2} = kT^{2} \frac{\partial \langle E \rangle}{\partial T} = kT^{2} C_{V}.$$

C_v is the heath capacity at constant volume



"random walk-like term"

Brownian displacement:

$$<\Delta R^2>=2Dt$$

$$(\Delta E)^2 = kT^2C_V$$
. (fluctuation-dissipation theorem)

$$(\Delta E)^2 = \langle E^2 - 2E\langle E \rangle + \langle E \rangle^2 \rangle = \langle E^2 \rangle - 2\langle E\langle E \rangle \rangle + \langle \langle E \rangle^2 \rangle$$

$$= \langle E^2 \rangle - \langle E \rangle^2.$$

How large are these fluctuations?

$$(\Delta E)^2 = kT^2C_V$$
. (fluctuation-dissipation theorem)

How does ΔE scale with the number of particles ΔN ? Lets normalize E:

$$\frac{\Delta E}{\langle E \rangle} = \sqrt{kT^2} \frac{\sqrt{C_V}}{\langle E \rangle}.$$

$$\downarrow \qquad \qquad C_V \propto N$$

$$< E > \propto N$$

$$\frac{\Delta E}{\langle E \rangle} \propto \frac{\sqrt{N}}{N} = \frac{1}{\sqrt{N}},$$

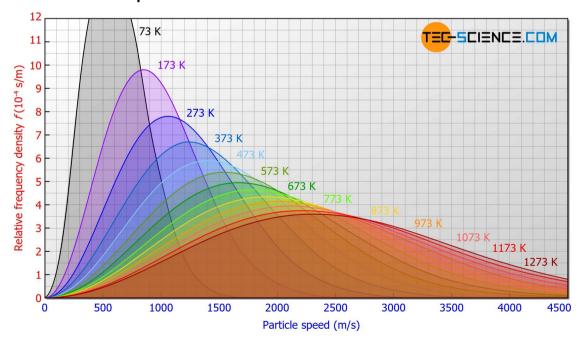
Energy fluctuations scale with $\sqrt{(1/N)}$. Fluctuations vanish if N approaches ∞

N is "countable" in simulations thus strong fluctuations.

Definition of temperature

$$\sum_{i=1}^{N} \frac{|\mathbf{p}_i|^2}{2m_i} = \frac{k_b T}{2} \left(3N - N_c\right) \longrightarrow T = \sum_{i=1}^{N} \frac{|\mathbf{p}_i|^2}{m_i k_B N_{df}}$$
$$3N - N_c = N_{df}$$

Individual velocities of particles follow the Maxwell-Boltzmann distribution!



 Initializing temperature T in a simulation: Assign the velocity of particles based the on the M-B distribution corresponding to T.

Controlling temperature: Velocity scaling

$$\Delta T = T_0 - T(t)$$

$$T = \sum_{i=1}^{N} \frac{|\mathbf{p}_i|^2}{m_i k_{\mathrm{B}} N_{df}}$$

 T_0 is the target temperature, T(t) is the instantaneous temperature.

scaling parameter
$$\lambda$$

$$\Delta T = \frac{1}{2} \sum_{i=1}^{\infty} 2 \frac{m_i \left(\lambda v_i\right)^2}{N_{df} k_B} - \frac{1}{2} \sum_{i=1}^{\infty} 2 \frac{m_i v_i^2}{N_{df} k_B}$$

$$\Delta T = \left(\lambda^2 - 1\right) T(t)$$

$$\lambda = \sqrt{T_0/T(t)}$$

 Velocity-scaling suppresses the characteristic thermal fluctuations which are present in a finite ensemble

Controlling temperature: Langevin dynamics

$$m_i \mathbf{a}_i = \mathbf{F}_i - (\mathbf{v}_i) + (\mathbf{R}_i)^t$$

The gain and loss of heat must be in balance (Fluctuation-dissipation theorem). This balance is temperature dependent and thus acts a thermostat.

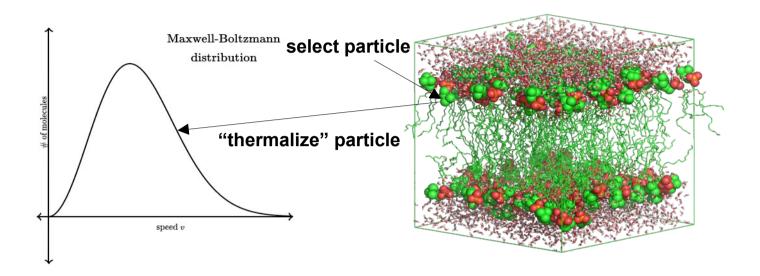
This balance is (only) satisfied when:

$$\langle \mathbf{R}_i(t) \rangle = 0$$

$$\langle \mathbf{R}_i(t) \mathbf{R}_i(t') \rangle = 2k_B T \gamma \delta(t - t')$$

- Explicit knowledge of the number of degrees of freedom in the system is not required to control the temperature.
- Sampling of phase-space is excellent.
- Canonical ensemble is well reproduced: $(\Delta E)^2 = kT^2C_V$.
- Dynamics is (of course) not conserved!

Controlling temperature: Andersen-Thermostat



- Stochastic thermostat: Reassigns the velocities of randomly selected particles using the M-B distribution.
- The number of particles selected and frequency of the velocity reassignment determines the coupling strength with the heat bath.
- An Boltzmann distribution as input gives a Boltzmann distribution as output.
- This still does not conserve dynamics or properties related to dynamics (e.g., diffusion constants)!

Controlling temperature: Berendsen Thermostat

Main idea: Model the change in T as a first order decay:

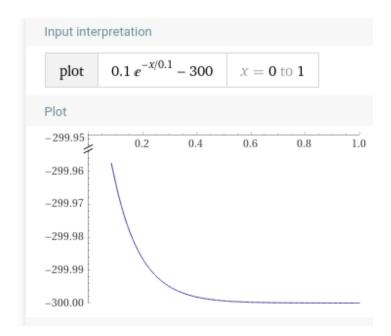
$$\frac{dT(t)}{dt} = \frac{1}{\tau} \left(T_0 - T(t) \right)$$

with, being a coupling constant (1/time), T₀ the target temperature. The solution is an exponential decay (weak coupling):



Herman Berendsen

$$T(t) = \tau C e^{-t/\tau} + T_0$$



Controlling temperature: Velocity scaling

$$\Delta T = T_0 - T(t)$$

$$T = \sum_{i=1}^{N} \frac{|\mathbf{p}_i|^2}{m_i k_{\mathrm{B}} N_{df}}$$

 T_0 is the target temperature, T(t) is the instantaneous temperature.

scaling parameter
$$\lambda$$

$$\Delta T = \frac{1}{2} \sum_{i=1}^{\infty} 2 \frac{m_i \left(\lambda v_i\right)^2}{N_{df} k_B} - \frac{1}{2} \sum_{i=1}^{\infty} 2 \frac{m_i v_i^2}{N_{df} k_B}$$

$$\Delta T = \left(\lambda^2 - 1\right) T(t)$$

$$\lambda = \sqrt{T_0/T(t)}$$

 Velocity-scaling suppresses the characteristic thermal fluctuations which are present in a finite ensemble

Controlling temperature: Berendsen Thermostat

$$\frac{dT(t)}{dt} = \frac{1}{\tau} \left(T_0 - T(t) \right)$$

In discrete time this takes the form:

$$\Delta T = \frac{\Delta t}{\tau} (T_0 - T(t))$$

$$\lambda = \sqrt{T_0/T(t)}$$
 GROMACS fast, flexible & free Leap-frog scheme
$$\lambda^2 = 1 + \frac{\Delta t}{\tau} \{ \frac{T_0}{T(t-\frac{\Delta t}{2})} - 1 \}$$

If _ goes to infinity then the ensemble approaches NVE, if _ equal dt then velocity scaling is recovered. Typically _ is 0.1 ps.

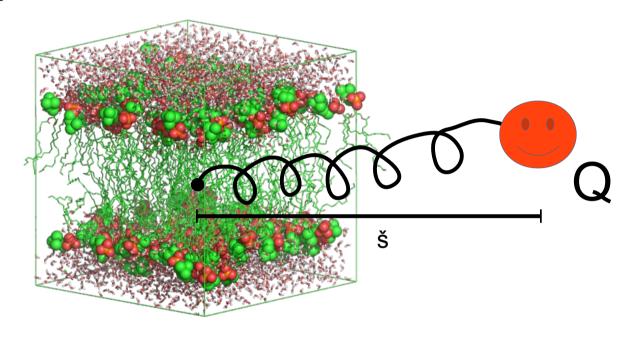
Controlling temperature: Berendsen Thermostat

The Berendsen thermostat (still) does not produce the canonical ensemble!!!!

(Though in Lennard-Jones fluids, it closely approaches it.)

$$(\Delta E)^2 = kT^2 C_V.$$

The main idea is to couple the heat bath and energy of the system by a spring with a virtual length s, a mass Q, and velocity ds/dt. This combined system is coined "the extended system":



Now let § act as an effective time-scaling parameter (§ ~ 1):

$$d\tilde{t} = \tilde{s}dt$$

This yields the following set of extended variables:

$$\tilde{\mathbf{r}} = \mathbf{r}, \dot{\tilde{\mathbf{r}}} = \tilde{s}^{-1}\dot{\mathbf{r}}, \tilde{s} = s \text{ and } \dot{\tilde{s}} = \tilde{s}^{-1}\dot{s}$$

$$\tilde{\mathbf{r}} = \mathbf{r}, \dot{\tilde{\mathbf{r}}} = \tilde{s}^{-1}\dot{\mathbf{r}}, \tilde{s} = s \text{ and } \dot{\tilde{s}} = \tilde{s}^{-1}\dot{s}$$

The Lagrangian of the extended ensemble is given by:

$$\mathcal{L} = \sum_i \frac{m_i}{2} \tilde{s}^2 \dot{\tilde{\mathbf{r}}}_i^2 - U(\tilde{\mathbf{r}}) + \frac{1}{2} Q \dot{\tilde{s}}^2 - g k_b T_0 \ln \tilde{s}$$
 kinetic energy heat bath
$$g = N_{df} + 1$$

$$\mathcal{L} = \sum_{i} \frac{m_{i}}{2} \tilde{s}^{2} \dot{\tilde{\mathbf{r}}}_{i}^{2} - U(\tilde{\mathbf{r}}) + \frac{1}{2} Q \dot{\tilde{s}}^{2} - g k_{b} T_{0} \ln \tilde{s}$$

$$\tilde{\gamma} \qquad \tilde{\mathcal{S}}$$
Euler-Lagrange:
$$\frac{\partial}{\partial t} \frac{\partial L}{\partial \dot{q}} = \frac{\partial L}{\partial q}$$

$$\ddot{\tilde{\mathbf{r}}}_{i} = \frac{\tilde{\mathbf{F}}_{i}}{m_{i}\tilde{s}^{2}} - \frac{2\dot{\tilde{s}}\dot{\tilde{\mathbf{r}}}_{i}}{\tilde{s}},$$

$$\ddot{\tilde{s}} = \frac{1}{Q\tilde{s}} \left(\sum_{i} m_{i}\tilde{s}^{2}\dot{\tilde{\mathbf{r}}}_{i}^{2} - gk_{b}T_{0} \right)$$

• Although these equations sample a microcanonical ensemble (constant energy) in the extended system, the real system (r, dr/dt) samples a canonical esemble!

The real system is retrieved via the transformation:

$$s = \tilde{s}, \ \dot{s} = \tilde{s}\dot{\tilde{s}}, \ \ddot{s} = \tilde{s}^{2}\ddot{\tilde{s}} + \tilde{s}\dot{\tilde{s}}^{2},$$

$$\mathbf{r} = \tilde{\mathbf{r}}, \ \dot{\mathbf{r}} = \tilde{s}\dot{\tilde{\mathbf{r}}}, \ \ddot{\mathbf{r}} = \tilde{s}^{2}\ddot{\tilde{\mathbf{r}}} + \tilde{s}\dot{\tilde{\mathbf{r}}}^{2}$$

$$\gamma = \frac{\dot{s}}{s}$$

$$\ddot{\mathbf{r}}_{i} = \frac{\mathbf{F}_{i}}{m_{i}} - \gamma \mathbf{r}_{i} ,$$

$$\dot{\gamma} = \frac{-k_{B}N_{df}}{Q} T(t) \left(\frac{g}{N_{df}} \frac{T_{0}}{T(t)} - 1 \right) .$$

heat bath acceleration

If Q goes to infinity the NVE ensemble is obtained. Small Q values (tight coupling)
may result in high-frequency oscillations. Q should be large. If Q is large then the
canonical ensemble is only obtained when the ensemble averaging occurs over longer
time-scales.

 The main limitation of the NH thermostat is its oscillatory behavior: True energy fluctuations are stochastic in nature!

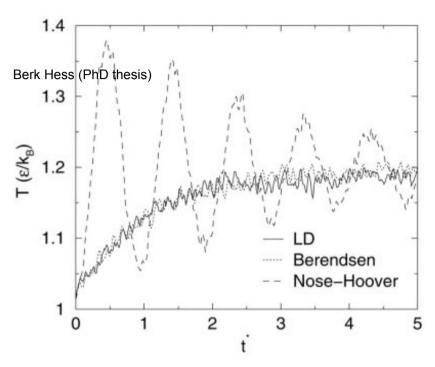
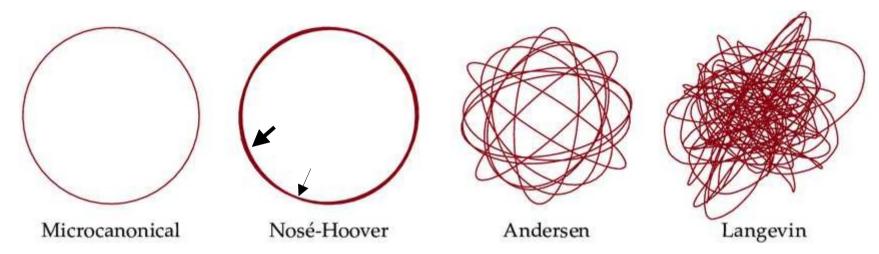


Figure 2. Response of three different thermostats to a temperature jump from 1 to 1.2 for a system of 8000 Lennard–Jones particles. Time and temperature are in reduced units (from Hess⁶¹).

- The Nose-Hoover thermostat has a very slow response! (e.g., Use Berendsen to first equilibrate system then switch to Nose-Hoover)
- The Nose-Hoover infact becomes nonergodic in small, solid systems (resonance with normal modes in the system).

Controlling temperature: A summary

Phase-space of the harmonic oscillator:



Energy is constant Temperature is not.

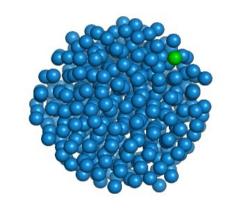
Ergodicity problems in small, stiff systems. Excellent conservation of dynamics.

Superior sampling. Dynamics is lost.

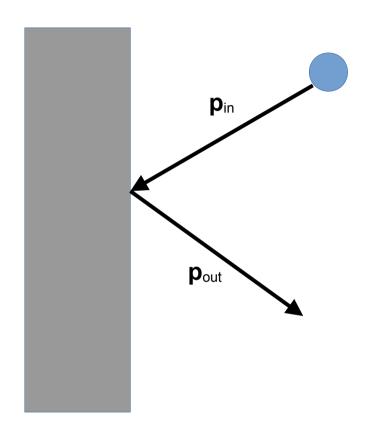
The sampling resembles a superposition of multiple NVE ensembles. Dynamics is distorted.

• All thermostats come with pro's and con's. Picking the most suitable thermostat depends on the application. Dynamics?: Nose-Hoover. Non-equilibrium? (weak-coupling schemes: Berendsen & co). Sampling efficiency only? Andersen and Langevin.

The famous "flying ice cube" artifact



Controlling pressure: Definition of pressure



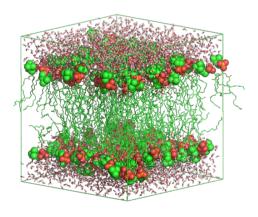
$$\Delta \mathbf{p} = \mathbf{p}_{out} - \mathbf{p}_{in}$$

The force on the wall is:

$$\mathbf{F} = -\sum_{collisions} \frac{\Delta \mathbf{p}}{\Delta t}$$

The pressure on the wall is:

$$P = F/A$$



• This classic "wall definition" of pressure is ambiguous in periodic boundary conditions, though can be effective in the presence of an actual boundary potential. In that case, the force on the wall F is precisely known (local pressure).

Controlling pressure: Clausius virial theorem

Lets start with Newton again.....

$$m \dot{\mathbf{r}_i} = \mathbf{F}_i$$
 multiply both sides by \mathbf{r}_i

$$\mathbf{r}\ddot{\mathbf{r}} = \frac{d}{dt} \left(\mathbf{r}\dot{\mathbf{r}} \right) - \dot{\mathbf{r}}^2 \,,$$

Substitution now yields

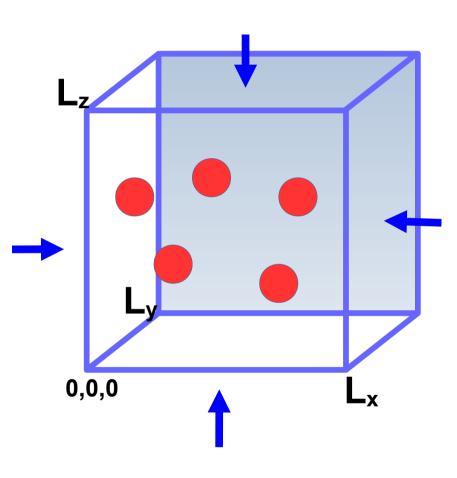
$$\frac{d}{dt}(m_i \mathbf{r}_i \dot{\mathbf{r}}_i) - m\dot{\mathbf{r}}_i^2 = \mathbf{F}_i \mathbf{r}_i$$

In this notation Fr is a scalar product: $F_xr_x + F_yr_y + F_zr_z$

Controlling pressure: Clausius virial theorem

Controlling pressure: Definition of pressure

Assume the system is under an externally applied pressure P. Then the total virial W is split into two contributions: $\langle W \rangle = \langle W^{ext} \rangle + \langle W^{int} \rangle_{\rm K}$



$$\langle W^{ext}
angle = \sum_{Walls} r_i F_i
ightarrow r_{i,w} \sum_{Wall} F_i = r_w F_{wall}$$
 average force on the wall A and A average force on the wall A average force on the wall A and A are average force on the wall A and A are average force on the wall A and A are average force on the wall A and A are average force on the wall A and A are average force on the wall A and A are average force on the wall A and A are average force on the wall A and A are average force on the wall A and A are average force on the wall A and A are average force

particles not (!)

Controlling pressure: Definition of pressure

$$\langle W^{int} \rangle := \Xi$$

Note that the average kinetic energy Ekin is 3/2NK_BT. Therefore, this yields:

$$PV = Nk_BT + \frac{1}{3}\Xi$$
 ideal gas term

$$\begin{split} \sum_{k=1}^{N} \mathbf{F}_k \cdot \mathbf{r}_k &= \sum_{k=1}^{N} \sum_{j=1}^{N} \mathbf{F}_{jk} \cdot \mathbf{r}_k = \sum_{k=2}^{N} \sum_{j=1}^{k-1} \left(\mathbf{F}_{jk} \cdot \mathbf{r}_k + \mathbf{F}_{kj} \cdot \mathbf{r}_j \right) \\ & \text{single sum} \\ &= \sum_{k=2}^{N} \sum_{j=1}^{k-1} \left(\mathbf{F}_{jk} \cdot \mathbf{r}_k - \mathbf{F}_{jk} \cdot \mathbf{r}_j \right) = \sum_{k=2}^{N} \sum_{j=1}^{k-1} \mathbf{F}_{jk} \cdot \left(\mathbf{r}_k - \mathbf{r}_j \right) \\ & \text{double sum} \end{split}$$

• Disclaimer: These derivations used the "single virial". In periodic boundary conditions the single virial requires additional corrections (no conservation of angular momentum), i.e., "book keeping" of force contributions that crossed the periodic boundaries.

Controlling pressure: Berendsen barostat

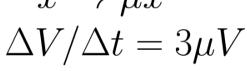
Main idea: Model the change in P as a first order decay:

$$\frac{dP}{dt} = \frac{P_0 - P(t)}{\tau_P}$$

with $_{\perp}$ being a coupling constant (1/time), P_0 the target pressure.

Scale coordinates and box over Δt:

$$x \to \mu x$$
$$\Delta V/\Delta t = 3\mu V$$



The relation ship between change in pressure and volume is given by:

$$\frac{dP}{dt} = -\frac{1}{\kappa_T V} \frac{dV}{dt} = -\frac{3\mu}{\kappa_T} \text{ isothermal compressibility Its dimension is } \frac{dV}{dt} = -\frac{3\mu}{\kappa_T} \text{ isothermal compressibility Its dimension is } \frac{dV}{dt} = -\frac{3\mu}{\kappa_T} \text{ isothermal compressibility Its dimension is } \frac{dV}{dt} = -\frac{3\mu}{\kappa_T} \frac{dV}{dt} = -\frac{3\mu}{\kappa_T}$$

Herman Berendsen

Simulation box and coordinates are thus (isotropically) scaled as:

$$\mu = 1 - \frac{\kappa_T \Delta t}{3\tau_P} (P_0 - P(t))$$

Controlling pressure: Other Barostats

 Andersen Barostat. Note that this is an 'extended' barostat rather than a stochastic one like 'his' thermostat! It could be coined "Noose-Hover" barostat;-):

$$\frac{d\mathbf{r}_{i}}{dt} = \frac{\mathbf{p}_{i}}{m_{i}} + \frac{1}{3}\mathbf{r}_{i}\frac{d\ln V}{dt},$$

$$\frac{d\mathbf{p}_{i}}{dt} = -\sum \hat{\mathbf{r}}_{ij}U'(\mathbf{r}_{ij}) - \frac{1}{3}\mathbf{p}_{i}\frac{d\ln V}{dt},$$

$$\frac{Md^{2}V}{dt^{2}} = p_{0} + \left(\frac{2}{3}\sum \frac{\mathbf{p}_{i}^{2}}{2m_{i}} - \frac{1}{3}\sum \mathbf{r}_{ij}U'(r_{ij})\right)/V.$$

The Parrinello-Rahman barostat is an anisotropic extension on the Anderson barostat.

