

TMSCA MIDDLE SCHOOL SCIENCE TEST #1© OCTOBER 19, 2019

GENERAL DIRECTIONS

- 1. About this test:
- A. You will be given 40 minutes to take this test.
- B. There are 50 problems on this test.
- 2. All answers must be written on the answer sheet/Scantron form/Chatsworth card provided. If you are using an answer sheet be sure to use **BLOCK CAPITAL LETTERS**. Clean erasures are necessary for accurate grading.
- 3. If using a Scantron answer form, be sure to correctly denote the number of problems not attempted.
- 4. You may write anywhere on the test itself. You must write only answers on the answer sheet.
- 5. You may use additional scratch paper provided by the contest director.
- 6. All problems have **ONE** and **ONLY ONE** correct [BEST] answer. There is a penalty for all incorrect answers.
- 7. On the back of this page is a copy of the periodic table of the elements as well as a list of some potentially useful information in answering the questions.
- 8. A simple scientific calculator with the following formulas is sufficient for the science contest: +, -, %, $^{\wedge}$, $\log x$, e^{x} , $\ln x$, y^{x} , $\sin x$, \sin^{-x} , $\cos x$, \cos^{-x} , $\tan x$, \tan^{-x} , with scientific notation and degree/radian capability.

The calculator must be silent, hand-held and battery operated. The calculator cannot be a computer or cannot have built-in or stored functionality that provides scientific information and cannot have communication capability. If the calculator has memory, it must be cleared. Each student may bring one spare calculator. **NO GRAPHING CALCULATORS ARE PERMITTED.**

- 9. All answers within \pm 5% will be considered correct.
- 10. All problems answered correctly are worth **FIVE** points. **TWO** points will be deducted for all problems answered incorrectly. No points will be added or subtracted for problems not answered.
- 11. In case of ties, percent accuracy will be used as a tie breaker.

1A 1	Periodic Table of the Elements																
1 H	2A 2											за 13	4A 14	^{5A} 15	6A 16	^{7А} 17	2 He
3 Li 6.94	4 Be _{9.01}											5 B 10.81	6 C 12.01	7 N 14.01	8 O 16.00	9 F 19.00	10 Ne 20.18
11 Na 22.99	12 Mg _{24.31}	3B 3	4B 4	5B 5	6B 6	7В 7	8	—8B—	10	1B 11	2B 12	13 Al _{26.98}	14 Si _{28.09}	15 P 30.97	16 S 32.07	17 Cl 35.45	18 Ar 39.95
19 K 39.10	20 Ca 40.08	21 Sc 44.96	22 Ti 47.87	23 V 50.94	24 Cr 52.00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 Ni 58.69	29 Cu 63.55	30 Zn 65.38	31 Ga _{69.72}	32 Ge 72.64	33 As 74.92	34 Se _{78.96}	35 Br 79.90	36 Kr 83.80
37 Rb 85.47	38 Sr 87.62	39 Y 88.91	40 Zr 91.22	41 Nb _{92.91}	42 Mo _{95.94}	43 Tc (98)	44 Ru 101.07	45 Rh 102.91	46 Pd 106.42	47 Ag 107.87	48 Cd 112.41	49 In 114.82	50 Sn 118.71	51 Sb 121.76	52 Te 127.60	53 126.90	54 Xe 131.29
55 Cs 132.91	56 Ba 137.33	57 La 138.9	72 Hf 178.49	73 Ta 180.95	74 W 183.84	75 Re 186.21	76 Os 190.23	77 r 192.22	78 Pt 195.08	79 Au 196.97	80 Hg _{200.59}	81 TI 204.38	82 Pb 207.20	83 Bi _{208.98}	Po (209)	85 At (210)	86 Rn (222)
87 Fr (223)	88 Ra (226)	89 Ac (227)	104 Rf (261)	105 Db (262)	106 Sg (266)	107 Bh (264)	108 Hs (277)	109 Mt (268)	110 Ds (281)	111 Rg (281)	112 Cn (285)	113 Nh (286)	114 FI (289)	115 Mc (289)	116 Lv (293)	117 Ts (293)	118 Og (294)

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dν	Но	l Er	Tm	Yb	Lu
140.1	140.9	144.2	(145)	150.4	152.0	157.3	158.9	162.5	164.9	167.3	168.9	173.0	175.0
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
232.0	231.0	238.0	(237)	(244)	(243)	(247)	(247)	(251)	(252)	(257)	(258)	(259)	(262)

OTHER USEFUL INFORMATION

Acceleration of gravity at Earth's surface, $g = 9.81 \text{ m/s}^2$

Avogadro's Number, N = 6.02 x 10²³ molecules/mole

Planck's constant, $h = 6.63 \times 10^{-34} \text{ J} \cdot \text{s}$

Planck's reduced constant, $\hbar = h/2\pi = 1.05 \text{ X } 10^{-34} \text{ J} \bullet \text{s}$

Standard temperature and pressure (STP) is 0°C and I atmosphere

Gram molecular volume al STP = 22.4 liters

Velocity of light, $c = 3.0 \times 10^8 \text{ m/sec}$

Absolute zero= 0 K = -273.15°C

Gas constant, R = 1.986 col/K•mole = 0.082 liter•otm/K•mole

One Faraday= 96,500 coulombs (9 .65 x 10⁴ C)

Dulong and Pelil's constant= 6.0 amu•cal/gram•K

Electron rest mass, $m_e = 9.11 \times 10^{-31} \text{ kg}$

Atomic mass unit, $m_u = 1.66 \times 10^{-21} \text{ kg}$

Boltzmann constant, $k_B = 1.38 \times 10^{-23} \text{ J/K}$

Permittivity of free space ε_0 = 8.85 x 10^{-12} C²/N•m²

Permeability of free space $\mu_0 = 4\pi \times 10^{-7} \text{ T} \cdot \text{m/A}$

1 Atmosphere= $1.02 \times 10^5 \text{ N/m}^2 = 760 \text{ Torr} = 760 \text{ mmHg}$

1 Electron Volt - 1.6 x 10⁻¹⁹ Joules

Charge of on electron" -1.6 x 10^{-19} coulombs (C)

1 horsepower (hp) = 746 W = 550 ft•lb/s

Neutron Moss= 1.008665 au

Proton Mass= 1.007277 au

1 au= 931.5 MeV

1 calorie= 4.184 Joules (J)

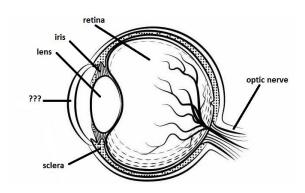
Specific heal of water= 4.18 J/g• °C

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1.	The abnormal rise of water caused by a storm's winds is called what? A. wind shear
	B. hurricane
	C. water spout
	D. storm surge
2.	What provides the energy to drive the hydrological cycle?
	A. the ocean B. the Earth C. water D. the sun
3.	A seismometer is used to detect earthquakes. This tool is paired with what two other important items to create a seismograph?
	A. rope and printer
	B. recording device and timer
	C. protractor and compass
	D. balance and meter stick
4.	Jones family made it to the hospital after Grandpa Jones fell off the ladder at their home in
	City A. What was their approximate average speed? City A. Hospital
	A. 60 km/h
	B. 76 km/h
	C. 30 km/h Family left City A at 10:00am and arrived at the hospital at 10:40am.
	D. 1.25 km/h
5.	Scientists believe the earliest fossils known and found on Earth are what?
	A. stromatolites B. trilobites C. both A and B D. coprolites
6.	The SI unit for measuring force is noted by which of these symbols?
	A. N B. C C. F D. T
7.	Alfred Wegener suggested that a single supercontinent existed in Earth's past. When did he
	believe that this supercontinent began to break apart in to smaller landmasses?
	A. early Paleozoic B. end of Cenozoic C. early Proterozoic D. early Mesozoic
8.	In Planck's constant of the SI units, what does the "J" and "s" stand for?
	A. J – Jeroboam s - sabin
	B. J - Jenner s - science
	C. J - Jayhe s - samsil
	D. J – Joules s - seconds
9.	Clues: English scientist, wrote the book, Micrographia 1665, discovered and named "cells"
	Who is this person?

A. Mendel B. Newton C. Hooke D. Leuwenhoek

- 10. James and his grandpa were exploring in the woods. They came across a place with acorns scattered all across the ground. James looked up and saw a big tree. Most likely the tree was what?
 - A. apple
- B. walnut
- C. oak
- D. maple



- 11. "???" is pointing to what part of the eye?
 - A. cornea
- B. pupil
- C. macula
- D. aqueous humor
- 12. One behavior that turtles participate in to help them control their body temperature is called what?
 - A. wiggling their front flippers
 - B. competition
 - C. resting on bottom of pond
 - D. basking
- 13. El Niño is a weather pattern that occurs in what ocean?
 - A. Indian
- B. Atlantic
 - C. Pacific
- D. Arctic
- 14. Which human body systems help with transporting oxygen and carbon dioxide?
 - A. Digestive and Endocrine
 - B. Integumentary and Skeletal
 - C. Circulatory and Respiratory
 - D. Endocrine and Digestive
- 15. Astronaut Chris Hadfield wrote about EVA in his book about experiences in space. What is EVA?
 - A. his Aunt
 - B. Energy Economical Association
 - C. Ergophobia Vaccine Application
 - D. Extravehicular Activity

- 16. Which out of the statements below is incorrect?
 - A. When water evaporates, it changes from a liquid to a gas.
 - B. When water condenses, it changes from a gas to a liquid.
 - C. When water precipitates, it changes from a liquid to a solid.
 - D. When water transpires, it changes from a liquid to a gas
- 17. Clues: born in England, early interest in meteorology, proposed that all substances were made of atoms in early 1800's Who is this person?
 - A. John Dalton
 - B. Ernest Rutherford
 - C. J.J. Thomson
 - D. Niels Bohr
- 18. Choose the statement below that is considered true by scientists.
 - A. Turtles are strictly herbivores.
 - B. Turtle shells are really modified ribs.
 - C. Turtles are endotherms.
 - D. Turtles lay a tiny clutch of eggs.
- 19. The lab team was working on classifying several unknown substances into metals, nonmetals, and metalloids by using their physical properties. They made a chart with a checklist to organize their investigation.

Substance	Shiny?	Solid?	Conductor of heat and electricity?	Ductile?	Malleable?
A1		✓			
A2	✓	✓	✓	✓	✓
A3					
A4	✓	✓	✓, depending		
A5					

Out of these substances, which statement below would most likely be false?

- A. A2 is most likely not a metal.
- B. A5 is definitely not a metal
- C. A4 may be a metalloid
- D. A3 and A5 are probably nonmetals
- 20. Research has shown that a plant contains what hormone that is extensively involved in stem and root growth?
 - A. abscisic acid
 - B. ethylene
 - C. indole acetic acid
 - D. testosterone

- 21. Which of these are physical properties used to identify minerals?
 - A. density
- B. flammability
- C. reactivity with an acid
- D. Both A and B
- 22. Group 2A on the Periodic Table is made up of elements called what?
 - A. alkaline Earth metals
 - B. noble gases
 - C. halogens
 - D. alkali metals
- 23. Cattle can be observed with a large white bird, the cattle egret, hanging out nearby.

 Most of the time, these organisms have a symbiotic relationship in which both benefit.

 Which type of relationship would this be?
 - A. There is no relationship.
 - B. It is an example of commensalism.
 - C. It is a parasitic relationship.
 - D. It is a mutualistic relationship.



- 24. Clues: astronaut, died tragically in Apollo I fire, first American to walk in space Who is this person?
 - A. Ed White
 - B. Roger Chaffee
 - C. Neil Armstrong
 - D. Gus Grissom
- 25. Which of the following statements is false when discussing chemical equations?
 - A. The products are the substances that result from the change.
 - B. The reactants are the substances that undergo the change.
 - C. This symbol \rightarrow stands for "yields".
 - D. Each symbol must have a coefficient.
- 26. This chart shows diseases caused by fungi in humans. Which is most likely obtained by direct contact, not by breathing in spores?
 - A. Blastomycosis
 - B. Coccidioidomycosis
 - C. Tinea pedis
 - D. Aspergillosis

Fungal Diseases
Blastomycosis
Coccidioidomycosis
Tinea pedis
Aspergillosis

- 27. What science lab tool would be used to measure liquid in milliliters?
 - A cup marked in ounces
 - B. ruler
 - C. balance scale
 - D. graduated cylinder

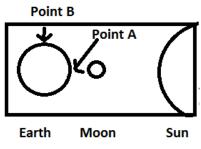
28. Which of the following human health hazards are spread by rodents? A. Zika virus B. Ebola virus C. Hantavirus D. Lyme disease 29. "Horsepower" is a term developed by James Watt to describe how a horse worked compared to his steam engines. Using the useful Horsepower Watts Automobile information chart, what would Automobile #3 420 Automobile #1 have in Watts? 270 Automobile #2 A. 325 W Automobile #3 325 B. .436 W ? Automobile #4 250 C. 450,242 W D. 242,450 W 30. Why do water pipes sometimes break when the temperature outside is below freezing? A. when water is cooled to 4 degrees C and below, it will expand which can break the pipes B. when water freezes, it will contract which can break the pipes C. when water is cooled to 4 degrees Celsius and below, it chemically changes D. frozen water pipes only break because of cracks or faulty pipes 31. What is passed down unchanged from generation to generation and is inherited only from the human's mother? A. eye color B. mitochondrial DNA C. ability to taste PTC D. the height gene 32. Clues: a scientist, makes predictions, studies the atmosphere and its current conditions What type of scientist is this person? A. climatologist B. botanist C. meteorologist D. seismologist 33. Solar winds from the sun can cause disturbances in which of the following? A. tides and ocean waves B. Earth air currents including the jet stream C. earthquakes, tsunamis, hurricanes, volcanos D. satellites, spacecraft, and the Earth's electric-power grid 34. Which of the following is in order from shortest to longest distance? A. kilometer, mile, 5,400 feet B. mile, kilometer, 5,400 feet C. 5,400 feet, mile, kilometer D. mile, 5,400 feet, kilometer 35. Which of these do NOT belong in the Eukarya Domain?

D. Spirochetes

C. Ferns

A. Amoebas B. Yeasts

- 36. When the gravitational pull of the sun and the moon are combined, there are extreme high and low tides. In this diagram, which of these tides would occur at point A and point B?
 - A. Point A high Point B low
 - B. Point A low Point B high
 - C. Point A and B would have the same high tides.
 - D. Point A and B would have the same low tides.



*Drawing is not to scale

- 37. In chemistry, "redox" is short for what?
 - A. reduction-oxidation reaction
 - B. Raoult's law
 - C. red-oxidation element
 - D. redundant oxygen
- 38. Clues: plant type, produces seeds enclosed in an ovary, flowering plant What type of plant is this?
 - A. gymnosperm
- B. angiosperm
- C. bryophytes
- D. pteridophytes
- 39. Jason and Jack were working on Science Fair Projects. Jason was measuring the dissolved oxygen levels at several local water locations. Jack was observing the color of soils in locations around his state. Both made tables of their data collection. What type of data was collected in each project?
 - A. Jason qualitative
- Jack quantitative
- B. Jason and Jack both quantitative
- C. Jason quantitative Jack qualitative
- D. Jason and Jack both qualitative
- 40. Cirrus clouds look like which of the following?
 - A. wispy, thin curls
 - B. thin, flat sheets
 - C. a UFO (flying saucer)
 - D. puffy with flat bottoms
- 41. Which part of the eye shown above controls the diameter of the pupil and helps to define the color of the eye?
 - A. retina
- B. iris
- C. cornea
- D. sclera
- 42. Scientists measure the population of Monarch butterflies overwintering in Mexico by the number of hectares of clustering butterflies. What is a hectare?
 - A. a unit equal to 10,000 m²
 - B. a unit equal to 50 square miles
 - C. a unit equal to 100 square acres
 - D. none of the above

43.	If you were to take an eyedropper and place drops of water on a piece of wax paper in close proximity of each other, the drops will join together because of what property of water? A. cohesion B. electromagnetic property C. high specific heat D. able to change into all 3 states of matter								
44.	Complete this an A. Late Jurass	• •	nosaurus Rex i arly Triassic		e Cretaceous as S Quaternary	_	rus is to what? leogene		
45.	Which part of the "white of the eye A. retina		c. cornea	embran D. sc		ater part i	s called the		
46.	On the last strike What science con to cause the ball A. the magnu	ncept helps to e to curve as it w	explain how the vas thrown?	e pitche		o the base	eball was able		
47.	Gold is to "Au" a	as Silver is to v	vhat?						
	A. Fe	B. Ag	C. Sn	D. Sl					
48.	Diethyltoluamide chemical?	e is commonly	found in insec	t repella	ants. What abbre	eviation is	s used for this		
	A. DDT	B. DEET	C. DTL	D. D	IE				
49.	Which of the foll A. yeast	lowing would r B. lichens	not be consider C. mushroor		fungus? D. toadstool				
50.	The metric system A. England	m was develop B. France	ed in what cou C. Spain	ntry ard D. R					

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1. D	18. B
2. D	19. A
3. B	20. C

36.	A

3	7	,	4	1
J	,	•	1	7