

# JIANA ZHENG

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## RESEARCH INTEREST

Nanomaterials, Electrocatalyst, Perovskites

## EDUCATION

University of Science and Technology of China (USTC), China

Hefei, China, 08/2020-06/2024

*School of the Gifted Young*

*BS in Applied Physics (Condensed Matter Physics Track)*

- GPA 3.6/4.3 (86.56/100)

- Selected Scholarships & Awards

National Second-class Prize, China Undergraduate Mathematical Contest in Modeling (CUMCM) 11/2022

Provincial Gold Award, National Business Plan Competition 05/2022

Outstanding Contribution Award at School of the Gifted Young Student Union 03/2022

Bronze award, Outstanding Student Scholarship 09/2021

Bronze award, Outstanding Freshman Scholarship 09/2020

## RESEARCH EXPERIENCE

### ➤ Interest 1: Conversion of small molecules through electrocatalytic methods

RA in Hefei National Laboratory for Physical Sciences at Microscale, Advisor: Prof. Jie Zeng

USTC, China

### **Training : Synthesize Amorphous Cu Nanoparticles for Efficient CO<sub>2</sub> Electroreduction to Liquid fuels** 08/2021-11/2021

- Repeated the experiment published on *Adv. Mater.* 2018, 30, 1–7
- Synthesized amorphous Cu (a-Cu) nanoparticles using tannic acid at room temperature, and crystallographic Cu (c-Cu) nanoparticles using NaBH<sub>4</sub> and PVP
- Determined the concentration of as-synthesized Cu via ICP, gained the experience in how to use GC and NMR to analyze the yields of gas and liquid products, respectively
- Learned the basic research paradigm of electrocatalysis

### **Project 1: Explore distance effect of single atoms on the stability of cobalt oxide OER catalysts** 05/2022-01/2023

- Synthesized the Ir-Cu<sub>0.3</sub>Co<sub>2.7</sub>O<sub>4</sub> single atom catalyst (SACs) through high-temperature pyrolysis of metal-organic frameworks (MOFs)
- Estimated average Ir-Ir distances on the exposed surface of SACs through HAADF-STEM image, and identified Ir distribution in catalyst by EDX elemental mapping
- Tested the electrocatalytic performance and the stability in acidic environment of each catalyst, and discovered the optimal Ir-Ir distances (~0.6nm) for stable acidic OER
- DFT calculation indicated that shorter Ir-Ir distances (~0.56nm at minimum) will inhibit the dissolution of cobalt oxide by increasing the migration energies of Co atom on the surface, which agreed well with our experimental results.

### **Project 2: Electrosynthesis of Oxalate using CO<sub>2</sub> over Pb-based Catalyst** 12/2021-06/2023

- Oxidized the Pb foils in air followed by H<sub>2</sub> reduction in tubular furnace to obtain large uniform high-index lead facets with more catalytic active sites
- Selected Et<sub>4</sub>NBF<sub>4</sub>-DMF as the aprotic electrolyte system to inhibit hydrogen evolution reaction (HER), thereby improving the selectivity of C-C coupling process
- Chose Zn foil as a sacrificial anode to precipitate the generated oxalate ion during the reaction, thus assisted the desorption of C<sub>2</sub>O<sub>4</sub><sup>2-</sup> and managed to maintain a high selectivity of CO<sub>2</sub>-to-C<sub>2</sub>O<sub>4</sub><sup>2-</sup> conversion
- Achieved a higher Faradic efficiency of oxalate (~90%) and a higher current (>100mA vs -3.0V) than previously reported
- Gained a deeper understanding of CO<sub>2</sub> reduction mechanism

### **Project 3: Electrosynthesis of $\text{NH}_2\text{OH}$ via plasma-electrochemical cascade pathway using $\text{N}_2$ and $\text{H}_2\text{O}$** 12/2022-06/2023

- Prepared the working electrode by magnetron sputtering Bi NPs onto carbon fiber paper
- Used plasma discharge device to convert  $\text{N}_2$  and  $\text{O}_2$  into  $\text{NO}_x$ , and  $\text{NO}_x$  was then absorbed into water to form  $\text{NO}_3^-$
- Adopted an excess amount of  $\text{CHOCOOH}$  to capture as-electroreduced  $\text{NH}_2\text{OH}$  into glyoxylic acid oxime
- The Bi NPs catalyst exhibit a high FE (73.4%) and selectivity (99.3%) towards  $\text{NH}_2\text{OH}$
- Project 2 and 3 were aimed to realize full electrosynthesis of glycine ( $\text{CO}_2-(\text{COOH})_2-\text{CHOCOOH}-\text{NH}_2\text{CH}_2\text{COOH}$ )

### ➤ Interest 2: Synthesis of functional nanomaterials (including metal & perovskites nanostructures)

#### **Project 1: Continuous Controllable Mass Production of Monodispersed Cu NCs in fluid device** 02/2022-10/2022

*Independent Project in Hefei National Laboratory for Physical Sciences at Microscale, Advisor: Prof. Jie Zeng USTC, China*

- Synthesized mono-dispersed Cu nanocrystals (NCs) via seed-growth mechanism using copper acetate as copper source, ascorbic acid as reductant, and PVP as surfactant
- Improved the purity and uniformity of Cu nanoparticles by decoupling the growth step from nucleation in a flow reactor
- Tuned the concentration of precursor and adjusted the ratio of copper source involved in each step to produce Cu nanoparticles of different sizes
- Tested the stability of continuous synthesis of Cu NCs in an experimental flow chemistry device
- Characterized the morphology of as-synthesized Cu NCs via SEM
- **Achievements:** Won the Provincial Gold Award in the “Challenge Cup” National Business Plan Competition and the **Bronze Award** in “Qingfeng Cup” Innovation and Entrepreneurship Competition of USTC

#### **Project 2: Employ Au nanocages/ZIF series MOFs as host carrier to encapsulate perovskite NCs** 07/2023-Present

*Independent Project, Advisor: Prof. Ou Chen, Department of Chemistry Brown University, United States*

- Synthesized Au nanocages using Ag nanocubes as templates via galvanic replacement
- Imparted functional nanostructures (like Ag nanocubes and QDs) in ZIF-8 nanoparticles
- Try to use zwitterionic ligands (like lecithin) and ligand-exchange methods to bridge the gap between nanoparticles dispersible in polar and non-polar solvents, thereby loading perovskites in cage-like structure for enhanced stability

#### **First-principles Calculations of Electron-phonon Coupling Phenomenon in GaAs and Si** 10/2022-12/2022

*Undergraduate Researcher, Advisor: Prof. Jin Zhao, Department of Physics USTC, China*

- Obtained skills to run a calculation task on remote supercomputers through personal terminal
- Carried out elementary electron and phonon calculations with DFT and DFPT using Quantum Espresso

### **UPCOMING PUBLICATIONS**

#### **Distance effect of single atoms on the stability of cobalt oxide catalysts for acidic oxygen evolution**

Zhirong Zhang<sup>1†</sup>, Chuanyi Jia<sup>2†</sup>, Peiyu Ma<sup>3†</sup>, Chen Feng<sup>1</sup>, Jin Yang<sup>1</sup>, Junming Huang<sup>1</sup>, **Jiana Zheng<sup>1</sup>**, Ming Zuo<sup>1</sup>, Shiming Zhou<sup>1\*</sup>, Jie Zeng<sup>1,4\*</sup>

(Nature Communications 23-43865; Current stage: Manuscript under consideration)

#### **Electrosynthesis of hydroxylamine via plasma-electrochemical cascade pathway using the air and water as raw materials**

Xiangdong Kong<sup>1,3</sup>, Jie Ni<sup>1,3</sup>, Zhimin Song<sup>1</sup>, Zhengwu Yang<sup>1</sup>, **Jiana Zheng<sup>1</sup>**, Zifan Xu<sup>1</sup>, Lang Qin,<sup>1</sup> Hongliang Li<sup>1</sup>, Zhigang Geng<sup>1,\*</sup>, Jie Zeng<sup>1,2,\*</sup>

(Nature Energy 2309202; Current stage: Manuscript under consideration)

### **CONTEST EXPERIENCE**

#### **Contest Project: Azimuth-only Passive Positioning of UAV in Formation Flight**

09/2022

*Core member, responsible for building mathematical models and writing papers*

- Created the model in Python and figured out solutions through geometric knowledge and iterative algorithm
- Organized and wrote an academic essay according to solving threads and computational simulation results
- **Achievement:** Won the National Second-class Prize in the 2022 China Undergraduate Mathematical Contest in Modeling

## RELEVANT COURSEWORK

### Synthesis of Cu-based Nanomaterials for Electrocatalytic Reduction of CO<sub>2</sub>

(Course: College Physics—Research Experiment)

09/2022-01/2023

Project Leader, Advisor: Dr. Rucheng Dai

USTC, China

- Designed and supervised the experiment as the project leader
- Prepared the gas-diffusion working electrode with porous carbon paper and Cu NPs (as the electrocatalyst, which can be produced in the previous Controllable Mass Production project through the flow reactor)
- Tested the electrochemical performance of Cu-based materials and discovered catalytic preferences of each catalyst
- Analyzed the constituent of liquid products via NMR, gas products via GC, and calculated the Faraday efficiency of each kind of product afterwards

## TECHNICAL PROFICIENCIES

### Computer Skills:

- Database: Scopus, Web of Science, and Google Scholar
- Basic software: Endnote, Origin and MATLAB
- Codes: Python, C, and Linux

### Research Skills:

- Proficient in conducting electrochemical experiments using H-cell, flow-cell and software like CHI660E, CHI1140E
- Proficient in many characterization methods, including XRD, SEM, TEM, UV/Vis and fluorescence spectrometer
- Proficient in analytical methods like NMR, IC, GC, and ICP
- Skilled use of software for data analysis, including DigitalMicrograph for TEM/STEM, MestReNova for NMR and Image J (for microscopy photo processing)