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## Roshan Jaiswal-Ferri

%Aero 402 Homework 4: 10/27/25

## Workspace Prep

```
warning off
format long      %Allows for more accurate decimals
close all;       %Clears all
clear all;        %Clears Workspace
clc;             %Clears Command Window
```

## Question 2: Non-Ideal Conditions

```
g2 = 1.4;
r2 = 297;
tc2 = 298;

g = 1.2;
R = 260; %J/kg-k
Tc = 1650; %combustion temp
tb = 10000;
F = 40; %N
g0 = 9.81; %at sea level

PcPe = linspace(1,100);
PePc = linspace(0,1);

CF = C_f(g,PePc, 1, 0);
Cstar = 1009.945158;
mdotv = F./(Cstar.*CF);
Isp = F./(mdotv.*g0);

eps = expansionRatio(g,PcPe);

Vtank = linspace(0,1);
Ptank = 1./Vtank; %make all constants = 1 for simplicity

% non ideal
```

---

```

mdot = F/(Cstar*1.4); %1.4 is the C_f chosen from the knee of the plot
disp(num2str(mdot));
ISP = F/(mdot*g0);
disp(num2str(ISP));
tb = 40000/F;
mp = mdot*tb;
disp(num2str(mp));

% Plots

% % This first plot is not what the question asks for (that ones next) but
% % this makes much more sense, especially because as the Pe/Pc ratio
% % approaches zero the C_f approaches the ideal case
% figure
% grid on
% hold on
% plot(PePc,CF)
% title('C_F vs Pe/Pc')
% xlabel('C_F')
% ylabel('Pe/Pc')

figure
grid on
hold on
plot(PcPe, C_f(g,1./PcPe, 1, 0))
plot(PcPe, C_f(g2,1./PcPe, 1, 0))
title('C_F vs Pc/Pe')
xlabel('C_F')
ylabel('Pc/Pe')
legend('Hydrazine','Cold Gas')

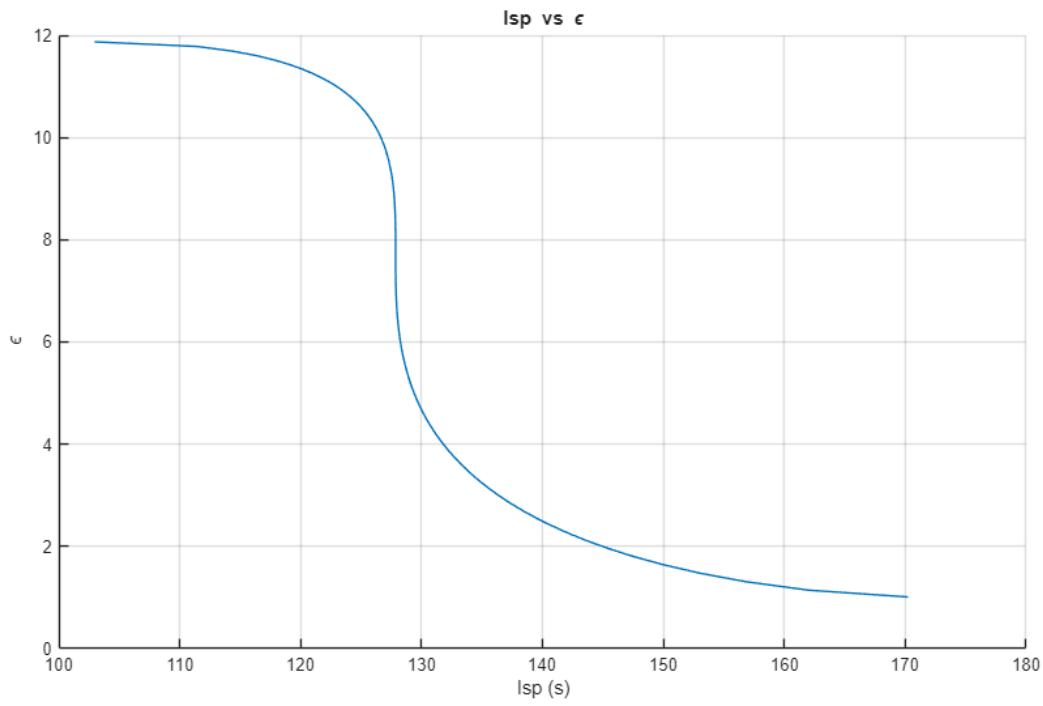
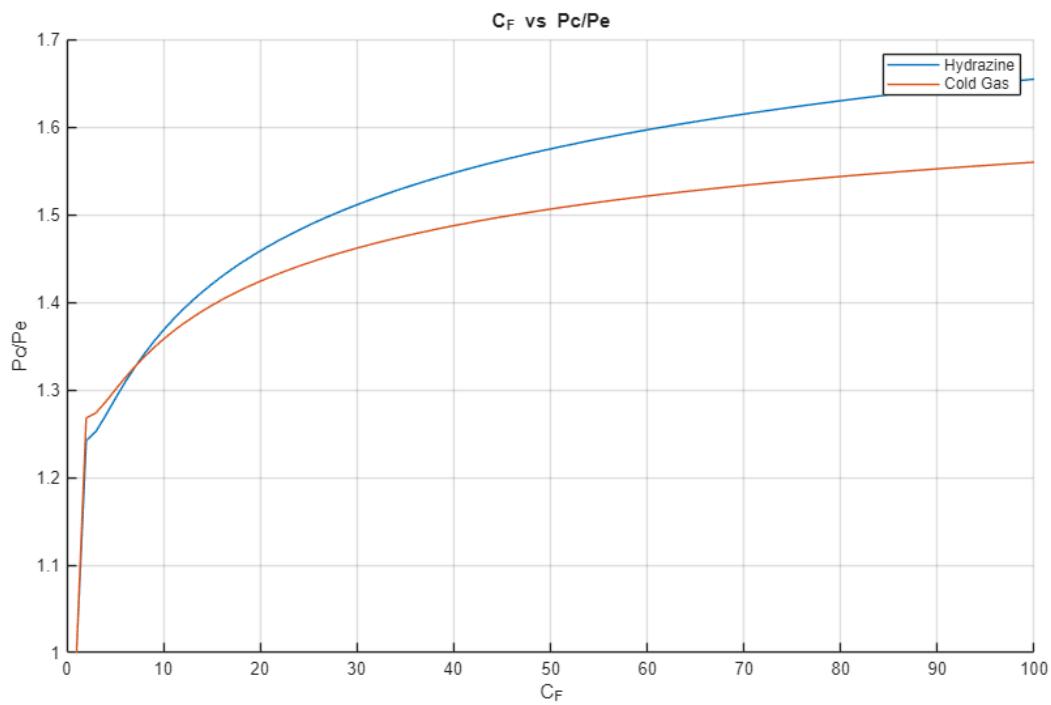
figure
grid on
hold on
plot(Isp,eps)
title('Isp vs \epsilon')
xlabel('Isp (s)')
ylabel('\epsilon')

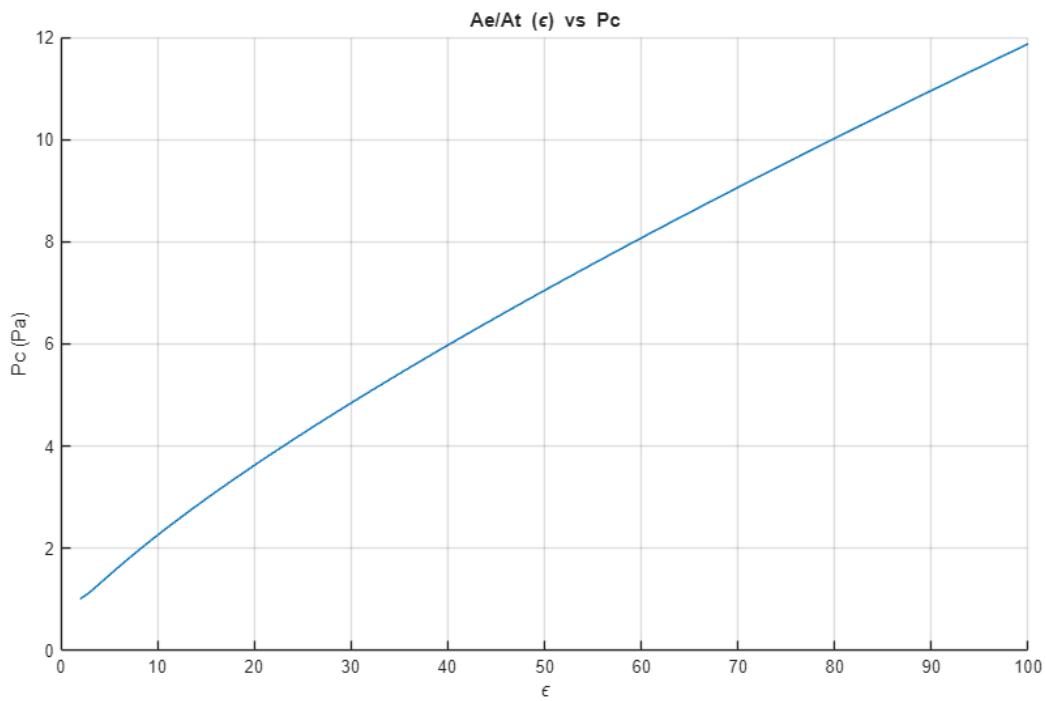
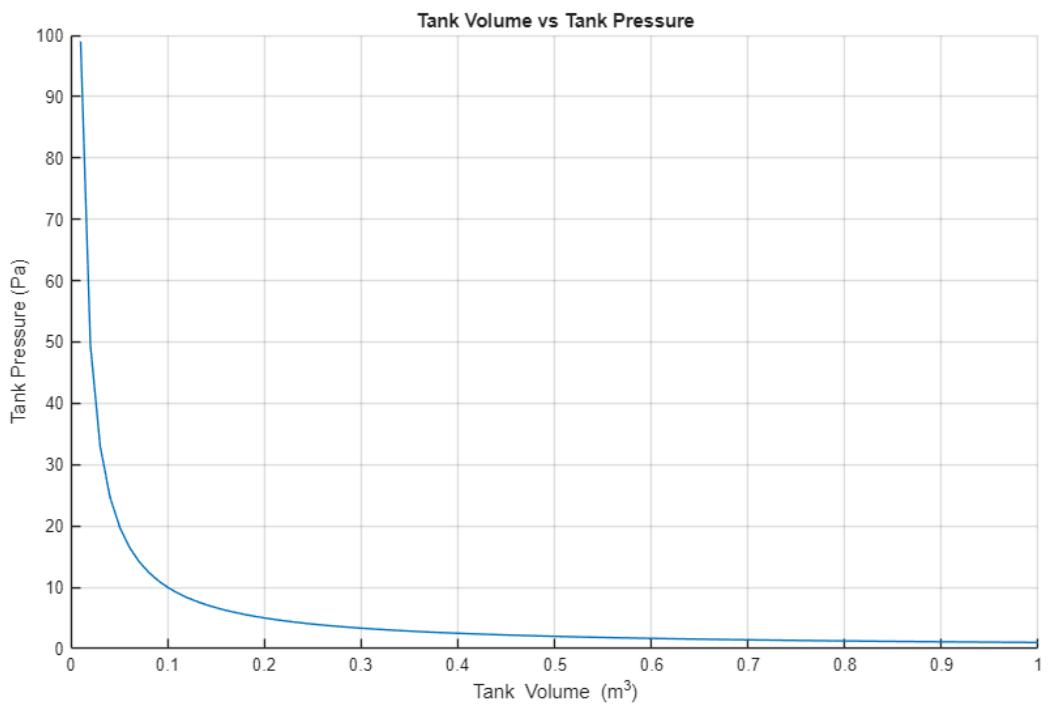
figure
grid on
hold on
plot(Vtank,Ptank)
title('Tank Volume vs Tank Pressure')
xlabel('Tank Volume (m^3)')
ylabel('Tank Pressure (Pa)')

figure
grid on
hold on
plot(PcPe,eps)
title('Ae/At (\epsilon) vs Pc')
xlabel('\epsilon')
ylabel('Pc (Pa)')

```

---





## Question 3:

```
mdot = 0.05; %nitrogen, h2o2, hydrazine
```

---

```

gamman2 = 1.4;
gammah2 = 1.25;
gammahy = 1.2;

Tcn2 = 300;
Tch2 = 1300;
Tchy = 1650;

Rn2 = 297;
Rh2 = 290;
Rhy = 260;

epsilon = 20;

[PcPen2, cstarn2, cfn2, Ispn2, thrustn2] = performance(gamman2, Rn2, Tcn2,
epsilon, mdot);
[PcPeh2, cstarh2, cfh2, Isph2, thrusth2] = performance(gammah2, Rh2, Tch2,
epsilon, mdot);
[PcPehy, cstarhy, cfhy, Isphy, thrusthy] = performance(gammahy, Rhy, Tchy,
epsilon, mdot);

Propellant = {'Nitrogen (N2)'; 'Hydrogen (H2)'; 'Hydrazine (Hy)'};
Gamma = [gamman2; gammah2; gammahy];
Tc = [Tcn2; Tch2; Tchy];
R = [Rn2; Rh2; Rhy];
PcPe = [PcPen2; PcPeh2; PcPehy];
cstar = [cstarn2; cstarh2; cstarhy];
cf = [cfn2; cfh2; cfhy];
Isp = [Ispn2; Isph2; Isphy];
Thrust = [thrustn2; thrusth2; thrusthy];

results = table(Propellant, Gamma, Tc, R, PcPe, cstar, cf, Isp, Thrust);

disp('-----');
disp(' Propellant Performance Comparison ');
disp('-----');
disp(results);

%c = cStar(1.4,297,298)

```

## Question 4:

```

% a) CEA Printout:
%
*****
**%
%
% NASA-GLENN CHEMICAL EQUILIBRIUM PROGRAM CEA2, FEBRUARY 5, 2004
% BY BONNIE MCBRIDE AND SANFORD GORDON
% REFS: NASA RP-1311, PART I, 1994 AND NASA RP-1311, PART II, 1996
%
%
*****

```

---

```

**%
%
%
%
%
% ### CEA analysis performed on Tue 28-Oct-2025 01:12:05
%
% # Problem Type: "Rocket" (Infinite Area Combustor)
%
% prob case = _____ 8869 ro equilibrium
%
% # Pressure (1 value):
% p,atm= 2
% # Supersonic Area Ratio (1 value):
% supar= 20
%
% # You selected the following reactants:
% reac
% name H2O2(L)           wt%=100.0000
%
% # You selected these options for output:
% # short version of output
% output short
% # Proportions of any products will be expressed as Mass Fractions.
% output massf
% # Heat will be expressed as siunits
% output siunits
%
% # Input prepared by this script:/var/www/sites/cearun/cgi-bin/CEARUN/
prepareInpu
% tFile.cgi
%
% ### IMPORTANT: The following line is the end of your CEA input file!
% end
%
%
%
%
%
% THEORETICAL ROCKET PERFORMANCE ASSUMING EQUILIBRIUM
%
% COMPOSITION DURING EXPANSION FROM INFINITE AREA COMBUSTOR
%
% Pin =    29.4 PSIA
% CASE = _____
%
%          REACTANT          WT FRACTION          ENERGY          TEMP
%          (SEE NOTE)          KJ/KG-MOL          K
% NAME      H2O2 (L)          1.0000000   -187780.000
272.740
%
% O/F=    0.00000  %FUEL=  0.000000  R,EQ.RATIO= 0.500000  PHI,EQ.RATIO=
0.000000
%
```

---

---

```

%                               CHAMBER      THROAT      EXIT
% Pinf/P                  1.0000     1.8079    284.05
% P, BAR                   2.0265     1.1209    0.00713
% T, K                     1274.54    1130.83    347.32
% RHO, KG/CU M            4.3364-1   2.7034-1   5.6024-3
% H, KJ/KG                 -5520.56   -5781.47   -7039.39
% U, KJ/KG                 -5987.88   -6196.10   -7166.74
% G, KJ/KG                 -19324.1   -18028.6   -10800.9
% S, KJ/ (KG) (K)          10.8302    10.8302    10.8302
%
% M, (1/n)                 22.676     22.676     22.676
% (dLV/dLP)t              -1.00000   -1.00000   -1.00000
% (dLV/dLT)p               1.0001     1.0000     1.0000
% Cp, KJ/ (KG) (K)         1.8460     1.7848     1.4318
% GAMMAS                   1.2479     1.2586     1.3442
% SON VEL, M/SEC           763.7      722.4      413.7
% MACH NUMBER              0.000      1.000      4.213
%
% PERFORMANCE PARAMETERS
%
% Ae/At                    1.0000    20.000
% CSTAR, M/SEC              1037.7     1037.7
% CF                        0.6961     1.6796
% Ivac, M/SEC                1296.4     1816.0
% Isp, M/SEC                  722.4     1742.9
%
%
% MASS FRACTIONS
%
% H2O                      0.52962   0.52963   0.52963
% *OH                      0.00002   0.00000   0.00000
% *O2                      0.47036   0.47037   0.47037
%
% * THERMODYNAMIC PROPERTIES FITTED TO 20000.K
%
% NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL
OXIDANTS

% b)
% The results partially match my results from the previous question. It is
% possible they mismatch because CEA specifically calculates and accounts
% for a 2atm input pressure where my calcs do not. CEA also includes more
% advanced/specific chemistry that my method ignores. But the Cf values are
% extremely close (i got 1.78 and CEA got 1.67). The Cstar values are also
% close, i got 933 and cea got 1037.

% c)
%
*****
** 
%
% NASA-GLENN CHEMICAL EQUILIBRIUM PROGRAM CEA2, FEBRUARY 5, 2004
% BY BONNIE MCBRIDE AND SANFORD GORDON
% REFS: NASA RP-1311, PART I, 1994 AND NASA RP-1311, PART II, 1996

```

---

---

```

%
%
*****
**%
%
%
%
%
% ### CEA analysis performed on Tue 28-Oct-2025 01:10:32
%
% # Problem Type: "Rocket" (Infinite Area Combustor)
%
% prob case = _____ 6230 ro equilibrium
%
% # Pressure (1 value):
% p,atm= 2
% # Supersonic Area Ratio (1 value):
% supar= 20
%
% # You selected the following reactants:
% reac
% name CH6N2(L),MMH      wt%=100.0000
%
% # You selected these options for output:
% # short version of output
% output short
% # Proportions of any products will be expressed as Mass Fractions.
% output massf
% # Heat will be expressed as siunits
% output siunits
%
% # Input prepared by this script:/var/www/sites/cearun/cgi-bin/CEARUN/
prepareInpu
% tFile.cgi
%
% ### IMPORTANT: The following line is the end of your CEA input file!
% end
%
%
%
%
%
% THEORETICAL ROCKET PERFORMANCE ASSUMING EQUILIBRIUM
%
% COMPOSITION DURING EXPANSION FROM INFINITE AREA COMBUSTOR
%
% Pin =    29.4 PSIA
% CASE = _____
%
%          REACTANT           WT FRACTION      ENERGY      TEMP
%          (SEE NOTE)        KJ/KG-MOL       K
% NAME      CH6N2 (L),MMH      1.0000000      54200.000
298.150
%
```

---

---

```

% O/F= 0.00000 %FUEL= 0.000000 R,EQ.RATIO= 0.000000 PHI,EQ.RATIO=
0.000000
%
% CHAMBER THROAT EXIT
% Pin/P 1.0000 1.7649 174.21
% P, BAR 2.0265 1.1482 0.01163
% T, K 961.67 893.65 548.83
% RHO, KG/CU M 3.2447-1 2.0176-1 3.9155-3
% H, KJ/KG 1176.43 837.74 -1071.71
% U, KJ/KG 551.87 268.62 -1368.81
% G, KJ/KG -13604.2 -12897.5 -9507.06
% S, KJ/(KG) (K) 15.3698 15.3698 15.3698
%
% M, (1/n) 12.802 13.056 15.359
% MW, MOL WT 10.973 11.351 15.359
% (dLV/dLP)t -1.06895 -1.07498 -1.00014
% (dLV/dLT)p 1.7639 1.8870 1.0016
% Cp, KJ/(KG) (K) 8.5440 9.6574 2.3681
% GAMMAS 1.2013 1.1902 1.2973
% SON VEL,M/SEC 866.2 823.0 620.8
% MACH NUMBER 0.000 1.000 3.416
%
% PERFORMANCE PARAMETERS
%
% Ae/At 1.0000 20.000
% CSTAR, M/SEC 1220.4 1220.4
% CF 0.6744 1.7375
% Ivac, M/SEC 1514.5 2260.6
% Isp, M/SEC 823.0 2120.4
%
%
% MASS FRACTIONS
%
% CH4 0.13924 0.16362 0.34821
% C2H6 0.00001 0.00000 0.00000
% *H2 0.09618 0.09006 0.04373
% NH3 0.00052 0.00045 0.00016
% *N2 0.60761 0.60766 0.60791
% C(gr) 0.15644 0.13819 0.00000
%
% * THERMODYNAMIC PROPERTIES FITTED TO 20000.K
%
% NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL
OXIDANTS
%
% i)
% MMH has a higher exhaust velocity and Isp than H2O2. This is because the
% molecular weight of the products of MMH are much smaller than H2O2,
% which then scales with exit velocity. Even though H2O2 burns hotter it
% is too heavy to accelerate fast enough & exit vel matters more
%
% ii)
% hydrazine may perform better but its reactants are very toxic compared

```

---

---

```
% to H2O2. H2O2 can decompose into water vapor and oxygen and H2O2 can
% also be cheaper.
```

## Functions

```
function CF = C_f(gamma, pe_pc, Ae_At, p0_pc)
CF = sqrt( (2.*gamma.^2./(gamma-1)) .* (2./(gamma+1)).^((gamma+1)./
(gamma-1)) ...
.* (1 - pe_pc.^((gamma-1)./gamma)) ) ...
+ (pe_pc - p0_pc) .* Ae_At;
end

function epsilon = expansionRatio(gamma, PcPe)

epsilon = ((2./(gamma+1)).^(1./(gamma-1))) .* (PcPe.^((1./gamma)) .* ...
(((gamma+1)./(gamma-1)) .* (1 - PcPe.^((1-gamma)./gamma))).^(-0.5));

end

function c = cStar(gamma, R, Tc)
top = sqrt(gamma*R*Tc);
base = 2/(gamma + 1);
bottomroot = sqrt( base^((gamma+1)/(gamma-1)) );

c = top / (gamma * bottomroot);

end

function [PcPe, cstar, cf, Isp, thrust] = performance(gamma, R, Tc, epsilon,
mdot)

g = 9.81;
syms PcPe
eq1 = epsilon == ((2./(gamma+1)).^(1./(gamma-1))) .* (PcPe.^((1./
gamma)) .* ...
(((gamma+1)./(gamma-1)) .* (1 - PcPe.^((1-gamma)./gamma))).^(-0.5));
sol = solve(eq1, PcPe);
PcPe = double(sol);
PcPe = max(real(PcPe(PcPe>1 & imag(PcPe)==0)));

cstar = cStar(gamma, R, Tc);
cf = C_f(gamma,1/PcPe, 20, 0 );
thrust = mdot*cstar*cf;
Isp = thrust/(mdot*g);

end

0.02829
144.1308
28.2901
```