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## Roshan Jaiswal-Ferri

%Aero 402 Homework 4: 10/27/25

## Workspace Prep

```
warning off
format long           %Allows for more accurate decimals
close all;            %Clears all
clear all;            %Clears Workspace
clc;                  %Clears Command Window
```

## Question 2: Non-Ideal Conditions

```
g2 = 1.4;
r2 = 297;
tc2 = 298;

g = 1.2;
R = 260; %J/kg-k
Tc = 1650; %combustion temp
tb = 10000;
F = 40; %N
g0 = 9.81; %at sea level

PcPe = linspace(1,100);
PePc = linspace(0,1);

CF = C_f(g,PePc, 1, 0);
Cstar = 1009.945158;
mdotv = F./(Cstar.*CF);
Isp = F./(mdotv.*g0);

eps = expansionRatio(g,PcPe);

Vtank = linspace(0,1);
Ptank = 1./Vtank; %make all constants = 1 for simplicity

% non ideal
```

---

```

mdot = F/(Cstar*1.4); %1.4 is the C_f chosen from the knee of the plot
disp(num2str(mdot));
ISP = F/(mdot*g0);
disp(num2str(ISP));
tb = 40000/F;
mp = mdot*tb;
disp(num2str(mp));

% Plots

% % This first plot is not what the question asks for (that ones next) but
% % this makes much more sense, especially because as the Pe/Pc ratio
% % approaches zero the C_f approaches the ideal case
% figure
% grid on
% hold on
% plot(PcPe,CF)
% title('C_F vs Pe/Pc')
% xlabel('C_F')
% ylabel('Pe/Pc')

figure
grid on
hold on
plot(PcPe, C_f(g,1./PcPe, 1, 0))
plot(PcPe, C_f(g2,1./PcPe, 1, 0))
title('C_F vs Pc/Pe')
xlabel('C_F')
ylabel('Pc/Pe')
legend('Hydrazine','Cold Gas')

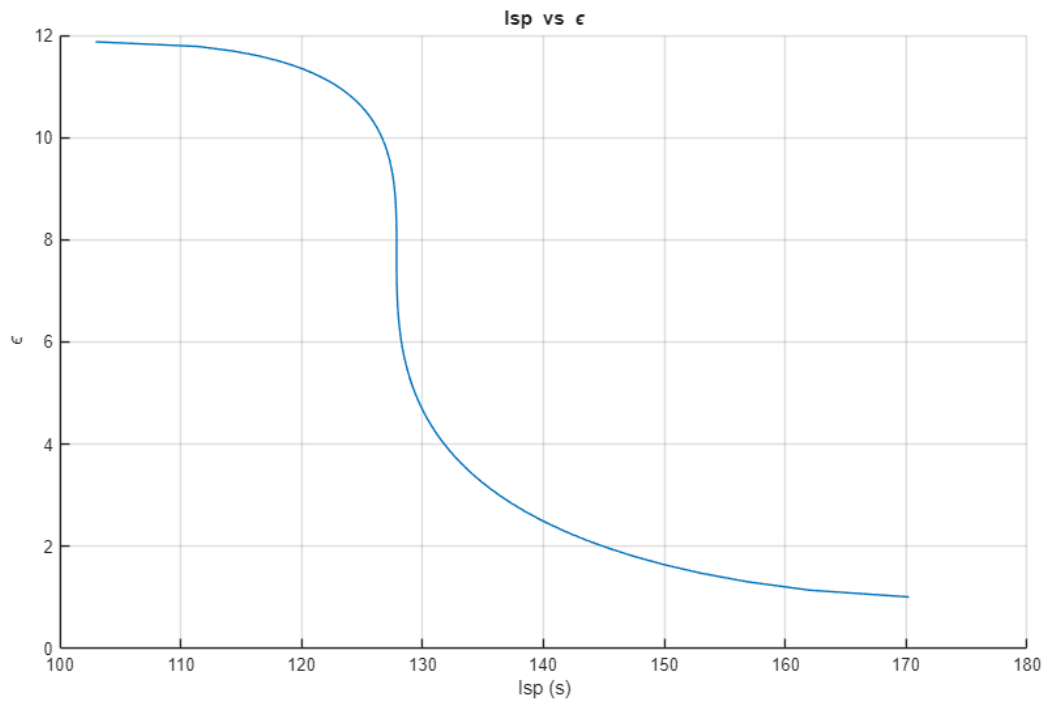
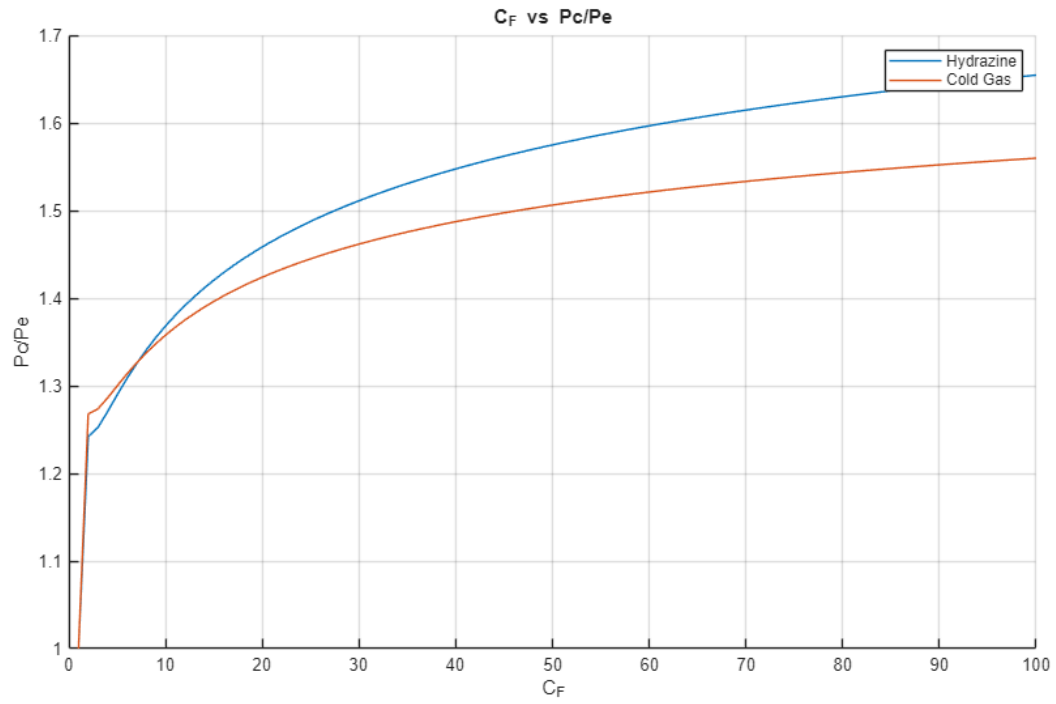
figure
grid on
hold on
plot(Isp,eps)
title('Isp vs \epsilon')
xlabel('Isp (s)')
ylabel('\epsilon')

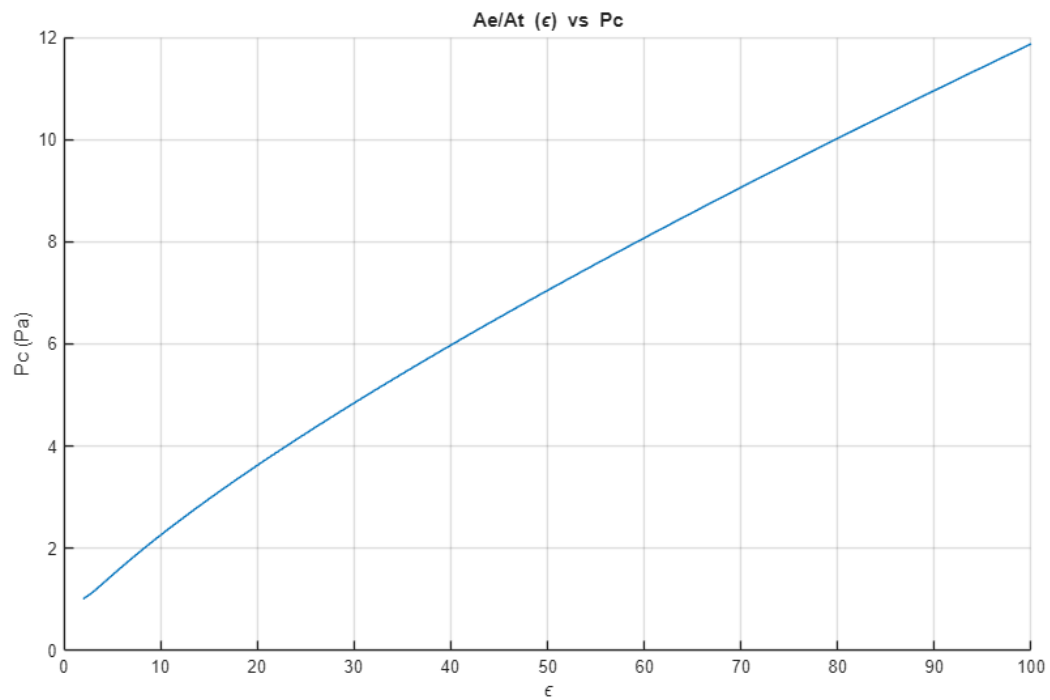
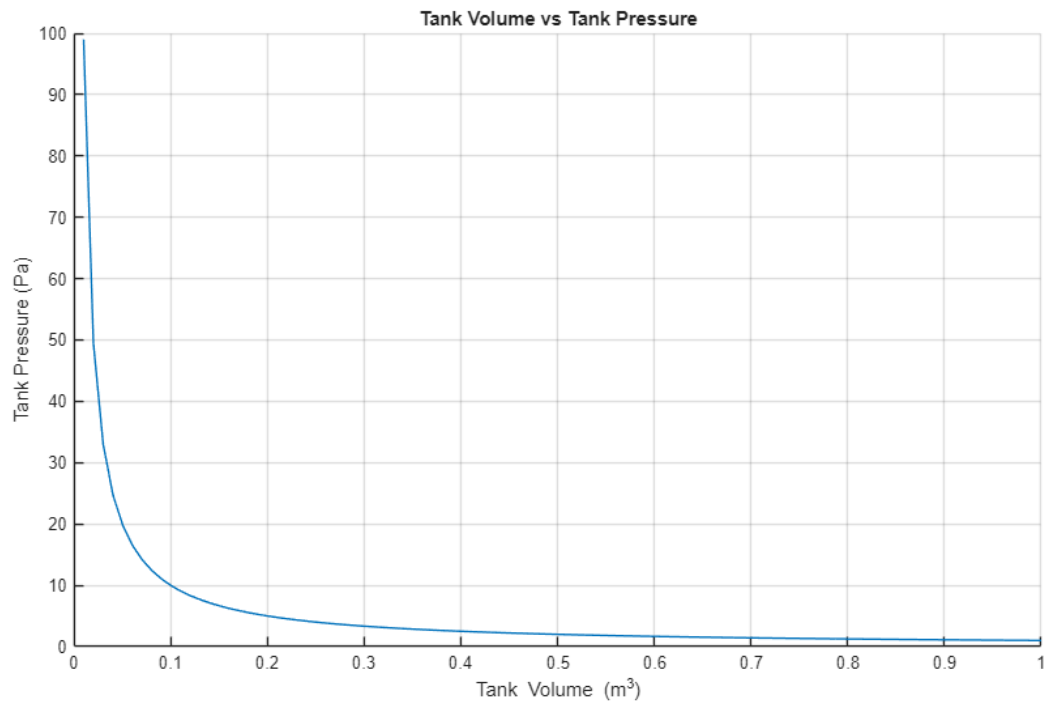
figure
grid on
hold on
plot(Vtank,Ptank)
title('Tank Volume vs Tank Pressure')
xlabel('Tank Volume (m^3)')
ylabel('Tank Pressure (Pa)')

figure
grid on
hold on
plot(PcPe,eps)
title('Ae/At (\epsilon) vs Pc')
xlabel('\epsilon')
ylabel('Pc (Pa)')

```

---





### Question 3:

`mdot = 0.05; %nitrogen, h2o2, hydrazine`

---

```

gamman2 = 1.4;
gammah2 = 1.25;
gammahy = 1.2;

Tcn2 = 300;
Tch2 = 1300;
Tchy = 1650;

Rn2 = 297;
Rh2 = 290;
Rhy = 260;

epsilon = 20;

[PcPen2, cstarn2, cfn2, Ispn2, thrustn2] = performance(gamman2, Rn2, Tcn2,
epsilon, mdot);
[PcPeh2, cstarh2, cfh2, Isph2, thrusth2] = performance(gammah2, Rh2, Tch2,
epsilon, mdot);
[PcPehy, cstarhy, cfhy, Isphy, thrusthy] = performance(gammahy, Rhy, Tchy,
epsilon, mdot);

Propellant = {'Nitrogen (N2)'; 'Hydrogen (H2)'; 'Hydrazine (Hy)'};
Gamma = [gamman2; gammah2; gammahy];
Tc = [Tcn2; Tch2; Tchy];
R = [Rn2; Rh2; Rhy];
PcPe = [PcPen2; PcPeh2; PcPehy];
cstar = [cstarn2; cstarh2; cstarhy];
cf = [cfn2; cfh2; cfhy];
Isp = [Ispn2; Isph2; Isphy];
Thrust = [thrustn2; thrusth2; thrusthy];

results = table(Propellant, Gamma, Tc, R, PcPe, cstar, cf, Isp, Thrust);

disp('-----');
disp(' Propellant Performance Comparison ');
disp('-----');
disp(results);

%c = cStar(1.4,297,298)

```

## Question 4:

```

% a) CEA Printout:
%
*****
**
%
%           NASA-GLENN CHEMICAL EQUILIBRIUM PROGRAM CEA2, FEBRUARY 5, 2004
%           BY   BONNIE MCBRIDE AND SANFORD GORDON
%           REFS: NASA RP-1311, PART I, 1994 AND NASA RP-1311, PART II, 1996
%
%
*****

```



---

```

%          CHAMBER    THROAT      EXIT
% Pinf/P      1.0000    1.8079    284.05
% P, BAR      2.0265    1.1209    0.00713
% T, K       1274.54   1130.83    347.32
% RHO, KG/CU M  4.3364-1  2.7034-1  5.6024-3
% H, KJ/KG     -5520.56 -5781.47 -7039.39
% U, KJ/KG     -5987.88 -6196.10 -7166.74
% G, KJ/KG     -19324.1 -18028.6 -10800.9
% S, KJ/(KG) (K)  10.8302  10.8302  10.8302
%
% M, (1/n)      22.676   22.676   22.676
% (dLV/dLP)t   -1.00000 -1.00000 -1.00000
% (dLV/dLT)p    1.0001   1.0000   1.0000
% Cp, KJ/(KG) (K)  1.8460   1.7848   1.4318
% GAMMAS       1.2479   1.2586   1.3442
% SON VEL,M/SEC   763.7    722.4    413.7
% MACH NUMBER     0.000    1.000    4.213
%
% PERFORMANCE PARAMETERS
%
% Ae/At                1.0000   20.000
% CSTAR, M/SEC         1037.7   1037.7
% CF                   0.6961   1.6796
% Ivac, M/SEC          1296.4   1816.0
% Isp, M/SEC           722.4    1742.9
%
%
% MASS FRACTIONS
%
% H2O                  0.52962   0.52963   0.52963
% *OH                  0.00002   0.00000   0.00000
% *O2                  0.47036   0.47037   0.47037
%
% * THERMODYNAMIC PROPERTIES FITTED TO 20000.K
%
% NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL
OXIDANTS

% b)
% The results partially match my results from the previous question. It is
% possible they mismatch because CEA specifically calculates and accounts
% for a 2atm input pressure where my calcs do not. CEA also includes more
% advanced/specific chemistry that my method ignores. But the Cf values are
% extremely close (i got 1.78 and CEA got 1.67). The Cstar values are also
% close, i got 933 and cea got 1037.

% c)
%
%*****
**
%
%          NASA-GLENN CHEMICAL EQUILIBRIUM PROGRAM CEA2, FEBRUARY 5, 2004
%          BY   BONNIE MCBRIDE AND SANFORD GORDON
%          REFS: NASA RP-1311, PART I, 1994 AND NASA RP-1311, PART II, 1996

```

---



---

% O/F= 0.00000 %FUEL= 0.000000 R,EQ.RATIO= 0.000000 PHI,EQ.RATIO= 0.000000

%

%

	CHAMBER	THROAT	EXIT
% Pinf/P	1.0000	1.7649	174.21
% P, BAR	2.0265	1.1482	0.01163
% T, K	961.67	893.65	548.83
% RHO, KG/CU M	3.2447-1	2.0176-1	3.9155-3
% H, KJ/KG	1176.43	837.74	-1071.71
% U, KJ/KG	551.87	268.62	-1368.81
% G, KJ/KG	-13604.2	-12897.5	-9507.06
% S, KJ/(KG) (K)	15.3698	15.3698	15.3698

%

% M, (1/n)	12.802	13.056	15.359
% MW, MOL WT	10.973	11.351	15.359
% (dLV/dLP)t	-1.06895	-1.07498	-1.00014
% (dLV/dLT)p	1.7639	1.8870	1.0016
% Cp, KJ/(KG) (K)	8.5440	9.6574	2.3681
% GAMMAS	1.2013	1.1902	1.2973
% SON VEL, M/SEC	866.2	823.0	620.8
% MACH NUMBER	0.000	1.000	3.416

%

% PERFORMANCE PARAMETERS

%

% Ae/At	1.0000	20.000
% CSTAR, M/SEC	1220.4	1220.4
% CF	0.6744	1.7375
% Ivac, M/SEC	1514.5	2260.6
% Isp, M/SEC	823.0	2120.4

%

%

% MASS FRACTIONS

%

% CH4	0.13924	0.16362	0.34821
% C2H6	0.00001	0.00000	0.00000
% *H2	0.09618	0.09006	0.04373
% NH3	0.00052	0.00045	0.00016
% *N2	0.60761	0.60766	0.60791
% C(gr)	0.15644	0.13819	0.00000

%

% \* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

%

% NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

% i)

% MMH has a higher exhaust velocity and Isp than H2O2. This is because the  
% molecular weight of the products of MMH are much smaller than H2O2,  
% which then scales with exit velocity. Even though H2O2 burns hotter it  
% is too heavy to accelerate fast enough & exit vel matters more

% ii)

% hydrazine may perform better but its reactants are very toxic compared

---

```
% to H2O2. H2O2 can decompose into water vapor and oxygen and H2O2 can
% also be cheaper.
```

## Functions

```
function CF = C_f(gamma, pe_pc, Ae_At, p0_pc)
CF = sqrt( (2.*gamma.^2./(gamma-1)) .* (2./(gamma+1)).^((gamma+1)./(
(gamma-1)) ...
    .* (1 - pe_pc.^((gamma-1)./gamma)) ) ...
    + (pe_pc - p0_pc) .* Ae_At;
end

function epsilon = expansionRatio(gamma, PcPe)

epsilon = ((2./(gamma+1)).^(1./(gamma-1))) .* (PcPe.^(1./gamma)) .* ...
    (((gamma+1)./(gamma-1)) .* (1 - PcPe.^((1-gamma)./gamma))).^(-0.5);

end

function c = cStar(gamma, R, Tc)
    top = sqrt(gamma*R*Tc);
    base = 2/(gamma + 1);
    bottomroot = sqrt( base.^((gamma+1)/(gamma-1)) );

    c = top / (gamma * bottomroot);

end

function [PcPe, cstar, cf, Isp, thrust] = performance(gamma, R, Tc, epsilon,
mdot)

    g = 9.81;
    syms PcPe
    eq1 = epsilon == ((2./(gamma+1)).^(1./(gamma-1))) .* (PcPe.^(1./
gamma)) .* ...
        (((gamma+1)./(gamma-1)) .* (1 - PcPe.^((1-gamma)./gamma))).^(-0.5);
    sol = solve(eq1, PcPe);
    PcPe = double(sol);
    PcPe = max(real(PcPe(PcPe>1 & imag(PcPe)==0)));

    cstar = cStar(gamma, R, Tc);
    cf = C_f(gamma,1/PcPe, 20, 0 );
    thrust = mdot*cstar*cf;
    Isp = thrust/(mdot*g);

end

0.02829
144.1308
28.2901
```