## **ELEMENTARY CHEMISTRY**

## CHEM 1110

## Fall 2022

Instructor:Matthew RowleyOffice Hours:MWF 10:00 am - 11:00 amTelephone:(435) 586-7875TR 1:00 pm - 2:00 pmEmail:matthewrowley1@suu.eduOffice:SC-220Please include the course number in the subject line of all correspondence.

## **Tentative Schedule**

This class will meet on Mondays, Wednesdays, and Fridays from 11:00 am - 11:50 am in room 130 of the science center (SC):

For the best lecture experience, read the indicated textbook chapter *before* each lecture.

	Date	Topic	Chapter
Week 1	W, Aug. 31	Chemistry: The Central Science	1.1–1.2
	F, Sep. 2	Elements and the Periodic Table	1.3–1.6
Week 2	M, Sep. 5	Labor Day – No Class!	
	W, Sep. 7	Measuring Physical Quantitites	1.7–1.9
	F, Sep. 9	Numbers and Math in Chemistry	1,10-1,12
Week 3	M, Sep. 12	Temperature, Heat, and Derived Units	1.13-1.14
	W, Sep. 14	Atoms, Elements, and Isotopes	2.1-2.3
	F, Sep. 16	Atomic Weight, Periodic Table, and Atomic Structure	2.4-2.6
Week 4	M, Sep. 19	Electron Configuration	2.7-2.9
	W, Sep. 21	Catch-up/Review Day - Midterm Exam 1 (Ch. 1–2)	
	F, Sep. 23	Ions and Ionic Bonds	3.1-3.4
Week 5	M, Sep. 26	Ionic Compounds	3.5-3.7
	W, Sep. 28	Naming Ionic Compounds	3.8-3.11
	F, Sep. 30	Molecular Compounds	4.1-4.3

	Date	Topic	Chapter		
Week 6	M, Oct. 3	Covalent Bonds and Molecules	4.4-4.7		
	W, Oct. 5	Molecular Structure	4.8-4.9		
	F, Oct. 7	Polarity and Binary Molecular Compounds	4.10-4.11		
Week 7 M, Oct. 10 Catch-up		Catch-up/Review Day - Midterm Exam 2 (Ch. 3	p/Review Day - Midterm Exam 2 (Ch. 3–4)		
	W, Oct. 12	<b>Balancing Chemical Reactions</b>	5.1-5.3		
	F, Oct. 14	Classes of Chemical Reactions	5.4-5.6		
Week 8	M, Oct. 17	Fall Break – No Class!			
	W, Oct. 19	Redox Reactions	5.7-5.8		
	F, Oct. 21	Chemical Calculations I	6.1-6.3		
Week 9	M, Oct. 24	Chemical Calculations II	6.4-6.5		
	W, Oct. 26	Chemical Reactions: Energy and Rates	7.1-7.3		
	F, Oct. 28	Chemical Reactions: Equilibrium	7.4-7.6		
Week 10	M, Oct. 31	Equilibrium Equations	7.7-7.9		
	W, Nov. 2	Catch-up/Review Day - Midterm Exam 3 (Ch. 5	<del>-</del> 7)		
	F, Nov. 4	Gases and Kinetic Molecular Theory	8.1–8.3		
Week 11	M, Nov. 7	Pressure and Gas Laws	8.4-8.7		
	W, Nov. 9	Gas Laws	8.8-8.11		
	F, Nov. 11	Liquids and Solids	8.12-8.15		
Week 12	M, Nov. 14	Solutions	9.1–9.4		
	W, Nov. 16	Solubility and Dilution	9.5-9.9		
	F, Nov. 18	Ions in Solution: Electrolytes	9.10-9.13		
Week 13	M, Nov. 21	Thanksgiving Break – No Class!			
	W, Nov. 23	Thanksgiving Break – No Class!			
	F, Nov. 25	Thanksgiving Break – No Class!			

	Date	Topic	Chapter
Week 14	M, Nov. 28	Acids and Bases	10.1–10.5
	W, Nov. 30	Acids and Bases – Calculations	10.6–10.10
	F, Dec. 2	<b>Buffers and Titrations</b>	10.11-10.14
		5 Catch-up/Review Day - Midterm Exam 4 (Ch. 8–10)	
Week 15	M, Dec. 5	Catch-up/Review Day - Midterm Exam	4 (Ch. 8–10)
Week 15	M, Dec. 5 W, Dec. 7	Catch-up/Review Day - Midterm Exam Nuclear Chemistry	14 (Ch. 8–10) 11.1–11.5
Week 15	, 0		- (