Quiz 1.2 – Matter, Properties, and Change

Name: Key
Question 1
Indicate whether the following are <i>physical</i> or <i>chemical</i> changes:
 Cutting a sheet of paper physical A metal tool rusting in the rain <u>Chemical</u> Ice melting in a glass of water <u>Physical</u> Sugar dissolving in a glass of water <u>physical</u> A battery discharging as it is being used <u>Chemical</u>
Question 2
Indicate whether each property is physical or chemical, and intensive or extensive
o Density <u>Physical</u> <u>intensive</u> o Flammability <u>Chemical</u> <u>intensive</u> o Mass <u>physical</u> <u>extensive</u> o Acidity <u>Chemical</u> <u>intensive</u> o Pressure <u>physical</u> <u>intensive</u>
Question 3
Classify each of these substances as one of the following:
Pure Element Pure Compound Homogeneous Mixture Heterogeneous Mixture
 Salt Water Homogeneous Minture Helium Gas fure Element Sugar Cubes fure Compound Silicon Crystal fure Element Marble Rye Bread Helerogeneous Minture The Air Around Us Homogeneous Minture