

CHEM 1210 (Rowley)

Exam 2 (Chapters 3-4) Fall 2025

Formulas

$$C_1 V_1 = C_2 V_2$$

$$\%m/m = \frac{m_A}{m_{total}} 100\%$$

$$\%v/v = \frac{V_A}{V_{solution}} 100\%$$

$$\%Yield = \frac{\text{actual yield}}{\text{theoretical yield}} 100\%$$

$$\%m/v = \frac{g_A}{V_{solution}} 100\%$$

$$ppm = \frac{m_A}{m_{total}} 10^6 ppm$$

$$ppb = \frac{m_A}{m_{total}} 10^9 ppb$$

Constants

$$R = 8.314 \frac{J}{mol K}$$

$$R = 0.08206 \frac{L \text{ atm}}{mol K}$$

Feel free to use this page as scratch paper, but final work for questions must be shown *in the question's section* to count for credit.

Periodic Table of the Elements