Rowley CHEM 1220 Midterm 4

Formulas

$$S = k_B \ln W$$

$$\Delta S_{universe} = \Delta S_{sys} + \Delta S_{surr}$$

$$\Delta G = \Delta H - T \Delta S$$

$$\Delta G = \Delta G^{\circ} + RT \ln Q$$

$$E^{\circ}_{cell} = E^{\circ}_{cathode} - E^{\circ}_{anode}$$

$$\Delta G^{\circ} = -nFE^{\circ}$$

$$E^{\circ} = \frac{RT}{nF} \ln K$$

$$\frac{N}{N_0} = \left(\frac{1}{2}\right)^{\frac{t}{t_{1/2}}}$$

Constants

$$R = 8.314 \frac{J}{mol \ K}$$

$$R = 0.08206 \frac{L \ atm}{mol \ K}$$

$$k_B = 1.380649 \times 10^{-23} \frac{J}{K}$$

$$N_A = 6.022 \times 10^{23} mol^{-1}$$

$$F = 96,485 \frac{C}{mol}$$

$$c = 2.998 \times 10^{8} \frac{m}{s}$$

$$\begin{split} \Delta S_{rxn}^{\circ} &= \sum_{i,products} \nu_i S_i^{\circ} - \sum_{j,reactants} \nu_j S_j^{\circ} \\ \Delta S_{surr} &= \frac{-q_s ys}{T} \\ \Delta G^{\circ} &= \Delta H^{\circ} - T \Delta S^{\circ} \\ \Delta G^{\circ} &= -RT \ln K \\ \frac{C_1 V_1}{n_1} &= \frac{C_2 V_2}{n_2} \\ \Delta G &= -nFE \\ E &= E^{\circ} - \frac{RT}{nF} \ln Q \end{split}$$

 $E = mc^2$

18 VIIIA 8A	Helium	Neon 20.180	Ar	Argon 39.948	ځ	Krypton 83.798	54	Xe non 131.294	98	&	Radon 222.018	18	Oganesson [294]						
8	17 VIIA 7A	Fluorine 18.998	J	Chlorine 35.453	Br	Bromine 79.904		Iodine 126.904		At	Astatine 209.987	<u>_</u>	Fennessine [294]	ì	_	Lutetium 174.967	103	L L Lawrencium [262]	
	16 VIA 6A	Oxygen 15.999	, S	32.066	Se	Selenium 78.971	53	Te Tellurium 127.6	1 85	&	Polonium [208.982]	16	LV Livermorium [293]		χ γ	Ytterbium 173.055	102	Nobelium 259.101	
	15 VA 5A	Nitrogen 14,007	T	90.974	As	Arsenic 74.922		Sb Antimony 121.760	8		Bismuth 208.980	11	Moscovium Li		Ę	Thulium 168.934	101	Mendelevium 258.1	
	14 IVA 4A	Carbon 12,011	Si	Silicon 28.086	Ge	Germanium 72.631	5.	Sn Tin .	83		Lead 207.2	4 . E	Flerovium N		。 点	Erbium 167.259	90	Fermium 257.095	
	13 IIIA 3A	6 Boron 10.811	A 4	Aluminum 26.982	eg.		20	Indium	82	F	Thallium 204.383	13 11	Nihonium F		.	Holmium 164.930		Einsteinium [254]	
nts		Ŋ	-	28 28	Zn	Zinc 65.38		Cadmium 112.414	8	Hg			Copernicium P		۵ٍ	Dysprosium 162.500		Californium 251.080	
Table of the Elements			Ξ:	18 18	7	Copper 63.546	48	Ag Silver 0	8		Gold 196.967	F	Roentgenium Co		် ရ	Terbium 158.925		Berkelium 247.070	
the E			5	۽ ح	Ë		47	Pd Palladium 106.42	79		Platinum 195.085	=	DS Darmstadtium Roi [281]		В	Gadolinium 157.25		C urium 247.070	
ole of			6	8 8 8	9	Cobalt 58.933	46	Rhodium P. 102.906	78		192.217	1	Meitnerium Dar [278]			Europium 151.964		Amanicium 243.061	
=			∞	7	j	Iron 55.845	45	Ruthenium R	77		Osmium 190.23	109	Hassium Me		SH	Samarium 150.36		Plutonium 244.064	
Periodic			7	78 78		Manganese 54.938	4	Technetium Ru	9/		Rhenium C 186.207	¥ .	Bohrium F		, Ba	Promethium 144.913		Neptunium 237.048	
				VIB 68 35	4	Chromium Ma 51.996	43	Molybdenum Tec	75		Tungsten R 183.84 1	7	Seaborgium B		U	_	92 9	Uranium 238.029	
M 12				VB 58		Vanadium Ch 50.942	42	Niobium Mol	74		Tantalum TL 180.948		DD Sea		, P	Ε	•	Protactinium 231.036	
CHE			4	1VB 4B	4	Titanium Va 47.867	4	Zirconium N 91.224	73		Hatnium Ta 178.49 1	-	Rutherfordium Di		Çe Ce		90	Thorium 232.038	
- \(\)				38 38	i .	Scandium Tir 44.956 4	4	Yttrium Zir 88.906	72		Lanthanum Hi 138.905 1	2	AC Actinium Ruth 227.028		Saries	3	Q Chinist	Series	
Rowley - CHEM 1220	2 IIA 2A	Be Beryllium 9.012		Magnesium 24.305	J	Calcium Sca 40.078 44	<u>&</u>	Strontium Yt 87.62 88	57		Barium Lant 137.328 13	8	Ra Radium Act 2226.025 22		Lant	3	*	, s	
- ₹ 4	Hydrogen II	Lithium Bery 6.941		Sodium Magr 22.990 24	1	Potassium Cal 39.098 40	88	Rubidium Stro 85.468 83	26		Cesium Bai 132.905 137	88	Fr Racium Rac 223.020 226						
- 25	Hydra 1.0	Lithi	² Z	22.9	2	Potas 39.(37	Rubic 85.4	55	U)	Ces 132.	87	Franc 223.v						

2017 Todd Helmenstine sciencenotes.org