Quiz 17.5 – Saturation and Formation Reactions

Name:
Question 1
Copper(II) ions will form a Cu(NH ₃) ₄ ²⁺ complex ion with $K_f=1.7\times 10^{13}$ Find [Cu ²⁺], [NH ₃], and [Cu(NH ₃) ₄ ²⁺] for a solution which is $0.200~N$ in Cu ²⁺ and $0.500~N$ in NH ₃
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Question 2
Consider the formation of the complex ion ${\rm Ag(CN)_2^-}$, with $(K_f=1.0\times 10^{21})$ Find $[{\rm Ag^+}]$, $[{\rm CN^-}]$, and $[{\rm Ag(CN)_2^-}]$ in a solution prepared by mixing $25.00~ml$ of $0.075~M$ AgNO $_3$ with $40.00~ml$ of $0.100~M$ NaCN

Question 3

 PbI_2 is a sparingly soluble salt with $K_{sp}=9.8\times 10^{-9}$, while PbI_4^{2-} is a complex ion with $K_f=3.0\times 10^4$ Find $[Pb^{2+}]$, $[I^-]$, and $[PbI_2^{2-}]$ if excess $PbI_2(s)$ is placed in pure water

God Says Yes to Me

By Kaylin Haught

I asked God if it was okay to be melodramatic and she said yes I asked her if it was okay to be short and she said it sure is I asked her if I could wear nail polish or not wear nail polish and she said honey she calls me that sometimes she said you can do just exactly what you want to Thanks God I said And is it even okay if I don't paragraph my letters Sweetcakes God said who knows where she picked that up what I'm telling you is Yes Yes Yes