# 1210 Review Quiz

Name:
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# Question 1

Give the number of protons, electrons, and neutrons in each atom or ion

	$e^-$	$p^+$	n
<sup>27</sup> Al			
<sup>127</sup> I <sup>-</sup>			
<sup>40</sup> Ca <sup>2+</sup>			

# Question 2

A coffee-cup calorimeter contains  $150.0\ g$  of pure water. When  $5.75\ g$  of ammonium nitrate are added to the water, they dissolve and the solution temperature drops by  $2.83\ ^{\circ}C$ . What is the enthalpy of solvation for ammonium nitrate?

# Question 3

Give the name or formula for each of the following compounds. For molecular compounds, draw the correct Lewis structure:

Name	Formula	Lewis Structure
Calcium Phosphate		
	$\mathrm{Mn_2S}_3$	
Dinitrogen Trioxide		-
	SF <sub>6</sub>	
Nitric acid		-
	$H_2SO_3$	

### **Question 4**

Consider the thermite reaction:

$$\underline{\hspace{0.5cm}} \operatorname{Fe_2O_3(s)} + \underline{\hspace{0.5cm}} \operatorname{Al(s)} \longrightarrow \underline{\hspace{0.5cm}} \operatorname{Fe(l)} + \underline{\hspace{0.5cm}} \operatorname{Al_2O_3(s)} \qquad \Delta H_{rxn} = -547.5 \, \frac{kJ}{mol}$$

Note the very exothermic enthalpy of this reaction, which leads to the iron product being molten.

- o First, balance this reaction
- $\circ$  If 2.5~g of Al react with 5.0~g of  $\mathrm{Fe_2O_3}$  , how many g of molten iron will be produced?

 $\circ$  How many J of heat will be released?

# Question 5

Predict the products and write a balanced chemical equation for the reaction between sulfuric acid and aqueous calcium hydroxide.

#### When I Heard the Learn'd Astronomer

### By Walt Whitman

When I heard the learn'd astronomer,

When the proofs, the figures, were ranged in columns before me,

When I was shown the charts and diagrams, to add, divide, and measure them,

When I sitting heard the astronomer where he lectured with much applause in the lecture-room,

How soon unaccountable I became tired and sick,

Till rising and gliding out I wander'd off by myself,

In the mystical moist night-air, and from time to time,

Look'd up in perfect silence at the stars.