

## Quiz 3.1 – Entropy

Name: \_\_\_\_\_

**Carnot cycle**

Consider a heat engine based around the Carnot cycle. Sketch this cycle on a p/V diagram, labeling the states in the process as A, B, C, and D

Tell which direction around this cycle operates as a heat engine, and which direction operates as a heat pump

- A-B-C-D:
- A-D-C-B:

Fill in the table below for the cycle when operating as a heat engine. Use generic variables ( $C_V$ ,  $T_H$ ,  $T_C$ ,  $V_A$ ,  $V_B$ , etc.)

Step	$w$	$q$	$\Delta U$	$\Delta S$
A				
B				
C				
D				
net (A-D)				

A car engine is a type of heat engine, and burns gasoline burns at about  $600^\circ\text{C}$ . If the ambient temperature is  $25^\circ\text{C}$ , what is the thermodynamic maximum efficiency a car engine can achieve?

**Measuring molar entropy**

He has  $T_{\text{boil}} = 4.25 \text{ K}$  and  $\Delta H_{\text{vap}} = 83 \frac{\text{J}}{\text{mol}}$ . The isobaric heat capacity for liquid helium is very complex, but can be approximated as  $C_p(l) \approx 7.4 \times 10^{-3} T^3 \frac{\text{J}}{\text{mol K}}$ . The isobaric heat capacity for gaseous He is simply  $C_p(g) = \frac{5}{2} R$ . Use these data to calculate the molar entropy for He gas at room temperature, and compare it to the value given in our textbook appendix.

**Irreversibility in Mechanical Systems**

Consider a spring which obeys Hook's law:  $F = -kx$  where  $x$  is the displacement away from equilibrium and  $k = 650 \frac{\text{N}}{\text{m}}$ . The acceleration due to gravity is  $9.80665 \frac{\text{m}}{\text{s}^2}$ .

- Calculate the equilibrium displacement if a  $10 \text{ kg}$  weight is placed on the spring

Considering the same weight-on-a-spring in Problem 1:

- Calculate the work done by the falling weight.

How much work would be done if instead the spring was stretched reversibly to the same equilibrium displacement. Bonus – Explain the discrepancy!

The spring-weight system will lose kinetic energy through friction with the air until it rests at its equilibrium position. What is  $\Delta S_{\text{universe}}$  for both the reversible and irreversible processes if they are done at room temperature ( $25^\circ \text{C}$ )?

*Who Has Seen the Wind?*

By Christina Rossetti

Who has seen the wind?  
Neither I nor you:  
But when the leaves hang trembling,  
The wind is passing through.

Who has seen the wind?  
Neither you nor I:  
But when the trees bow down their heads,  
The wind is passing by.