3610 Midterm Exam 3 Equations and Constants

K

Gas Constant Values			
8.314	$\frac{J}{mole\ K}$	0.08314	$\frac{Lbar}{moleK}$
0.08206	$\frac{Latm}{moleK}$	8.314	$\frac{m^3 Pa}{mole K}$
Boltzmann Constant Values			
1 381 × 1	$0^{-23} \frac{J}{}$	0 6950	cm^{-1}

$$\begin{array}{rcl} & & & \\ \hline 1\,L\,atm & = & 101.325\,J \\ \\ 1\,atm & = & 1.01325\,bar \\ \\ 1\,atm & = & 760\,torr \\ \\ 1\,atm & = & 101,325\,Pa \end{array}$$

$$\Delta G_{mix} = nRT \left(\chi_A \ln \chi_A + \chi_B \ln \chi_B \right)$$

$$\Delta S_{mix} = -nR \left(\chi_A \ln \chi_A + \chi_B \ln \chi_B \right)$$

$$p_A = \chi_A p_A^{\star}$$

$$p_B = \chi_B K_B$$

$$\mu_A = \mu_A^{\star} + RT \ln \chi_A$$

$$K_b = \frac{RT_b^{\star 2}}{\Delta H_{vap}}$$

$$\Delta T_b = K_b \chi_B = K_b C_B$$

$$K_f = \frac{RT_f^{\star 2}}{\Delta H_{freeze}}$$

$$\Delta T_f = K_f \chi_B = K_f C_B$$

$$\frac{n_g}{n_{total}} = \left| \frac{x_B - z_B}{x_B - y_B} \right|$$

$$\frac{n_l}{n_{total}} = \left| \frac{z_B - y_B}{x_B - y_B} \right|$$

$$\ln \gamma_A = \xi \chi_B^2 \qquad \ln \gamma_B = \xi \chi_A^2$$

$$\mu_B = \mu_B^{\ominus} + RT \ln \gamma_B \chi_B$$

$$I = \frac{1}{2} \sum_{ions} z_i^2 \left(\frac{c_i}{c^{\Theta}}\right) \qquad c^{\Theta} = 1 \, molal$$

$$\log \gamma_{\pm} = -\frac{A|z_{+}z_{-}|\sqrt{I}}{1 + B\sqrt{I}} + CI$$

$$\ln \frac{K_2}{K_1} = -\frac{\Delta H^{\Theta}}{R} \left(\frac{1}{T_2} - \frac{1}{T_1} \right)$$

$$\Delta G = \Delta G^{\Theta} + RT \ln Q$$

$$E_{cell} = E_{cathode} - E_{anode}$$

$$\Delta G^{\Theta} = -RT \ln K_{eq}$$

$$\frac{\mathrm{d}E_{cell}^{\Theta}}{\mathrm{d}T} = \frac{\Delta S^{\Theta}}{\nu F}$$

$$K_p = K_C \left(RT \right)^{\Delta n}$$

$$\nu_C E_C^{\Theta} = \nu_A E_A^{\Theta} + \nu_B E_B^{\Theta}$$

$$E_{cell} = E_{cell}^{\Theta} - \frac{RT}{\nu F} \ln Q$$

$$J_{energy} = -\kappa \frac{\mathrm{d}T}{\mathrm{d}z}$$

$$\Delta H^{\Theta} = -\nu F \left(E_{cell}^{\Theta} - T \frac{\mathrm{d} E_{cell}^{\Theta}}{\mathrm{d} T} \right)$$

$$J_{matter} = -D \frac{\mathrm{d}\mathcal{N}}{\mathrm{d}z}$$

$$J_{energy} = -\kappa \frac{\mathrm{d}T}{\mathrm{d}z}$$

$$J_{matter} = -D \frac{\mathrm{d}\mathcal{N}}{\mathrm{d}z}$$
 $J_{energy} = -\kappa \frac{\mathrm{d}T}{\mathrm{d}z}$ $J_{x-momentum} = -\eta \frac{\mathrm{d}v_x}{\mathrm{d}z}$

$$D = \frac{1}{3}\lambda v_{mean} = \frac{1}{3} \left(\frac{k_B T}{\sigma p}\right) \left(\frac{8RT}{\pi M}\right)^{1/2} \qquad \qquad \eta = \frac{1}{3}\lambda v_{mean} m \mathcal{N} = \frac{pMD}{RT}$$

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$$\eta = \eta_0 e^{E_a/RT}$$

$$\kappa = \frac{1}{3} \lambda v_{mean} \nu \mathcal{N} k_B = \frac{\nu p D}{T}$$
 $\nu = \frac{1}{2} N_{D.o.F.}$

$$G = \frac{1}{R} = \kappa \frac{A}{l}$$

$$u = \frac{ze}{f}$$

$$\Lambda_m = \frac{\kappa}{c} = (z_+ u_+ \nu_+ + z_- u_- \nu_-) F$$

$$f = 6\pi \eta a$$

$$u = \frac{ze}{f}$$

$$s = uE = u\frac{\Delta V}{m}$$

$$c(x,t) = \frac{n_0}{A\sqrt{\pi Dt}}e^{-x^2/4Dt}$$

$$x_{rms} = \sqrt{2Dt}$$

$$F=96,485\frac{C}{mol}$$

$$e = 1.60217662 \times 10^{-19} C$$