3610 Midterm Exam 1 Equations and Constants

Gas	Constant	V_2	اانود
Ctas	Constain	va	mes

8.314
$$\frac{J}{mole K}$$
 0.08314 $\frac{L \, bar}{mole \, K}$

$$0.08206 \quad \frac{L \ atm}{mole \ K} \qquad \qquad 8.314 \quad \frac{m^3 \ Pa}{mole \ K}$$

Boltzmann Constant Values

$$1.381 \times 10^{-23}$$
 $\frac{J}{K}$ 0.6950 $\frac{cm^{-1}}{K}$

$$1 L atm = 101.325 J$$

$$1 atm = 1.01325 bar$$

$$1 atm = 760 torr$$

$$1 atm = 101,325 Pa$$

$$\begin{split} H &= U + pV \\ \mathrm{d}U &= \mathrm{d}q + \mathrm{d}w \\ \mathrm{d}w &= -p_{external} \mathrm{d}V \\ p &= \frac{nRT}{V - nb} - a \frac{n^2}{V^2} \\ v_{rms} &= \left(\frac{3RT}{M}\right)^{1/2} \\ \Delta U_V &= C_V \Delta T \\ z &= \sigma v_{ret} \mathcal{N} \\ \Delta H_{rxn}(T_2) &= \Delta H_{rxn}(T_1) + \int_{T_1}^{T_2} \Delta C_p \mathrm{d}T \\ \Delta H_{rxn}^\circ &= \sum_{products} \nu_i \Delta H_{f,i}^\circ - \sum_{reactants} \nu_j \Delta H_{f,j}^\circ \\ v_i T_i^c &= V_f T_f \\ dU &= \left(\frac{\partial U}{\partial V}\right)_T \mathrm{d}V + \left(\frac{\partial U}{\partial T}\right)_V \mathrm{d}T = \pi_T \mathrm{d}V + C_V \mathrm{d}T \\ \mathrm{d}H &= \left(\frac{\partial H}{\partial p}\right)_T \mathrm{d}p + \left(\frac{\partial H}{\partial T}\right)_p \mathrm{d}T = -\mu C_p dp + C_p \mathrm{d}T \end{split}$$

18 VIIIA 8A 8A 8A 8A	Helium 4.003	o Ne	Neon 20.180	18 Ar	Argon 39.948	36	궃	Krypton 83.798	54	Xe	Xenon 131.294	98	R	Radon 222.018	118	0	Oganesson [294]
ţ	VIIA VI	L	Fluorine 18.998	17 C	Chlorine 35.453	35	Br	Bromine 79.904	53	-	Iodine 126.904	85	At	Astatine 209.987	117	Z	Tennessine [294]
ý	VIA 6A	٥ _	Oxygen 15.999	16	Sulfur 32.066	34	Se	Selenium 78.971	52	P	Tellurium 127.6	84	Po	Polonium [208.982]	116	2	Livermorium [293]
ŭ.	5 V S		Nitrogen 14.007	15 D	Phosphorus 30.974	33	As	Arsenic 74.922	51	Sp	Antimony 121.760	83	<u>.</u>	Bismuth 208.980	115	Σ	Moscovium [289]
;	1V 4 4 A	U	Carbon 12.011	¹⁴ S	Silicon 28.086	32	Ge	Germanium 72.631	20	Sn	Tin 118.711	82	Pb	Lead 207.2	114	ᄑ	Flerovium [289]
ć	IIA 3A	ص	Boron 10.811		Aluminum 26.982	_	Вa	Gallium 69.723	49	I I	Indium 114.818	81	F	Thallium 204.383		Z	Nihonium [286]
ents					11B 2B		Zu	Zinc 65.38	48	<u>გ</u>	Cadmium 112.414	80	Ē	Mercury 200.592	Ė	5	Copernicium [285]
Table of the Elements				7	8 8	29	3	Copper 63.546	7	Ag	Silver 107.868	6	Au	Gold 196.967	111	Rd	Roentgenium [280]
f the				0		28	Ż	Nickel 58.693	46	Pd	Palladium 106.42	8	ፚ	Platinum 195.085	10	Ds	Ę
ble o				6			ဝိ	Cobalt 58.933	4		Rhodium 102.906	7	Ļ	Iridium 192.217	109	Ĭ	Meitnerium D
				00		27	Fe	Iron 55.845	4		Ruthenium 101.07	1	Os	Osmium 190.23	Ė	H	_
Periodic				7	VIIB 7B	56	Z	Manganese 54.938	44	ည	Technetium F 98.907	92	Re	Rhenium 186.207	Т	Bh	Bohrium [264]
				g	VIB 6B		Ċ	Ε	43	Θ	Molybdenum 95.95	75	>	Tungsten 183.84	-		Seaborgium [266]
				ın	VB 5B		>	Vanadium C 50.942	42		Niobium Mc 92.906	74	اط م	c	_		
				4	IVB 48		ï	Titanium V 47.867	41		Zirconium 91.224	73	Ŧ	Hafnium 178.49	4 105	R F	Rutherfordium [261]
				m	III B 3 B	22	Sc	Scandium T 44.956	40	>	Yttrium Zi 88.906	71 72		_	89-103 104		Rut
•	IA 2A	Be	Beryllium 9.012		Magnesium 24.305	21	e B		39	<u>'</u>	Strontium Y 87.62 8	57-71	Ba	Barium 137.328	-68	Ra	Radium 226.025
		4				20	U		38			26	<u>m</u>		88	~	
- 4 5 T	Hydrogen 1.008	:	Lithium 6.941	1 Z	Sodium 22.990	19	¥	Potassium 39.098	37	Rb	Rubi d iun 85.468	22	S	Cesium 132.905	87	Ï	Francium 223.020

	57	58	59	09	61	62	63	64	65	99	29	89		70	71
Lanthanide Series	E	Ö	P	P Z		Sm	Eu	рg	q T	۵	유	山	٦	Ab	P _L
	Lanthanum	Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Holmium	Erbium		Ytterbium	Lutetium
	138.905	140.116	140.908	144.243		150.36	151.964	15/.25	158.925	162.500	164.930	16/.259	- 1	1/3.055	1/4.96/
	89	06	91	92	93	94	95	96	97				101	102	103
Actinide	Ac	두	Pa		2 Z	Pu	Am	5	路	უ	Es	Fm	δ Σ	°Z	۲
	Actinium	Thorium	Protactinium	Uranium	Neptunium	Plutonium	Americium	Curium	Berkelium				Mendelevium	Nobelium	Lawrencium
	227.028	232.038	231.036	238.029	237.048	244.064	243.061	247.070	247.070				258.1	259.101	[562]

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