

# 3610 Final Exam Equations and Constants

Gas Constant Values			
8.314	$\frac{J}{mole\ K}$	0.08314	$\frac{L\ bar}{mole\ K}$
0.08206	$\frac{L\ atm}{mole\ K}$	8.314	$\frac{m^3\ Pa}{mole\ K}$
Boltzmann Constant Values			
$1.381 \times 10^{-23}$	$\frac{J}{K}$	0.6950	$\frac{cm^{-1}}{K}$

Conversions		
1 L atm	=	101.325 J
1 atm	=	1.01325 bar
1 atm	=	760 torr
1 atm	=	101,325 Pa

$$F = 96,485 \frac{C}{mol}$$

$$e = 1.60217662 \times 10^{-19} C$$

$$H = U + pV$$

$$Z = \frac{pV}{nRT} = \left(1 + \frac{B}{V_m} + \frac{C}{V_m^2} + \dots\right)$$

$$dU = \delta q + \delta w$$

$$C_{V,m} = \frac{1}{2} R \cdot n_{D.o.F}$$

$$\delta w = -p_{external} dV$$

$$C_{p,m} = C_{V,m} + R$$

$$p = \frac{nRT}{V - nb} - a \frac{n^2}{V^2}$$

$$v_{mean} = \left(\frac{8RT}{\pi M}\right)^{1/2}$$

$$v_{rms} = \left(\frac{3RT}{M}\right)^{1/2}$$

$$\Delta H_p = C_p \Delta T$$

$$\Delta U_V = C_V \Delta T$$

$$pV = nRT$$

$$z = \sigma v_{rel} \mathcal{N}$$

$$v_{rel} = \sqrt{2} v_{mean}$$

$$\mathcal{N} = \frac{N}{V} = \frac{p}{k_BT}$$

$$\lambda = \frac{v_{rel}}{z} = \frac{k_BT}{\sigma p}$$

$$\Delta H_{rxn}(T_2) = \Delta H_{rxn}(T_1) + \int_{T_1}^{T_2} \Delta C_p \mathrm{d}T$$

$$\Delta H \approx \Delta U + \Delta n_{gas}RT$$

$$\Delta H_{rxn}^{\circ} = \sum_{products} \nu_i \Delta H_{f,i}^{\circ} - \sum_{reactants} \nu_j \Delta H_{f,j}^{\circ}$$

$$\pi_T = \left(\frac{\partial U}{\partial V}\right)_T$$

$$p_iV_i^{\gamma}=p_fV_f^{\gamma}\qquad \gamma=\frac{C_{p,m}}{C_{V,m}}$$

$$\alpha = \frac{1}{V} \left(\frac{\partial V}{\partial T}\right)_p$$

$$V_iT_i^c = V_fT_f^c \qquad c = \frac{C_{V,m}}{R}$$

$$\kappa_T = -\frac{1}{V} \left(\frac{\partial V}{\partial p}\right)_T$$

$$\mathrm{d}U = \left(\frac{\partial U}{\partial V}\right)_T \mathrm{d}V + \left(\frac{\partial U}{\partial T}\right)_V \mathrm{d}T = \pi_T \mathrm{d}V + C_V \mathrm{d}T$$

$$\mathrm{d}H = \left(\frac{\partial H}{\partial p}\right)_T \mathrm{d}p + \left(\frac{\partial H}{\partial T}\right)_p \mathrm{d}T = -\mu C_p \mathrm{d}p + C_p \mathrm{d}T$$

$$G = H - TS$$

$$A = U - TS$$

$$\mathrm{d}G = -S\mathrm{d}T + V\mathrm{d}p$$

$$\left(\frac{\partial \Delta G/T}{\partial T}\right)_p = -\frac{\Delta H}{T^2}$$

$$S = k_B \ln \mathcal{W}$$

$$C_V = \frac{1}{2}R \cdot n_{D.o.F}$$

$$\frac{q_H}{q_C} = -\frac{T_H}{T_C}$$

$$C_p = C_V + R$$

$$\eta = \frac{w}{q_H}$$

$$\mathrm{d}S_{system} = \frac{\mathrm{d}q_{reversible}}{T}$$

$$S(T) = S(0) + \int_0^T \frac{C_p}{T} \mathrm{d}T + \sum_{transitions} \frac{\Delta H_{trs}}{T_{trs}}$$

$$\ln \frac{p_2}{p_1} = \frac{-\Delta H}{R} \left( \frac{1}{T_2} - \frac{1}{T_1} \right)$$

$$\eta = 1 - \frac{T_C}{T_H}$$

$$\mathrm{d}S_{surrounding} = -\frac{\mathrm{d}q_{sys}}{T}$$

$$\Delta S_T = nR \ln \frac{V_f}{V_i}$$

$$\Delta S_V = C_V \ln \frac{T_f}{T_i}$$

$$\Delta S_p = C_p \ln \frac{T_f}{T_i}$$

$$\Delta S_{trs} = \frac{\Delta H_{trs}}{T}$$

$$G(p_2) = G(p_1) + nRT \ln \frac{p_2}{p_1}$$

$$F = C - P + 2$$

$$\Delta S(T_2) = \Delta S(T_1) + \int_{T_1}^{T_2} \frac{\Delta C_p}{T} \mathrm{d}T$$

$$\left(\frac{\partial \mu}{\partial T}\right)_p = -S_m$$

$$\frac{\mathrm{d}p}{\mathrm{d}T} = \frac{\Delta H}{T\Delta V_m}$$

$$\left(\frac{\partial \mu}{\partial p}\right)_T = V_m$$

$$\Delta G_{mix} = nRT\left(\chi_A \ln \chi_A + \chi_B \ln \chi_B\right)$$

$$\Delta S_{mix} = -nR\left(\chi_A \ln \chi_A + \chi_B \ln \chi_B\right)$$

$$p_A = \chi_A p_A^\star$$

$$p_B = \chi_B K_B$$

$$\mu_A = \mu_A^\star + RT \ln \chi_A$$

$$K_b = \frac{RT_b^{\star 2}}{\Delta H_{vap}}$$

$$\Delta T_b=K_b\chi_B=K_bC_B$$

$$K_f=\frac{RT_f^{\star 2}}{\Delta H_{freeze}}$$

$$\Delta T_f=K_f\chi_B=K_fC_B$$

$$\frac{n_g}{n_{total}}=\left|\frac{x_B-z_B}{x_B-y_B}\right|$$

$$\frac{n_l}{n_{total}}=\left|\frac{z_B-y_B}{x_B-y_B}\right|$$

$$\ln \gamma_A = \xi \chi_B^2 \qquad \ln \gamma_B = \xi \chi_A^2$$

$$\mu_B=\mu_B^{\ominus}+RT\ln\gamma_B\chi_B$$

$$I=\frac{1}{2}\sum_{ions}z_i^2\left(\frac{c_i}{c^{\ominus}}\right)\qquad c^{\ominus}=1\;molal$$

$$\log \gamma_{\pm} = -\frac{A\,|z_+z_-|\sqrt{I}}{1+B\sqrt{I}} + CI$$

$$\ln \frac{K_2}{K_1} = -\frac{\Delta H^\ominus}{R}\left(\frac{1}{T_2} - \frac{1}{T_1}\right)$$

$$\Delta G = \Delta G^\ominus + RT \ln Q$$

$$E_{cell}=E_{cathode}-E_{anode}$$

$$\Delta G^\ominus = -RT \ln K_{eq}$$

$$\frac{\mathrm{d} E_{cell}^\ominus}{\mathrm{d} T} = \frac{\Delta S^\ominus}{\nu F}$$

$$K_p=K_C\left(RT\right)^{\Delta n}$$

$$\nu_C E_C^\ominus = \nu_A E_A^\ominus + \nu_B E_B^\ominus$$

$$E_{cell}=E_{cell}^\ominus-\frac{RT}{\nu F}\ln Q$$

$$\Delta H^\ominus = -\nu F \left(E_{cell}^\ominus - T \frac{\mathrm{d} E_{cell}^\ominus}{\mathrm{d} T}\right)$$

$$J_{matter}=-D\frac{\mathrm{d}\mathcal{N}}{\mathrm{d}z}\qquad J_{energy}=-\kappa\frac{\mathrm{d}T}{\mathrm{d}z}\qquad J_{x-momentum}=-\eta\frac{\mathrm{d}v_x}{\mathrm{d}z}$$

$$D=\frac{1}{3}\lambda v_{mean}=\frac{1}{3}\left(\frac{k_BT}{\sigma p}\right)\left(\frac{8RT}{\pi M}\right)^{1/2}$$

$$\eta=\frac{1}{3}\lambda v_{mean}m\mathcal{N}=\frac{pMD}{RT}$$

$$\eta = \eta_0 e^{E_a/RT}$$

$$\kappa = \frac{1}{3} \lambda v_{mean} \nu \mathcal{N} k_B = \frac{\nu p D}{T} \qquad \nu = \frac{1}{2} N_{D.o.F.}$$

$$G=\frac{1}{R}=\kappa\frac{A}{l}$$

$$\Lambda_m = \frac{\kappa}{c} = \left(z_+ u_+ \nu_+ + z_- u_- \nu_- \right) F$$

$$f=6\pi\eta a$$

$$u=\frac{ze}{f}$$

$$s=uE=u\frac{\Delta V}{m}$$

$$c(x,t)=\frac{n_0}{A\sqrt{\pi Dt}}e^{-x^2/4Dt}$$

$$x_{rms}=\sqrt{2Dt}$$

$$v=\frac{\mathrm{d}\left[\mathrm{A}\right]}{\nu_A\mathrm{d}t}=\frac{1}{V}\frac{\mathrm{d}\xi}{\mathrm{d}t}$$

$$\log\left(\frac{v_2}{v_1}\right)=m\log\left(\frac{[\mathrm{A}]_2}{[\mathrm{A}]_1}\right)$$

$$\chi = \chi_0 e^{-t/\tau}$$

$$\tau = \frac{1}{k_r + k_r'}$$

$$k=Ae^{-\frac{E_a}{RT}}$$

$$\log\left(\frac{k_2}{k_1}\right)=-\frac{E_a}{R}\left(\frac{1}{T_2}-\frac{1}{T_1}\right)$$

$$v_{\mathrm{Lind.}-\mathrm{Hinsh.}}=\frac{k_a k_b [A]^2}{k_b+k_a' [A]}$$

$$\tau_0 = \frac{1}{k_F + k_{IC} + k_{ISC}}$$

$$\phi_{F,0} = \frac{k_F}{k_F + k_{IC} + k_{ISC}} = k_F \tau$$

$$\frac{\phi_0}{\phi} = 1 + \tau_0 k_Q [Q]$$

$$\eta_T = 1 - \frac{\phi_F}{\phi_{F,0}} = \frac{R_0^6}{R_0^6 + R^6}$$

$$\frac{1}{v} = \frac{1}{v_{max}} + \left(\frac{K_M}{v_{max}}\right)\frac{1}{[\mathrm{S}]_0}$$

$$v = P\sigma v_{rel} N_A^2 e^{-\frac{E_a}{RT}}[A][B]$$

$$k_d = \frac{8RT}{3\eta}$$

$$k \propto e^{\Delta S^\ddagger/R} e^{\Delta H^\ddagger/RT}$$

# Periodic Table of the Elements

1 1A 1A	2 IIA 2A	3 IIIB 3B	4 IVB 4B	5 VB 5B	6 VIB 6B	7 VIIB 7B	8 VIII 8	9 VIII 9	10 VIII 10	11 IB 1B	12 IIB 2B	13 IIIA 3A	14 IVA 4A	15 VA 5A	16 VIA 6A	17 VIIA 7A	18 VIIIA 8A
1 <b>H</b> Hydrogen 1.008	2 <b>He</b> Helium 4.003	3 <b>Li</b> Lithium 6.941	4 <b>Be</b> Beryllium 9.012	5 <b>B</b> Boron 10.811	6 <b>C</b> Carbon 12.011	7 <b>N</b> Nitrogen 14.007	8 <b>O</b> Oxygen 15.999	9 <b>F</b> Fluorine 18.998	10 <b>Ne</b> Neon 20.180	11 <b>Na</b> Sodium 22.990	12 <b>Mg</b> Magnesium 24.305	13 <b>Al</b> Aluminum 26.982	14 <b>Si</b> Silicon 28.086	15 <b>P</b> Phosphorus 30.974	16 <b>S</b> Sulfur 32.066	17 <b>Cl</b> Chlorine 35.453	18 <b>Ar</b> Argon 39.948
19 <b>K</b> Potassium 39.098	20 <b>Ca</b> Calcium 40.078	21 <b>Sc</b> Scandium 44.956	22 <b>Ti</b> Titanium 47.867	23 <b>V</b> Vanadium 50.942	24 <b>Cr</b> Chromium 51.996	25 <b>Mn</b> Manganese 54.938	26 <b>Fe</b> Iron 55.845	27 <b>Co</b> Cobalt 58.933	28 <b>Ni</b> Nickel 58.693	29 <b>Cu</b> Copper 63.546	30 <b>Zn</b> Zinc 65.38	31 <b>Ga</b> Gallium 69.723	32 <b>Ge</b> Germanium 72.631	33 <b>As</b> Arsenic 74.922	34 <b>Se</b> Selenium 78.971	35 <b>Br</b> Bromine 79.904	36 <b>Kr</b> Krypton 83.798
37 <b>Rb</b> Rubidium 85.468	38 <b>Sr</b> Strontium 87.62	39 <b>Y</b> Yttrium 88.906	40 <b>Zr</b> Zirconium 91.224	41 <b>Nb</b> Niobium 92.906	42 <b>Mo</b> Molybdenum 95.95	43 <b>Tc</b> Technetium 98.907	44 <b>Ru</b> Ruthenium 101.07	45 <b>Rh</b> Rhodium 102.906	46 <b>Pd</b> Palladium 106.42	47 <b>Ag</b> Silver 107.868	48 <b>Cd</b> Cadmium 112.414	49 <b>In</b> Indium 114.818	50 <b>Sn</b> Tin 118.711	51 <b>Sb</b> Antimony 121.760	52 <b>Te</b> Tellurium 127.6	53 <b>I</b> Iodine 126.904	54 <b>Xe</b> Xenon 131.294
55 <b>Cs</b> Cesium 132.905	56 <b>Ba</b> Barium 137.328	57-71 Lanthanide Series	72 <b>Hf</b> Hafnium 178.49	73 <b>Ta</b> Tantalum 180.948	74 <b>W</b> Tungsten 183.84	75 <b>Re</b> Rhenium 186.207	76 <b>Os</b> Osmium 190.23	77 <b>Ir</b> Iridium 192.217	78 <b>Pt</b> Platinum 195.085	79 <b>Au</b> Gold 196.967	80 <b>Hg</b> Mercury 200.592	81 <b>Tl</b> Thallium 204.383	82 <b>Pb</b> Lead 207.2	83 <b>Bi</b> Bismuth 208.980	84 <b>Po</b> Polonium [208.982]	85 <b>At</b> Astatine 209.987	86 <b>Rn</b> Radon 222.018
87 <b>Fr</b> Francium 223.020	88 <b>Ra</b> Radium 226.025	89-103 Actinide Series	104 <b>Rf</b> Rutherfordium [261]	105 <b>Db</b> Dubnium [262]	106 <b>Sg</b> Seaborgium [266]	107 <b>Bh</b> Bohrium [264]	108 <b>Hs</b> Hassium [269]	109 <b>Mt</b> Meitnerium [278]	110 <b>Ds</b> Darmstadtium [281]	111 <b>Rg</b> Roentgenium [280]	112 <b>Cn</b> Copernicium [285]	113 <b>Nh</b> Nihonium [286]	114 <b>Fl</b> Flerovium [289]	115 <b>Mc</b> Moscovium [289]	116 <b>Lv</b> Livermorium [293]	117 <b>Ts</b> Tennessine [294]	118 <b>Og</b> Oganesson [294]
57 <b>La</b> Lanthanum 138.905	58 <b>Ce</b> Cerium 140.116	59 <b>Pr</b> Praseodymium 140.908	60 <b>Nd</b> Neodymium 144.243	61 <b>Pm</b> Promethium 144.913	62 <b>Sm</b> Samarium 150.36	63 <b>Eu</b> Europium 151.964	64 <b>Gd</b> Gadolinium 157.25	65 <b>Tb</b> Terbium 158.925	66 <b>Dy</b> Dysprosium 162.500	67 <b>Ho</b> Holmium 164.930	68 <b>Er</b> Erbium 167.259	69 <b>Er</b> Erbium 167.259	70 <b>Tm</b> Thulium 168.934	71 <b>Yb</b> Ytterbium 173.055	72 <b>Lu</b> Lutetium 174.967	73 <b>Lu</b> Lutetium 174.967	74 <b>Lu</b> Lutetium 174.967
89 <b>Ac</b> Actinium 227.028	90 <b>Th</b> Thorium 232.038	91 <b>Pa</b> Protactinium 231.036	92 <b>U</b> Uranium 238.029	93 <b>Np</b> Neptunium 237.048	94 <b>Pu</b> Plutonium 244.064	95 <b>Am</b> Americium 243.061	96 <b>Cm</b> Curium 247.070	97 <b>Bk</b> Berkelium 247.070	98 <b>Cf</b> Californium 251.080	99 <b>Es</b> Einsteinium [254]	100 <b>Fm</b> Fermium 257.095	101 <b>Md</b> Mendelevium 258.1	102 <b>No</b> Nobelium 259.101	103 <b>Lr</b> Lawrencium [262]	104 <b>Lr</b> Lawrencium [262]	105 <b>Lr</b> Lawrencium [262]	106 <b>Lr</b> Lawrencium [262]