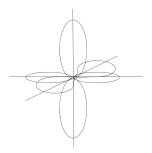
C orbitals

$$\frac{1l}{1S} \frac{1}{2S} \frac{1}{2P_X} \frac{1}{2P_Y} \frac{1}{2P_Z}$$

These are 2sp³ hybridized orbitals



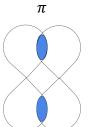


Bonding

Sigma bonds are stronger than pi bonds

σ





Methane forms a tetrahedral shape

