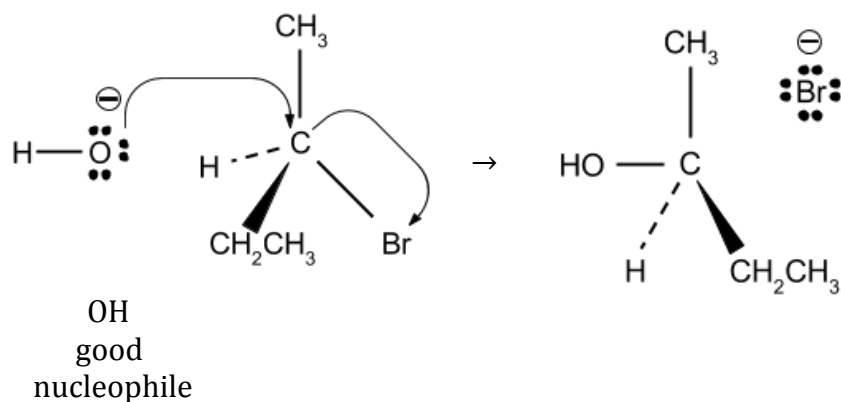


S_N2 Stereo Chemistry



Beginning R - 2 - bromo butane

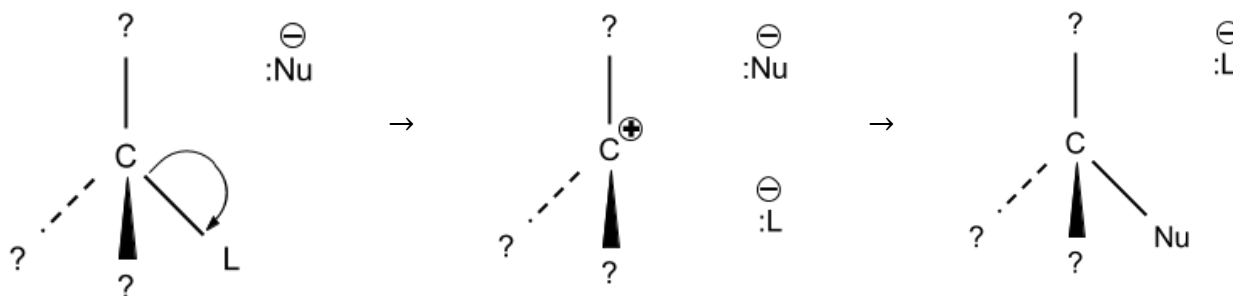
Ending S - 2 - butanol

All S_N2 reactions with a chiral C change their handedness between the beginning and ending stages.

R changes to S

S changes to R

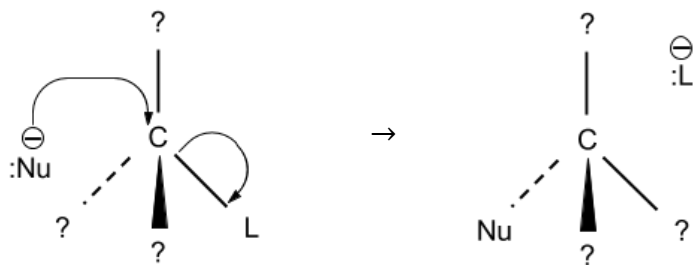
S_N1



S_N1 reactions can occur with a tertiary and secondary carbocation

But very unlikely with a primary carbocation

S_N2



S_N2 reactions occur when nucleophile is strong and when carbocation has very low steric hindrance
atoms attached to carbocation do not block nucleophile from attacking

Thus S_N2 reactions can occur with a central C such as a methyl or a primary carbocation

Summary of S_N1 and S_N2 basic difference

A weak nucleophile and many C atoms protecting the central C then S_N1, since :Nu cannot attack

A strong nucleophile and few or no C atoms protecting the central C then S_N2, since :Nu can attack