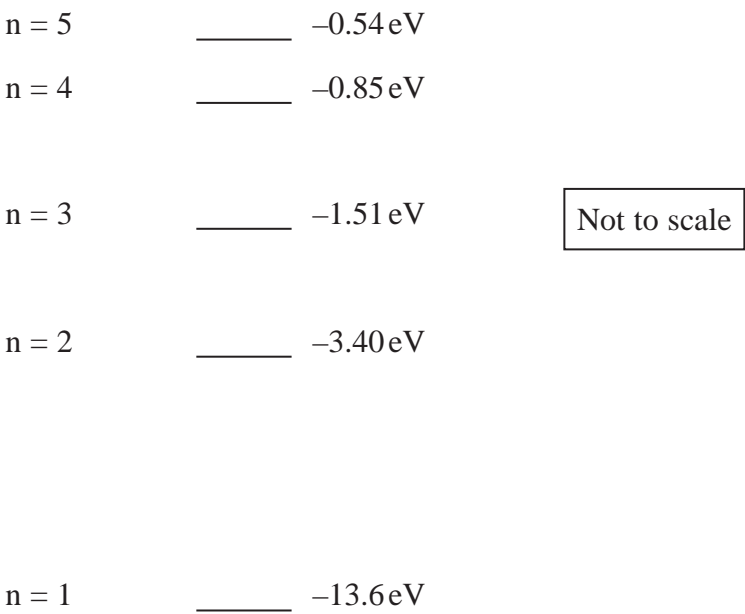


18 In 1864, William Huggins and William Miller used dark lines in the spectrum of the Sun to identify elements in the Sun’s atmosphere.

- *(a) Explain how gases in the Sun’s atmosphere cause dark lines in the spectrum corresponding to different elements.
- (6)

(b) The diagram shows some energy levels of a hydrogen atom.



The absorption spectrum for hydrogen includes a set of lines that all derive from transitions involving the n = 2 energy level. One of these lines is known as the hydrogen-alpha line.

Deduce the transition involved in the formation of the hydrogen-alpha line.

wavelength of hydrogen-alpha line = 656.46 nm

(4)

- (c) In 1868, William Huggins analysed light from the star Sirius A. The wavelength of the hydrogen-alpha line for light from Sirius A was slightly different from the hydrogen-alpha line observed from a source in a laboratory.

Huggins suggested that this difference could be explained using the Doppler effect and could be used to determine the speed and direction of the star’s motion relative to the Earth.

- (i) Assess Huggins’s suggestion.
- (3)

- (ii) Sirius A has a component of velocity away from the Earth of 5.5 km s^{−1}.

The wavelength of the hydrogen-alpha line observed from a source in the laboratory is 656.46 nm.

Calculate the wavelength of the hydrogen-alpha line as seen in the spectrum of Sirius A.

(2)

Wavelength =