

A student made measurements to determine if some gold coins were made from pure gold. The coins that were available to the student are shown below.



(Source: © Bjoern Wylezich/Shutterstock)

- (a) The student used digital calipers to measure the thickness  $t$  and the diameter  $d$  of one of the coins.
- (i) Calculate the volume  $V$  of the coin, and the percentage uncertainty in  $V$ .

$$\begin{aligned} t &= 1.54 \text{ mm} \\ d &= 22.16 \text{ mm} \end{aligned}$$

(7)

$V =$  .....

Percentage uncertainty in  $V = \dots\dots\dots$

- (ii) The student measured the mass of the coin using an electronic balance. The balance had a resolution of 0.1 g.

Assess whether the coin could be made from pure gold.

density of pure gold =  $1.93 \times 10^4 \text{ kg m}^{-3}$   
mass of coin = 11.2 g

(4)

- (b) The student's experimental method could have been improved.

Explain two changes the student could have made to the experimental method.

(4)