- The volume of the gas is decreased to $\frac{2}{3}V$ and the temperature increased to $\frac{6}{5}T$. Which of the following is the new pressure of the gas?
- \square A $\frac{5}{9}p$

2 A sample of an ideal gas has pressure p, volume V and absolute temperature T.

 \boxtimes **B** $\frac{4}{5}p$

 $\square \quad \mathbf{C} \quad \frac{5}{4}p$ $\square \quad \mathbf{D} \quad \frac{9}{5}p$