PHYS 1512: 14 (Final Week)

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Equations

$$B.E. = (\Delta m)c^2$$
 (Binding Energy) (1)

$$r \approx (1.2 * 10^{-15} m) A^{\frac{1}{3}}$$
 (atomic radius) (2)

$$A = N + Z$$
 (Atomic Mass number) (3)

$$A = A_o e^{-\lambda t} \quad \text{(Radioactive decay)} \tag{4}$$

$$N = N_o e^{-\lambda t}$$
 (Radioactive decay) (5)

$$\lambda = \frac{0.693}{T_{1/2}} \quad \text{(Half-life)} \tag{6}$$

Homework Question Chp 31 #12

Conservation laws

The binding energy of a nucleus is 359.5 MeV. What is the mass defect of the nucleus in atomic mass units?

** 1u = 931.5 MeV

Homework Question Chp 31 #35

DECAY

Strontium $_{38}Sr^{90}$ has a half-life of 29.1 yr. It is chemically similar to calcium, enters the body through the food chain, and collects in the bones. Consequently, $_{38}Sr^{90}$ is a particularly serious health hazard. How long (in years) will it take for 99.9467% of the $_{38}Sr^{90}$ released in a nuclear reactor accident to disappear?

Homework Question Chp 31 #4

Radius

By what factor does the nucleon number of a nucleus have to increase in order for the nuclear radius to increase by a factor of 4?

Homework Question Chp 32 #24

Electric potential energy in motion

The energy released by each fission within the core of a nuclear reactor is 2.00 102 MeV. The number of fissions occurring each second is 1.20 1017. Determine the power (in watts) that the reactor generates.

$$1 J = 1.6 * 10^{-19} eV$$

Homework Question Chp 32 #32

Electric potential energy in motion

In one type of fusion reaction a proton fuses with a neutron to form a deuterium nucleus:

$${}_{1}^{1}H + {}_{1}^{0}n \implies {}_{1}^{2}H + \gamma$$

The masses are 1_1H (1.0078 u), 0_1n (1.0087 u), and 2_1H (2.0141 u). The γ -ray photon is massless. How much energy (in MeV) is released by this reaction?