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# Crack-Free Monolayer Graphene Interlayer for Improving Perovskite Crystallinity and Energy Level Alignment in Efficient Inverted Perovskite Solar Cells

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Inverted planar perovskite solar cells (PSCs) have intrigued great promise in negative hysteresis, simple fabrication process, and flexible substrate implementation, in which the shared hole transport materiel is  $\text{NiO}_x$ . However, the low-temperature solution processing of the  $\text{NiO}_x$  film is usually accompanied by defect formation, which deteriorates the perovskite quality and device performance. Meanwhile, the energy-level offset between the  $\text{NiO}_x$  and perovskite films is relatively large, limiting the interfacial charge transport. To suppress those setbacks, a defect-free monolayer graphene sheet is transferred onto the  $\text{NiO}_x$  film surface as a template for van der Waals epitaxial growth of perovskite films for the first time, leading to enhancing crystallinity with a large grain size of perovskite layer, 0.20 eV energy level offset drop, and accelerating charge transfer for the devices. Finally, the power conversion efficiency of 19.21% without hysteresis is achieved, exceeding 18.35% of the control device.

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# 1. Introduction

Halide perovskite materials have exceptional photoelectric properties, such as high absorption coefficient, appropriate direct bandgap, long carrier diffusion length, and low exciton binding energy.<sup>[1,2]</sup> Typically, perovskite solar cells (PSCs) have boosted certified power conversion efficiency (PCE) up to 25.7%, [3] being rendered comparable with current commercial solar cells. In PSCs, perovskite film is sandwiched between the electron-transport layer (ETL) and hole-transport layer (HTL).[4] In perovskite film, the electron-hole pairs are generated after capturing photons, then separated and transported into the ETL and HTL, respectively, and finally collected through both two electrodes as photocurrent.<sup>[5]</sup> In

the aforementioned working process, the perovskite crystallinity and energy level alignment are critical to the photovoltaic parameters of PSCs, such as open-circuit voltage ( $V_{\rm OC}$ ), and short-circuit current ( $I_{\rm SC}$ ), and fill factor (FF).

According to the device architectures, the PSCs can be categorized in planar and mesoporous types on whether using the mesoporous scaffolds. The highly efficient PSCs are usually achieved on metal-oxide-based mesoporous architecture. [6] However, the mesoporous scaffolds are commonly fabricated with requiring high-temperature annealing, which increases the cost of production and hinders the progress of flexible modules. While the inverted planar PSCs reveal great promise in negative hysteresis, simple fabrication process, and flexible substrate implementation. [7] In the inverted planar device, the HTL, besides being a major role in hole extraction, serves as the landing surface for the growing perovskite layer. Thus, the perovskite crystallinity is crucially affected by the surface of HTL. Meanwhile, the less energy level offset at the HTL and perovskite interface is essential for reducing the loss of  $V_{\rm OC}$  and enhancing charge transfer.

In inverted planar PSCs, poly(3,4-ethylenedioxythiophene): poly(styrenesulfonate) (PEDOT:PSS), poly(triarylamine) (PTAA), and nickel oxide (NiO<sub>x</sub>) are the most frequently materials of HTL.<sup>[8]</sup> Though PEDOT:PSS was first applied as HTL in inverted PSCs, its natural hygroscopicity and acidity induce instability and give rise to low performance. In addition, the highest occupied



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molecular orbital (HOMO) level of PEDOT:PSS (5.0 eV) cannot match well with the valence band maximum (VBM) of the perovskite layers, which strongly limits the theoretical Voc. Compared with the PEDOT:PSS, PTAA HTL in an inverted device has better stability and hole transferability, but the material synthesis needs a more complex process and high cost, limiting the large scale utilization. [9] The inorganic NiO<sub>x</sub> HTL has attracted much attention due to its low-cost, good transparency in the visible and infrared region, good hole transport ability, and excellent stability. The record PCE is over 22% of the MAPbI<sub>3</sub>-based inverted PSCs with NiO<sub>x</sub> HTL. [10–12]

To date, the  $\mathrm{NiO}_x$  HTL in inverted PSC has been prepared by various methods, such as magnetron sputtering, electron beam deposition, pulsed laser deposition, and solution process. [13] Among them, the solution process to fabricate  $\mathrm{NiO}_x$  HTL has been considered a promising approach due to its simple fabrication, low cost, and high yield. However, the solution-processed  $\mathrm{NiO}_x$  HTL is subjected to the natural randomly distributed defects and intrinsic conductivity. [14] In serious cases, the natural defects influence the perovskite nucleation and crystallinity randomly. Meanwhile, the HOMO level of  $\mathrm{NiO}_x$  HTL is 5.3 eV, resulting in a large offset of energy level in  $\mathrm{MAPbI}_3$ -based inverted PSCs. [15] Therefore, interface engineering has been considered an effective way for the solution-processed  $\mathrm{NiO}_x$  HTL to achieve higher efficiency of PSCs, which can improve the conductivity, modify the energy level, and reduce the surface defects for perovskite crystallization. [16–18]

Monolayer graphene (MLG) is a strictly 2D crystal of carbon atoms that exhibits excellent conductivity, extraordinary optical transmittance, high flexibility, and stability. [19,20] In addition, high-quality crack-free MLG has a super-clean surface and homogeneousness with a six-member ring structure composed of carbon atoms. [21] A prior study indicated that MLG induces the crystallization of P3HT films, where the particular crystalline mosaic with  $\pi$ - $\pi$  stacking oriented homogeneously at various angles inside the film favors the creation of a continuous pathway of interconnected crystallites for charge transport. [22] Meanwhile, due to the large special surface area and highly delocalized electrons, MLG can regulate energy levels at the interface and produce efficient charge transfer when in contact with a semiconductor. [23]

In this article, a large-area sheet of crack-free MLG was transferred onto the surface of NiO<sub>x</sub> film for enhancement of inverted planar PSCs. The defect-free ultra-clean MLG was conducted as a template for van der Waals epitaxial growth of perovskite films by solution process for the first time. The crystallinity of perovskite film was enhanced, and grain size increased about three times. An obvious drop of 0.20 eV of energy level offset was achieved for the NiO<sub>x</sub>-based hole transport layer, which accelerated the charge transport and extraction for the devices. In the device structure of fluorine-doped tin oxide (FTO)/NiO<sub>x</sub>/MLG/MAPbI<sub>3</sub>/PCB61M/PCP/Ag, PCE of 19.21% without hysteresis was achieved with the improvement in parameters of Voc (1.093 V), Jsc (22.6 mA cm<sup>-2</sup>), and FF (77.8%), exceeding 18.35% (PCE) of the control device.

#### 2. Results and Discussions

#### 2.1. Crack-Free MLG Transfer on NiOx

The schematic diagram of transferring crack-free MLG on  $NiO_x$  film is illustrated in **Figure 1a**. The MLG-modified  $NiO_x$  film is

fabricated as follows: NiOx nanoparticle aqueous dispersion is spin-coated on FTO conductive glass. MLG is synthesized on Cu foil by chemical vapor deposition (CVD), then the MLG is crack freely transferred onto NiOx film with the sacrificial mechanically reinforcing polymer film, which is low-temperature processed. [24] The top-view scanning electron microscope (SEM) image of MLG-modified NiO<sub>x</sub> film verified the crack-freely transfer of MLG in Figure 1b. Meanwhile, the Raman spectrum also confirmed the MLG-modified NiO<sub>x</sub> film, shown in Figure 1c. In the Raman spectrum of NiO<sub>x</sub>/MLG film, the peak of the G-band is the characteristic peak of graphene, resulting from the in-plane vibration of sp<sup>2</sup> carbon atoms.<sup>[25]</sup> The symmetric peak of the 2D-band is sharp with full width at half-maximum about 35 cm<sup>-1</sup>, indicating the monolayer graphene. The inconspicuous peak of the D-band suggests few defects in the graphene layer. Besides, the peak located at 1085.2 cm<sup>-1</sup> is assigned to 2 longitudinal optical (2LO) phonon modes of NiO<sub>x</sub>: [26] The morphology is investigated via SEM. Figure S1a and S1b, Supporting Information, exhibits uniform and smooth surfaces of both NiOx and NiOx/MLG films, which is considered to contribute to full coverage and crystallization of the MAPbI3 film. [27] In addition, after covering MLG on NiO<sub>x</sub> film, the decreased roughness will be beneficial to fabricating perovskite film in high quality (shown in Figure S1c-f, Supporting Information).

#### 2.1.1. MAPbI<sub>3</sub> Crystallization Modification by MLG

In inverted PSC, the  $\mathrm{NiO}_x$  layer is in front of perovskite film under illumination. Therefore, the transparency of  $\mathrm{NiO}_x$  should be high to ensure MAPbI<sub>3</sub> absorbs the light and generates carriers. Comparing the UV–vis absorption and transmittance of  $\mathrm{NiO}_x$  and  $\mathrm{NiO}_x/\mathrm{MLG}$  film, the  $\mathrm{NiO}_x/\mathrm{MLG}$  film displays a slightly stronger absorption than  $\mathrm{NiO}_x$  film (**Figure 2a**). However, given that MLG is an atomic layer with high optical transmittance, it would not influence the substrate transparency significantly, which is demonstrated in the transmittance spectra of  $\mathrm{NiO}_x$  and  $\mathrm{NiO}_x/\mathrm{MLG}$  films (Figure 2b). At the same time, the UV–vis absorption of MAPbI<sub>3</sub> is slightly enhanced after MLG-modified  $\mathrm{NiO}_x$  as substrate (see Figure 2c), which also indicates a higher quality of MAPbI<sub>3</sub> obtained on  $\mathrm{NiO}_x/\mathrm{MLG}$ .

In the meanwhile, the MLG modification has changed certain properties of NiOx film. The typical X-ray diffraction (XRD) patterns of MAPbI<sub>3</sub> on NiO<sub>x</sub> and NiO<sub>x</sub>/MLG substrate are presented in Figure 3a. The main strong peaks of both films at 14.2°, 28.5°, and 32.0° correspond to the (110), (220), and (310) planes, respectively, [28] which are indexed to the perovskite structure. All the peaks are attributed to MAPbI<sub>3</sub> and FTO, suggesting that a pure phase of MAPbI<sub>3</sub> is obtained. The crystallization of MAPbI<sub>3</sub> is improved when deposited on NiOx/MLG, as it shows stronger diffraction peaks than those on NiOx. It is essential to achieve high photovoltaic performance with high-quality perovskite films. Figure 3b,c displays surface SEM images of NiO<sub>x</sub>-MAPbI<sub>3</sub> and NiO<sub>x</sub>/MLG-MAPbI<sub>3</sub>, which show the MAPbI<sub>3</sub> with dense, smooth, and pinhole-free. By comparison, the grain size of MAPbI<sub>3</sub> on NiO<sub>x</sub>/MLG is more uniform and slightly larger than that on NiOx, which agrees with the result of XRD that the MLG modified NiO<sub>x</sub> improves the crystallization of MAPbI<sub>3</sub>.



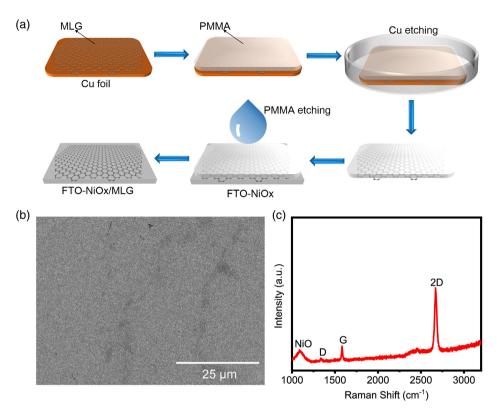


Figure 1. a) Schematic diagram of the monolayer graphene (MLG) crack-free transfer to fluorine-doped tin oxide (FTO)-NiO<sub>x</sub> substrate; b) top-view scanning electron microscope (SEM) image of FTO-NiO<sub>x</sub>/MLG; c) Raman spectrum of NiO<sub>x</sub>/MLG film.

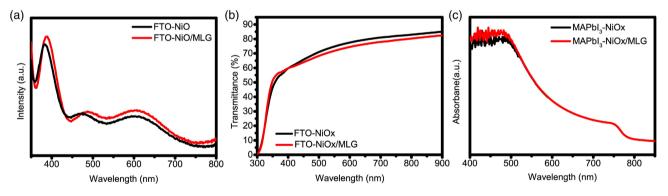


Figure 2. a) UV–vis absorption and b) transmittance spectra of  $NiO_x$  and  $NiO_x/MLG$  films on FTO; c) UV–vis absorption spectra of MAPbI<sub>3</sub> films on  $NiO_x$  and  $NiO_x/MLG$ .

#### 2.1.2. Hole Transportation Improvement by MLG

To inspect the effect of MLG modification on the energy level of NiO $_x$  film, ultraviolet photoelectron spectroscopy (UPS) is carried out. The details that calculate the work function (WF) and valence band maximum (E $_{VBM}$ ) are illustrated in Figure 4a for NiO $_x$  and 4b for NiOx/MLG. Briefly, the energy of the ultraviolet source (Helium I $\alpha$ ) is 21.22 eV (as the wavelength of Helium I $\alpha$  is 58.13 nm). [29] WF is calculated via 21.22 minus the high binding energy cutoff. Thus, WF $_{NiOx}$  = 4.05 eV and WF $_{NiOx/MLG}$  = 4.42 eV. For E $_{VBM}$ , it is obtained by totaling WF and the low binding energy cut-off,

which  $E_{VBM,\ NiOx} = 5.16\,\text{eV}$  and  $E_{VBM,\ NiOx/MLG} = 5.36\,\text{eV}$ . The results of UPS demonstrate that with the modification of MLG, the WF and  $E_{VBM}$  of HTL are increased that closer to the  $E_{VBM}$  of perovskite. The shift of  $E_{VBM}$  indicates that it improves hole extraction and better band alignment between HTL and MAPbI $_3$  valence band, which will discuss below details. The hole extraction of HTL is the key factor to affect the photovoltaic performance of PSCs. Steady-state photoluminescence (PL) measurement is usually implemented to evaluate the hole extraction ability of the thin films. Figure 4c illustrates the PL spectra of MAPbI $_3$  on different substrates with an excitation wavelength 500 nm irradiated from the MAPbI $_3$  film side. On

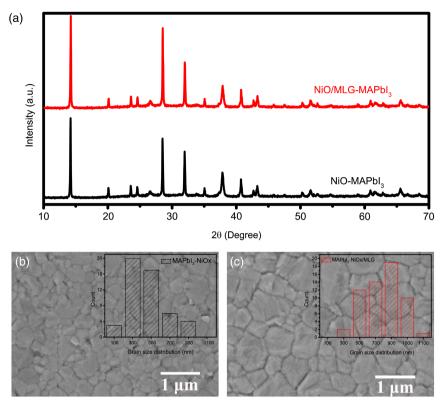


Figure 3. a) Typical X-ray diffraction (XRD) pattern of the MAPbI<sub>3</sub> films on NiO<sub>x</sub> and NiO<sub>x</sub>/MLG; Top-view SEM images and of MAPbI<sub>3</sub> films on: b) NiO<sub>x</sub> and c) NiO<sub>x</sub>/MLG.

the bare glass, MAPbI<sub>3</sub> appears to have a strong luminescence peak at ≈775 nm, because of serious carrier recombination. It is well-known that surface modification is a facile and effective way to enhance the contact and transport of carries between perovskite and ETL/HTL. To further investigate the hole transfer dynamics between MAPbI<sub>3</sub> and HTL, PL spectra of MAPbI<sub>3</sub> on  $NiO_x$  and  $NiO_x/MLG$  are compared. First, the peak intensities decrease obviously after importing HTL, implying the improvement of the extracting charge carriers. Then, the peak intensity of MAPbI<sub>3</sub> on NiO<sub>x</sub>/MLG is slightly lower than that of MAPbI<sub>3</sub> on NiO<sub>x</sub>, which can confirm that NiO<sub>x</sub>/MLG extract holes more efficiently. From the corresponding time-resolved PL (TRPL) spectra of MAPbI<sub>3</sub> on HTL shown in Figure 4d, the average fitting carrier life-time ( $\tau_{avo}$ ) for the sample based on NiO<sub>x</sub> is 35 ns and for that based on NiO<sub>x</sub>/MLG is 20 ns, which listed the fitting parameters in Table S1, Supporting Information. The slightly shorter  $\tau_{avg}$ demonstrates the improved hole extraction property, which can lead to higher Voc in PSCs. To further explore the hole extraction and transport mechanism, we draw the schematic diagram of MAPbI3 and HTL interface under illumination in Figure 4e. The HTL plays the role of an energy barrier to inhibit the recombination of holes. Hence, the band alignment for HTL and MAPbI<sub>3</sub> is essential for the photovoltaic performance. <sup>[30]</sup> The energy barrier for hole transfer from MAPbI3 to HTL is eliminated, because of the reduced valence band offset from 0.24 to 0.04 eV. Consequently, the carrier recombination is suppressed, and the hole extraction is enhanced, which improves device Voc.<sup>[31]</sup> When the carriers transfer from MAPbI<sub>3</sub> to HTL, they could lose part of free energy owing to the misaligned energy band, which would cause additional  $V_{\rm oc}$  loss. Furthermore, the misalignment of band energy also leads to the accumulation of photogenerated charges that could make against the charge extraction. The more suited the valence band between MAPbI and HTL is, the easier holes are extracted. Moreover, the amount of holes transferred to HTL is increased. Therefore, the hole extraction and transfer are enhanced after MLG modified the NiO $_x$  layer. It can be expected that the Voc of the device based on NiO $_x$ /MLG will be higher than that on NiO $_x$ .

#### 2.1.3. Device Architecture and Photovoltaic Performance

**Figure 5**a shows the device architecture that consists of FTO-glass, hole transport layer (MLG modified  $NiO_x$ ), perovskite layer (MAPbI<sub>3</sub>), electron transport layer (PCBM), interfacial layer (BCP), and contact (Ag). Given that the  $NiO_x$  is prepared via a low-temperature process, defects exist in the film inevitably. Worse, the defects usually cause the poor interface contact between  $NiO_x$  film and MAPbI<sub>3</sub> layer, leading to poor crystallinity and morphology of MAPbI<sub>3</sub> and even the low light absorption and charge recombination. Therefore, the insertion of MLG in the interface between  $NiO_x$  film and MAPbI<sub>3</sub> layer can shield the defects in  $NiO_x$  film, overcoming the above-mentioned obstacles to obtain a high-quality MAPbI<sub>3</sub> layer. Figure 5b displays the mechanism of MLG shielding the defects in  $NiO_x$  film.

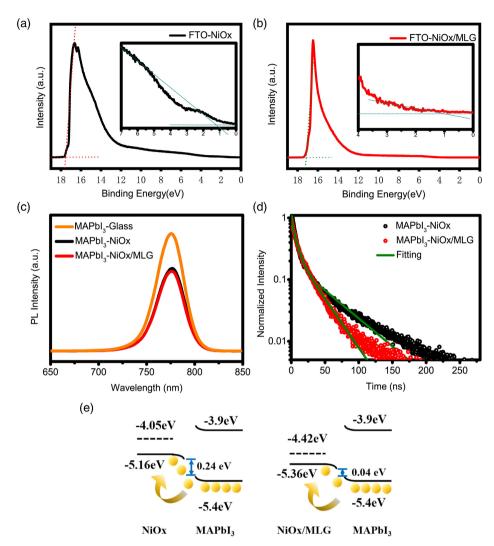


Figure 4. Ultraviolet photoelectron spectroscopy (UPS) spectra of: a)  $NiO_x$  and b)  $NiO_x/MLG$  on FTO; c) Steady-state photoluminescence (PL) spectra of MAPbI<sub>3</sub> films on glass,  $NiO_x$ , and  $NiO_x/MLG$ ; d) time-resolved PL (TRPL) spectra of MAPbI<sub>3</sub> on  $NiO_x$  and  $NiO_x/MLG$ ; e) Schematic diagram of hole transportation at the interface between  $NiO_x$  (left),  $NiO_x/MLG$  (right) and MAPbI<sub>3</sub> film under illumination.

The energy level of every layer is summarized in Figure 5c, in which the energy level of  $\mathrm{NiO}_x$  and  $\mathrm{NiO}_x/\mathrm{MLG}$  is calculated from UPS shown earlier, and of the other layers (FTO, MAPbI<sub>3</sub>, PCBM, BCP, and Ag) refers to the published works.<sup>[33]</sup> It is distinctly illustrated that the valence band of  $\mathrm{NiO}_x/\mathrm{MLG}$  is closer to it of MAPbI<sub>3</sub> than that of  $\mathrm{NiO}_x$ , which is conductive to hole extraction. The cross-sectional SEM images of devices based on  $\mathrm{NiO}_x$  and  $\mathrm{NiO}_x/\mathrm{MLG}$  are compared in Figure 5d,e, distinguished hierarchical in the device. In the meanwhile, the cross-sectional SEM images also agree with the morphology SEM result that the MAPbI<sub>3</sub> on  $\mathrm{NiO}_x/\mathrm{MLG}$  displays larger grains.

Additionally, as another measure of MAPbI<sub>3</sub> film quality, trapstate density ( $n_{\rm trap}$ ) can be calculated quantitatively by the space charge-limited current (SCLC) technique. As the modification of MLG is located between the hole transport layer and MAPbI<sub>3</sub>, the hole  $n_{trap}$  would be affected. The dark J-V curves of the hole-only devices based on NiO<sub>x</sub> and NiO<sub>x</sub>/MLG are measured and fitted in **Figure 6**a,b, with the structure of FTO/NiO<sub>x</sub>/(MLG/) perovskite/spiro-OMeTAD/Ag. According to the dependence of current on the applied voltage, the plot can be divided into three regions 1) the ohmic region at low bias showing a linear relationship between the current and the electric field; 2) the child region at high bias; 3) trap-filling limit (TFL) region exhibiting a sharp increase in the current. In the TFL region, traps are filled until trap-filling limit voltage ( $V_{\rm TFL}$ ) as bias increases, [34] in which  $n_{\rm trap}$  can be calculated by Equation (1)

$$n_{trap} = \frac{2\varepsilon\varepsilon_0 V_{TFL}}{eL^2} \tag{1}$$

where e is the elementary charge (1.6 ×  $10^{-14}$  C), L is the thickness of the perovskite layer (455 nm from the cross-sectional SEM),  $\varepsilon_0$  is the vacuum permittivity ( $\approx 8.85 \times 10^{-14}$  F cm<sup>-1</sup>),  $\varepsilon$  is the relative dielectric constant of the perovskite layer (adopt at  $\approx 28.8$ ). The  $V_{\rm TFL}$  value for the device base on NiO<sub>x</sub> and NiO<sub>x</sub>/MLG is 0.794 and 0.494 V, respectively. The calculated

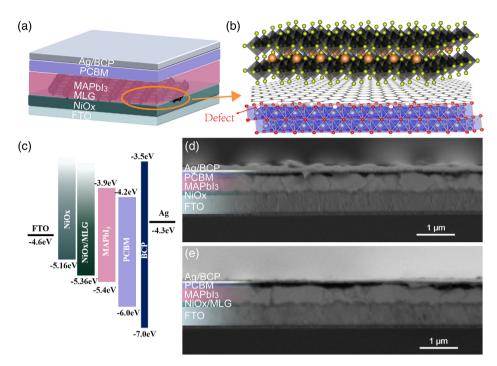


Figure 5. a) Schematic illustration of the inverted perovskite solar cell (PSC) based on MLG; b) Schematic diagram of MLG inserting the interface between  $NiO_x$  film and MAPbI<sub>3</sub> layer; c) Energy level diagram of each layer in the device; Cross-sectional SEM image of the real device based on d)  $NiO_x$  and e)  $NiO_x/MLG$ .

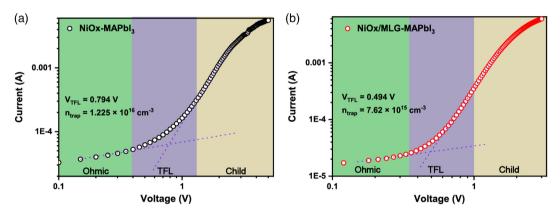


Figure 6. Space charge-limited current (SCLC) versus voltage of hole-only device based on: a) NiO<sub>x</sub>-MAPbI<sub>3</sub> and b) NiO<sub>x</sub>/MLG-MAPbI<sub>3</sub>.

hole  $n_{\rm trap}$  for the device base on NiO<sub>x</sub>/MLG is  $7.62 \times 10^{15}$  cm<sup>-3</sup>, which is much lower than that based on NiO<sub>x</sub> ( $1.225 \times 10^{16}$  cm<sup>-3</sup>). Therefore, the SCLC result agrees with the aforementioned result that the quality of MAPbI<sub>3</sub> is enhanced after introducing MLG between NiO<sub>x</sub> and MAPbI<sub>3</sub>.

Photovoltaic performance is the most important and intuitive evaluation criterion for PSCs. We analyze the photovoltaic performance of the device based on  $\mathrm{NiO}_x$  and  $\mathrm{NiO}_x/\mathrm{MLG}$  comprehensively and present it in **Figure 7**. For inverted PSCs, hysteresis behavior is known to be negligible. The J-V curves of the champion devices based on  $\mathrm{NiO}_x$  and  $\mathrm{NiO}_x/\mathrm{MLG}$  HTL are exhibited in Figure 7a,b. Both devices based on  $\mathrm{NiO}_x$  and  $\mathrm{NiO}_x/\mathrm{MLG}$  show negative hysteresis, which also demonstrates the good carrier transport at every

layer. It is demonstrated that the photovoltaic performance is remarkably boosted after using MLG modifying NiO $_x$  HTL. The photovoltaic parameters from J-V curves for both devices are listed inset in Figure 7a. The champion device based on NiO $_x$  obtains a PCE of 18.35%. While with the modification of MLG, the device PCE reaches 19.21%. By comparison, the improvement of photovoltaic parameters mainly comes from the enhancement of  $V_{\rm oc}$ , attributed to the alignment of energy levels. The detailed statistical photovoltaic parameters of PSCs based on NiO $_x$  and NiO $_x$ /MLG are compared and listed in Figure 7c, which summarized the photovoltaic parameters of devices in Table S2 and S3, Supporting Information. It is obviously shown that the parameters are enhanced with the modification of MLG, and distributed in

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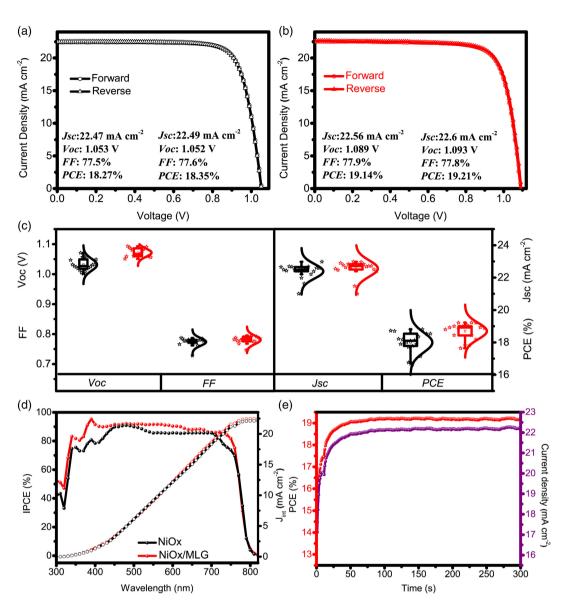


Figure 7. Hysteresis characteristics of the corresponding champion devices based on: a)  $NiO_x$  and b)  $NiO_x/MLG$ ; c) Statistics of photovoltaic statistic parameters from the devices based on  $NiO_x$  and  $NiO_x/MLG$ ; d) incident photon-current conversion efficiency (IPCE) spectra and e) stabilized power output curves of the champion device based on  $NiO_x/MLG$ .

a narrow range, suggesting excellent reproducibility. The improvements of both  $V_{\rm oc}$  and FF mean the MLG modification of  ${\rm NiO}_x$  decreases the recombination effectively at the  ${\rm NiO}_x$ –MAPbI $_3$  interface and the enhancement of carrier transport. Moreover, the incident photon-current conversion efficiency (IPCE) is carried out to study the photon to current conversion efficiency and integrate the current density of both devices based on  ${\rm NiO}_x$  and  ${\rm NiO}_x/{\rm MLG}$ . As shown in Figure 7d, both the devices display the charge collection over the broad spectral range from 350-800 nm with average efficiency of over 85%. As expected, the  ${\rm NiO}_x/{\rm MLG}$ -based device shows higher IPCE, which is probably on account of the enhanced charge transfer and balanced photocurrent collection. [35] The integrated photocurrent densities for each device

are 22.47 and 22.13 mA cm $^{-2}$ , respectively, which are slightly lower than those attained from J-V curves. Given that the IPCE measurements are carried out without one-sun illumination, the integrated current densities from IPCE curves are usually lower than that obtained from J-V curves. [36] Finally, stabilized power output test for the device based on NiO $_x$ /MLG is carried out by holding the voltage at the maximum power point as long as 300 s ( $V_{\rm max} = 0.9138$  V), which is shown in Figure 7e. The stabilized PCE and  $J_{\rm sc}$  constantly stay along the time at about 19.18%, and 22.27 mA cm $^{-2}$ , which are comparable to the PCE and  $J_{\rm sc}$  obtained from J-V curves of a device based on NiO $_x$ /MLG. In the meanwhile, the results of stabilized power output test for the device based on NiO $_x$  are shown in Figure S2, Supporting Information.



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#### 3. Conclusion

In summary, MLG was used for  $\mathrm{NiO}_x/\mathrm{perovskite}$  interface modification. After modification, a 0.1 eV drop of the HOMO level was obtained. Perovskite layer with high crystallinity and large grain was made on the  $\mathrm{NiO}_x/\mathrm{adenine}$  substrate. Improved contact, accelerated charge transport, and extraction were achieved in the HTL/perovskite interface after adenine modification. Compared with the control  $\mathrm{NiO}_x$  device, an enhancement of Voc and Jsc was observed after adenine modification. Moreover, a significant improvement in device stability was achieved in adenine-modified devices. Therefore, this work provides a novel interface engineering strategy between  $\mathrm{NiO}_x$  and perovskites, which is helpful for minimizing the voltage deficit and improving PSC stability.

# 4. Experimental Section

*Materials*: Unless otherwise indicated, all materials were used as received and purchased from Sigma-Aldrich. MAPbl<sub>3</sub> film was fabricated using Pbl<sub>2</sub>:MAI(1:1)-dimethylformamide complex (99.99%, TCI) and dimethylsulfoxide (DMSO) solvent from Acros Organics.

Fabrication of Crack-Free Monolayer Graphene: First, the crack-free MLG was deposited on Cu foil by the CVD method, according to our previous article. [24] The Cu foil was annealed at 1000 °C under a CO2 atmosphere to clean the surface, and after eliminating the CO2, CH4 was added to deposit MLG. Next, PMMA with a thickness of 100–200 nm was coated on the exposed surface of MLG to reinforce the mechanical strength. Following, the Cu foil was etched in a 0.2 M Na2S2O8 aqueous solution, leaving the PMMA-coated MLG film floating on the aqueous solution. Subsequently, the floating PMMA/MLG film was salvaged on the surface of the FTO-NiO2 substrate. Finally, the PMMA was removed by dissolving in acetone to expose the clean MLG surface.

Fabrication of Perovskite Solar Cells: FTO glass substrates (Nippon sheet glass) were sequentially cleaned with the detergent solution, acetone, and ethanol in an ultrasonic bath. After UV-O3 treatment, NiOx thin layer was loaded on FTO glass by spin-coating at 4000 rpm for 30 s with 20 mg mL $^{-1}$  NiOx aqueous dispersion and dried at 150 °C for 10 min. Here, the 20 mg mL $^{-1}$  NiOx aqueous dispersion was fabricated as our reported article. First, 10 m NaOH solution was dropped-wise into 0.5 m NiCl2·6H2O solution until the pH value reached 10. Then, the obtained green precipitation was centrifuged, washed three times with deionized water, and dried at 80 °C overnight. After being ground, the green powder was annealed at 270 °C for 2 h to obtain NiOx nanoparticles. The NiOx nanoparticles were dispersed in deionized water by ultrasound for the concentration of 20 mg mL $^{-1}$ . Before use, it was filtrated with a 0.45  $\mu$ m polytetrafluoroethylene (PTFE) filter.

The large-area crack-free MLG was coated on  $NiO_x$  film as the aforementioned method, which was then treated in UV-O<sub>3</sub> for 10 min to improve the surface wettability. For 1.8 m MAPbI<sub>3</sub> precursor solution, the MAPbI<sub>3</sub> single crystal was dissolved in DMSO solution, which was spin-coated on the substrates at 1000 rpm for 10 s and continuously at 4000 rpm for 30 s. During the second step,  $600\,\mu\text{L}$  of chlorobenzene was poured at the 20 s to rush the film for extracting the DMSO. The MAPbI<sub>3</sub> films were crystallized on a hot plate at 100 °C for 1 h. After cooling down, 20 mg mL<sup>-1</sup> PCB61M solution was spin-coated at 1500 rpm for 30 s as an electron transfer layer. Following, saturated bathocuproine (BCP) isopropanol solution was spin-coated on PC61BM film at 6000 rpm for 30 s. Finally, silver top electrodes with a thickness of 100 nm were thermal evaporated to complete finish the fabrication of the PSCs.

Characterization: XRD measurement was carried out using a Bruker D8 Advance diffractometer in an angle range of  $2\theta = 10^{\circ}$  to 70°. The morphology was characterized using an SEM (ZEISS Merlin) and atomic force microscope (AFM, Bruker). Raman characterization was carried out using

a Renishaw micro-Raman spectroscope (532 nm, 2.33 eV, ×100 objective) and analyzed by MATLAB. Absorbance and transmittance spectra of NiO<sub>x</sub> and NiO<sub>x</sub>/MLG films on FTO were measured with an integrating sphere using UV/Vis/NIR spectroscopy (PerkinElmer Lambda). UPS were taken on an AXIS SUPRA instrument (Kratos Analytical). PL dynamics were measured using time-resolved single photon counting (TRSPC), which is incorporated into the Fluorolog-312 spectrofluorometer. The SCLC plots were obtained using a digital source meter (Keithley 2400). The current density-voltage (/-V) characteristics of the solar cells were measured using Keithley 2400 under AM 1.5G irradiation (100 mW  ${\rm cm}^{-2}$ ) from commercial solar simulators (Oriel, 450 W Xenon, AAA class). The light intensity was calibrated with a Si reference cell equipped with an IR-cutoff filter (KG5, Newport). The voltage scan rate was  $10 \text{ or } 25 \text{ mV s}^{-1}$ . The active area of solar cells is 0.16 cm<sup>2</sup> by a mask to fix the effective area and reduce the influence of the scattered light. Before measurement, there was no preconditioning such as light soaking or forward voltage bias. External quantum efficiency (EQE) was measured by IQE200B (Oriel) without bias light.

# **Supporting Information**

Supporting Information is available from the Wiley Online Library or from the author.

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# **Conflict of Interest**

The authors declare no conflict of interest.

## **Data Availability Statement**

The data that support the findings of this study are available in the supplementary material of this article.

### Keywords

hole transport layers, monolayer graphene, NiOx, perovskite solar cells

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- T. Zhu, L. Shen, S. Xun, J. S. Sarmiento, Y. Yang, L. Zheng, H. Li,
  H. Wang, J. L. Bredas, X. Gong, Adv. Mater. 2022, 34, e2109348.
- [2] J. J. Yoo, G. Seo, M. R. Chua, T. G. Park, Y. Lu, F. Rotermund, Y. K. Kim, C. S. Moon, N. J. Jeon, J. P. Correa-Baena, V. Bulovic, S. S. Shin, M. G. Bawendi, J. Seo, *Nature* 2021, 590, 587.
- [3] M. Kim, J. Jeong, H. Lu, T. K. Lee, F. T. Eickemeyer, Y. Liu, I. W. Choi, S. J. Choi, Y. Jo, H.-B. Kim, S.-I. Mo, Y.-K. Kim, H. Lee, N. G. An, S. Cho, W. R. Tress, S. M. Zakeeruddin, A. Hagfeldt, J. Y. Kim, M. Gratzel, D. S. Kim, *Science* 2022, 375, 302.
- [4] F. Yu, J. Liu, J. Huang, P. Xu, C.-H. Li, Y.-X. Zheng, H. Tan, J.-L. Zuo, Sol. RRL 2021, 6, 2100906.



- [5] R. Hu, L. Chu, J. Zhang, X. A. Li, W. Huang, J. Power Sources 2017, 361, 259
- [6] H. Min, D. Y. Lee, J. Kim, G. Kim, K. S. Lee, J. Kim, M. J. Paik, Y. K. Kim, K. S. Kim, M. G. Kim, T. J. Shin, I. S. Seok, *Nature* 2021, 598, 444.
- [7] D. Yang, X. Zhang, K. Wang, C. Wu, R. Yang, Y. Hou, Y. Jiang, S. Liu, S. Priya, *Nano Lett.* **2019**, *19*, 3313.
- [8] F. Li, X. Deng, F. Qi, Z. Li, D. Liu, D. Shen, M. Qin, S. Wu, F. Lin, S. H. Jang, J. Zhang, X. Lu, D. Lei, C. S. Lee, Z. Zhu, A. K. Jen, J. Am. Chem. Soc. 2020. 142, 20134.
- [9] M. Degani, Q. An, M. Albaladejo-Siguan, Y. J. Hofstetter, C. Cho, F. Paulus, G. Grancini, Y. Vaynzof, Sci. Adv. 2021, 7, eabj7930.
- [10] Y. M. Chang, C. W. Li, Y. L. Lu, M. S. Wu, H. Li, Y. S. Lin, C. W. Lu, C. P. Chen, Y. J. Chang, ACS Appl. Mater. Interfaces 2021, 13, 6450.
- [11] S. Wang, Y. Li, J. Yang, T. Wang, B. Yang, Q. Cao, X. Pu, L. Etgar, J. Han, J. Zhao, X. Li, A. Hagfeldt, Angew. Chem., Int. Ed. 2022, e202116534.
- [12] J. Zhang, J. Yang, R. Dai, W. Sheng, Y. Su, Y. Zhong, X. Li, L. Tan, Y. Chen, Adv. Energy Mater. 2022, 12, 2103674.
- [13] N. Tiwari, H. Arianita Dewi, E. Erdenebileg, R. Narayan Chauhan, N. Mathews, S. Mhaisalkar, A. Bruno, Sol. RRL 2021, 6, 2100700.
- [14] F. Ma, Y. Zhao, J. Li, X. Zhang, H. Gu, J. You, J. Energy Chem. 2021, 52, 393.
- [15] S. Teo, Z. Guo, Z. Xu, C. Zhang, Y. Kamata, S. Hayase, T. Ma, ChemSusChem 2019, 12, 518.
- [16] Y. Du, C. Xin, W. Huang, B. Shi, Y. Ding, C. Wei, Y. Zhao, Y. Li, X. Zhang, ACS Sustainable Chem. Eng. 2018, 6, 16806.
- [17] W. Chen, Y. Zhou, L. Wang, Y. Wu, B. Tu, B. Yu, F. Liu, H. W. Tam, G. Wang, A. B. Djurisic, L. Huang, Z. He, Adv. Mater. 2018, 30, e1800515.
- [18] Q. Wang, C. C. Chueh, T. Zhao, J. Cheng, M. Eslamian, W. C. H. Choy, A. K. Jen, ChemSusChem 2017, 10, 3794.
- [19] H. Su, T. Wu, D. Cui, X. Lin, X. Luo, Y. Wang, L. Han, Small Methods 2020. 4, 2000507.
- [20] K. Gong, J. Hu, N. Cui, Y. Xue, L. Li, G. Long, S. Lin, Mater. Des. 2021, 211, 110170.

- [21] K. Chung, K. Lee, Y. Tchoe, H. Oh, J. Park, J. K. Hyun, G.-C. Yi, Nano Energy 2019, 60, 82.
- [22] T. Gatti, F. Lamberti, P. Topolovsek, M. Abdu-Aguye, R. Sorrentino, L. Perino, M. Salerno, L. Girardi, C. Marega, G. A. Rizzi, M. A. Loi, A. Petrozza, E. Menna, Sol. RRL 2018, 2, 1800013.
- [23] J. Zhang, J. Fan, B. Cheng, J. Yu, W. Ho, Sol. RRL 2020, 4, 2000502.
- [24] S. Huang, M. Dakhchoune, W. Luo, E. Oveisi, G. He, M. Rezaei, J. Zhao, D. T. L. Alexander, A. Zuttel, M. S. Strano, K. V. Agrawal, Nat. Commun. 2018, 9, 2632.
- [25] C. Klein, D. Cohen-Elias, G. Sarusi, Heliyon 2018, 4, e01030.
- [26] N. Mironova-Ulmane, A. Kuzmin, I. Steins, J. Grabis, I. Sildos, M. Pärs, J. Phys.: Conf. Ser. 2007, 93, 012039.
- [27] X. Xia, Y. Jiang, Q. Wan, X. Wang, L. Wang, F. Li, ACS Appl. Mater. Interfaces 2018, 10, 44501.
- [28] R. Hu, R. Zhang, Y. Ma, W. Liu, L. Chu, W. Mao, J. Zhang, J. Yang, Y. Pu, X. A. Li, Appl. Surf. Sci. 2018, 462, 840.
- [29] R. Hu, C. Ge, L. Chu, Y. Feng, S. Xiao, Y. Ma, W. Liu, X. A. Li, M. K. Nazeeruddin, J. Energy Chem. 2021, 59, 581.
- [30] X. Gong, Q. Sun, S. Liu, P. Liao, Y. Shen, C. Gratzel, S. M. Zakeeruddin, M. Gratzel, M. Wang, Nano Lett. 2018, 18, 3969.
- [31] Q. Jiang, L. Zhang, H. Wang, X. Yang, J. Meng, H. Liu, Z. Yin, J. Wu, X. Zhang, J. You, Nat. Energy 2016, 2, 1.
- [32] M. Stolterfoht, P. Caprioglio, C. M. Wolff, J. Márquez, J. Nordmann, S. Zhang, D. Rothhardt, U. Hrmann, Y. Amir, A. J. E. Redinger, E. Science, *Energy Environ. Sci.* 2019, 12, 2778.
- [33] Z. Zhou, X. Li, M. Cai, F. Xie, Y. Wu, Z. Lan, X. Yang, Y. Qiang, A. Islam, L. Han, Adv. Energy Mater. 2017, 7, 1700763.
- [34] R. Hu, Y. Zhang, S. Paek, X.-X. Gao, X. A. Li, M. K. Nazeeruddin, J. Mater. Chem. A 2020, 8, 8058.
- [35] W. S. Yang, B. W. Park, E. H. Jung, N. J. Jeon, Y. C. Kim, D. U. Lee, S. S. Shin, J. Seo, E. K. Kim, J. H. J. S. Noh, *Science* 2017, 356, 1376
- [36] W. Chen, L. Xu, X. Feng, J. Jie, Z. He, Adv. Mater. 2017, 29, 1603923.