

IT Number:

Machine Number:

Lab:



SLIIT
ACADEMY

Sri Lanka Institute of Information Technology

Higher Diploma
in
Information Technology

Mid Examination
Year 1, Semester 2 (2019)

January Intake

Version A

Object Oriented Programming (JAVA)
(IT1108)

Duration: 1 Hour

Instructions to Candidates:

- ◆ This is a closed book examination.
- ◆ This paper contains 2 questions on 3 pages without the cover page.
- ◆ Create a Directory on the **HOME** directory with your **IT Number** and save all your programs in the folder.
- ◆ Read all questions before answering.
- ◆ The total marks obtainable for this examination is **30**.
- ◆ Use only the **VI** as the text editor.

Question 1**(20 Marks)**

- a) Write a java method called *calculateMSpeed* which calculate the root-mean-square speed (v_p) of a gas molecule. The above method returns the calculated answer as a float value. The method takes two double values (M-the molar mass of the gas in kilograms per mole, T-temperature in Kelvin) as parameters.

The formula to calculate the root-mean-square speed (v_p) is as follows:

$$\text{root-mean-square speed of a gas molecule } (v_p) = \sqrt{\left(\frac{3RT}{M}\right)} \times 10^3$$

v_p = root-mean-square speed of a gas molecule.

M = the molar mass of the gas in kilograms per mole.

T = temperature in Kelvin.

R = molar gas constant ($8.3144598 \text{ m}^2 \text{ kg s}^{-2} \text{ K}^{-1} \text{ mol}^{-1}$)

- b) Write a main method to take the molar mass of the gas in kilograms per mole and temperature in Kelvin as user inputs and calculate the root-mean-square speed of a gas molecule by calling *calculateMSpeed*.
- c) Based on the calculated root-mean-square speed, determine the category of the speed as follows:

Root-Mean-Square of a gas molecule.	Category of the Speed
0-500	Slow
501-1000	Medium
greater than or equal to 1001	Fast

Hint: Use the scanner class to get the inputs from the user

Sample Input:

Enter molar mass of the gas in kilograms per mole: 32000

Enter the Temperature in kelvin: 298

Sample Output:

root-mean-square speed of a gas molecule: 481.9597707

rounded answer: 481.96

Category of the Speed: Slow

Note: -round the root-mean-square speed to two decimal places.

Marking Criteria

Criteria	Marks	Marks Obtained
Proper class structure and main method	2	
Importing the Scanner class and their usage	2	
Reading user input	2	
Calling the method with correct parameters	2	
correct method header with correct parameters marks	2	
Rounding the answer	4	
Calculate the answer	2	
Deciding the answer based on the calculation using nested if else statements	2	
Final output with compiled ByteCode	2	
Total Marks	20	

Question.2**(10 Marks)**

Write a java program to takes a greeting message and the number of repetitions, using command line arguments. Print your greeting message, according to the number of repetitions, in uppercase letters

Sample Input: Greeting Message - "welcome"

Number of repetitions 3

Sample Output:

WELCOME

WELCOME

WELCOME

Marking Criteria

Criteria	Marks	Marks Obtained
Proper class structure and main method	2	
Usage of the command line arguments	2	
Conversions of datatypes	2	
Printing the message with correct case.	3	
Final output with compiled ByteCode	1	
Total Marks	10	