

Coulomb's Law Lab

PHY-222-AC01

Jasper RUNCO

August 25, 2020

1 Theory

Coulomb's law describes a mathematical expression for the interaction of electrically charged objects. The electrical charge generates a force pair, F_E , with a magnitude that has been experimentally determined to such a degree that it is accepted as a universal law.

1.1 Definitions

Electrically charged object - Matter with more or less electrons than protons is negatively or positively charged respectively.

Force pair - The principal of Newton's third law in which every action force has a corresponding reaction force equal in magnitude and opposite in direction.

Coulomb's law -

$$F_E = k \frac{q_1 q_2}{r^2}$$

r - The distance between the charged objects measured in meters.

$q_1 q_2$ - The sum of the electrical charges of each object, measured in Coulombs.

k - The electric constant, the experimentally determined proportionality constant with a value of $k = 9.0 \times 10^9 \frac{Nm^2}{C^2}$

2 Objectives

First Objective

Experimentally confirm Coulomb's law.

Second Objective

Study how distance and charge affect the electric force.

Third Objective

Experimentally determine the value of the electric constant, k.

3 Experimental Data

3.1 Part One

Table 1:

$q_1 = 2\mu C$		$q_2 = 4\mu C$	
$\mathbf{r(cm)}$	$r^2(m^2)$	$\frac{1}{r^2}(\frac{1}{m^2})$	$F_E(N)$
10	1.0×10^{-2}	1×10^2	7.190
9	8.1×10^{-3}	1.2×10^2	8.877
8	6.4×10^{-3}	1.6×10^2	11.234
7	4.9×10^{-3}	2.0×10^2	14.674
6	3.6×10^{-3}	2.8×10^2	19.972
5	2.5×10^{-3}	4.0×10^2	28.760
4	1.6×10^{-3}	6.3×10^2	44.938
3	9.0×10^{-4}	1.1×10^3	79.889

3.2 Part Two

Table 2:

$q_1 = 5\mu C$	$r = 6cm$
$q_2(\mu C)$	$F_E(N)$
10	124.827
9	112.344
8	99.862
7	87.379
6	74.896
5	62.414
4	49.931
3	37.448

4 Data Analysis

4.1 Part One

4.1.1 F_E with respect to r

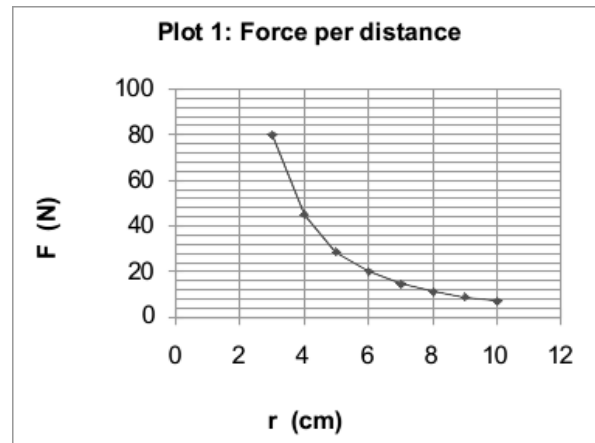


Figure 1: Force per distance

Comments on Figure 1: As the charged objects' distance (r) tends towards zero from the positive direction, the force (F) tends towards infinity. As r tends towards positive infinity, F tends towards zero.

4.1.2 F_E with respect to $\frac{1}{r^2}$ to find k

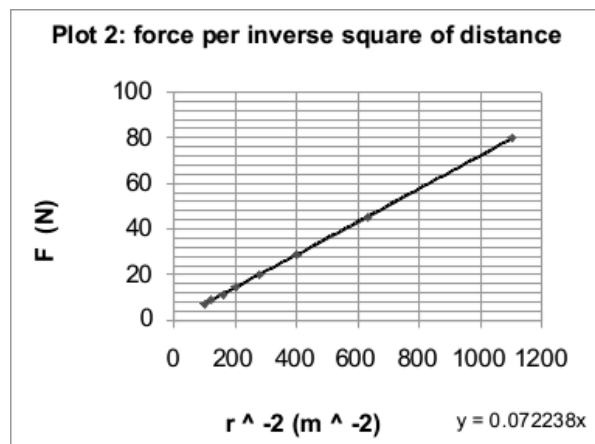


Figure 2: Force per inverse square of distance

Coulomb's Law - $F_E = k \frac{q_1 q_2}{r^2}$

Linear regression of Figure 2 data - $y = 0.072238x$

$$q_1 = 2\mu C$$

$$q_2 = 4\mu C$$

$$\begin{aligned}
 F_E &= k \frac{(2\mu C)(4\mu C)}{r^2} \implies \\
 F_E(N) &= k (8 \times 10^{-12} C^2) \left(\frac{1}{r^2} \right) \left(\frac{1}{m^2} \right) \\
 y &= 0.072238x \implies \\
 F_E &= 0.072238 \left(\frac{1}{r^2} \right) \\
 0.072238 \left(\frac{1}{r^2} \right) (N) &= k (8 \times 10^{-12} C^2) \left(\frac{1}{r^2} \right) \left(\frac{1}{m^2} \right) \\
 k &= \frac{0.072238 \left(\frac{1}{r^2} \right) (N)(m^2)}{8 \times 10^{-12} C^2 \left(\frac{1}{r^2} \right)} \\
 k &= \boxed{9.02975 \times 10^9} \left(\frac{Nm^2}{C^2} \right)
 \end{aligned}$$

4.1.3 Percent error in k

$$\begin{aligned}\% \text{ error} &= \left| \frac{\text{theoretical} - \text{experimental}}{\text{theoretical}} \right| \times 100\% \\ \% \text{ error} &= \left| \frac{\left(9.0 \times 10^9 \frac{Nm^2}{C^2}\right) - \left(9.02975 \times 10^9 \frac{Nm^2}{C^2}\right)}{\left(9.0 \times 10^9 \frac{Nm^2}{C^2}\right)} \right| \times 100\% \\ \% \text{ error} &= 0.330556\%\end{aligned}$$

4.2 Part Two

4.2.1 F_E with respect to q_2

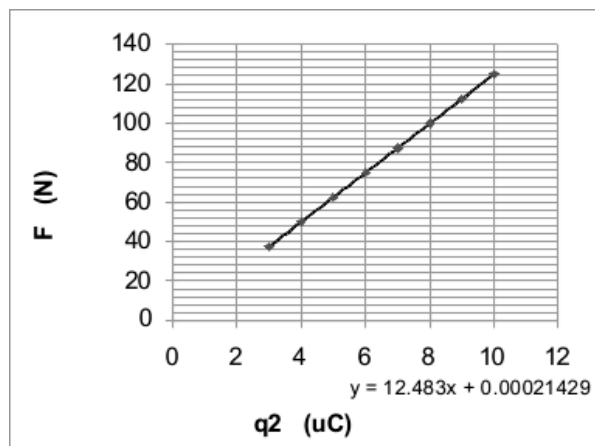


Figure 3: Force per magnitude charge on q_2

Comments on Figure 3: There is a positive linear relationship between the magnitude of charge on one object $q_2(\mu C)$ and electric force $F_E(N)$. Given the precision of F_E measurement, it appears to approach a lower bounds of $F_E = 0$ at $q_2 = 0$.

4.2.2 F_E with respect to q_2 to find k

Coulomb's Law - $F_E = k \frac{q_1 q_2}{r^2}$

Linear regression of Figure 3 data - $y = 12.483x + 0.00021429$

q_1 - $5\mu C$

r - $6cm$

$$\begin{aligned}
F_E &= k \frac{(5\mu C)(q_2)}{(6cm)^2} \implies \\
F_E(N) &= kq_2(\mu C) \left(\frac{5 \times 10^{-6}C}{3.6 \times 10^{-3}m^2} \right) \\
y &= 12.483x + 0.00021429 \implies \\
F_E &= 12.483q_2 + 0.00021429 \\
(12.483q_2 + 0.00021429)(N) &= kq_2(\mu C) \left(\frac{5 \times 10^{-6}C}{3.6 \times 10^{-3}m^2} \right) \implies \\
k &= \frac{(12.483q_2 + 0.00021429)(3.6 \times 10^{-3}m^2)(N)}{(q_2)(5 \times 10^{-6}C)(10^{-6}C)} \implies \\
k &= \boxed{8.98776 \times 10^9 \left(\frac{Nm^2}{C^2} \right)} + 1.542888 \times 10^5 q_2^{-1} \left(\frac{Nm^2}{C^2} \right)
\end{aligned}$$

4.2.3 Percent error in k

$$\begin{aligned}
\% \text{ error} &= \left| \frac{\text{theoretical} - \text{experimental}}{\text{theoretical}} \right| \times 100\% \\
\% \text{ error} &= \left| \frac{\left(9.0 \times 10^9 \frac{Nm^2}{C^2} \right) - \left(8.98776 \times 10^9 \frac{Nm^2}{C^2} \right)}{\left(9.0 \times 10^9 \frac{Nm^2}{C^2} \right)} \right| \times 100\% \\
\% \text{ error} &= 0.136\%
\end{aligned}$$

5 Results and Conclusions

6 Discussion of Experimental Uncertainty