

DAY 4

# PHYSICS

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## BASIC ELECTRICITY

### Basic Electrical Concept

- ↳ Electricity
- ↳ Magnetism
- ↳ Optics
- ↳ NEM
- ↳ Periodic Motion
- ↳ Heat & Thermodynamics

### Review

- There are two types of charge: -ve & +ve
- most objects are electrically neutral;  $\Rightarrow$  equal no. of -ve & +ve charges (net charge = 0)
- An object becomes charged by adding or removing electrons
- An e<sup>-</sup> carries -ve charge of magnitude  $e = 1.602 \times 10^{-19} C$

### Law of Charges

- Like charges repel & opposite charges attract

### Law of charge conservation

- The total charge of an isolated system is strictly conserved
- conductors → materials where some e's can move freely
- insulators → materials where none of the charges can move freely

## ELECTROMAGNETISM

Electricity

Magnetism

Optics

### Fundamental Concepts of electromagnetism

- Charge
- Force
- field
- potential
- current
- electric circuit
- magnetic field
- induction
- alternating current
- waves
- reflection
- refraction
- image
- interference
- diffraction

# Electromagnetism

Electromagnetism → fundamental force of nature  
→ dominant force in a vast range of natural & technological phenomena

- The electromagnetic force is solely responsible for the structure of matter, organic, or inorganic.
- Physics, chemistry, biology, material science
- The operation of most technological devices is based on electromagnetic forces. From lights, motors, and batteries, to communication and broadcasting systems, as well as microelectric devices
- Engineering.

## History of Electric Charge

600 BC	→ Greeks → attractive properties of amber when rubbed
1600 AD	→ electric bodies attract as well as repel
1735 AD	→ du Fay: Two distinct types of electricity
1750 AD	→ Franklin: +ve. & -ve charges
1770 AD	→ Coulomb: "Inverse square law"
1890 AD	→ J.J. Thompson: Quantization of electric charge - "electron"

## Transfer of Charge

Some materials attract electrons more than other.

~~Like charges repel~~ Like charges repel:  $\leftarrow \oplus \rightarrow \quad \leftarrow \ominus \rightarrow$   
unlike charges attract  $\oplus \rightarrow \leftarrow \ominus$

# The Importance of the ATOM

## • Atom Structure

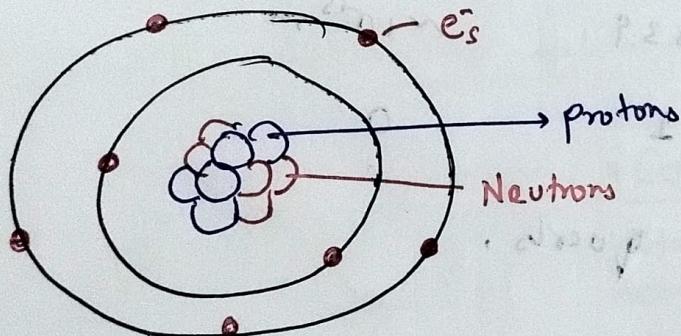
- ↳ Made up of  $e^-$ ,  $p^+$ , and neutron.
- ↳ nucleus contain protons ( $p^+$ ) & neutrons.
- ↳ shells contain the  $e^-$ , which orbit the nucleus
- ↳ Building blocks of matter

→ The atomic structure of a material will help to determine the ease of current flow.

- ↳ Atoms can be charged
- ↳ Positive
- ↳ Negative
- ↳ Neutral

⇒ Law of Charges :- Like charges Repel ; unlike charges attract

- ↳ A material that has an excess of  $e^- \Rightarrow -ve$  charge
- ↳ A material that has fewer electrons  $\Rightarrow +ve$  charge



THE ATOM :- made of +ve charges in Nucleus & surrounded by cloud of  $e^-$ .

$$\begin{aligned} \text{Proton} &\rightarrow +e^+ \\ \text{Electron} &\rightarrow (-e^-) \\ [e = 1.602 \times 10^{-19} C] \end{aligned}$$

- Atoms are normally neutral,  $\rightarrow$  no  $e^-$  = no  $p^+$
- charges balance, atom has no net charge
- $e^-$  easily removable from atom
- removal of  $e^- \rightarrow +ve$  charged atom

## Electric Charge Summary

- ↳ property of matter → SI unit: Coulomb (C)
- ↳ can be (-ve) or (+ve).
- ↳ +ve charge → nuclei, e's orbit around nucleus
- ↳ add "q" to  $e^-$  → -ve charge
- ↳ remove "q" from  $e^-$  → +ve charge
- ↳ like charges repel, opposite attract
- ↳ charges travel in conductors, not in insulators
- ↳ force of attraction OR repulsion  $\propto \frac{1}{r^2}$

## Charge is Quantized

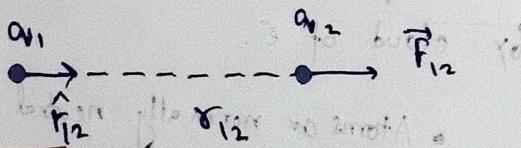
$$Q = ne$$

$$e = 1.6 \times 10^{-19} \text{ C}$$

	charge	mass	diameter
electron	-e	1	0
proton	+e	1.676	$\sim 10^{-15} \text{ m}$
neutron	0	1.679	$\sim 10^{-15} \text{ m}$
positron	+e	1	0

protons, neutrons → made up of quarks.

## Coulomb's Law



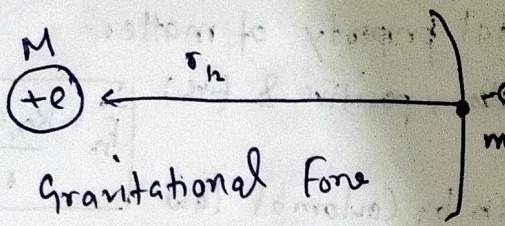
$$\vec{F}_{12} = \frac{k q_1 q_2}{r_{12}^2} \hat{r}_{12}$$

$$k = \frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ Nm}^2/\text{C}^2$$

$$\epsilon_0 = \text{permittivity of free space}$$

$$= 8.85 \times 10^{-12} \text{ C}^2/\text{Nm}^{-1}$$

# GRAVITATIONAL AND ELECTRIC FORCES IN HYDROGEN ATOM



$$\vec{F}_g = G \frac{Mm}{r_{12}^2} \hat{r}$$

$$F_g = 3.6 \times 10^{-47} \text{ N}$$

$$m = 9.1 \times 10^{-31} \text{ kg}$$

$$M = 1.7 \times 10^{-27} \text{ kg}$$

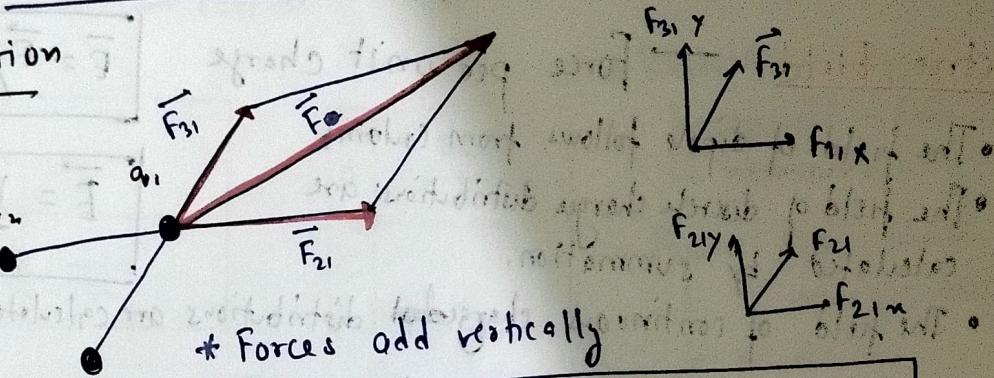
$$r_{12} = 5.3 \times 10^{-11} \text{ m}$$

electric force

$$\vec{F}_e = \left( \frac{1}{4\pi\epsilon_0} \right) \frac{qe}{r_{12}^2} \hat{r}$$

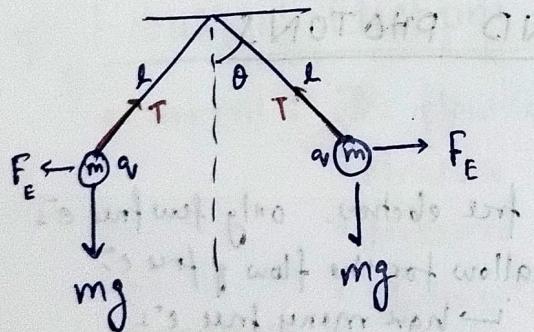
$$F_e = 3.6 \times 10^{-8} \text{ N}$$

Superposition Principle



$$\vec{F} = (F_{21x} + F_{31x}) \hat{x} + (F_{21y} + F_{31y}) \hat{y}$$

Example: Electricity balancing gravity



Conductors, insulators and dielectrics

Materials in which charge is free to move Conductors  
Materials in which charge isn't free to move Insulators

- Insulators generally contain molecular dipoles, which torque and force in E.F.
- Such materials are called dielectrics

# Even if molecules aren't intrinsically dipoles, they acquire induced dipole moments as a result of electric forces stretching the molecule.

# Alignment of molecular dipoles reduce an externally applied field

## Summary

**Electric Charge** is a fundamental property of matter.

- Charge comes in two varieties, positive & negative
- Charge → conserved
- Force b/w 2 charges is given by Coulomb's law

$$\vec{F}_{12} = \frac{kq_1 q_2}{r^2} \hat{r}$$

- Electric force obeys Superposition Principle,  
↳ forces due to individual charges sum vectorially.

**Electric Field** : Force per unit charge

$$\vec{E} = \vec{F}/q$$

- The field of dipole follows from Coulomb
- The field of discrete charge distributions are calculated by summation.
- The field of continuous charged distributions are calculated by integration

$$\vec{E} = \frac{kq}{r^2} \hat{r}$$

→ A point charge experiences a force in an E.F.

→ A dipole experiences a torque in an EF & affects it if the field is not uniform.

$$\vec{F} = q\vec{E}$$

## ELECTRICAL MATERIALS AND PHOTONS

### Electrical Materials Used :

- Insulators : inhibit the flow of free electrons, only few free e<sup>-</sup>s
- Conductors : Materials → readily allow for the flow of free e<sup>-</sup>s  
↳ have many free e<sup>-</sup>s
- Semiconductors : more free e<sup>-</sup>s than insulator but less than conductor

### Photons :

- The basic unit of light energy
- Light → stream of tiny particles of energy → photons
- Used in Solar Photovoltaics and helps with the creation of electricity
- P-N Junction could be considered the heart of solar cell

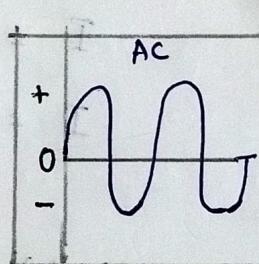
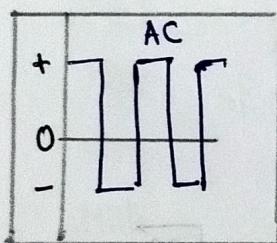
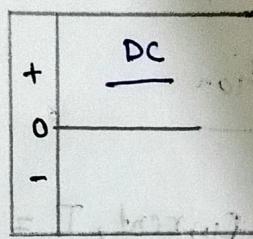
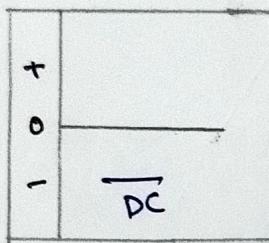
## # Electricity for Renewables

### • Direct Current - DC

- current flows in one direction only
- car battery
- photovoltaic cells

### • Alternating Current - AC

- current flows in one direction then the other and alternates back and forth.
- This is what is used in one's home single phase.
- can be transformed



- Resistance: opposition to current flow
- ↳ Measured in ( $\Omega$ )
- ↳ ohmmeter

- Power: Rate of work or energy consumption

- ↳ Watts
- ↳ rate at which energy is used in a circuit

## # Flow of Electrons through Conductors

- Current: The flow of  $e^-$  through a conductor

- Measure in Ampere

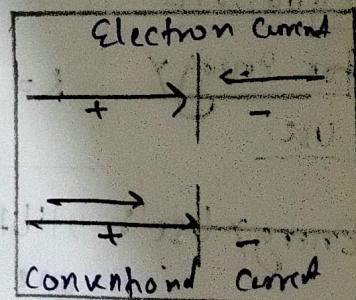
- Measured with Amp. Meters

- Electron Current flow

- ↳ Negative to positive flow

- Conventional Current flow

- ↳ Positive to negative flow



Understanding  
the circuit

- A practical circuit contains the four following characteristics:

- ↳ Load: A load consumes power

- ↳ A switch to control load for safety

- ↳ Power supply: Photovoltaic source to produce potential

- ↳ A path of  $e^-$  to flow - Wires in circuit

## # Ohm's Law

- It requires one volt to push one ampere of current through one ohm of resistance in a DC circuit
- It is a proportion that shows how voltage (V), current (I) and resistance (R) are related in a circuit.

### • Examples:

$$V = IR$$

$$I = \frac{V}{R}$$

$$R = \frac{V}{I}$$

$V = 12V$ ,  $R = 3\text{ ohms}$ , then

$$I = \frac{12V}{3\text{ ohms}} = 4\text{ amps}$$

### • Identifying the Equation

Power,  $P = IE = VI$  Current,  $I = \sqrt{\frac{P}{R}}$

$$P = I^2 \cdot R$$

$$P = \frac{E^2}{R}$$

$$I = \frac{P}{E}$$

$$I = \frac{E}{R}$$

Resistance,  $R = E/I$  Voltage (V) or  $E = \sqrt{PR}$

$$E = IR$$

## Equations for Energy Use

$$R = \frac{P}{I^2}$$

$$E = P/I$$

$$R = \frac{E^2}{P}$$

- Energy Use: utility companies sell energy consumption. Any power one can generate on one's own is money saved

$$\bullet \text{Energy} = \text{Power} \times \text{Time}$$

• To determine energy consumed, multiply watts times hours of operation.

$$\bullet \text{Energy Consumed} = \text{Watts} \times \text{Hours} = \text{Watt Hours (Wh)}$$

Q- drawing 0.71 amp current with 120 V. time = 4 hours. Energy used?

$$\bullet P = V \times I = 0.71A \times 120V = 85.2 \text{ Watts}$$

$$\bullet \text{Energy} = P \times T = 85.2 \text{ watts} \times 4 \text{ hours} = 340.8 \text{ Whrs}$$

- ## # Conversions
- Necessary conversions one needs to know  
In electrical production Kilowatt-hrs (kWh)  
and megawatt-hour (MWh)

- To convert watts to kilowatts, divide by 1000.
- $1\text{kWh} = 1000 \text{ watts for 1 hr}$
- $1\text{MWh} = 1000,000 \text{ watt, used for 1 hr.}$
- Example: if a circuit uses 11500 Wh of energy, a power company will charge the following:

$$\frac{11,500 \text{ Wh}}{1000} = 11.5 \text{ kWh}$$

## # Engineering notation with metric prefixes

Prefix	Symbol	Value
Tera	T	one Trillion ( $10^{12}$ )
Giga	G	one Billion ( $10^9$ )
Mega	M	one million ( $10^6$ )
Kilo	k	one thousand ( $10^3$ )
Milli	m	one thousandth ( $10^{-3}$ )
Micro	$\mu$	One millionth ( $10^{-6}$ )
Nano	n	one billionth ( $10^{-9}$ )
Pico	p	one trillionth ( $10^{-12}$ )

$$• 1000 \text{ W} = 1 \text{ kW} = \frac{1000}{1000} \text{ kW}$$

$$• 1500,000 \text{ W} = 1.5 \text{ MW} = \frac{1500,000}{10^6} \text{ MW}$$

$$• 0.032 \text{ A} = 3.2 \text{ mA}$$

$$• 0.00007 \text{ A} = 70 \text{ micro Amp.} = 70 \text{ mA}$$

## Capacitors

- device used to store electrical charge  
(a surprisingly useful thing to do in circuit)
- consist of a pair of conducting plates separated by an insulator. The insulator is called a dielectric and is often air, paper, or oil.

## Capacitance & Applications

- Capacitance is the ability of a system to store electric charge.
- depends on plate area, separation dist., and dielectric material
- $C = \frac{K\epsilon_0 A}{d}$
- Application : — Energy storage (e.g. camera flash)  
— Filtering signals in circuits  
— Timing and tuning in radios

## Capacitance

measure of extent to which a capacitor can store charge is called its capacitance.

$$C = \frac{Q}{V}$$

$$\begin{cases} C \rightarrow \text{Farad} \\ Q \rightarrow \text{Coulomb} \\ V \rightarrow \text{Volts} \end{cases}$$

In reality → Net charge on cap. = 0,  
+ve charge = -ve charge.  
Q → excess +ve charge on the +ve plate

## CONCLUSION

- Protons and neutrons do not move from atom to atom
- Current → movement of  $e^-$
- Insulators → resist flow of  $e^-$ , conductors → allow flow of  $e^-$ , semiconductor → b/w both.
- Photon → energy particles from sun used in produc<sup>n</sup> of electricity in PV cell.
- Voltage, current, resistance → used in circuit calc. (Ohms law) design
- Energy consumed → Watt-hrs, (PV systems)
- electrical has four requirement, including switch, load, power supply & conductor
- circuit → only 2 diff items : a switch to pass power or a load to consume power