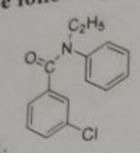
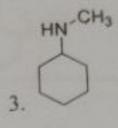
期\_中\_试卷 (A) 也 考试形式\_\_闭卷\_\_

2019 年 4 月

\_专业\_产用在学

Part 1. Give the CCS name or structural formula for each of the following compounds. (10 points)





N-甲基环己胺

N.N. Z基苯基氨基间氧苯酰氨

乙酸邻种酸甲酯

6. 溴化十二烷基三乙基铵

H, C2-N+-C2H5. Br-

R-2-溴麸酸

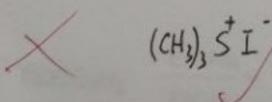
对羟基苯胺

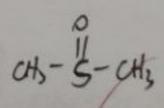
7. LDA

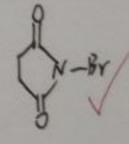
8. 碘化三甲基锍

9. DMSO

10. NBS







Part 2. Choose the most suitable key for each problem. Mark the corresponding letter in the bracket after each one. (10 points)

1. 脂肪的碱水解称为

A. 酯化 B. 皂化 C. 还原 D. 水解

A. Rank the hydrolysis rate of the following esters under alkaline

a 
$$CO_2E1$$

b.  $CO_2E1$ 

c.  $CO_2E1$ 

c.  $CO_2E1$ 

d.  $CO_2E1$ 

A.  $a > b > c > d$ 

B.  $c > a > b > d$ 

under alkaline

No.

No.

C. d>a>c>b D. d>b>a>c

3. Which of the following name reaction is usually used to prepare 1,5-dicarbonyl compounds?

A.克莱门森反应 B. 威廉姆森反应 C. 狄克曼反应 D. 麦克尔加战

. Which compound has the strongest basicity?

5. 酰胺的 Hoffmann 重排反应经过的主要活性中间体是 A. 碳正离子 B. 碳负离子 C. 碳烯 D. 氮烯

Part 3. Please complete the following reaction and write the structure of the major product  $(2' \times 10 = 20)$ :

2. 
$$COCI$$
  $CH_2N_2$   $COCHN_1$   $Ag_2O$   $CH_2Og_1$   $CH_$ 

part 4. Synthesis  $(5' \times 4 = 20)$ :

」. 由 CH₃COOC₂H₅ 合成

CHOCCHI CHONGS CH3-E-CHICOCCHI CHOCK

CH- 2 (2) Heat

2. Preparation of the compound shown below using benzene as the starting materials.

3. 以苯甲醛、丙二酸二乙酯为主要原料合成

CH(CO) EH,

CH(CO)

0= (1) OH (1) OH (1) H\*

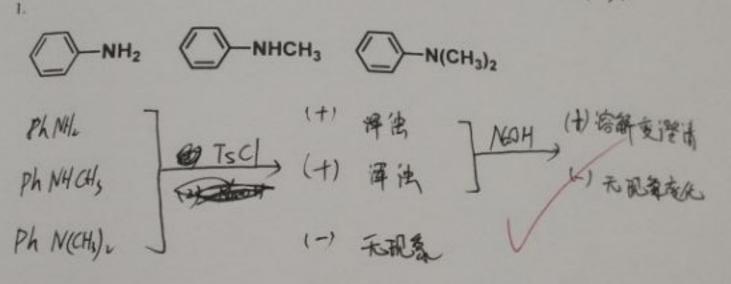
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4. Please using S-containing compound to prepare: сн₃сн₂снёсн₂сн₃ он 1901 CH CH- A1 - E CH. CH, part 5. Suggest the possible mechanism of the following reactions, deduce the product first if it is necessary. (5'×3): COOH + C2H5OH H2SO4 -HS OF C-O-CLHS -HT OF C-O-CLHS

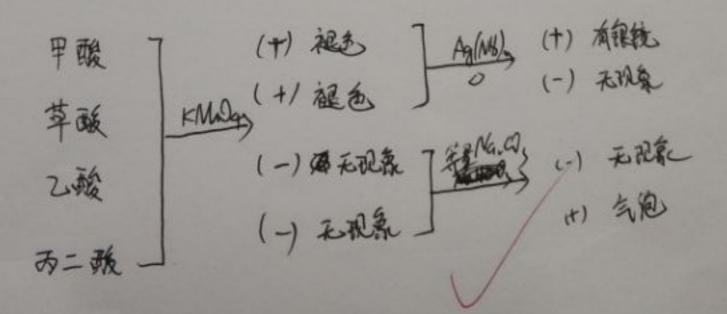
-HS OF C-O-CLHS -HT OF C-O-CLHS 2. D-C-NH<sub>2</sub> Br<sub>2</sub> NaOH NH D-G-NH. NOW, D-G-NH-Br 10-H)  $\left[ \begin{array}{c} \left[ \begin{array}{c} \mathcal{E} - \dot{\mathcal{N}} \end{array} \right] \longrightarrow \begin{array}{c} \mathcal{N} - \mathcal{C} = 0 \end{array} \right] \longrightarrow \begin{array}{c} \mathcal{N} - \mathcal{C} = 0 \\ \mathcal{N} \longrightarrow \mathcal{N} - \mathcal{C} = 0 \end{array}$ - (OL) >- NH,

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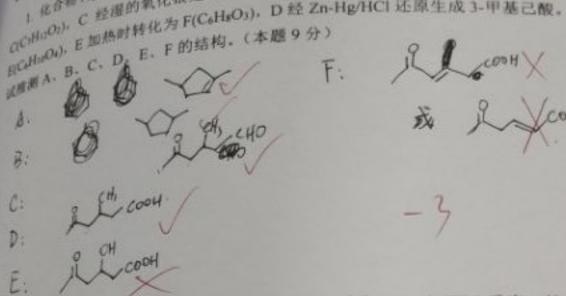
## Part 6. Identify the following compounds in chemical ways (7'):



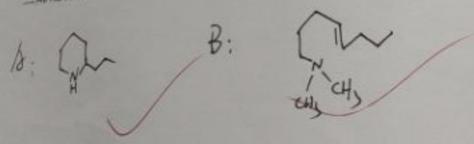
## 2. 甲酸,乙酸,草酸,丙二酸



part 7. Structures  $(6' \times 3 = 18)$ : 1. C 经湿的氧化银处理后生成 D(C2H12O2), D 可发生确从 1. 化合理 A(C-Hill) 加州 化银处理后生成 D(C-HillO3), D 可发生碘仿反应生成 a(C-HillO3), C 经湿的氧化银处理后生成 D(C-HillO3), D 经 Zn-Hg/HC1 还原生成 3 (C-HillO3), E加热时转化为 F(C6HillO3), D 经 Zn-Hg/HC1 还原生成 3 (C-HillO3), E加热时转化为 F(C6HillO3), D 经 Zn-Hg/HC1 还原生成 3 (C-HillO3), D (C-HillO3) (C)HoOn), C 整面の \*\* (C6HsO3), D 经 Zn-Hg/HC1 还原生成 3-甲基己酸。 B(6HsO4), E加热时转化为 F(C6HsO3), C を題 9 分 )



2. 化合物(A)CaH<sub>17</sub>N, 其核磁共振谱无双重峰, 它与 2 mol CH<sub>3</sub>I 反应, 然后 与 Agr O(湿)作用,接着加热则生成一个中间体(B),其分子式为 C10H21N。(B) 进一步甲基化后与湿的 Ag2O 作用,加热则生成三甲胺、1,5-辛二烯和1,4-辛 二烯混合物。写出化合物 A 和 B 的结构式。(本题 4 分)



3. 化合物 A和 B. 分子式均为 C4H8O2, 它们的谱在 1735 cm 附近都有强吸 收,它们的 HNMR 谱数据如下:

A: 1.3 (t, 3H), 2.0 (s, 3H), 4.1 (q, 2H) ppm-

B: 1.2 (t, 3H), 2.3 (q, 2H), 3.7 (s, 3H) ppm. 试推测 A、B 的结构。(本题 5 分)