

Lecture 2

Introduction to x-rays

Reading material for x-ray production

1. Smith and Webb: Chap. 2, pages 34 - 42.
2. Hendee: Chap. 2 (pages 12 - 16) for atomic physics concepts; Chap. 5 for x-ray production

Attempt the worked out problems in each chapter.

X-rays

- The oldest diagnostic imaging technique (image taken in 1895).
- Led to the first Nobel prize in physics in 1901



For the interesting story behind the discovery, read:

<https://www.nobelprize.org/prizes/physics/1901/perspectives/>

www.wikipedia.org

What are x-rays?

- Made of photons with no mass or charge. Can't be deflected by electric or magnetic fields.
- Travels in vacuum with a speed of $\sim 3 \times 10^8$ m/s.
- Energy (E) = $h\nu = hc/\lambda$;
h: Planck's constant (6.626×10^{-34} Joule-sec)

(540-1650 KHz) (88-108 MHz) Microwave

AM Radio FM

1 GHz 100 GHz

Infrared far near

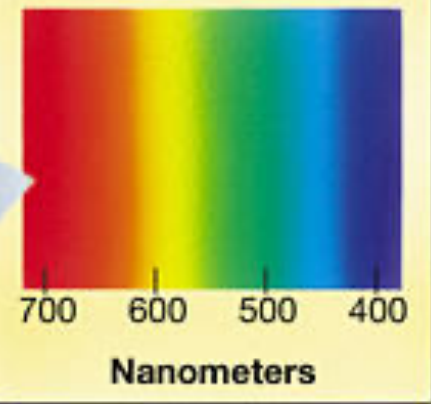
100 microns 1

Visible

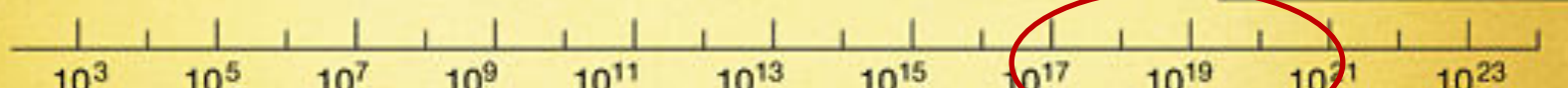
Ultraviolet near far

"Soft" X rays "Hard"

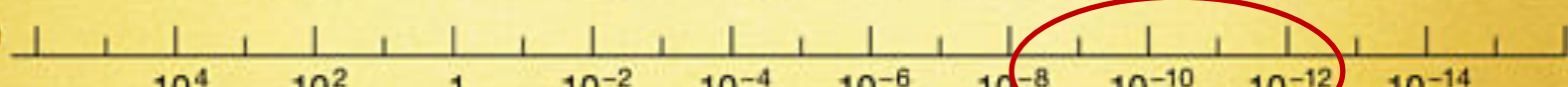
Gamma rays



Frequency (Hertz)



Wavelength (meters)



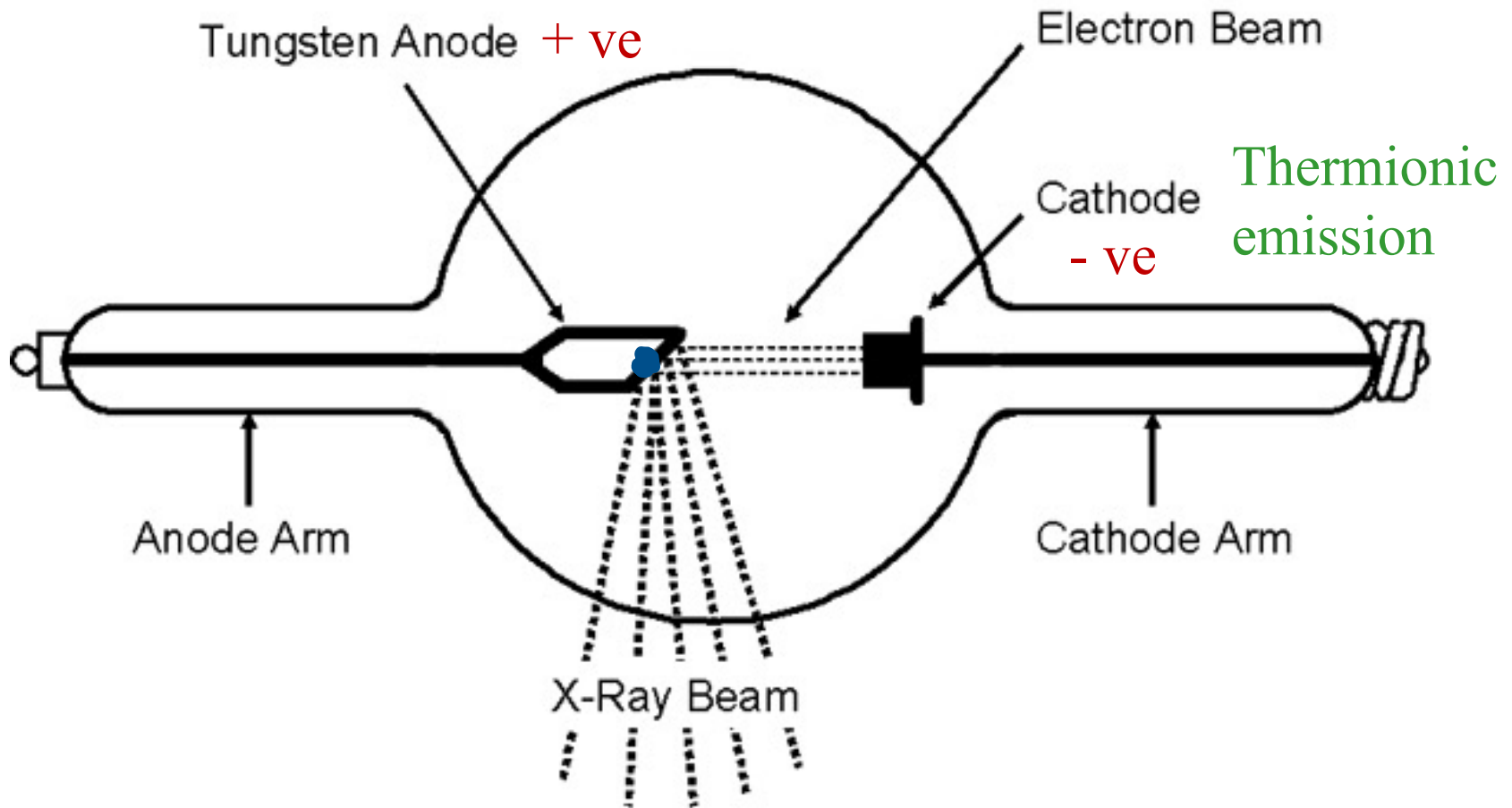
Scale



How are x-rays generated?

X-ray production: Coolidge tube

X-ray production: Coolidge tube



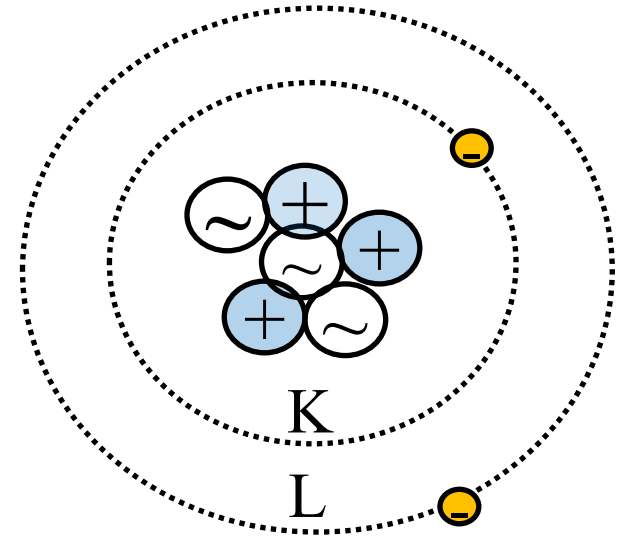
Some of the kinetic energy of electrons hitting the target is converted into x-ray photons; the rest is dissipated as heat.

X-rays are of two kinds

1. Characteristic X-rays
2. Bremsstrahlung (translates in English as “braking radiation”)

What happens inside the target (anode)?

- Electrons have discrete energy levels
- Binding energy: energy input needed to remove electron from atom. Higher binding energy for electrons in inner shells.
- Electron will release energy when it moves from higher to lower energy level.



Shells: K, L,...