Estimate the bicarbonates of well water based on its PH value using polynomial regression model.

Manual calculations: degree => 2

Step 1: Initialize m,= 1, m2=1, c=-1, epoch = 1, N=0.1, ns=2.

step 1: iter = 1

step 3: 1 = 1

Step 4;
$$\frac{\partial E}{\partial m_i} = -(4i - m_2 n_i^2 - m_1 n_i - C) n_i$$

= $-(154 - (4 - 6)^2 - 4 - 6 + 1)(4 - 6)$
= $-(92 \cdot 64)(4 \cdot 6)$
= $-704 \cdot 064$

$$\frac{\partial \epsilon}{\partial m_2} = -(y_i - m_2 n_i)^2 - m_i n_i - c) n_i^2$$

$$= -(157 - (7.6)^2 - 7.6 + 1) (7.6)^2$$

$$= -5350.8864$$

$$\frac{\partial \epsilon}{\partial c} = -(9i - m_2 \pi i^2 - m_1 \pi i - c)$$

$$= -(157 - (7.6)^2 - 7.6 + i)$$

$$= -92.64$$

Step 5:
$$\Delta m_{1} = -\eta \frac{\partial \epsilon}{\partial m_{1}} = -(0.1)(-704.064)$$

$$= 70.4064$$

$$\Delta m_{2} = -\eta \frac{\partial \epsilon}{\partial m_{2}} = -(0.1)(-5350.8864)$$

$$= 535.08864$$

$$\Delta C = -\eta \frac{\partial \epsilon}{\partial C} = -(0.1)(-92.64)$$

$$= 9.264$$