

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
466	Oxalic acid	$\text{H}^\cdot + \text{HO}_2\text{CO}_2\text{H} \longrightarrow$		3.3×10^5	Average of 2 values.	
			1	3.8×10^5	e-r.; esr; Decay of spin polarization, compared with 2-PrOH(7D). no H abstr. [730053]; rel. to $k(\text{H}^\cdot + \text{BzOH})$.	710003
			1	3×10^6	γ -r.; C.k. with 2-PrOH(7D); cor. for $k(\text{e}_{\text{aq}}^- + \text{H}_2\text{C}_2\text{O}_4) = 2.5 \times 10^{10}$; rel. to $k(\text{H}^\cdot + \text{BzOH})$.	710017