No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
272	Acid Red 265 dianion	. OH + AR $-265 \longrightarrow$	6.4	1.7×10^{10}	Average of 10 valuesr.; C.k.; obs. $G(-dye)$; rel. to $k(.OH + SCN-)$.	750188
			6.4	2.2×10^{10}	-r.; C.k.; obs. $G(-dye)$; rel. to $k(.OH + PhOH)$.	750188
			6.4	1.9×10^{10}	-r.; C.k.; obs. $G(-dye)$; rel. to $k(.OH + 1-PrOH)$.	750188
			6.4	1.7×10^{10}	-r.; C.k.; obs. $G(-dye)$; rel. to $k(.OH + (CH_3)_2CHCH_2OH)$.	750188
			6.4	1.9×10^{10}	-r.; C.k.; obs. $G(-dye)$; rel. to $k(.OH + tert-BuOH)$.	750188
			6.4	1.6×10^{10}	-r.; C.k.; obs. $G(-dye)$; rel. to $k(.OH + MeOH)$.	750188
			6.4	1.4×10^{10}	-r.; C.k.; obs. $G(-dye)$; rel. to $k(.OH + EtOH)$.	750188
			6.4	1.9×10^{10}	-r.; C.k.; obs. $G(-dye)$; rel. to $k(.OH + BuOH)$.	750188
			6.4	1.7×10^{10}	-r.; C.k.; obs. G(-dye); rel. to k(.OH + $C_2H_5CH(OH)CH_3$).	750188
			6.4	1.3×10^{10}	-r.; C.k.; obs. G(-dye); rel. to k(.OH + glucose).	750188