No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
528 Acid Chrome Blue trianion	$e_{aq}^- + ACB \longrightarrow addn \cdot$	$7 1.1 \times 10^{10}$	$\gamma\text{-r.};$ C.k. in soln. contg. EtOH; obs. G(-dye) in aerated soln.; reference rate not given; rel. to $k(\mathrm{e_{aq}}^-+\mathrm{O_2}).$	