

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
222	Tin(II) Ion	$\cdot\text{OH} + \text{Sn}^{2+} \longrightarrow \text{OH}^- + \text{Sn}^{3+}$	0.8	2×10^9	γ -r.; C.k.; rel. to $k(\cdot\text{OH} + \text{Fe}^{2+})$.	590007