No.	Compound name	Reaction equation	рН	Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$	Comments	Reference
1159	Sulfanilate Ion	$.  \mathrm{OH} + 4 \text{-} \mathrm{H}_{2} \mathrm{NC}_{6} \mathrm{H}_{4} \mathrm{SO}_{3}{}^{-} \longrightarrow$	7	$2.9 \times 10^{9}$	p.r.; P.b.k. at 270 nm.	730094