No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
671 Biuret	${ m e_{aq}}^- + { m H_2NCONHCONH_2} \\ { m H_2NC(O-)NHCONH_2}$	$\longrightarrow$ 10.3	$2.5 \times 10^8$	p.r.; D.k. at 700 nm in soln. contg. ${\sim}0.11$ mol $\mathrm{L}^{-1}$ tert-BuOH.	730091