No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1194 Thiamine cation	$.  \mathrm{OH} + \mathrm{Thm}^+ \longrightarrow$	$4.7  3.0 \times 10^9$	p.r.; C.k. (p.b.k. gave $3.2 \times 10^9$ ); rel. to $k(.OH + SCN^-)$ .	771034