No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
359	Fluoroacetate Ion	$H^{\bullet} + FCH_2CO_2^- \longrightarrow H_2 + {}^{\bullet}CHFCO_2^-$	8.5	6.5×10^5	γ -r.; C.k.; practically no F [−] formed; rel. to $k(H^{\bullet} + CD_3OH)$.	670050