No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
106	Nitrous oxide	$H \cdot {}^{+}N_{2}O \longrightarrow \cdot OH + N_{2}$	alk.	2.1×10^6	γ -r.; C.k.; rel. to H · $^+$ Fe(CN) $_6$ ³⁻ .	680693