No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
136 Ethylenediaminenickel(II) Ion	$. \operatorname{OH} + \operatorname{Ni}(\operatorname{en})_n^{2+} \longrightarrow \operatorname{OH}^-$ $\operatorname{Ni}(\operatorname{en})_n^{3+}$	+ 10.0 7	7.2×10^9	p.r.; C.k. in O_2 -satd. soln.; cor. for $OH + en$; also detd. at pH 8.0, 8.5, and 9.0; rel. to k(.OH + SCN ⁻).	720461