

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
393	3-Bromopropionate Ion	$\cdot\text{OH} + \text{BrCH}_2\text{CH}_2\text{CO}_2^- \longrightarrow$	8.5	2.3×10^8	γ -r.; C.k.; rel. to $k(\cdot\text{OH} + \text{EtOH})$.	670050