No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1186	2,2,6,6-Tetramethyl-4- piperidone N-oxyl	$^{\bullet}\mathrm{OH} + \mathrm{TAN} \longrightarrow$		3.9×10^9	Average of 4 values.	761067
			4.5, 9.2	4.0×10^9	p.r.; Condy.; may be both addn. and electron transfer.	761067
			nat.	3.4×10^9	p.r.; C.k.; cor. for H; obs. Phe abs. at 320 nm; rel. to $k(\cdot OH + \text{Phe})$.	723021
			7	4.1×10^{9}	p.r.; D.k. at 230 nm.	710061
			10.5	4.1×10^{9}	p.r.; C.k.; cor. for CO_3^- + TAN; rel. to $k(\cdot OH + CO_3^{2-})$.	710061