No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
219	$\label{eq:linear_state} $$ \{ alpha, \S alpha, $		5.8-7.9	$3.7 \times 10^{10}$	p.r.; D.k. at 510 nm $(e_{aq}^{-})$ as well as 420 nm $(\text{Fe}^{III})$ and p.b.k. at 440 nm $(\text{Fe}^{II})$ in soln. contg. (5-50) $\times 10^{-6}$ mol L <sup>-1</sup> Fe <sup>III</sup> ; p $K_a$ FePFP $(\text{H}_2\text{O})_2^{5+} = 3.9$ .	86A154