No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
192 Hydrofluoric acid	$e_{aq}^- + HF \longrightarrow HF^-$	5.02	$6 \times 10^7$	X-r.; Calcd. from data in [620021] cor. for I, where HF $_2^-$ is 95% of the acid present, and the pK of HF and HF $_2^-$ differ by 0.6; pK HF $\sim$ 3 (2 HF $\rightarrow$ HF $_2^-$ + H <sup>+</sup> )	670099