No. Compound name	Reaction equation	nЦ	ate constant $L \operatorname{mol}^{-1} \operatorname{s}^{-1}$ Comments	Reference
198 Rhodium(III) chloride	$\cdot \mathrm{OH} + \mathrm{RhCl_3} \longrightarrow \mathrm{OH}^- + \mathrm{RhCl_3}^+$	$< 1  2.9 \times 1$	$0^9$ p.r.; P.b.k. at 580 nm in 5 mol L <sup>-1</sup> HCl.	771050