

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
151	Tris(glycinato)nickelate(II) Ion	$\cdot\text{OH} + \text{Ni}(\text{Gly})_3 \longrightarrow \text{H}_2\text{O} + \text{Ni}(\text{Gly})_2(\text{NH}_2\text{CHCO}_2)^-$	6.6	$\sim 2 \times 10^8$	p.r.; P.b.k. in soln. contg. $2 \times 10^{-3}$ mol L <sup>-1</sup> Ni <sup>2+</sup> and $6 \times 10^{-3}$ mol L <sup>-1</sup> glycine.	81A128