No.	Compound name	Reaction equation	рН	Rate constant $(L \text{mol}^{-1} \text{s}^{-1})$	Comments	Reference
57	${\bf Trioxal atocobal tate (III)\ Ion}$	$\begin{array}{cccc} \mathrm{H} & \cdot & ^{+}\mathrm{Co}(\mathrm{C}_{2}\mathrm{O}_{4})_{3}{}^{3-} & \longrightarrow & \mathrm{H}^{+} & + \\ \mathrm{Co}(\mathrm{C}_{2}\mathrm{O}_{4})_{2}{}^{2-} & + & \mathrm{O}_{2}\mathrm{CCO}_{2}{}^{-} \end{array}$	- 7	6.8×10^{8}	v-r.; C.k.; rel. to $k(H \cdot + NO_2^-)$.	670498