No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
399	3-(p-Hydroxyphenyl)propionic acid	$\mathrm{H}^{\bullet} + \mathrm{HOC_6H_4CH_2CH_2CO_2H}$ -4- $\mathrm{HOC_6H_5(CH_2)_2CO_2H}$	→ 0.0	4.0×10^9	p.r.; P.b.k. (H adduct) in soln. contg. 1 mol l^{-1} tert-BuOH.	730003
			2.0-2	$3 ext{ } 1.7 \times 10^9$	p.r.; C.k.; p.b.k. at 320 and 330 nm (H adduct); rel. to $k({\rm H^{\cdot}}$ + ${\rm O_2}).$	690445