No.	Compound name	Reaction equation	ŀ	эΗ	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
461	$\label{eq:continuous} \begin{split} & \operatorname{Tricarbonatodioxouranate}(\operatorname{VI}) \\ & \operatorname{Ion} \end{split}$	$e_{aq}^{-} + UO_2(CO_3)_3^{4-}$ $UO_2(CO_3)_3^{5-}$	<i>→</i> 1	11.4	1.3×10^{10}	p.r.; D.k. in soln. contg. 0.1 mol L^{-1} tert-BuOH and 0.01 mol L^{-1} carbonate and 0.4-7 \times 10 ⁻⁴ mol L^{-1} U(VI).	81A148