No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments	Reference
91	Nitrilotriacetatoferrate(II) Ion	$.OH + FeNTA^- \longrightarrow OH^- + FeNTA$	6.0	5.0×10^9	P.r.; P.b.k.; at pH 6.2 $k = 2.5 \times 10^9$ and at pH 11 $k = 1.1 \times 10^9$.	78A436