No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
68	$\begin{array}{c} \operatorname{Bis}(\operatorname{ethylenediamine})\operatorname{copper}(\operatorname{II})\\ \operatorname{ion} \end{array}$	$. \mathrm{OH} + \mathrm{Cu(en)_2}^{2+} \longrightarrow \mathrm{Cu^{II}(en)_2}$	6.5	3.0×10^9	p.r.; P.b.k.; at pH 10.2 and 11.2 $k = 5.0 \times 10^9$ and 8.0×10^9 , resp.	710775