No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
389	5-Bromoorotate Ion	$. \mathrm{OH} + 5 \mathrm{-BrUCO_2}^- \\ 5 \mathrm{-BrU(OH)} - 6 \mathrm{-CO_2}^-$	<i>─</i> 7	$3 \times 10^9$	p.r.; P.b.k. at 340 nm; $pK = 2.4, 7.3$ .	730002
		0 = 0 (0 = 1)		7	$6.8 \times 10^9$	p.r.; C.k.; rel. to