No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
439	4-Chlorobenzoate ion	$\begin{array}{l} \mathrm{.OH} \ + \ 4\mathrm{-ClC}_{6}\mathrm{H}_{4}\mathrm{CO}_{2}^{-} \\ \mathrm{4-ClC}_{6}\mathrm{H}_{4}\mathrm{(OH)}\mathrm{CO}_{2}^{-} \end{array}$	—→ 6-9.4	$5.0 \times 10^{9}$	p.r.; p.b.k. at 346 nm; cor. for (OH + OH) and (H + aromatic).	080304