No.	Compound name		Reaction equation		рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
120		kis(4-N- atomangane	· OH + (OH) ₂ MnTMpyP ³⁺ s((OH) ₂ MnTMpyP-OH] ³⁺		11.0	2.7×10^{10}	p.r.; C.k.; $pK_a = 8.0, 10.6$; rel. to $k_{\cdot OH + Fe(CN)_6^{4-}}$.	84A426
	ion				12.4	3.5×10^9	p.r.; D.k.	84A120