

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
108	Mercury(II) bromide	$\cdot\text{OH} + \text{HgBr}_2 \longrightarrow \text{HgBrOH} + \text{Br}\cdot$		$> 9 \times 10^8$	p.r.; P.b.k. at 355 nm in soln. contg. HgBr ₂ ; obs. transient may be HgBr ₃ .	761087