

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
307	3-Aminobenzoate ion	$\cdot\text{OH} + 3\text{-H}_2\text{NC}_6\text{H}_4\text{CO}_2^- \longrightarrow 3\text{-H}_2\text{NC}_6\text{H}_4(\text{OH})\text{CO}_2^-$		$8.8 \times 10^9$	p.r.; C.k. in soln. contg. $5 \times 10^{-3}$ mol L <sup>-1</sup> carbonate; rel. to $k(\cdot\text{OH} + \text{CO}_3^{2-})$ .	80A202