No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
475 Phenol	$H^{\bullet} + C_6H_5OH \longrightarrow C_6H_6OH$		$1.7 \times 10^9$ $1.2 \times 10^9$	Average of 3 values. p.r.; P.b.k. at 335 nm.	86A331
		7	$2.1 \times 10^{9}$	p.r.; esr; Decay of H signal.	71U303
		2.0	$1.8 \times 10^9$	p.r.; P.b.k. at 330 nm.	670122