No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
841	Isopropyl acetate	$.  \mathrm{OH} + \mathrm{CH_3CO_2CH}(\mathrm{CH_3})_2 \longrightarrow$	6-7	$4.5 \times 10^8$	p.r.; C.k., at pH 2.0 $k = 7 \times 10^7$ ; rel. to k(.OH + SCN <sup>-</sup> ).	050387