No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments	Reference
106	Mercury(I) cyanide	$.\,OH + HgCN \longrightarrow$		3.1×10^9	p.r.; Estd. from d.k. of transient HgCN in soln. contg. 10^{-3} mol L^{-1} $Hg(CN)_2$ with no additives.	751203