No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
267	Acetyl peroxide	\cdot OH + (CH ₃ CO) ₂ O ₂ \longrightarrow	7	6.0×10^7	p.r.; C.k. in soln. contg. 2×10^{-3} mol L ⁻¹ SCN ⁻ ; rel. to $k(\cdot OH + SCN^-)$.	81A374