No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
284 -Chloropropionate Ion	$\begin{array}{ccc} \mathrm{H} & \cdot & ^{+}\mathrm{ClCH_{2}CH_{2}CO_{2}}^{-} \\ \cdot & \mathrm{CH_{2}CH_{2}CO_{2}}^{-} + \mathrm{HCl} \end{array}$	$\longrightarrow$ 8.5	$8.9 \times 10^7$	-r.; C.k.; rel. to $k(H + 2\text{-PrOH})$ .	670050