No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
128	Hydrazinium Ion	$\cdot \mathrm{OH} + \mathrm{H_2NNH_3}^+ \longrightarrow \mathrm{H_2O} + \cdot \mathrm{N_2H_4}^+$	6	1.0×10^{9}	p.r.; C.k.; rel. to $k(\cdot OH + SCN^{-})$	720003