No.	Compound name	Reaction equation	рН	Rate constant $(L  \text{mol}^{-1}  \text{s}^{-1})$	Comments	Reference
1217	Trichloroethylene	$\begin{array}{lll} .\mathrm{OH} & + & \mathrm{ClCH}{=}\mathrm{CCl}_2 \\ .\mathrm{CCl}_2\mathrm{CHClOH} \end{array}$	<del></del>	$4.2 \times 10^9$	Average of 2 values.	710709
			$\sim 6.5$	$4.0 \times 10^9$	p.r.; P.b.k. (condy.) (Cl <sup>-</sup> ); (CHClOHCCL <sub>2</sub> $\rightarrow$ H <sup>+</sup> + Cl <sup>-</sup> + CCl <sub>2</sub> CHO).	710709
				$4.3\times10^9$	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$ .	710709