No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
232	Benzophenone	$H^{\bullet} + (C_6H_5)_2CO \longrightarrow C_6H_5COC_6H_5$	1	6.1×10^9 6.6×10^9	Average of 2 values. p.r.; P.b.k. at 390 nm in soln. contg. 0.1 mol $\rm L^{-1}$ tert-BuOH.	751125
			1.0	5.6×10^9	$\ensuremath{\text{p.r.}}; \ensuremath{\text{P.b.k.}}$ (H adduct); tert-BuOH as OH scavenger.	720171