No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
553 Trimethylacetate Ion	$\mathrm{H}^{\cdot} + (\mathrm{CH_3})_3 \mathrm{CCO_2}^- \longrightarrow \mathrm{H_2}$ $\mathrm{CH_2C}(\mathrm{CH_3})_2 \mathrm{CO_2}^-$	+	1.9×10^6	γ -r.; C.k.; rel. to $k(H^{\cdot} + (CH_3)_2CDOH)$.	660422