No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
103	Ammonium Ion	$H^{\cdot} + NH_4^{+} \longrightarrow$	7	$< 4 \times 10^4$	$\gamma\text{-r.};$ No reaction; c.k. in soln. contg. 0.1 mol L <sup>-1</sup> (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> and 1.6 × 10 <sup>-2</sup> mol L <sup>-1</sup> N <sub>2</sub> O; rel. to $k(\mathrm{H^{-}} + \mathrm{DCO_2^{-}}).$	640095