No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
132 1,2,3-Trimethylbenzene	$O^{-} + C_6H_3(CH_3)_3 \longrightarrow OH^{-} - C_6H_3(CH_3)_2(C^{\cdot}H_2)$	+ 13 2	$.1 \times 10^9$	p.r.; P.b.k. at 263 nm in N2O-satd. soln. contg. 0.5 mol $\rm L^{-1}$ NaOH.	751009