No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1509	Vinylidene chloride	$e_{aq}^- + H_2C = CCl_2 \longrightarrow Cl^- + CH_2\dot{C}Cl$	$\sim 6.5$	$2.3\times10^{10}$	p.r.; D.k. at 720 nm in soln. contg. 0.5 mol $\rm L^{-1}$ $\it tert\text{-BuOH}.$	710709