No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
95 Triiodine Ion	$H^{\cdot} + I_3^{-} \longrightarrow I_2^{\cdot -} + H^{+} + I^{-}$	$1.5 4.5 \times 10^{10}$ 2.92 $1.72 - 2.92$	$\gamma\text{-r.};$ C.k.; obs. G(H ₂); rel. to $k(\mathrm{H^{\cdot}}+\mathrm{glucose}).$ 1×10^{10} 2.1×10^{10}	690338 phot.; C.k.; data phot.; C.k.; data