No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
81	Chloroiron(III) Ion	$H^{\bullet} + FeCl^{2+} \longrightarrow H^{+} + Fe^{2+} + Cl^{-}$	0.8	7×10^9 0.8	b; e.d.; C.k. in HCl; rel. to $k(H^{\bullet} + Fe^{3+})$. 4.8×10^9	660345 b; γ -r.; C.k. in I