No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
629 Eosin dianion	$. OH + C_{20}H_8Br_4O_5{}^{2-} \longrightarrow$	$10.5 1.3 \times 10^{10}$	p.r.; C.k. in air-satd. soln.; cor. for presence of HCO_3^- ; rel. to $k(.OH + CO_3^{2-})$.	f 670038