No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
325	Neptunate(V) ion	$e_{aq}^- + Np(V) \longrightarrow Np(IV)$	~ 14	$5.0 \times 10^{9}$	p.r.; D.k. at 650 nm in soln. contg. 1.0 mol $L^{-1}$ OH <sup>-</sup> ; $k=6.1,6.6,$ and $3.2\times10^9$ at 2.0, 5.0, and 7.5 mol $L^{-1}$ OH <sup>-</sup> , resp.	