No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
338	$({\bf Aqua}) {\bf pentachloroosmate}({\bf IV}) \\ {\bf ion}$	$\begin{array}{l} e_{\mathrm{aq}}^{} + \mathrm{OsCl}_{5}(\mathrm{H}_{2}\mathrm{O})^{-} \\ \mathrm{OsCl}_{5}(\mathrm{H}_{2}\mathrm{O})^{2-} \end{array}$	 →	4.4×10^{10}	p.r.; D.k. at 700 nm in N2-satd. soln. contg. 1 mol $\rm L^{-1}$ tert-BuOH and Na2SCl6; ionic strength effect on addition of NaClO4 showed that the complex was aquated; k cor. for I .	77A219