

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
1084	Promasine, conjugate acid	$\cdot\text{OH} + \text{PzH}^+ \longrightarrow \text{OH}^- + [\text{PzH}]_2^{\bullet+}$	5.5	3.7×10^9	p.r.; P.b.k.; reaction also involves OH addn. to S which leads to cation radical, addn. to ring and H abstr.	79A060