

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
1371	Pyridoxine	$\text{e}_{\text{aq}}^- + \text{PH} \longrightarrow \cdot\text{PH}_2$	6.8	$2.2 \times 10^{10}$	p.r.; D.k. at 700 nm in soln. contg. 0.1 mol L <sup>-1</sup> tert-BuOH; $pK_a = 5.0, 8.97$ ; at pH 11.0 $k = 2.5 \times 10^9$ .	751024