No.	Compound name	Reaction equation	pH	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
19	Pentaammineaquacobalt(III) Ion	$H^{\bullet} + Co(NH_3)_5^{3+} Co(NH_3)_5H_2O^{2+}$	$\longrightarrow$ 1	$<7.0\times10^5$	-r.; C.k.; rel. to $k(H^{\bullet} + EtOH)$ .	660010
		- 1 ( 0)0 2 -		1.8-3.9	$4.5 \times 10^5$	a.e.d.; Detd. fro