No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
1235	Trimethylamine	$.  \mathrm{OH} + (\mathrm{CH_3})_3 \mathrm{N} \longrightarrow .  \mathrm{CH_2N}(\mathrm{CH_3})_2 + \\ \mathrm{H_2O} + (\mathrm{CH_3})_3 \mathrm{N}^{\bullet}$	12	$1.3\times10^{10}$	p.r.; C.k.; extrapolated value from pH study; product anal. see [86A113]; rel. to k(.OH + SCN $^-$ ).	710585