No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
265	Manganese(II) Ions	$e_{aq}^{-} + Mn^{2+} \longrightarrow Mn^{+}$	2.	0×10^7	p.r.; D.k. at 600 nm in soln. contg. 0.05 mol $\rm L^{-1}$ MeOH.	771011