

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
421	Butyrate Ion	$\cdot\text{OH} + \text{n-C}_3\text{H}_7\text{CO}_2^- \longrightarrow$	9	2×10^9	γ -r.; C.k. with RNO; rel. to k($\cdot\text{OH} + \text{EtOH}$).	660423