No.	Compound name	Reaction equation	рН	Rate constant $(L \text{mol}^{-1} \text{s}^{-1})$	Comments	Reference
276	Acrylamide	$\begin{array}{ccc} .\mathrm{OH} & + & \mathrm{H_2C}{=}\mathrm{CHCONH_2} & & \longrightarrow \\ \mathrm{CH_3COHCONH_2} & & \end{array}$		5.9×10^9	Average of 4 values.	
			7-7.2	4.7×10^9	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$.	80A240
			nat.	5.3×10^9	p.r.; C.k.; rel. to $k(.OH + Fe(CN)_6^{4-})$.	710578
				6.8×10^{9}	p.r.; P.b.k. at 390 nm.	710578
			~12	6.6×10^9	p.r.; C.k. at pH 10.9 and 12.9; rel. to $k(.OH + CO_3^{2-})$.	700052