No.	Compound name	Reaction equation	pH	Rate constant $(\text{L mol}^{-1}\,\text{s}^{-1})$	Comments	Reference
327	Hydroxydioxoneptunium(VI) Ion	$\text{e}_{\text{aq}}^{-} + \text{NpO}_2\text{OH}^{+} \longrightarrow \text{NpO}_2\text{OH}$		2.9×10^{10}	p.r.; Extrapolated from k detd. in soln. at pH 2.5 (condy.) and at 5.3 and 5.8 [761084, 76A248]; $pK = 5.45$.	83A071

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