No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments	Reference
35	Aconitate Ion	$O^{\cdot-} + {}^-O_2CCH{=}C(CO_2{}^-)CH_2CO_2{}^- \longrightarrow$ $OH^- + {}^-O_2CCH{=}C(CO_2{}^-)\cdot CHCO_2{}^-$	14	$\sim 1 \times 10^8$	p.r.; P.b.k. (allylic radical).	751003