No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
252 2-Bromopropionate ion	$\mathrm{H}\cdot^{+}\mathrm{CH_{3}CHBrCO_{2}}^{-}\longrightarrow\mathrm{HBr}$ $\mathrm{CH_{3}CHCO_{2}}^{-}$	8.5	1.3×10^9	γ -r.; C.k.; rel. to $k(H \cdot + 2\text{-PrOH})$.	670050