No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1192	Theobromine	$.\mathrm{OH} + \mathrm{C_7H_8N_4O_2} \longrightarrow$	5.8	$\times 10^9$	p.r.; P.b.k.	86R158