No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
259	Butane	$\mathrm{H\cdot}^{+}\mathrm{n\text{-}C_{4}H_{10}}\longrightarrow$	1	3.6×10^7	esr.; esr; Decay of spin polarization, compared with 2-PrOH(7D); rel. to $k({\rm H}^{+}{\rm BzOH}).$	710003