No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
434	Stannate(II) Ion	$e_{aq}^- + Sn(II) \longrightarrow$	11	3.4×10^9	p.r.; D.k.; counterions Cl ⁻ , Na ⁺ ; in 1 mol L ⁻¹ NaOH $k=6.2\times10^9$	650047