No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
234	Xenon(VI) trioxide	$.\mathrm{OH} + \mathrm{XeO}_{3} \longrightarrow \mathrm{HOOXeO}_{2}$	8-9 ~	$\sim 1 \times 10^7$	p.r.; P.b.k. at 400-450 nm; reported $k = 10^7 - 10^8$ .	82A160