No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1008 4-Nitrophenoxide Ion	$.\mathrm{OH}\ +\ 4\text{-NO}_2\mathrm{C}_6\mathrm{H}_4\mathrm{O}^-\ 4\text{-HOC}_6\mathrm{H}_4\mathrm{NO}_3^- + \mathrm{H}^+$	$\longrightarrow$ 9 $7.6 \times 10^9$	$\gamma$ -r.; C.k.; rel. to $k$ (.OH + RNO).	72G837