No.	Compound name	Reaction equation	р	Н	Rate constant $(L  \text{mol}^{-1}  \text{s}^{-1})$	Comments	Reference
901	Methyl acetate	$. OH + CH_3CO_2CH_3$ $CH_3CO_2\dot{C}H_2 + H_2O$	<i>─</i> 6	-7 1.5	$2 \times 10^8$	p.r.; C.k., at pH 2.0 $k = 2.2 \times 10^8$ ; rel. to $k(.OH + SCN^-)$ .	650387