No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1474	Trichlorofluoromethane	$e_{aq}^- + CCl_3F \longrightarrow Cl^- + \cdot CFCl_2$	$\sim 6$	$1.6\times10^{10}$	$\gamma\text{-r.};$ C.k. with H <sup>+</sup> as well as N <sub>2</sub> O, elec. condy.; rel. to $k(e_{\rm aq}{}^- + {\rm N_2O}).$	710026