No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
278 Acrylic acid	$.\mathrm{OH} + \mathrm{H_2C} \mathrm{=} \mathrm{CHCO_2H} \longrightarrow$	$1 1.5 \times 10^9$	Fenton; C.k.: $k(.OH+MeOH)/k(.OH+Fe^{2+}) = 4.3$; rel. to $k(.OH+MeOH)$.	739341