No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
855	Maleic acid	$.  \mathrm{OH}  +  \mathrm{HO_2CCH}{=} \mathrm{CHCO_2H}  - \\ \mathrm{HO_2CCHOHCHCO_2H}$	→ 4-10.5	$6.0\times10^9$	p.r.; P.b.k. at 280-400 nm; $k$ independent of wavelength and pH.	85A487