No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
573 o-Xylene	$H^{\bullet} + C_6H_4(CH_3)_2$ 1,2- $C_6H_4(CH_3)_2$	$\longrightarrow 1 \qquad 2.0 \times 10^9$	p.r.; P.b.k. in soln. contg. 0.1-0.5 mol L^{-1} tert. BuOH and HClO ₄ .	751009