No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
233	Mercury(II) ions	$e_{\rm aq}^- + {\rm Hg}({\rm II}) \longrightarrow {\rm Hg}({\rm I})$	14	$7.1 \times 10^9$	p.r.; D.k. at 720 nm; transient spectrum $(\mathrm{Hg_I})$ formed varies with pH.	83A178