No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
226	Gallium(III) Ions	$e_{aq}^- + Ga(III) \longrightarrow Ga(II)$	11.7-13.3 4		p.r.; C.k.; counterion ClO_4^- ; $k_{\text{obs}} = (0.7-2.0) \times 10^9$; $pK = 10.3, 11.7 \text{ for } \text{Ga(OH)}_4^- \rightarrow \text{Ga(OH)}_6^{3-}$; rel. to $k(\text{e}_{\text{aq}}^- + \text{H}^+)$; k cor. for I.	79A190