No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
423	Caffeine	$.\mathrm{OH} + \mathrm{C_8H_{10}N_4O_2} \longrightarrow$	6.9	9×10^9	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$.	85R089