No. Compo	ound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
885 2-Meth	oxybenzoate Ion	$.OH + 2-CH_3OC_6H_4CO_2^ CH_3OC_6H_4(OH)CO_2^-$	\rightarrow ~ 7 5.4	$\times 10^9$	p.r.; C.k.; 1×10^{-1} mol L ⁻¹ solute; rel. to $k(.OH + SCN^{-})$.	771007