

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
1111	Isocitrate Ion	$\text{e}_{\text{aq}}^- + \text{O}_2\text{CCH}_2\text{CH}(\text{CO}_2^-)\text{CHOHCO}_2^- \rightarrow \text{H}_2\text{C}(\text{CO}_2^-)\text{CH}(\text{CO}_2^-)\text{CHOHCO}_2^-$	11, 13	2×10^7	γ -r.; C.k.; rel. to $k(\text{e}_{\text{aq}}^- + \text{ClCH}_2\text{CO}_2^-)$.	660160