No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1473	Trichloroethylene	$e_{aq}^- + ClCH = CCl_2 \longrightarrow Cl^- + C_2HCl_2$	1	$.9 \times 10^{10}$	p.r.; D.k. at 720 nm in soln. contg. 0.5 mol $\rm L^{-1}$ tert-BuOH.	710709