No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
236	Hydrogen perxenate(VIII) Ion	$.\mathrm{OH} + \mathrm{HXeO_8}^{3-} \longrightarrow$	11-13	3.9×10^{8}	p.r.; P.b.k. assuming pK of OH = 11.9.	82A160