No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1143	8-Mercaptopropionate ion	$\begin{array}{l} e_{\rm aq}^- \ + \ \mathrm{HSCH_2CH_2CO_2}^- \\ \cdot \mathrm{CH_2CH_2CO_2}^- + \mathrm{HS}^- \end{array}$	\longrightarrow 7.4	5.0×10^9	p.r.; D.k. at 700 nm in soln. contg. \sim 0.1 mol L ⁻¹ tert-BuOH; $pK_a = 4.3, 10.3$; at pH 13.0 $k \approx 2.8 \times 10^8$.	730090