No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
995	Nitrobenzene	$. \mathrm{OH} + \mathrm{C}_{6}\mathrm{H}_{5}\mathrm{NO}_{2} \longrightarrow \mathrm{HOC}_{6}\mathrm{H}_{5}\mathrm{NO}_{2}$	7	3.2×10^9	p.r.; P.b.k. at 410 nm; cor. for (OH + OH) and (H + aromatic).	680304
				4.7×10^{9}	p.r.; P.b.k. at 410 nm.	670458
				3.5×10^9	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$	670458