No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
857	Maleic hydrazide, conjugate base	$. OH + MH^{-} \longrightarrow$	11	4.6×10^9 4.2×10^9	Average of 2 values. p.r.; P.b.k. at 430 nm; 33% addn, 67% semiquinone formn.	83A166 83A166
				4.9×10^{9}	p.r.; C.k.; rel. to $k(.OH + MeOH)$.	83A166