No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments	Reference
482	Tetracyanozincate(II) ion	$\mathrm{e_{aq}^{-}} + \mathrm{Zn(CN)_4^{2-}} \longrightarrow$	10	1.8×10^8	p.r.; D.k.; soln. contains 0.1 mol $\mathrm{L^{-1}}$ $\mathrm{CN^{-}}$.	850047