No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
94	Ferrocenylacetate Ion	$\begin{array}{cccc} .\mathrm{OH} \ + \ \mathrm{FcCH_2CO_2}^- & \longrightarrow & \mathrm{OH}^- \ + \\ \mathrm{Fc}^+\mathrm{CH_2CO_2}^- \end{array}$	9.0 8.	7×10^9	p.r.; P.b.k. at 625 nm in 0.025 mol $\rm L^{-1}$ borax buffer contg. (0.2-0.8) $\times 10^{-3}$ mol $\rm L^{-1}$ substrate.	84A460