

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
1248	Tyramine, negative ion	$\cdot\text{OH} + {}^-\text{OC}_8\text{H}_4\text{CH}_2\text{CH}_2\text{NH}_2 \longrightarrow$	11.2	$1.5 \times 10^{10}$	p.r.; C.k.; $pK_a = 9.5, 10.8$ ; rel. to $k(\cdot\text{OH} + \text{SCN}^-)$ .	730003