No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
10	Iron(II) Ion	$O^{-} + Fe^{2+} \longrightarrow OH^{-} + Fe^{3+}$	4.4-4.8 3	$3.8 \times 10^9$	p.r.; C.k.; rel. to $k(O^{-} + \text{EtOH})$ .	710137