No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
119 Oxygen	$\mathrm{H^{\cdot}} + \mathrm{O_2} \longrightarrow \mathrm{HO_2^{\cdot}}$	2.1×10^{10}	Selected value.	640043
		$1.2 2.1 \times 10^{10}$	p.r.; P.b.k. at 240 nm.	640043