No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1055	Phenoxide Ion	$. \mathrm{OH} + \mathrm{C}_6\mathrm{H}_5\mathrm{O}^- \longrightarrow \mathrm{HOC}_6\mathrm{H}_5\mathrm{O}^-$	10.7 9	0.6×10^9	γ -r.; C.k.; measured $^{14}\mathrm{CO}_2$; rel. to M(.OH + BzO ⁻).	650099