No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
127	Hydrazine	$.\mathrm{OH} + \mathrm{H_2NNH_2} \longrightarrow .\mathrm{NHNH_2} + \mathrm{H_2O}$	10	1.4×10^{10}	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$.	720003