No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
219 Aspartate monoanion	$H^{-+} Asp^{-} \longrightarrow$	$7   2.7 \times 10^6$	$\varepsilon$ -r.; esr; Decay of spin polarization, compared with EtOH. in phosphate buffered soln.; rel. to $k(\mathrm{H^{.+}BzOH})$ .	