No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments
279	Permanganate Ion	$e_{aq}^{} + MnO_4^{} \longrightarrow MnO_4^{2-}$		2.6×10^{10}	Average of 2 values.
			3×10^{10}	p.r.; D.k.	650044
		7.0	2.2×10^{10}	p.r.; D.k.; at pH 13 $k = 3.7 \times 10^{10}$.	640046