

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
438	3-Chlorobenzoate Ion	$\cdot\text{OH} + \text{3-ClC}_6\text{H}_4\text{CO}_2^- \longrightarrow \text{3-ClC}_6\text{H}_4(\text{OH})\text{CO}_2^-$	7	$5.3 \times 10^9$	$\gamma$ -r.; C.k.; obs. $G(\text{acetone})$ ; rel. to $k(\cdot\text{OH} + 2\text{-PrOH})$ .	740167