

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
344	Hydrogen phosphate Ion	e _{aq} ⁻ + HPO ₄ ²⁻ →	9-12.3	1.4 × 10 ⁵	p.r.; D.k.; <i>pK_a</i> (H ₃ PO ₄) = 2.12, 7.21, 12.32; <i>I</i> = 0.6.	731049