No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1030	Oxalate ion	$.OH + ^-O_2CCO_2^- \longrightarrow$	6	7.7×10^6	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$.	710041