No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
906	Methylammonium Ion	$^{\bullet}$ OH + CH <sub>3</sub> NH <sub>3</sub> $^{+}$ $\longrightarrow$	4	$3.6 \times 10^7$	p.r.; C.k., also detd. at pH 2 and 7; rel. to $k(\cdot OH + SCN^-)$	700371
				6-8	$9.1\times10^7$	p.r.; C.k.; pH st