

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
1492	Tyramine, negative ion	$\text{e}_{\text{aq}}^- + \text{OC}_8\text{H}_4\text{CH}_2\text{CH}_2\text{NH}_2^- \longrightarrow$	11.2	5.8×10^7	p.r.; D.k. at 690 nm in soln. contg. $\sim 0.1 \text{ mol L}^{-1}$ <i>tert</i> -BuOH;	730003