No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
323		$e_{aq}^{-} + NpO_2(CO_3)_2^{3-}$ $Np(CO_3)_6^{6-}$	$\longrightarrow$	$2.5\times10^9$	p.r.; D.k. in soln. contg. 0.1-5 mol L <sup>-1</sup> K <sub>2</sub> CO <sub>3</sub> and $2 \times 10^{-4}$ mol L <sup>-1</sup> Np(V). Calcd. $k = 4.5 \times 10^{8}$ for $I = 0$ [751177].	741170