No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
213	Sulfatoiron(III) Ion	$e_{aq}^- + FeSO_4^+ \longrightarrow$	2.1-2.9 ~	$\sim 2 \times 10^{10}$	$\gamma\text{-r.};$ C.k.; soln. contg. 0.5 mol L $^{-1}$ Na ₂ SO ₄ in H ₂ SO ₄ ; rel. to k(e_{aq}^- + H $^+$).	710203