No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
415	Malate Ion	$H^{\cdot} + -O_2C(CH_2)_3CH(OH)CO_2^{-} \longrightarrow H_2 + -O_2C(CH_2)_2COHCO_2^{-}$	5.5	5.5×10^7	p.r.; Unpublished data, A. Shafferman and G. Stein.	741007