No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
131	Hydroxylaminedisulfonate Ion	$.OH + HON(SO_3)_2^{2-}$ $NO(SO_3)_2^{2-} + H_2O$	→ 8.4-8.8	$5.7 \times 10^7$	e-r; esr; Obs. buildup of ON(SO3-)2; $k$ is ten times faster at pH 11.5-12 and probably is cn. dependent; rel. to $k(.\mathrm{OH} + \mathrm{PhH}).$	680471 6806