No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
452	Nicotinic acid, conjugate acid	$H^{\cdot} + 3$ -py $H^{+}CO_{2}H \longrightarrow 3$ -py $H_{2}^{+}CO_{2}H$	1	5×10^8	p.r.; P.b.k. (H adduct); tert-BuOH as OH scavenger.	710582