No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
26	Ruthenate(VI) Ion	$O^{-} + RuO_4^{2-} \longrightarrow OH^- + RuO_4^-$	> 13 2	2.4×10^9	γ -r.; C.k.; rel. to $k(O^{-} + NO_2^{-})$.	680063