No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
34	Carbon dioxide	$e_{aq}^- + CO_2 \longrightarrow$	7	$7.7 \times 10^9$	p.r.; D.k. at 578 nm in pres. of $10^{-3}$ mol $L^{-1}$ MeOH, $[CO_2] = 50\text{-}200 \times 10^{-6}$ mol $L^{-1}$ .	630073