No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
834 Dicyandiamide	$\mathbf{e_{aq}}^- + \mathrm{NCN} = \mathrm{C(NH_2)_2}$ $[\mathrm{H_2NC}(=\mathrm{NH})\mathrm{NHCN}]^-$	→ 3-4 1	1.2×10^{10}	$\gamma\text{-r.};$ C.k. in soln. contg. 0.05 mol L $^{-1}$ EtOH plus KClO ₄ ; obs. G(H ₂); rel. to $k(\mathrm{e_{aq}}^- + \mathrm{H}^+);$ $I=0.002.$	