No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
735	1-Chlorobutane	$e_{aq}^- + CH_3(CH_2)_3Cl \longrightarrow Cl^- + CH_2(CH_2)_2CH_3$		$4.8 \times 10^7$	p.r.; D.k.	771107
			9-10	$4.5 \times 10^{8}$	p.r.; D.k. at 600 nm.	700407
			7.28	$4.5\times10^8$	p.r.; D.k.	650018