No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
574 p-Xylene	$H^{+} + C_6H_4(CH_3)_2$ 1,4-C ₆ H ₆ (CH ₃) ₂	$\longrightarrow 1 \qquad 2.2 \times 10^0$	p.r.; P.b.k. in soln. contg. $0.1\text{-}0.5 \text{ mol } L^{-1}$ tert-BuOH and HClO ₄ .	751009