No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
613	Dimethyl sulfide	$. \mathrm{OH} + (\mathrm{CH_3})_2 \mathrm{S} \longrightarrow \mathrm{CH_3S}(\mathrm{OH}) \mathrm{CH_3}$	1.	$.9 \times 10^{10}$	p.r.; P.b.k. at 280 nm; .CH ₂ SCH ₃ and [(CH ₃) ₂ S] ₂ ⁺ formn. deduced by opt. and condy. studies.	751078