No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
254 2-Bromopropionic acid	$\mathrm{H}^{\cdot} + \mathrm{CH_3CH(Br)CO_2H} \longrightarrow \mathrm{HB}$ $\mathrm{CH_3}\dot{C}\mathrm{HCO_2H}$	3r + 1.0	$1.4\times10^9$	$\gamma$ -r.; C.k.; rel. to $k(H^{\cdot} + 2\text{-PrOH})$ .	670050