No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
19 Hydrogen arsenate(V) Ion	$e_{aq}^- + HAsO_4^{2-} \longrightarrow$	$11.0 1.9 \times 10^8$	p.r.; D.k.; counterion Na ⁺ ; at $I = 10^{-3}$, $k_{obs} = 2.0 \times 10^{8}$; $pK = 2.25, 6.77, 11.6$; k cor. for I .	680295