No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
149	Glycinatonickel(II) Ion	$\begin{array}{ccc} .\operatorname{OH} & + & \operatorname{Ni}(\operatorname{Gly})^+ & \longrightarrow \\ \operatorname{Ni}(\operatorname{NH_2CHCO_2})^+ & \end{array}$	$H_2O + 6.6 \sim$	$\sim 1.5 \times 10^8$	p.r.; P.b.k. in soln. contg. 2×10^{-3} mol L ⁻¹ Ni ²⁺ and 2×10^{-3} mol L ⁻¹ glycine.	81A128