

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
435	Chloroacetic acid	$\cdot\text{OH} + \text{ClCH}_2\text{CO}_2\text{H} \longrightarrow$	1	4.3×10^7	p.r.; C.k.; $\text{p}K_a = 2.85$; rel. to $k(\cdot\text{OH} + \text{SCN}^-)$.	650387