No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
88	Hemin	$H \cdot + Fe^{3+}$ heme addn · (to porphyrin)	<i>→</i> 7 >	3×10^{10}	p.r.; P.b.k. at 610 nm; intermed. formed decays to give Fe^{2+} heme.	771128