No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
514	Serine, conjugate acid	$H^{\cdot} + \operatorname{Ser} H^{+} \longrightarrow$	1	$1.2 \times 10^6$	$e$ -r.; esr; Decay of spin polarization, compared with EtOH; rel. to $k(\mathrm{H^{ ext{-}}} + \mathrm{BzOH})$ .	710040
			1	$3.6\times10^6$	$\gamma\text{-r.};$ C.k.; obs. $G(\mathrm{H}_2);$ rel. to $k(\mathrm{H}^{\cdot}+\mathrm{AA}).$	680343