No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$
460	Nitromethane	$\mathrm{H}^{+}\mathrm{CH_{3}NO_{2}}\longrightarrow$	_	$4.3\times10^7$
		1		e.r.; esr; Decay of spin polarization, compared with 2-PrOH(7D). no H abstr. [730053]; rel
			$4.6 \times 10^{7}$	r.; C.k.; obs. $G(H_2)$ in sulfuric acid soln.; rel. to $k(H \cdot {}^+EtOH)$ .