No. Com	mpound name	Reaction equation	рН	Rate constant $(L \text{mol}^{-1} \text{s}^{-1})$	Comments	Reference
830 Dich		$e_{aq}^- + (Cl_2C_6H_3CO)_2O_2$ $Cl_2C_6H_3CO^- + Cl_2C_6H_3CO_2$		5×10^9	p.r.; D.k. at 650 nm in deaerated soln. contg. 0.5 mol ${\rm L^{-2}}$ $tert$ -BuOH.	81A374