No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments	Reference
1087	Iminodiacetate Ion	$e_{aq}^{-} + HN(CH_2CO_2)_2^{2-} \longrightarrow [HN(CH_2CO_2)_2]^{\cdot 3-}$	10.5	1.2×10^7	p.r.; D.k. in soln. contg. 0.1 mol L^{-1} tert-BuOH and 0.01 mol L^{-1} IDA; at 0.25 mol L^{-1} and pH 8 $k = 1.7 \times 10^7$.	81A148