

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
219	Selenite(IV) Ion	$\cdot\text{OH} + \text{SeO}_3^{2-} \longrightarrow \text{OH}^- + \text{SeO}_3^{\bullet-}$	9.5	$3.5 \times 10^9$	p.r.; P.b.k. at 430 nm in SeO2 soln.	771173