No.	Compound name	Reaction equation	рН	Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$	Comments	Reference
13	Cadmium(II) Ion	$H^{\cdot +} Cd^{2+} \longrightarrow$	~ 7	$< 3 \times 10^5$	$\gamma$ -r.; No reaction. c.k. in soln. contg. $1-5\times 10^{-2}$ mol L <sup>-1</sup> CdSO <sub>4</sub> and $10^{-2}$ mol L <sup>-1</sup> EtOH, obs. G(H <sub>2</sub> ); rel. to $k(H\cdot + EtOH)$ .	