No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
200	Allyl alcohol	H <sup>•</sup> + H <sub>2</sub> C=CHCH <sub>2</sub> OH − H <sub>3</sub> CCHCH <sub>2</sub> OH	$\longrightarrow$ 1	$1.1\times10^9$	$\gamma$ -r; C.k.; rel. to $k(H^{\bullet} + HCO_2H)$ .	680343
~~~~	~~~~		$\sim 6$	$3.4\times10^9$	$\gamma$ -r; C.k.; rel. to $k(H^{\bullet} + 2 \text{-PrOH})$ .	630041