

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
1189	Tetrapropylammonium ion	$\cdot\text{OH} + (\text{CH}_3\text{CH}_2\text{CH}_2)_4\text{N}^+ \longrightarrow$		2.5×10^9	p.r.; C.k.; ave. of values rel. to $k(\cdot\text{OH} + \text{SCN}^-) = 1.1 \times 10^{10}$ and $k(\cdot\text{OH} + \text{Fe}(\text{CN})_4^-) = 1.1 \times 10^{10}$.	80A346