No.	Compound name	Reaction equation	рН	Rate constant $(L  \text{mol}^{-1}  \text{s}^{-1})$	Comments	Reference
1337	-Phenylpropionate Ion	$e_{aq}^- + C_6H_5CH_2CH_2CO_2^- \longrightarrow$	9-11.5	$1.3 \times 10^7$ $1.4 \times 10^7$	Average of 2 values. p.r.; D.k. at 720 nm in Ar-satd. soln. contg. 0.1 mol $\rm L^{-1}$ tert-BuOH and $\rm 10^{-4}\text{-}10^{-2}$ mol $\rm L^{-1}$ carboxylate ion.	85A158
			12.1	$1.1\times10^7$	p.r.; D.k.	650018