

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
392	2-Bromopropionate Ion	$\cdot\text{OH} + \text{CH}_2\text{BrCO}_2^- \longrightarrow$	8.5	$2.3 \times 10^8$	$\gamma$ -r.; C.k.; rel. to $k(\cdot\text{OH} + \text{EtOH})$ .	670050