No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments	Reference
1119	Pyridoxine	$.\,OH + PH \longrightarrow P\cdot\ + H_2O$	7.2	6.3×10^9	p.r.; P.b.k.; at pH 3.6 and 10.5 $k = 4.3$ and 7.4×10^9, resp.	751024