No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
335	Ozone	$e_{aq}^- + O_3 \longrightarrow O_3^{\bullet -}$	9	3.6×10^{10}	p.r.; C.k.; obs. abs. at 430 nm $({\rm O_3^{-}})$ in soln. contg. 0.15 mol L ⁻¹ HCO $_3^-$ and $\sim 10^{-4}$ mol L ⁻¹ ozone; rel. to $k({\rm e_{aq}}^- + {\rm O_2})$.	83A117