No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
165	${\bf Hexacyanoosmate(II)\ Ion}$	$OH + Os(CN)_6^{4-} \longrightarrow Os(CN)_6^{3-}$	H ⁻ +	9.6×10^9	Average of 3 values.	
				8×10^9 1.1×10^{10}	p.r.; C.k.; rel. to $k(.OH + 2\text{-}PrOH)$. p.r.; C.k.; rel. to $k(.OH + MeOH)$.	731031 731031
		~ 7	1.0×10^{10}	p.r.; P.b.k. at 410 nm.	731031	