No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
832	Isoamylammonium Ion	$. \mathrm{OH} + (\mathrm{CH_3})_2 \mathrm{CHCH_2CH_2NH_3}^+ \longrightarrow$	4	7.9×10^9	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$	700371