No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1232	Trimethylacetic acid	$^{\bullet}\mathrm{OH} + (\mathrm{CH_3})_3\mathrm{CCO_2H} \longrightarrow$	~ 2	6.5×10^{8}	${\rm Fe^{3+} + H_2O_2};$ Obs. effect of solute on ${\rm Fe^{3+}}$ catalyzed decomp. of hydrogen peroxide.; rel. to $k({}^{\bullet}{\rm OH} + {\rm H_2O_2}).$	769406