No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
306 2-Aminobenzoate ion	$\begin{array}{l} {\rm .OH} \ + \ 2\text{-}{\rm H}_{2}{\rm NC}_{6}{\rm H}_{4}{\rm CO}_{2}{}^{-} \\ {\rm 2\text{-}H}_{2}{\rm NC}_{6}{\rm H}_{4}{\rm (OH)}{\rm CO}_{2}{}^{-} \end{array}$	$\longrightarrow$ 6.2	$1.1\times10^{10}$	p.r.; P.b.k.; $pK_a = 2, 5$ .	741063