

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
434	Chloroacetate Ion	$\cdot\text{OH} + \text{ClCH}_2\text{CO}_2^- \longrightarrow$		$\sim 4.0 \times 10^8$	$\gamma$ -r.; C.k., obs. $G(\text{Cl}^-)$ : ratio of H abstr. to $\text{Cl}^-$ formn. = 4.0; rel. to $k(\cdot\text{OH} + \text{MeOH})$ .	690422