No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1073	Phosphoenolpyruvate Ion	$. OH+CH_2=C(CO_2^-)OPO(O^-)_2 \longrightarrow$	~ 7	8.9×10^{9}	$\gamma\text{-r.};$ C.k.; obs. $G(\text{inorg. phosphate});$ 8.4% oxidative dephosphorylation; rel. to $k(.\text{OH} + \textit{tert}\text{-BuOH}).$	753070