No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments	Reference
1165	Tetrabutylammonium Ion	$.\,\mathrm{OH} + [\mathrm{CH_3(CH_2)_3}]_4\mathrm{N}^+ \longrightarrow$		4.9×10^9	p.r.; C.k.; ave. of values rel. to $k(.\,\mathrm{OH} + \mathrm{SCN}^-) = 1.1 \times 10^{10}$ and $k(.\,\mathrm{OH} + \mathrm{Fe(CN)_6}^{4-}) = 1.1 \times 10^{10}$.	80A346