No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
465	Vanadyl(IV) ion	$e_{aq}^- + VO^{2+} \longrightarrow VO^+$	3.8-4.6	8×10^{10}	p.r.; D.k. in soln. contg. $(2-5) \times 10^{-4}$ mol L ⁻¹ VO(ClO ₄) ₂ and 0.01-0.05 mol L ⁻¹ EtOH or 0.12 mol L ⁻¹ tert-BuOH; $pK_a = 5.79$, estd. $k = 6 \times 10^{10}$ for $e_{aq}^- + \text{VOOH}^+$; cor. for charge on cation.	86A152