

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
1085	Promethazine, conjugate acid	$\cdot\text{OH} + \text{PZH}^+ \longrightarrow \text{PZH}_2^{2+}$		$9.2 \times 10^9$	Average of 3 values.	
			3.5	$9.7 \times 10^9$	p.r.; P.b.k. at 505 nm; 29% of the OH reacts to give the radical cation, the remainder may add or abstract H.	83A272
				$8.0 \times 10^9$	p.r.	83A392
			7	$1.0 \times 10^{10}$	p.r.; Unpubl. data.	741181