No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
488 L-Proline, conjugate acid	$H^{\cdot} + ProH^{+} \longrightarrow$		5.8×10^5	Average of 2 values.	
		1	7×10^5	e_{aq}^- ; esr; Decay of spin polarization, compared with EtOH; $pK_a = 1.99$; rel. to $k(H^{\cdot} + BzOH)$.	710040
		1	4.2×10^5	γ -r.; C.k.; rel. to $k(H^{\cdot} + AA)$.	680343