No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
878	Methanedisulfonate Ion, hydrogen	$\begin{array}{lll} .\mathrm{OH}+-\mathrm{O_3SCH_2SO_3H} &\longrightarrow & \mathrm{H_2O}+\\ -\mathrm{O_3SCHSO_3H} \end{array}$	1.05	2.5×10^7	p.r.; C.k.; calcd. from obs. rate and $pK_a=1.5;$ rel. to $k(.\mathrm{OH}+\mathrm{MeOH}).$	77A221