No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
493	Acetamide	${\rm e_{aq}}^- + {\rm CH_3CONH_2} \ {\rm CH_3C(O^-)NH_2}$	<del></del>	$4.0 \times 10^7$	Average of 2 values.	
			5.2	$4.5 \times 10^7$	p.r.; D.k. at 600 nm; $k$ increases with pressure 0-6.4 kbar $(6.4 \times 10^8 \text{ N/m}^2)$ .	720298
			9.2	$3.5 \times 10^7$	p.r.; D.k. in soln. contg. $0.5 \text{ mol } L^{-1}$ $tert$ -BuOH and borate buffer.	710414