No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
384 Bromoacetate Ion	$.  \mathrm{OH} + \mathrm{BrCH_2CO_2}^- \longrightarrow$	9 4	$4.6 \times 10^7$	$\gamma\text{-r.};$ C.k. with RNO; rel. to k(.OH + EtOH).	660423