No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
124 Ammonia	$.  \mathrm{OH} + \mathrm{NH_3} \longrightarrow .  \mathrm{NH_2} + \mathrm{H_2O}$	11.4	$9 \times 10^7$	p.r.; C.k. with benzoate ion and thiocyanate; reference values not given; $.{\rm OH}+{\rm NH_4}^+$ too slow to observe.	