No.	Compound name	Reaction equation	pH	Rate constant $(\text{L}\,\text{mol}^{-1}\,\text{s}^{-1})$	Comments	Reference
953	Formamide	$e_{aq}^{-} + \text{HCONH}_2 \longrightarrow$	6.3	1.8×10^7	p.r.; D.k. at 600 nm; k increases with pressure $\rightarrow$ 6.4 kbar (6.4×10^8 N/m^2).	720298
			9.2	6.3×10^7	p.r.; D.k. in soln. contg. 0.5 mol L^{-1} $tert$-BuOH and borate buffer.	710414
			.	$< 1 \times 10^6$	p.r.; No effect of 10^{-2} mol L^{-1} formamide on d.k.	670054