No. Compound name	Reaction equation	pH Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
954 Formate Ion	$e_{aq}^- + HCO_2^- \longrightarrow$	$\sim 10 < 1 \times 10^6$	p.r.; D.k. at 578 nm. in soln. contg. 1 mol $\rm L^-$ formate ion.	1 630073