No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
306	1,4,8,12-	$e_{aq}^- + NiL^{2+} \longrightarrow NiL^+$	5.5	1.2×10^{10}	p.r.; D.k. as well as p.b.k. in Ar-satd. soln. contg.	85A145
	${\it Tetraazacyclopentade canenickel (Interpretation of the content of the conten$	I)			$(1-10) \times 10^{-5} \text{ mol L}^{-1} \text{ nickel complex and } 0.1 \text{ mol}$	
	ion				L^{-1} tert-BuOH.	