No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments	Reference
232	Octadecatungstodiphosphate(IV,V). ion	$OH + [P_2W_{18}O_{62}-H]^{6-} \longrightarrow P_2W_{18}O_{62}{}^{6-} + H_2O$	1.5	2.7×10^9	p.r.; D.k. at 345 nm in deaerated soln.; 10-20% of reaction is deprotonation ($\rightarrow P_2W_{18}O_{62}{}^{6-} + H^+$), $k \sim 5 \times 10^4\,\mathrm{s^{-1}}$; $[P_2W_{18}O_{62}-H]^{6-}$ formed from H + $P_2W_{18}O_{62}{}^-$.	81A385