No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
135	Nickel(II) Ions	$. \mathrm{OH} + \mathrm{Ni}^{2+} \longrightarrow$	<	5×10^5	p.r.; Estd. from lack of effect of 0.25 mol L^{-1} Ni ²⁺ on yield of Cu ^{III} in O ₂ -satd. 2 ×10 ⁻³ mol L^{-1} Cu ²⁺ .	751027