No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
374	Glycerol	$H \cdot + HOCH_2CH(OH)CH_2OH \longrightarrow$		2.9×10^7	Average of 3 values. p.r.; C.k.: obs. Ag at 410 nm: contains <i>tert</i> -BuOH; k increases $0 \rightarrow 6.72 \times 10^8$ N m ⁻³ ; rel. to $k(\text{H}\cdot + \text{Ag}^+)$.	731053
			1	2.7×10^7	e.r.; esr; Decay of spin polarization, compared with EtOH; rel. to $k(H\cdot + BzOH)$.	710040
			1	3.3×10^7	p.r.; C.k.; p.b.k. at 313 nm (Ag ²⁺); rel. to $k(H\cdot + Ag^+)$.	670550