No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments
954	2-Methyl-2-propanol	$\begin{array}{ccccc} .\mathrm{OH}  +  (\mathrm{CH_3})_3\mathrm{COH} & \longrightarrow & \mathrm{H_2O}  + \\ .\mathrm{CH_2C}(\mathrm{CH_3})_2\mathrm{OH} & (95.7\%) & + \\ (\mathrm{CH_3})_3\mathrm{CO} \cdot (4.3\%) & & & \end{array}$		$6.0 \times 10^8$	Selected value. p.r.; 59, and 79°C; rel. to
			$6.6\times10^8$	p.r.; C.k.; obs. ABTS <sup>+</sup> formn. at 415 nm; rel. to $k(.OH + ABTS)$ .	82A196
	6			$6.0 \times 10^8$	p.r.; D.k. at 280 i BuOH.
	7			$7.6 \times 10^{8}$	
	nat.			$5.9 \times 10^{8}$	p.r.; C.k.; rel. to $k(.$
	7			$4.2 \times 10^{8}$	p.r.; C.k.; rel. to $k(.$