No.	Compound name	Reaction equation	рН	Rate constant $(L  \text{mol}^{-1}  \text{s}^{-1})$	Comments	Reference
432	Chloral hydrate	$\begin{array}{ll} .\mathrm{OH} +\mathrm{CCl_3CH(OH)_2} \\ \mathrm{CCl_3}\dot{C}(\mathrm{OH})_2 + \mathrm{H_2O} \end{array}$	$\longrightarrow$	$3.1\times10^9$	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$ .	730062