No. Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
130 Praseodymium(III) Ion	$H \cdot {}^{+} Pr^{3+} \longrightarrow$	1.9 <	1×10^3	γ -r.; No reaction; c.k. with CH ₃ OH; [method of 670099]; substrate causes little change in H ₂ yield on addn. of 10^{-2} - 10^{-1} mol L ⁻¹ Pr(III) to 10^{-4} mol L ⁻¹ MeOH.	