No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
340	Dimethyl sulfoxide	$H^{\bullet} + CH_3SOCH_3 \longrightarrow CH_3S(OH)CH_3$	1	$\sim 2.7 \times 10^6$	γ -r.; C.k.; obs. $G(\text{Dimethylsulfide})$; rel. to $k(\text{H}^{\cdot} + \text{Fe}(\text{CN})_6^{3-})$.	84A282
			~1	9.7×10^6	$\gamma\text{-r.};$ C.k.; obs. $G(\mathrm{H_2})$ and $G(\mathrm{CH_4});$ rel. to $k(\mathrm{H^{\cdot}}+\mathrm{EtOH}).$	81A159