

No.

10→Nitric oxide→ $\{ce{NO + \cdot OH -> H^+ + NO_2^-}\}$ → $2 \times 10^{10}$ →f.p.; (condy.); Estd. from redn. in hydrolysis of N<sub>2</sub>O<sub>3</sub> (from NO + NO<sub>2</sub>) in soln. contg.  
→→→7→ $1.0 \times 10^{10}$ →p.r.; P.b.k. at 220 nm (NO<sub>2</sub><sup>-</sup>).→730096  
→→→→ $1.7 \times 10^{10}$ →f.p.; C.k.; rel. to  $k_{\{ce{\cdot OH + NO_2^-}\}}$ .→707264