No. Compound nam	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
592 Dimethyl disulf	de $.OH + CH_3SSCH_3$ $[CH_3SSCH_3]^+$	$\longrightarrow$ OH <sup>-</sup> + $\sim 4$	$1.7 \times 10^{10}$	p.r.; P.b.k.; obs. radical cation form n. ( $\sim 50\%$ OH addn.); same rate constant by c.k. with SCN ^	751089