No.	Compound name	Reaction equation	рН	Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$	Comments	Reference
221	Samarium(II) Ion	$. OH + Sm^{2+} \longrightarrow OH^{-} + Sm^{3+}$	3-6	6.1×10^9 6.2×10^9	Average of 2 values. p.r.; D.k. (Sm ²⁺ formed in 5×10^{-2} mol L ⁻¹ Sm ³⁺ soln.)	731084
				6×10^9	p.r.; D.k. (Sm^{2+}) in Ar-satd. soln. contg. $Sm(III)$ sulfate.	720065