No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
328	Neptunate(VI) Ion	$e_{aq}^- + Np(VI) \longrightarrow Np(V)$	~ 13	2.3×10^{10}	p.r.; D.k. at 650 nm; $k=1.3,1.3,2.0,1.1,0.83,0.68 \times 10^{10}$ at 1.0, 2.0, 5.0, 7.5, 10.0, 12.5 mol L ⁻¹ OH ⁻ , resp.	731123