

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
1487	Trimethyl phosphate	$\text{e}_{\text{aq}}^- + (\text{CH}_3\text{O})_3\text{P}(\text{O}) \longrightarrow \cdot\text{CH}_3 + (\text{CH}_3\text{O})_2\text{PO}_2^-$		$2 \times 10^5$	p.r.	723008