No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
398	Butadiene	. OH + H ₂ C=CHCH=CH ₂ \longrightarrow	7.	0×10^9	p.r.; C.k.; obs. formn. of ${\rm I}_2^-$ at 400 nm; rel. to $k(.{\rm OH} + {\rm I}^-).$	670041