No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
99	Manganese(II) Ions	$H^{\bullet} + Mn^{2+} \longrightarrow H^{+} + Mn^{+}$	<0	6.6×10^8	p.r.; P.b.k. at 290 nm in 6 mol L^{-1} H_2SO_4 ; extrapolated from measurements at $\sim 200-250$ K.	751197