No.	Compound name	Reaction equation	рН	Rate constant $(L \text{mol}^{-1} \text{s}^{-1})$	Comments	Reference
711	Glycine, conjugate acid	$.OH + H_3N^+CH_2CO_2H$ $H_3N^+CHCO_2H + H_2O$	<i>→</i> 1	1.7×10^7	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$.	650387
		2 . 2 -		2.8-3	1.0×10^7	p.r.; C.k.; $pK_a =$