No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
137	1,4,7,10- Tetraazacyclotridecanenickel(II) Ion	$. \mathrm{OH} + \mathrm{NiL}^{2+} \longrightarrow \mathrm{NiL}^{3+}$	3.3	3.5×10^{9}	p.r.; P.b.k. in soln. contg. $1\text{-}10 \times 10^{-4} \text{ mol L}^{-1}$ NiL(ClO ₄) ₂ .	86A470