

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
1251	Tyrosine, dianion	$\cdot\text{OH} + \text{TyrO}^{2-} \longrightarrow$	11.2	$1.3 \times 10^{10}$	p.r.; C.k.; rel. to $k(\cdot\text{OH} + \text{SCN}^-)$ .	730003