No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
682 2-Bromo-5-nitrothiazole	$\begin{array}{l} e_{\mathrm{aq}}^{} + C_{3} \mathrm{HBrN}_{2} \mathrm{O}_{2} \mathrm{S} \\ C_{3} \mathrm{HBrN}_{2} \mathrm{O}_{2} \mathrm{S}^{-} \end{array}$	$\longrightarrow 7 \qquad 2.0 \times 10^{10}$	p.r.; D.k. at 600 nm in soln. contg. 0.5 mol $\rm L^{-1}$ tert-BuOH, as well as p.b.k.	761075