No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1100	Protoporphyrin IX, dimethyl ester	$.  \mathrm{OH} + \mathrm{C}_{36}\mathrm{H}_{38}\mathrm{N}_4\mathrm{O}_4 \longrightarrow$		$6.0 \times 10^{10}$	Average of 2 values.	
				$6.0 \times 10^{10}$	$\gamma\text{-r.};$ C.k. in air-satd. 2N H <sub>2</sub> SO <sub>4</sub> soln. contg. $1.5\times10^{-3}$ mol L $^{-1}$ tert-BuOH; rel. to $k(.\mathrm{OH}$ + tert-BuOH).	80A392
				$6.1 \times 10^{10}$	$\gamma$ -r.; C.k. in air-satd. 2N H <sub>2</sub> SO <sub>4</sub> soln. contg. $10^{-4}$ mol L <sup>-1</sup> thymine; rel. to $k(.OH + 5-MeU)$ .	80A392