No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
18	1-Hydroxyethyl	$CH_3CH_2OH + e_{aq}^- \longrightarrow C_2H_5OH$	7	6×10^9	$\gamma\text{-r.};$ Estd. from product yields (CH ₃ CHO, H ₂ O ₂) in soln. contg. 0.1 mol L ⁻¹ EtOH and 1.8-2.8 \times 10^{-5} mol L ⁻¹ oxygen.	