No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
79	Hydroxoiron(III) Ion	$H^{\bullet} + FeOH^{2+} \longrightarrow H^{+} + Fe^{2+}$		$0.2 \times 10^9$ $\sim 1 - 2$ 0.3	Average of 2 values. $1.2 \times 10^9$ $1.2 \times 10^9$	γ-r.; C.k. in HC