No.	Compound name	Reaction equation	pH	Rate constant ($L\,mol^{-1}\,s^{-1}$)
1300	Oxalate Ion	$e_{aq}^- + -O_2CCO_2^- \longrightarrow C_2O_4^{3-}$	7.6	3.1×10^7
			4.6×10^7	f.p.; D.k. at 650 nm in soln. contg. 0.0-0.25 mol L^{-1} oxalate ion
			1.7×10^7	p.r.; D.k. at pH 5 and 9 in soln. contg. 10^{-2} mol L^{-1} EtOH; assumed pK_1 = 1.25 and p