No.	Compound name	Reaction equation	рН	Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$	Comments	Reference
439	Methyl methacrylate	$H \cdot + H_2C = C(CH_3)CO_2CH_3 \longrightarrow addn \cdot$	1	$7.2 \times 10^8$	p.r.; esr; Calcd. from time profile of H in deaerated soln. contg. various solute concns. and $10^{-2}$ mol $\rm L^{-1}$ tert-BuOH.	83A153