No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
982	Nicotinate Ion	$. \mathrm{OH} + 3\mathrm{-pyCO_2}^- \longrightarrow 3\mathrm{-py(OH)CO_2}^-$	9.1	2.3×10^9	p.r.; P.b.k.	710582