No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
266	n-Butyric acid	$H' + CH_3CH_2CH_2CO_2H \longrightarrow H_2 + CH_3CH_2CH_2CO_2$	7.4	7.4×10^6	Average of 2 values.	
			1	7.9×10^6	e-r.; esr; Decay of spin polarization, compared with 2-PrOH(7D); rel. to $k(H^{\cdot} + BzOH)$.	710003
			1	6.8×10^{6}	$\gamma\text{-r.};$ C.k. with 2-PrOH(7D); rel. to k(H $^{\circ}$ + BzOH).	710017