No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
89	Mercury(I) Ion dimer	$\mathrm{H}\cdot^{+}\mathrm{Hg_{2}}^{2+}\longrightarrow\mathrm{H}^{+}+\mathrm{Hg_{2}}^{+}$	1.0	4.7×10^{9}	p.r.; P.b.k. at 300 nm; at pH 1.5-1.6 and 0, resp., $k = 4.1 \times 10^9$ and 4.6×10^9 .	79A063