No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1187	1,1,3,3-Tetramethylurea	$.  \mathrm{OH} + (\mathrm{CH_3})_2 \mathrm{NCON}(\mathrm{CH_3})_2 \longrightarrow$	~	$8 \times 10^9$	therm.; C.k.; Obs. ethylene generation in $\rm H_2O_2$ -methional, and methane generation in $\rm H_2O_2$ -DMSO. Rel. to $k(.\rm OH + MeOH) = 8.5 \times 10^5$ , $k(.\rm OH + PhOH) = 1.4 \times 10^{10}$ as well as EtOH, BuOH, BzO-and DMSO.	78M373