No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
776	Cumene hydroperoxide	$e_{aq}^- + C_6H_5C(CH_3)_2O_2H \longrightarrow OH^- + C_6H_5C(CH_3)_2O^-$		4.2×10^9	Average of 2 values.	
			7, 9	4.4×10^9	p.r.; D.k. at 600 nm in soln. contg. 1 mol ${\rm L}^{-1}$ tert-BuOH.	84A160
				4×10^9	p.r.; D.k. at 650 nm in deaer ated soln. contg. 0.5 mol $\rm L^{-1}$ $tert\text{-BuOH}.$	81A374