No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments
647	Ethylenediaminetetraacetate Ion	$\cdot\,\mathrm{OH} + \mathrm{H_2EDTA}^{2-} \longrightarrow \mathrm{H\ abstr.}$	4.0	4.0×10^8	p.r.; P.b.k. as well as c.k. with SCN^-.
		$\cdot\,\mathrm{OH} + \mathrm{HEDTA}^{3-} \longrightarrow \mathrm{H\ abstr.}$	9.0	2.0×10^9	