No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
358 Benzhydrylammonium Ion	$\begin{array}{l} .\mathrm{OH}  +  \left(\mathrm{C_{6}H_{5}}\right)_{2}\mathrm{CHNH_{3}}^{+} \\ \mathrm{HOC_{6}H_{5}CH(C_{6}H_{5})NH_{3}}^{+} \end{array}$	<i>─</i> 7	$6.9 \times 10^9$	p.r.; P.b.k. at 320 nm in $N_2O$ -satd. soln.	86A410