

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
369	Fumaric acid	$\text{H}^\bullet + \text{HO}_2\text{CCH}=\text{CHCO}_2\text{H} \longrightarrow \text{HO}_2\text{CCHCH}_2\text{CO}_2\text{H}$	0.7	7.0×10^8	p.r.; P.b.k. in soln. contg. $\sim 1 \text{ mol L}^{-1}$ tert-BuOH.	730097
				1	8×10^8	e-r.; esr; Decay