No.	Compound name	Reaction equation	pH	Rate constant ($\mathrm{L\,mol^{-1}\,s^{-1}}$)	Comments	Reference
314	Fluoronickel(II) Ion	$e_{aq}^{-} + NiF^{+} \longrightarrow$	8.5	$< 1.2 \times 10^{10}$	p.r.; D.k.; counterion F^{-}, $I = 10^{-1}$, $k_{obs} = 7.2 \times 10^{9}$. The real value for NiF^{+} is lower since the soln. contained 12% Ni^{2+}; k cor. for I.	680295