

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
866	Dimethyl fumarate	$\text{e}_{\text{aq}}^- + \text{CH}_3\text{O}_2\text{CCH}=\text{CHCO}_2\text{CH}_3 \longrightarrow$ $[(\text{CHCO}_2\text{CH}_3)_2]'^-$	9.2	$3.3 \times 10^{10}$	p.r.; D.k. at 700 nm in soln. contg. ~0.5 mol L <sup>-1</sup> tert-BuOH.	730097