

No.	Compound name	Reaction equation	pH
295	Nitrate ion	$\text{e}_{\text{aq}}^- + \text{NO}_3^- \longrightarrow \text{NO}_3^{2-}$	1.1×10^{10}
			8.6×10^9
6.4		1.0×10^{10}	p.r.; D.k.; k also detd. at 6.4 kbar (6.4×10^8 N/m ²), counterion Na ⁺ .
\{\}sim7		1.0×10^{10}	p.r.; D.k. at 550 nm, soln. contg. 10^{-3} mol L ⁻¹ EtOH and 0.01, 0.1 and 1 mol L ⁻¹ NaNO ₃ ; k cor. fo
			9×10^9
			9.3×10^9