No. Compound nam	ne Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
247 1-Bromo-1-chlor trifluoroethane	ro-2,2,2- $\mathrm{H}^{+}\mathrm{CF_{3}CHClBr}$ -	\longrightarrow 1	3.8×10^{8}	p.r.; C.k. in soln. contg. 10% $tert$ -BuOH; rel. to $k(H\cdot +Q)$.	83A195