No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
138	1,4,8,11- Tetraazacyclotetradecanenickel(II ion	$\begin{array}{ccc} .\operatorname{OH} + \operatorname{Ni}(\operatorname{cyclam})^{2+} & \longrightarrow & \operatorname{OH}^{-} \\) \operatorname{Ni}(\operatorname{cyclam})^{3+} \end{array}$	+ 6	5×10^9	p.r.; P.b.k. at 545 nm in soln. contg. 0.3 mol L ⁻¹ ClO ₄ ⁻ ; at pH 3.1 $k = \sim 2 \times 10^9$; $pK = 4$.	80A350