

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
155	Nitrilotriacetonickelate(II) ion	$\cdot\text{OH} + \text{NiNTA}^- \longrightarrow \text{H}_2\text{O} + \text{Ni}[(\text{-O}_2\text{CCH}_2)_2\text{NCHCO}_2^-]$		$1.3 \times 10^9$	Average of 2 values.	
			7	$1.2 \times 10^9$	p.r.; P.b.k. ( $\lambda_{\text{max}} = 290 \text{ nm}$ ).	80A194
			4-9	$1.4 \times 10^9$	p.r.; P.b.k.	78A436