No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
737	Chlorodifluoromethane	$e_{aq}^- + CHClF_2 \longrightarrow Cl^- + \cdot CHF_2$	2	$.9 \times 10^9$	$\gamma\text{-r.};$ C.k.; obs. G(Cl^-) relative to N2O, NO_3^- or SF_6; reference rates not given.	760157