No.	Compound name	Reaction equation	рН	Rate constant $(L  \text{mol}^{-1}  \text{s}^{-1})$	Comments	Reference
1009	4-Nitrophenyl-N-tert-butyl nitrone	$\begin{array}{l} \mathrm{.OH} \hspace{0.1cm} + \hspace{0.1cm} 4\mathrm{-NO_2}\mathrm{-PBN} \\ 4\mathrm{-NO_2}\mathrm{-PBN(OH)} \end{array}$	<del>&gt;</del>	$7.4 \times 10^9$	p.r.; C.k.; rel. to $k(.OH + SCN-)$ .	82A184