

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
55	3-Butenoate Ion	$\text{'O}^- + \text{CH}_2=\text{CHCH}_2\text{CO}_2^- \longrightarrow \text{OH}^- +$ 14		6.5×10^8	p.r.; C.k.; rel. to $k(\text{'O}^- + 3\text{-HX})$.	751003