No.	Compound name	Reaction equation pH	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1090	2-Propanol	$\begin{array}{llllllllllllllllllllllllllllllllllll$	$1.9\times10^9$	Selected value.	
			$2.3 \times 10^9$	p.r.; C.k.; also measurements at 39, 59, and 79°C; rel. to $k(.OH + SCN^{-})$ .	84A349
		6	$1.9 \times 10^9$	p.r.; C.k.; obs. ABTS.+ formn. at 415 nm; rel. to $k(.OH + ABTS)$ .	82A196
	nat.		$2.3 \times 10^9$	p.r.; C.k.; rel. to $k(.OH + Fe(CN)_6^{4-})$ .	710578
	2-10		$1.9\times10^9$	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$ .	680316
	7		$1.9 \times 10^9$	p.r.; C.k.; rel. to $k(.OH + I^{-})$ .	650010; 670041