No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
1007	4-Nitrophenol	$\begin{array}{l} .\mathrm{OH} \ + \ 4\text{-}\mathrm{O}_{2}\mathrm{NC}_{6}\mathrm{H}_{4}\mathrm{OH} \\ 4\text{-}\mathrm{HOC}_{6}\mathrm{H}_{4}\mathrm{NO}_{3}^{-} + \mathrm{H}^{+} \end{array}$	\longrightarrow 7	3.8×10^9	p.r.; P.b.k. at 290 nm, as well as d.k. at 400 nm; $pK_a=7.14$.	680303