No.	Compound name	Reaction equation	$\mathbf{p}$	$ \begin{array}{cc} \text{Rate constant} \\ (\text{L mol}^{-1}  \text{s}^{-1}) \end{array} $	Comments	Reference
1328	Phenylboric acid	$e_{aq}^{-} + C_6H_5B(OH)_2$ $[C_6H_5B(OH)_2]^{-}$	<del></del>	$1.8 \times 10^{6}$	p.r.; Transient obs. at 297 nm in neutral soln. decreases to ca. one-half at pH 5 and is not obs. in acid soln.; $pK_a = 8.84$ ; $k$ estd. by c.k.; rel. to $k(e_{aq}^- + H^+)$ .	