No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1134	Malonate ion	$e_{\rm aq}^- + CH_2(CO_2^-)_2 \longrightarrow$	1×10^7		γ -r.; C.k.; calcd. from pH study; $k_{obs} = 1.4 \times 10^7$ at pH 7.2; rel. to $k(e_{aq}^- + BrPhOH)$.	720027