No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
549 Trichlorofluoromethane	$\mathrm{H}\cdot^{+}\mathrm{CCl_{3}F}\longrightarrow$	$1 1.6 \times 10^6$	e-; $esr; Decay of spin polarization, compared with 2 PrOH (700); single means a BzOH).$	