No.	Compound name	Reaction equation	pН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments
358	${\it Tetrachloropalladate}({\it II}) {\it \ Ion}$	$e_{aq}^- + PdCl_4^{2-} \longrightarrow PdCl_4^{3-}$		$1.2 \times 10^{10}$ $1.1 \times 10^{10}$	Average of 2 values. p.r.; Soln. contg. 0.01 mol L <sup>-</sup>
		7.1	$1.2\times10^{10}$	p.r.; D.k.; counterion $K^+$ ; soln. contg. 0.1 mol $L^{-1}$ $Cl^-$ .	650047