No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
300	${\it Tetraamminenickel(II)\ Ion}$	$e_{aq}^- + Ni(NH_3)_4^{2+} \longrightarrow$	11.3	$8.4 \times 10^{9}$	p.r.; D.k. at 575 nm; soln. contains 0.67 mol $L^{-1}$ NH <sub>3</sub> and $10^{-3}$ – $10^{-4}$ mol $L^{-1}$ NiSO <sub>4</sub> ; no. of ligands in complex are unknown, Ni(NH <sub>3</sub> ) $_5^{2+}$ and Ni(NH <sub>3</sub> ) $_6^{3+}$ are also assumed to be present.	720460