No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
134	Nickel(I) Ion	$.\mathrm{OH} + \mathrm{Ni}^{+} \longrightarrow \mathrm{OH}^{-} + \mathrm{Ni}^{2+}$			$2\times10^{10}$	p.r.; Estd. from