No.	Compound name	Reaction equation		pH Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$	Commonts	Reference
84	Tris(1,10-phenanthroline) $iron(III)$ Ion	$H \cdot Fe(phen)_3^{3+} = Fe(phen)_3^{2+}$	H^+ +	4.5×10^9	Average of 2 values.	