No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
686	2-Bromophenoxide Ion	$e_{aq}^- + 2 - BrC_6H_4O^- \longrightarrow Br^- +$ ${}^{\bullet}C_6H_4O^-$		$2.1 \times 10^9$	Average of 2 values.	
			11.5	$2.3\times10^{9}$	p.r.; D.k. at 470 nm (as well as p.b.k. at 400 nm) in N <sub>2</sub> -purged soln. contg. $4\times10^{-4}$ mol L <sup>-1</sup> 2-bromophenol and 0.2 mol L <sup>-1</sup> tert-BuOH.	761068
			$\sim 11$	$1.9\times10^9$	p.r.; D.k.	640138