No.	Compound name	Reaction equation	рН	Rate constant $(L \text{mol}^{-1} \text{s}^{-1})$	Comments	Reference
4	Bromide ion	$H^{\bullet} + Br^{-} \longrightarrow HBr^{-}$	~ 7 2.8×10^7		γ -r.; C.k.; obs. $G(H_2)$; rel. to $k(H^{\bullet} + HCO_2^{-})$.	720264