No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
350	Lead(II) Ion	$e_{aq}^- + Pb^{2+} \longrightarrow Pb^+$	3.5	$5 \times 10^{10}$	$3.8 \times 10^{10}$ p.r.; D.k. at 578 nm in soln. contg. $10^{-2}$ mol $l^{-1}$ MeOH; pK 7.8, 9.4, 10.8	Average of 3 val 78A206
	7			$9 \times 10^{10}$ $9 \times 10^{10}$	p.r.; D.k. p.r.; D.k.; counterion $ClO_4^-$	650044 650047