No.	Compound name	Reaction equation	рН	Rate constant $(L \text{mol}^{-1} \text{s}^{-1})$	Comments	Reference
884	Methoxyacetate ion	$. \mathrm{OH} + \mathrm{CH_3OCH_2CO_2}^- \longrightarrow$	9	6.1×10^8	γ -r.; C.k. with RNO; rel. to $k(.OH + EtOH)$.	660423