No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
234	$\label{eq:tris} \begin{split} & \operatorname{Tris}(\text{ethylenediamine}) \operatorname{mercury}(\operatorname{II}) \\ & \operatorname{Ion} \end{split}$	$e_{\rm aq}^- + {\rm Hg(en)_3}^{2+} \longrightarrow$	11.2	1.6×10^{10}	p.r.; D.k. at 575 nm; counterion Cl $^-$; soln. contains 2×10^{-2} mol L $^{-1}$ ethylenediamine-diHCl.	690277