

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
235	Hydrogen xenate(VI) Ion	$\cdot\text{OH} + \text{HXeO}_4^- \longrightarrow \text{H}^+ + \text{HXeO}_6^{2-}$	11-13	$< 5 \times 10^7$	p.r.; P.b.k. at 320 or 600 nm; assuming $pK(\cdot\text{OH}) = 11.9$.	82A160