No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	deference
228 Uranium(III) ion	$. \mathrm{OH} + \mathrm{U}^{3+} \longrightarrow \mathrm{UOH}^{3+}$	$\sim 0.3 - 4.1 \times 10^8$	p.r.; D.k. at 350 nm in He-satd. soln. contg. 0.5 85 mol L^{-1} HClO ₄ , contg. (0.3-2.4) $\times 10^{-1}$ mol L^{-1} U ³⁺ ; inner-sphere mechanism.	5A122