No.	Compound name	Reaction equation	pH	Rate constant $(\mathrm{L\,mol^{-1}\,s^{-1}})$	Comments	Reference
462	Bisoxalatodioxouranate(VI) ion	$e_{aq}^{-} + UO_2(ox)_2{}^{2-} \longrightarrow UO_2(ox)_2{}^{3-}$	5.0	2.7×10^{10}	p.r.; D.k. in soln. contg. 0.1 mol $\mathrm{L^{-1}}$ $tert$-BuOH and 0.01 mol $\mathrm{L^{-1}}$ oxalate and 0.4–7 $\times$ 10^{-4} mol $\mathrm{L^{-1}}$ U(VI).	81A148