

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
988	Nitrilotriacetate ion	$\cdot\text{OH} + \text{NTA}^{3-} \longrightarrow \text{O}_2\text{CCHN}(\text{CH}_2\text{CO}_2^-)_2 + \text{H}_2\text{O}$	9.0	2.5×10^9	p.r.; P.b.k., as well as c.k. with SCN^- , at pH 4.0 $k = 7.5 \times 10^8$; $pK_a = 0.8, 1.8, 2.5, 9.6$.	78A436