No.	Compound name	Reaction equation	рН	Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$	Comments	Reference
708	tert-Butyl hydroperoxide	$e_{aq}^- + (CH_3)_3COOH \longrightarrow OH^- + (CH_3)_3CO\cdot$	8.2	1.6×10^{10}	p.r.; D.k. at 680 nm in N2-satd. soln. contg. 6.5 \times 10^{-2} mol $\rm L^{-1}$ 2-PrOH.	82A451
				6×10^9	p.r.; D.k. at 650 nm in deaer ated soln. contg. 0.5 mol $\rm L^{-1}$ $tert\textsc{-BuOH}.$	81A374