No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
214	Antimony(III) Ion	$OH + SbO^+ \longrightarrow$	0.4	2×10^9	$\gamma\text{-r.};$ Cited from [62G215]; rel. to $k(.\mathrm{OH}+\mathrm{Fe}^{2+}).$	760311