No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
154	$\begin{array}{c} \operatorname{Bis}(\operatorname{iminodiacetato})\operatorname{nickelate}(\operatorname{II}) \\ \operatorname{Ion} \end{array}$	$\begin{array}{ll} .\operatorname{OH} + \operatorname{Ni}(\operatorname{IDA})_2{}^2{}^- & \longrightarrow \operatorname{H}_2\operatorname{O} + \\ \operatorname{Ni}(\operatorname{O}_2\operatorname{CCH}_2\operatorname{NHCHCO}_2)(\operatorname{IDA})^2{}^- \end{array}$	7	1.0×10^9	p.r.; P.b.k.	81A023