No.	Compound name	Reaction equation	рН	Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$
115	Manganese(II) Ion	$.\mathrm{OH} + \mathrm{Mn}^{2+} \longrightarrow \mathrm{MnOH}^{2+}$	3, 9	9.0×10^{7} 2.6×10^{7}
		6.7	2.9×10^7	C.k. in soln. contg. EtOH, CCl ₄ and 0.1-0.3 mol L ⁻¹ Mn ²⁺ ; obs. Mn ^{III} at 255 nm (cor. 3.6×10^7