

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
1170	Tetraethylammonium Ion	$\cdot\text{OH} + (\text{C}_2\text{H}_5)_4\text{N}^+ \longrightarrow$		$1.3 \times 10^8$	p.r.; C.k.; ave. of values rel. to $k(\cdot\text{OH} + \text{SCN}^-) = 1.1 \times 10^{10}$ and $k(\cdot\text{OH} + \text{Fe}(\text{CN})_4^-) 1.1 \times 10^{10}$	80A346