No. Compound name	Reaction equation	pH Rate constan $(L \text{ mol}^{-1} \text{ s}^{-1})$. Comments	Reference
167 Hypophosphite ion	$. \mathrm{OH} + \mathrm{H_2PO_2}^- \longrightarrow \mathrm{OH}^- + \mathrm{H_2PO_2}^\bullet$	$10.7 1.8 \times 10^9$	p.r.; C.k.; rel. to $k(.OH + CO_3^{2-})$.	650190