No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
282	2-Chlorophenol	$H^{\bullet} + ClC_6H_4OH \longrightarrow 2\text{-HOC}_6H_5Cl$	2.0	$1.5 \times 10^9$	p.r.; P.b.k. at 340 nm in deoxygenated soln.; contg. $1-10\times 10^{-3}$ mol $\rm L^{-1}$ substrate and 0.5-2.5 mol $\rm L^{-1}$ tert-BuOH.	86A463