No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
63	${\bf Pentacyanonnitrosylchromate(II)} \\ {\bf Ion}$	$. OH + Cr(CN)_5 NO^{3-} \longrightarrow$	7	$7.9 \times 10^9$	$\gamma\text{-r.};$ C.k.; assumed $k(.\mathrm{OH}+\mathrm{CN}^-)=3.0\times10^9;$ rel. to $k(.\mathrm{OH}+\mathrm{RNO}).$	690531