

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
1130	Pyruvate Ion	$\cdot\text{OH} + \text{CH}_3\text{COCO}_2^- \longrightarrow$	9	3.1×10^7	γ -r.; C.k.; based on $k(\cdot\text{OH} + \text{RNO}) = k(\cdot\text{OH} + \text{Fe}(\text{CN})_6^{4-})$; rel. to $k(\cdot\text{OH} + \text{RNO})$.	670555