No.	Compound name	Reaction equation	рН	Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$	Comments	Reference
1263	Vinylidene chloride	$\cdot \operatorname{OH} + \operatorname{H}_2\operatorname{C} = \operatorname{CCl}_2 \longrightarrow \operatorname{CH}_2\operatorname{OH} \cdot \operatorname{CCl}_2$	$\sim 6.5$	$6.8 \times 10^9$	p.r.; C.k.; rel. to $k(\cdot OH + SCN^{-})$	710709