

No.	Compound name	Reaction equation	pH	Rate constant (L mol ⁻¹ s ⁻¹)	Comments	Reference
830	o-Iodohippurate Ion	$\cdot\text{OH} + 2\text{-IC}_6\text{H}_4\text{CONHCH}_2\text{CO}_2^- \longrightarrow$ $2\text{-IC}_6\text{H}_4(\text{OH})\text{CONHCH}_2\text{CO}_2^-$	~ 7	5.3×10^9	p.r.; C.k.; also detd. $k/k'(\cdot\text{OH} + \text{CO}_3^{2-}) = 12.8$; rel. to $k'(\cdot\text{OH} + \text{SCN}^-)$.	761191