No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
104 1,10-Phenanthroline	$O^{}$ + phen \longrightarrow phenOH	$13.6 7.4 \times 10^8$	p.r.; C.k.; rel. to $k(\cdot O^- + HCO_2^-)$.	78A207