No. Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
842 Isopropylamine	$. OH + (CH3)2CHNH2$ $(CH3)2\dot{C}NH2 + H2O$	$\longrightarrow$ 12.0 1	$3 \times 10^{10}$	p.r.; C.k.; value extrapolated from pH study; rel. to $k(.\mathrm{OH} + \mathrm{SCN}^-).$	710585