No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
75	Fluoride Ion	$H^{\bullet} + F^{-} \longrightarrow HF + e_{aq}^{-}$	8.3	$1.0 \times 10^4$	p.r.; Effect of H concn. on $e_{aq}^-$ decay in 1.0 mol L <sup>-1</sup> KF.	670115