

No.	Compound name	Reaction equation	pH	Rate constant (L mol <sup>-1</sup> s <sup>-1</sup> )	Comments	Reference
91	Mercury(II) hydroxide	$\text{H}^\bullet + \text{Hg}(\text{OH})_2 \longrightarrow \text{HgOH} + \text{H}^+$		$2.4 \times 10^9$	Average of 2 values.	
			$\sim 7$	$2.3 \times 10^9$	$\gamma$ -r.; C.k.; $8 \times 10^{-6}$ mol L <sup>-1</sup> HgCl <sub>2</sub> and $2 \times 10^{-5}$ mol L <sup>-1</sup> HCOONa; rel. to $k(\text{H}^\bullet + \text{HCO}_2^\bullet)$ .	660616
			$\sim 7$	$2.0 \times 10^9$	$\gamma$ -r.; C.k.; error in reported ratio ( $1.5 \times 10^3$ ) noted in [660616]; rel. to $k(\text{H}^\bullet + \text{EtOH})$ .	650192