No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
424 Antimonate(V) Ion	$e_{aq}^- + SbO_3^- \longrightarrow$	1.2×10^{10} $11.0 1.3 \times 10^{10}$ $11.0 1.2 \times 10^{10}$	Average of 2 values. p.r.; D.k.; counterion K^+ ; $I = 10^{-3}$. p.r.; D.k.	680295 650047