No.	Compound name	Reaction equation	рН	Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$	Comments	Reference
136	o-Xylene	$O^{-} + C_6H_4(CH_3)_2 \longrightarrow OH^- + 2-CH_3C_6H_4CH_2$	13	$1.8 \times 10^{9}$	p.r.; P.b.k. at 261 nm in N <sub>2</sub> O-satd. soln. contg. 0.5 mol $\rm L^{-1}$ NaOH.	751009