No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
46 Chlorocadmium(II) Ion	$e_{aq}^- + CdCl^+ \longrightarrow$	6.8 $1.1 \times 10^{10}$	p.r.; D.k.; contains 1.0 mol $L^{-1}$ $Cl^{-}$ as well as $CdCl_2(H_2O)_2$ and $CdCl_3(H_2O)^{-}$ , counterion $SO_4^{2-}$ .	650047