No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
13	Bromite ion	$^{\circ}\mathrm{OH} + \mathrm{BrO_2}^- \longrightarrow \mathrm{OH}^- + \mathrm{BrO_2}^-$	2	$.0 \times 10^9$	Average of 2 values.	