No.	Compound name	Reaction equation	рН	Rate constant $(\operatorname{L}\operatorname{mol}^{-1}\operatorname{s}^{-1})$	Comments	Reference
622	Benzene	$e_{aq}^- + PhH \longrightarrow [C_6H_6]^-$	9-11.5	$9.0\times10^6$	p.r.; D.k. at 720 nm in Ar-satd. soln. contg. 0.1 mol $\rm L^{-1}$ tert-BuOH and $\rm 10^{-1}$ - $\rm 10^{-2}$ mol $\rm L^{-1}$ benzene.	85A158
			11	$7.2 \times 10^6$	p.r.; D.k. at 600 nm in soln. contg. 2 mol L <sup>-1</sup> tert-BuOH; $k = 9.0 \times 10^6$ in soln. without tert-BuOH.	771012
			11, 13	$1.3 \times 10^7$	p.r.; D.k. at 600 nm in Ar-satd. soln. contg. 0.8 or 2 mol $\rm L^{-1}$ tert-BuOH or 0.6 mol $\rm L^{-1}$ isobutanol.	741090
				$1.3 \times 10^7$	p.r.; D.k. at 600 nm; also reported $k$ in micellar solutions; $k=1.1\times10^8$ in CTAB, $4\times10^6$ in SDS, $6\times10^6$ in Igepal CO-730.	710586