No. Compound name	Reaction equation	pH Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
1364 Pyridazine	$e_{aq}^{-} + C_4 H_4 N_2 \longrightarrow$	$7.0 2.2 \times 10^{10}$	p.r.; D.k. at 700 nm in Ar-satd. soln. contg. $tert$ -BuOH.	741127