No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
225	Titanium(III) Ions	$.\mathrm{OH} + \mathrm{Ti}^{3+} \longrightarrow \mathrm{OH}^{-} + \mathrm{Ti}^{4+}$	1	1.2×10^9	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$.	72G240