No.	Compound name	Reaction equation	рН	Rate constant $(L \text{ mol}^{-1} \text{ s}^{-1})$	Comments	Reference
130	Hydroxylammonium Ion	$.\mathrm{OH} + \mathrm{NH_3OH}^+ \longrightarrow .\mathrm{NH_2OH} + \mathrm{H_2O}$	4	$< 5.0 \times 10^{8}$	p.r.; C.k.; rel. to $k(.OH + SCN^{-})$.	710493