No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
218	Hydrogenselenite(IV) Ion	$. OH + HSeO_3^- \longrightarrow SeO_3^{2-} + H_2O$	5-6 6	$1.0 \times 10^9$ $2.9 \times 10^8$	$2.3 \times 10^8$ p.r.; P.b.k. at 420 nm. p.r.; P.b.k. at 430 nm in SeO2 soln.	Average of 2 val 80A335 771173