No.	Compound name	Reaction equation	рН	Rate constant $(L \operatorname{mol}^{-1} \operatorname{s}^{-1})$	Comments	Reference
967	Methyl sulfide Ion	$\cdot \mathrm{OH} + \mathrm{CH_3S^-} \longrightarrow \mathrm{OH}^- + \mathrm{CH_3S} \cdot$	11	6.0×10^{9}	p.r.; P.b.k.; obs. RSSR $^-$ at 410-420 nm; p $K=10.3.$	690553