No.	Compound name	Reaction equation	рН	Rate constant $(L  \text{mol}^{-1}  \text{s}^{-1})$	Comments	Reference
516	Deoxyuridine monophosphate	. OH + dUMP $\longrightarrow$ dUMP $-$ OH	$\sim 7$	$2.9\times10^9$	$\gamma\text{-r.};$ C.k.; obs. $G(\text{inorg. phosphate});$ 6.4% oxidative dephosphorylation; rel. to $k(.\text{OH}+tert\text{-BuOH}).$	753070