



Energy, work and power

Work & Energy

$$W = \Delta E$$

$$W = F \times d$$

Efficiency

$$(\%) \text{Efficiency} = \frac{\text{useful energy output}}{\text{total energy output}} \times 100\%$$

$$(\%) \text{Efficiency} = \frac{\text{useful power output}}{\text{total power output}} \times 100\%$$

Power

$$\text{Power} = \frac{\text{work done}}{\text{time taken}}$$

$$P = \frac{W}{t}$$

$$P = \frac{\Delta E}{t}$$

● Work

- **Work done is the amount of energy transferred**
- When a force moves an object through a distance, work is done on the object and energy is transferred.
- The force does “work” to move the object and energy is transferred mechanically from one store to another.

● Work

$$W = \Delta E$$
$$W = F \times d$$

where

W = work done (J)

ΔE = energy transferred(J)

F = force (N)

d = **distance moved in the direction of the force** (m)



● Efficiency

The efficiency of any device is defined as:

$$(\%) \text{Efficiency} = \frac{\text{useful energy output}}{\text{total energy output}} \times 100\%$$

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- Efficiency is given as **a percentage**, or **a decimal**.
- Total energy output = total energy input because of the conservation of energy.
- The efficiency can never be equal to or higher than 100%, since some energy is always wasted.

● Efficiency

Study question #1:

A toaster transfers 216000 J of energy electrically from the mains. 84000 J of energy is transferred to the bread's thermal energy store. Calculate the efficiency of the toaster.



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$$\text{efficiency} = \frac{\text{useful energy output}}{\text{total energy output}} \times 100\% = \frac{84000}{216000} \times 100\% = 38.888...\% = 39\%$$

● Efficiency

Study question #2:

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$$\text{Useful energy output} = 500 - 420 = 80 \text{ J}$$

$$\text{efficiency} = \frac{\text{useful energy output}}{\text{total enegy output}} \times 100\% = \frac{80}{500} \times 100\% = 16\%$$

● Efficiency

Study question #3:

An electrical kettle has an efficiency of 76%. 2500 J of energy is transferred from the mains to the kettle every second. When the kettle is full, it needs to transfer 418 000 J of energy to the thermal energy store of the water to boil it. How long does a full kettle need to be switched on for in order to boil the water? Use 2 methods.

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Method 1:

$$\begin{aligned} \text{Total thermal energy} &= \frac{418000}{0.76} = 550\,000\text{ J} \\ \frac{550000}{2500} &= 220\text{ s} \end{aligned}$$

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Method 2:

$$\begin{aligned} &\text{Usefully transferred to the thermal energy store of the water} \\ &= 2500 \times 0.76 = 1900\text{J} \end{aligned}$$

$$\frac{418000}{1900} = 220 \text{ s}$$