molsim tutorial

for the v. 0.9 series

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Februaray 2022

1 Introduction

molsim is a GNU Octave/Matlab toolbox for molecular dynamics simulations. molsim supports simulations of

- simple Lennard-Jones systems,
- molecular systems with bond, angle, and torsion potentials,
- confined flow systems, eg., Couette and Poiseuille flows,
- charged systems using shifted force and Wolf methods,
- dissipative particle dynamics systems,
- different ensembles,
- and more ...

molsim is basically a wrapper for the seplib library, which is a light-weight flexible molecular dynamics simulation library written in ISO-C99. seplib is CPU-based and offers shared memory parallization; this parallization is supported by molsim. seplib is also developed and maintained by this author, and the underlying algorithms are based on the books by Allen and Tildesley [1], Frenkel and Smit [2], Rapaport [3], and R. Sadus [4].

In this text

>>

symbolizes the GNU Octave/Matlab command prompt. This

\$

symbolizes the shell prompt.

Example scripts and functions to simulate different systems can be found under the package tests directory. It is highly recommended that the user's project starts from one of these .m-files, and the user then makes the necessary changes.

2 Installation

2.1 GNU Octave

GNU Octave's package manager offers a very easy installation. From the command prompt type on one single line

```
>> pkg install "https://github.com/jesperschmidthansen/molsim/ \
archive/refs/tags/v<version>.tar.gz"
```

to install the package. <version> can be for example 0.9.2 Check contact by

```
>> molsim('hello')
Hello.
```

In case this fails, check the path where molsim is install by

```
>> pkg list molsim
```

If the path is not in your GNU Octave search path add this using the addpath command. You can also download the tar.gz file manually from

```
https://github.com/jesperschmidthansen/molsim
```

and save it in some directory of your choice. From this directory enter GNU Octave and type

```
>> pkg install molsim-<version>.tar.gz
```

2.2 Matlab

From

https://github.com/jesperschmidthansen/seplib/

download and save the current release seplib-<version>.tar.gz in a directory of your choice. Unpack, configure and build the library

- \$ tar zxvf seplib-<version>.tar.gz
- \$ cd seplib
- \$./configure
- \$ make
- \$ cd octave

To build the mex-file enter Matlab

\$ matlab -nodesktop

and run the script buildmex

>> buildmex

Depending on the system this will build a molsim.mex;archtype; file. You can copy this file to a directory in your Matlab search path.

Note: Matlab compatibility is not guarantied. molsim will only be tested against very limited Matlab versions.

3 The basic interface strategy

This tutorial is not meant to introduce molecular dynamics; such introduction can be found in the books listed in the reference list. In brief, the basic idea is to solve the classical equation of motion of an ensemble of interacting particles. In the simplest form this means solving (numerically) Newton's second law

$$\frac{\mathrm{d}\mathbf{r}_i}{\mathrm{d}t} = \mathbf{v}_i \; , \quad \frac{\mathrm{d}\mathbf{p}_i}{\mathrm{d}t} = \mathbf{f}_i \tag{1}$$

where $\mathbf{r}_i, \mathbf{p}_i$ and \mathbf{f}_i are the particle position, momentum and force acting on the particle, respectively. In a standard simulation we solve this set of (non-linear) differential equations by (i) evaluating the forces, and (ii) from this integrate forward in time. The following pseudo code lists the basic idea

Listing 0

The molsim interface seeks to emulate this basic idea, and at the same time give the user coding flexibility and accessibility to the simulation quantities. In general, the molsim interface is on the form

```
molsim(<action>, <specifier>, <arguments>);
```

The action can be any particular action the user wishes to perform, for example, calcforce, integrate, and so on. The action is specified by the second argument; say, 1j specifies that action calcforce should apply the Lennard-Jones interaction. The specifier arguments are given in the final input and can be a single number, string, vector, or a sequence of these.

4 First quick example: The Lennard-Jones liquid

Listing 1 shows a script simulating a standard Lennard-Jones (LJ) system in the micro-canonical ensemble, where number of particles, volume, and total energy is conserved.

Listing 1

```
% Specify the LJ paramters
cutoff = 2.5; epsilon = 1.0; sigma = 1.0; aw=1.0;

% Set init. position and velocities 10x10x10 particles
% in box with lengths 12x12x12. Velocities set to default.
% Configuration stored in start.xyz.
molsim('set', 'lattice', [10 10 10], [12 12 12]);

% Load the configuration file
molsim('load', 'xyz', 'start.xyz');

% Main mol. simulation loop - 10 thousand time steps
```

```
for n=1:10000

% Reset forces etc
molsim('reset');

% Calculate force between particles of type A (default type)
molsim('calcforce', 'lj', 'AA', cutoff, sigma, epsilon, aw);

% Integrate forward in time - use leapfrog alogrithm
molsim('integrate', 'leapfrog');
end

% Free memory allocated
molsim('clear');
```

In Listing 1 no information is printed or saved, and admitted not very useful. Inside the main loop the user can call the print-action

```
if rem(n,100)==0
  molsim('print');
end
```

to print current iteration number, potential energy per particle, kinetic energy per particle, total energy per particle, kinetic temperature, and total momentum to screen every 100 time step.

Information can also be stored into variables for further analysis. For example, to get the system energies and pressure

```
[ekin, epot] = molsim('get', 'energies');
press = molsim('get', 'pressure');
and particle positions and velocities

x = molsim('get', 'positions');
v = molsim('get', 'velocities');
```

4.1 NVT and NPT simulations

Often you will not perform simulations in the micro-canonical ensemble, but under a desired temperature and/or pressure. One way to achieve this with molsim is to use simple relaxation algorithms. To simulate at temperature, say 2.2, you call the action 'thermostate' with specifier 'relax' after the integration step

```
molsim('thermostate', 'relax', 'A', 2.2, 0.01);
```

The last argument is the relaxation parameter; the higher value the faster relaxation. Notice that too large values makes the system unrealistically stiff; the best value is optimized via trail-and-error. There is also a Nosé-Hoover thermostate available with the interface

```
molsim('thermostate', 'nosehoover', 'A', 2.2, 10.0);
```

The last argument is here the thermostat mass and not the relaxation time. Again, you should choose this parameter with care.

To simulate at pressure, say 0.9, you call the action 'barostate' after the integration step,

```
molsim('barostate', 'relax', 0.9, 0.01, 'iso');
```

The choice of relaxation parameter, here 0.01, is again a matter of the specific system. The last argument tells the barostate to do an isotropic compression. If this is left out the barostate works by changing the system box length in the z-direction only (an-isotropic scaling); this is practical when doing sampling as two directions are fixed. You can use the barostate and the thermostate in the same simulation mimicking an NPT system.

5 The molsim force field

molsim supports simulations of confined, charged, and molecular systems. In general, the molsim force field is defined from the potential function

$$U(\mathbf{r}_i, r_{ij}, \dots) = U_{\text{lattice}} + U_{\text{vWaals}} + U_{\text{coloumb}} + U_{\text{bonds}} + U_{\text{angles}} + U_{\text{torsion}}$$
 (2)

The first term allows for a simulation of a fictitious fixed crystal arrangement, where the particles/atoms are tethered around a pre-set lattice site. The potential function is a harmonic spring type

$$U_{\text{lattice}} = \sum_{\text{sites}} \frac{1}{2} k_0 (\mathbf{r}_i - \mathbf{r}_0)^2 \tag{3}$$

where k_0 is the spring constant, \mathbf{r}_i is the position of particle/atom i, and \mathbf{r}_0 is the lattice site. Using this requires that the crystal sites are set: use molsim('set', 'virtualsites'); to set the current positions as crystal sites.

The short ranged van der Waals pair interaction is given via the standard Lennard-Jones potential

$$U_{\text{vWaals}} = \sum_{i,j \text{ pairs}} 4\epsilon \left[\left(\frac{\sigma}{r_{ij}} \right)^{12} - a_w \left(\frac{\sigma}{r_{ij}} \right)^6 \right]. \tag{4}$$

Here r_{ij} is the particle distance, ϵ and σ define the characteristic energy and length scales. The parameter a_w determines the weight of the attractive second term in the potential function.

The Coulomb potential is

$$U_{\text{coloumb}} = \sum_{i,j \text{ pairs}} \frac{q_i q_j}{r_{ij}}.$$
 (5)

Currently this long ranged interaction is evaluated using approximate shiftedforce or Wolf methods; this can be specified. Note: these two algorithms do not apply to confined systems.

Bonds are model via the harmonic spring potential

$$U_{\text{bonds}} = \sum_{\text{bonds}} \frac{1}{2} k_s (r_{ij} - l_0)^2 \tag{6}$$

 k_s is the spring constant and l_0 is the zero force bond length. Currently molsim does not support rigid bonds.

The angle potential is the cosine squared potential

$$U_{\text{angles}} = \frac{1}{2} \sum_{\text{angles}} k_{\theta} (\cos(\theta) - \cos(\theta_0))^2, \qquad (7)$$

where k_{θ} is the force amplitude, and θ_0 the zero-force angle. See Fig. 1 for the angle definition.

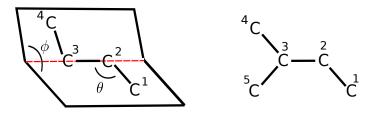


Figure 1: Illustration of the angle and torsion angle.

Finally, the torsion angle potential is the Ryckaert-Belleman potential

$$U_{\text{dihedral}} = \sum_{\text{dihedrals}} \sum_{n=0}^{5} c_n \cos^n(\pi - \phi).$$
 (8)

Here c_n are the six Ryckaert-Belleman coefficients, and ϕ is the torsion angle, see Fig. 1. Two illustrative examples are when only $c_1 \neq 0$:

- 1. If $c_1 > 0$ then the minimum energy torsion angle is $\phi = 0$; this is illustrated in the right-hand figure of a planar molecule with the torsion angle defined by the 1-2-3-5 bonds.
- 2. If $c_1 < 0$ then the minimum energy torsion angle is $\phi = \pi$; this is the torsion angle defined by the 1-2-3-4 bonds.

Importantly, you can have different types of particles with different charges, different bonds, angles, and torsion angles. It is thus possible to simulate mixtures, highly complex molecules, etc.

6 Molecular systems: Toluene

This example shows how to setup a simulation of model liquid toluene. The model of the molecule is a so-called united atomic unit (UAU) model. This means that each carbon group is represented by a single Lennard-Jones particle, thus, the toluene molecule is composed of seven Lennard-Jones particles, six forming the phenyl ring structure (particle indices 2-7) and one representing the methyl group (index 1), see Fig. 2.

To define the molecule geometry (or topology) we need different intramolecular interactions, i.e., bond, angle, and torsion angle potentials. Lennard Jones interactions between carbon groups in same molecule are excluded. The model is further simplified by using only two different bond types and one angle type. There are two different torsion angles, eg., 1-2-3-4 form one type of torsion angle, $\phi=\pi$, whereas 7-2-3-4 form torsion angle with $\phi=0$. We define the molecular model in two files; one with extension .xyz giving the carbon groups' positions for one molecule and one with extension .top defining bonds and angles in the molecule. You can find examples of these two files for different molecules under the resources directory. To setup the entire system (i.e. the ensemble of molecules) we copy the single molecule .xyz and .top to the current directory and use the set action; for example to simulate 500 molecules

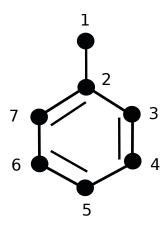


Figure 2: United atomic unit representation of toluene.

```
>> molsim('set', 'molconfig', 'toluene.xyz', 'toluene.top', ... 500, 0.05, 42)
```

The two last arguments are the molecular number density (keep very low initially and compress the system afterwards), and a seed for the random number generator. This generates a system start.xyz file and start.top file that can be loaded.

We exclude the molecular dipole moment, i.e., we do not apply any charges to the system.

We now only need the parameter values for the interaction potentials and we will simply take what is available in the literature [5] and convert them into MD reduced units. Listing 2 shows the resulting script

Listing 2

```
% Simulation parameters
temp0 = 4.969; dens0 = 1.96; dt = 0.001; nloops = 200000;

% Intra-molecular parameters
bondlength_0 = 0.4; bondlength_1 = 0.38; springconstant = 48910;
bondangle = 2.09; angleconstant = 1173;

torsionparam_0 = [0.0, 133.0, 0.0 0.0 0.0];
torsionparam_1 = [0.0, -133.0, 0.0 0.0 0.0];
```

```
% Load positions, set temp, remove intra-molecular pair-interaction etc
molsim('load', 'xyz', 'start.xyz');
molsim('load', 'top', 'start.top');
molsim('set','timestep', dt);
molsim('set', 'temperature', temp0);
molsim('set', 'exclusion', 'molecule');
% Main loop
for n=1:nloops
 molsim('reset')
 molsim('calcforce', 'lj', 'CC', 2.5, 1.0, 1.0, 1.0);
  molsim('calcforce', 'bond', 0, bondlength_0, springconstant);
  molsim('calcforce', 'bond', 1, bondlength_1, springconstant);
 molsim('calcforce', 'angle', 0, bondangle, angleconstant);
 molsim('calcforce', 'torsion', 0, torsionparam_0);
 molsim('calcforce', 'torsion', 1, torsionparam_1);
 molsim('thermostate', 'nosehoover', 'C', temp0, 10.0);
  molsim('integrate', 'leapfrog');
 molsim('compress', dens0);
end
```

Notice that

- molsim('set', 'exclusion', 'molecule'); ensures that van der Waals and Coulomb interactions are excluded if the particles are in same molecule. Exclusion can also be set for bonded particles using the 'bonded' argument.
- For the bond, angle, and torsion specifiers the first argument pertains to the type.

7 Sampling

The user can access the system configuration through the get action and from this perform data analysis via GNU Octave's or Matlab's built-in tools. molsim also offers some run-time data sampling. The different samplers can be initialized before the main loop using the sample action

```
molsim('sample', <sample specifier>, <arguments>);
```

For example, to sample the stress autocorrelation function with 200 sample points and over a sample time span window of 5.0 we write

```
molsim('sample', 'sacf', 200, 5.0);
```

The actual sampling is carried out by the specifier do; inside the main loop there must be a call

```
molsim('sample', 'do');
```

Typically this call is done just after the integration. Check the reference sheet for the list of available samplers.

8 Parallization

molsim offers two types of sheared memory parallization, namely,

- Loop parallization
- Task-block parallization

We here only document the first type; the task-block type will be included in later versions of the tutorial. To use loop parallization simply call the set action with argument omp

```
molsim('set', 'omp', <nthreads>);
```

where nthreads is the number of threads you will use. Typically you do not wish to use more threads than the number of cpu-cores¹. The call is placed anywhere before the main loop. Be aware that depending on your hardware and the particular system the parallization eefficiency quickly drop

¹not even with hyper-threading

as function of number of threads. To explore this we define the speed-up and efficiency by

speedup =
$$\frac{t_{\text{single}}}{t_{\text{parallel}}}$$
 and efficiency = $\frac{t_{\text{single}}}{t_{\text{parallel}}N_{\text{threads}}}$, (9)

where t_{single} is the single thread execution time and t_{parallel} the parallel execution time. The speed-up is plotted in Fig. 3 for a simple Lennard-Jones liquid simulation. The efficiency is also plotted in Fig. 3. The efficiency

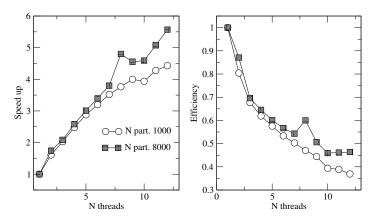


Figure 3: Left: Speed-up as function of number of threads. Right: Efficiency as function of number of threads. The test machine is a 72-core machine, and this result is not valid for a standard desktop.

quickly drops, even on this multi-core machine, and adding more threads (using more cpu cores) is not always optimal.

References

- [1] M. P. Allen and D. J. Tildesley, Computer Simulation of Liquids, (1989).
- [2] D. Frenkel and B. Smit, Understanding Molecular Simulation, (1996).
- [3] D. C. Rapaport, The Art of Molecular Dynamics Simulation, (1995).
- [4] R. J. Sadus, Molecular Simulation of Fluids. Theory, Algorithms and Object-Orientation, (1999).
- [5] J.S. Hansen Where is the hydrodynamic limit? Mol. Sim., 47:1391 (2021).

Reference sheet

Action	Specifier	Arguments	Output
load	xyz top	file name file name	
save		 type names file name 	
set	timestep temperature cutoff omp exclusion temperaturerelax compressionfactor types skin charges lattice molconfig	time step (0.005) temperature (1.0) Max. cut-off (2.5) No. of threads 'bonded' or 'molecule' relaxation time (0.01) compress factor (0.9999) particles types (vector string) buffer-skin neighblist atom charges (vector) 1: array $[N_x, N_y, N_z]$ 2: array $[L_x, L_y, L_z]$ 1: xyz file 2: top file 3: No. molecules 4: Crystal density 5: Random seed	

Action	Specifier	Arguments	Output
get	numbpart box		scalar 3-vector
	energies		2-vector (kin, pot)
	pressure		scalar
	velocities		$N_{\mathrm{part.}} \times 3\text{-matrix}$
	positions		$N_{\mathrm{part.}} \times 3$ -matrix
	forces		$N_{\mathrm{part.}} \times 3\text{-matrix}$
	types		$N_{ m part.}$ -string
	molpositions		$N_{ m mol} imes 3$ -matrix
calcforce	1j	1: part. types	
	- J	2: cutoff	
		$3: \sigma$	
		$4: \epsilon$	
		$5: a_w$	
	bond	1: type	
		2: bond length	
		3: spring constant	
	angle	1: type	
		2: zero angle force	
		3: force constant	
	torsion	1: type	
		2: tors. param	
	lattice	1: part. type	
		2: spring constant	
	dpd	1: part. types	
		2: cutoff	
		3: rep. parameter 4: σ	
		4. 0	

Action	Specifier	Arguments	Output
integrate	leapfrog dpd	λ	
thermostate	relax	 type target temperature relax time 	
	nosehoover	1: type 2: target temperature 3: themostate mass	
barostate	relax	1: target pressure 2: relax time 3: opt. iso	
compress		target density	
add	force tolattice	1: force vector 2: direction (0,1,2) 1: dx (scalar) 2: direction (0,1,2)	
clear			
reset			
print			

Action	Specifier	Arguments	Output
sample	vacf or mvacf	1: sample vector length 2: sample time span	
	sacf or msacf	 sample vector length sample time span 	
	msd	 sample vector length sample time span no. wavevectors 	
	hudmaaammalatiana an	4: particle type	
	hydrocorrelations or	1: sample vector length	
	mhydrocorrelations	2: sample time span 3: no. wavevectors	
	profiles	1: particle type	
	profitob	2: sample vector length	
		3: sample interval	
	do	o	
task	lj	 part. types cutoff σ 	
		$4: \epsilon$ $5: a_w$	
		6: block id	
	bond	1: type	
		2: bond length	
		3: spring constant	
		4: block id	
	angle	1: type	
		2: zero angle force	
		3: force constant	
		4: block id	
	torsion	1: type	
		2: tors. param	
		3: block id	
	coulomb	1: cutoff	
		3: block id	
	do	no. blocks	