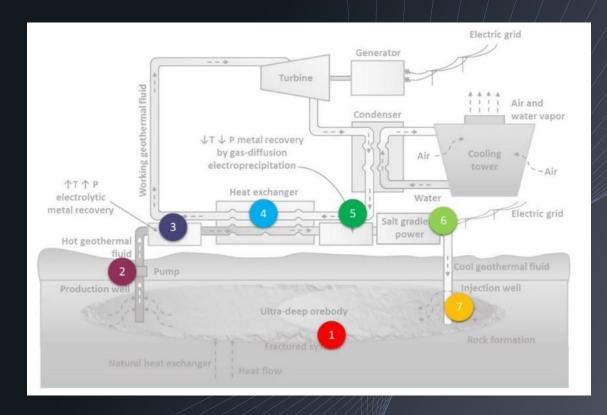
# Modeling Enhanced Geothermal Systems (EGS) for Energy and Metal Recovery

Samy Palaniappan, July 2019

## Objective

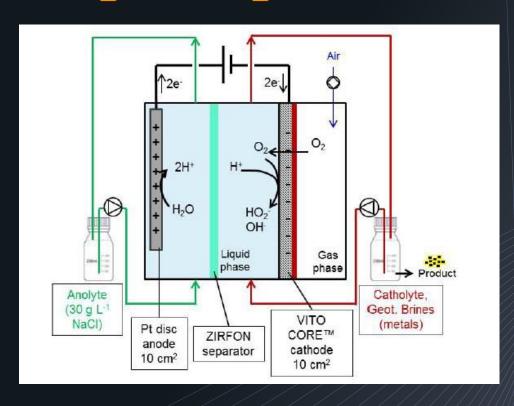
Creating a model pipeline to analyze data for valorizing metals and energy

# System Modeling



Focussed on processes 3 to 6

#### Electro-precipitation



#### Model Approach

Individual System
-Model Training
Logistic regression
was used to model
the system
components, using
experimental data.

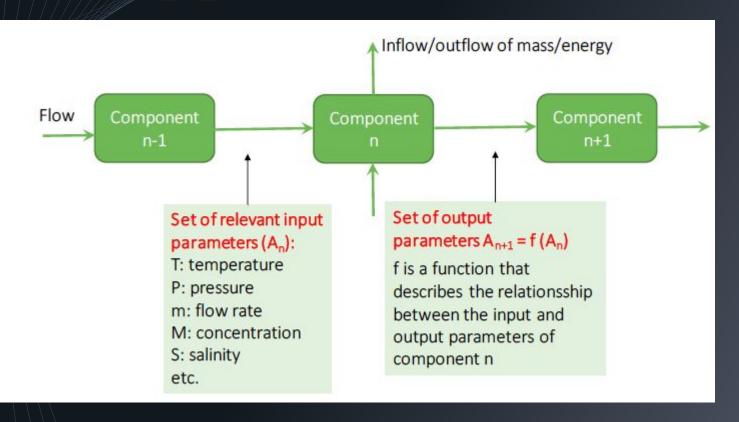
Testing Data
GenerationStochastic Model

Monte Carlo method was used to generate input data from a given distribution.

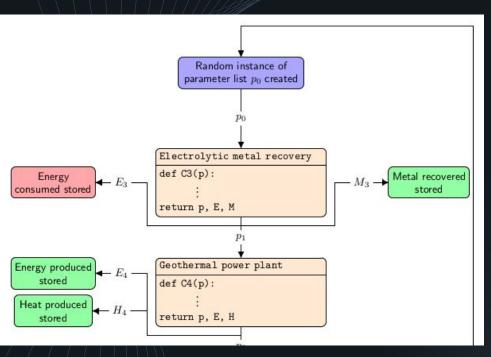
#### **Test Data**

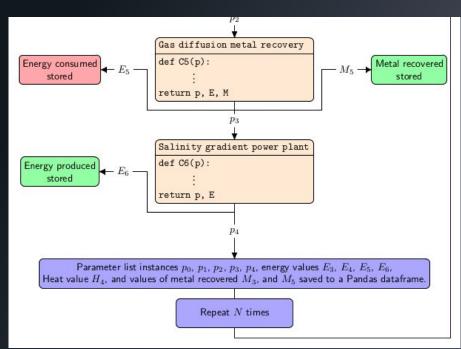
The model was tested with real brine data from 1. Mol, Belgium, Reykjavik, Iceland, Landau, Germany, Cornwall, UK, and Pannoni, Romania.

#### Model Approach

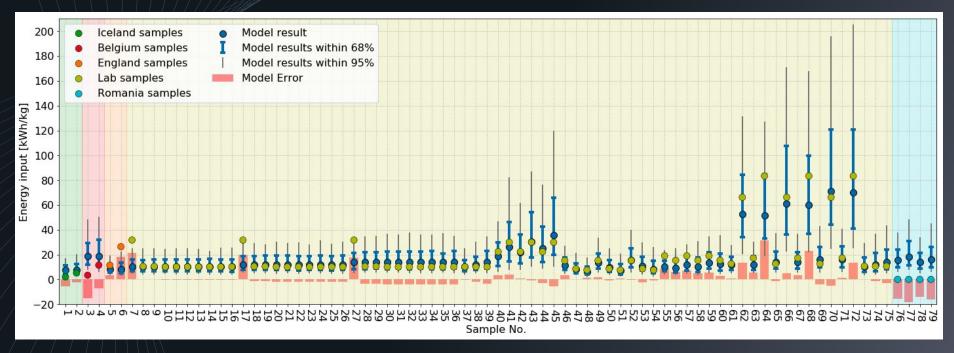


### Model Algorithm

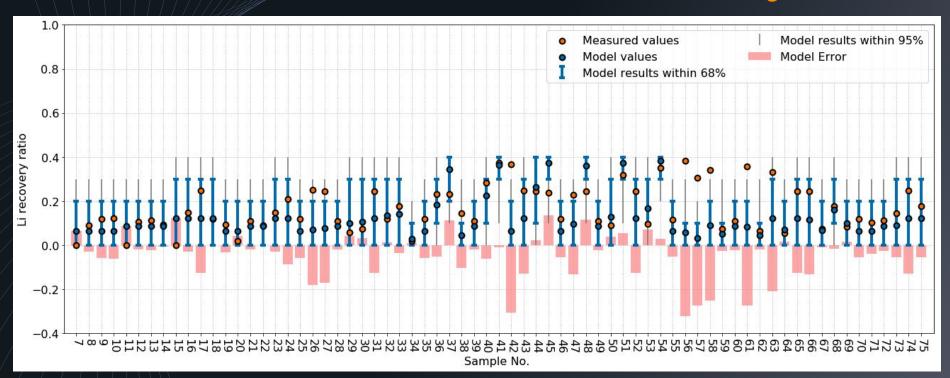




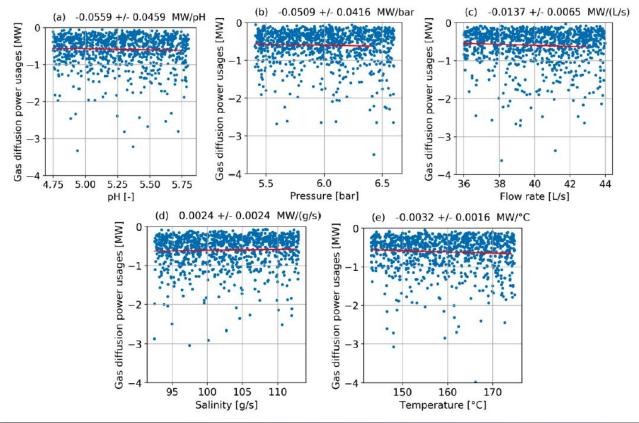
### Model Results - Energy (EP)



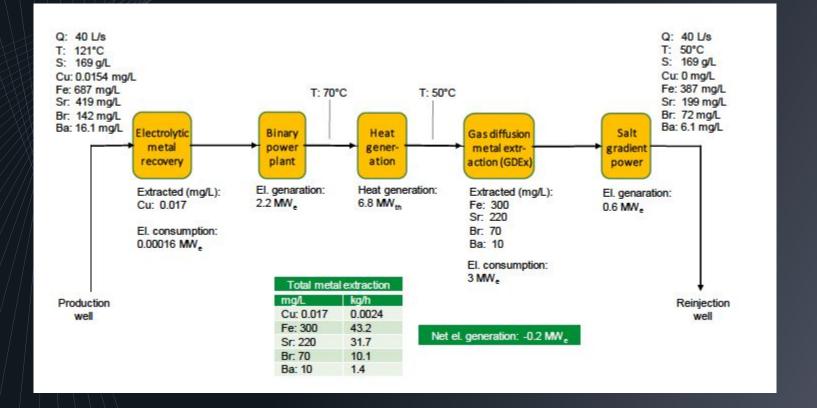
#### Model Results - Li Recovery



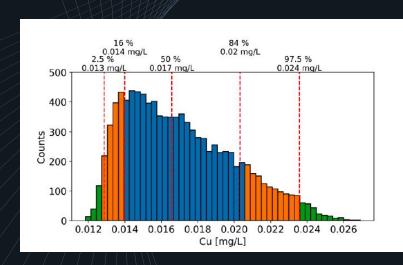
#### Model Results - Sensitivity study

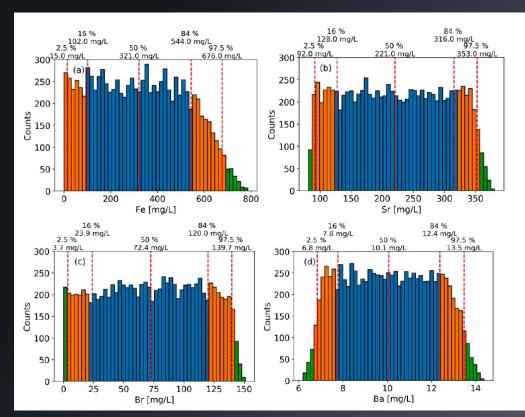


#### Results - Mol, Belgium

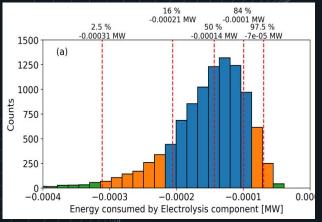


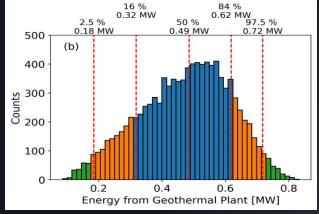
#### Results - Metals - Mol, Belgium

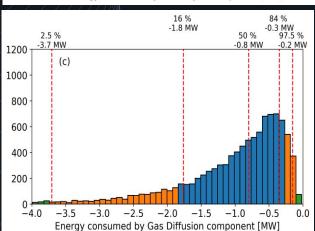


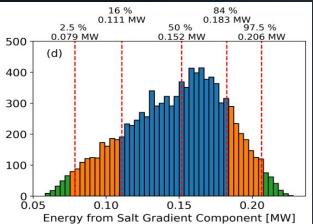


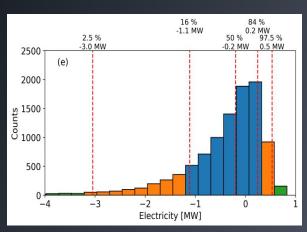
### Results - Energy - Mol, Belgium











#### Results - Summary

|         | Reykjanes | Landau | Balmatt | Cornwall | Romania |  |  |
|---------|-----------|--------|---------|----------|---------|--|--|
| Q (L/s) | 100       | 40     | 40      | 40       | 55      |  |  |
| T /°C)  | 150       | 123    | 121     | 175      | 140     |  |  |
| S (g/L) | 35        | 103    | 169     | 10.8     | 10.8    |  |  |

| ( / / / / / / XXXII            |       |       |            |        |        |  |  |  |  |
|--------------------------------|-------|-------|------------|--------|--------|--|--|--|--|
| El. generation MW <sub>e</sub> |       |       |            |        |        |  |  |  |  |
| Binary plant                   | 3.6   | 1.3   | 2.2        | 2.3    | 1.6    |  |  |  |  |
| Salt gradient plant            | 0.083 | 0.084 | 0.6        | 0.008  | 0.01   |  |  |  |  |
| Electrolysis comp.             | -0.3  | -0.12 | -0.00016   | -0.005 | -0.003 |  |  |  |  |
| Gas diffusion comp.            | -0.3  | -0.6  | -3         | -0.08  | -6     |  |  |  |  |
| Net el. generation             | 3.1   | 0.7   | -0.2       | 2.2    | -4.4   |  |  |  |  |
| Heatgangration                 | ,     | 2     | <i>c</i> o | 2.4    | 77     |  |  |  |  |

| Metal extracted   | mg/L  | kg/h  | mg/L  | kg/h  | mg/L  | kg/h  | mg/L | kg/h  | mg/L | kg/h |
|-------------------|-------|-------|-------|-------|-------|-------|------|-------|------|------|
| Cu – Copper       | 17    | 6.12  | 0.038 | 0.005 | 0.017 | 0.002 | 0.4  | 0.058 | 0.2  | 0.04 |
| As – Arsenic      | 0.11  | 0.039 | 9.7   | 1.4   |       |       |      |       |      |      |
| Ag –Silver        | 0.06  | 0.022 |       |       |       |       |      |       |      |      |
| Sb – Antimony     | 0.013 | 0.005 |       |       |       |       |      |       |      |      |
| Fe- Iron          | 40    | 14.4  | 40    | 5.8   | 300   | 43.2  |      |       | 0.7  | 0.14 |
| Br – Bromine      | 30    | 10.8  | 100   | 14.4  | 70    | 10.1  |      |       |      |      |
| Zn – Zink         | 5     | 1.8   |       |       |       |       |      |       | 1.3  | 0.26 |
| Sr – Strontium    |       |       | 230   | 33.1  | 220   | 31.7  | 13   | 1.9   | 200  | 40   |
| Li – Lithium      |       |       | 50    | 7.2   |       |       | 6    | 0.86  |      |      |
| Ba – Barium       |       |       |       |       | 10    | 1.4   |      |       | 5    | 1    |
| B – Boron         |       |       |       |       |       |       | 4    | 0.58  |      |      |
| Mn - Manganese    |       |       |       |       |       |       | 3.4  | 0.49  |      |      |
| Total metal extr. | 92    | 33    | 430   | 62    | 600   | 86    | 27   | 3.9   | 207  | 41   |

#### Conclusion

- 1. A model pipeline was constructed, which will output the metals recovered, and the net energy for power plant, upon putting in details about a geothermal fluid can be input
- 2. HTHP electrodeposition is feasible if brine is rich in noble metal content It consumes very little power for 1 kg of metal removed.
- 3. Electro-precipitation process consumes a lot of energy. However, this can be improved by better cell design.
- 4. Salinity gradient / Reverse electrodialysis produces enough energy to sustain Electrochemical process.
- 5. Among the brines studied, Reykjavik Iceland site is most suitable for pilot plant construction (metal value & net energy)

#### **Future Direction**

- 1. Improve cell design for EP.
- 2. Collect more data for SGP-RED and ED.
- 3. Try bootstrapped / ensemble models