Quantum Mechanics

Here we will simply introduce the underlying ideas of quantum theory and the wave-particle nature of matter. First we shall discuss the requirement of quantum mechanice. In order to do that we shall first discuss with the phenomena of black body radiation and its characteristis.

Black Body Radiation Spectrum

A black body absorbs all radialions incident upon it. It lechans as a perfect radiator when it is healed.

The radiation emitted by blackbody depends only on its demperature not on the nature of the material.

The experimetal spubrum of black body radiation and wavelength of black body radiation at different lemperature as shown below Experimental Observations Ex

1. The increase in temperature results dicrease in wavelength Im at which energy emitted is maximum.

AmT = const = 0.2896 was given by Wein's and is known as when's displacement-law.

2. Intensity of radiation intreases with intrease of wavelength at particular temperature and

7377277, 7377277, 7377277, 74737, 747, 7472, 7472

belones maximum at a particular wavelengt and then after intensity of radiation decreases with increase of wavelength. 3. In the radiation spectrum of a blackbody energy is not uniformly distribuled-4. In crease in l'emperature results un au uncrease in energy emission for all wavelengthe wavelengths. 5. At a particular lomperature for the range of wavelength considered, total energy emitted by the blackbody is represented by the area under each circle. Stefan's Law emitted from a swrface is proportional to the fourth power of its extrahile lump. If E is the radiant heat energy emitted from unit area in one selond and I Vien E = 074, Where of is constant of proportionality and her the Value 5.67x108 watt per moter 2 per K4 throsed explanetor of blackbody radiation Wren's radiation law W. wien proposed an empirical relationship between Ex and I for a lemp Tof black body Ex(T) dx = C; f(xT) dx, G is const. and f (AT) is a function of product AT.