# Notes on Electromagnetism

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# **Class Information**

#### Recommended Textbooks

- The Feynman Lectures on Physics
- Picasso Lezioni di fisica

### 1 Coulomb's Law

Coulomb's law describes the electrostatic force between two charged particles. The force  $\mathbf{F}_{12}$  exerted by a charge  $q_1$  on a charge  $q_2$  is given by:

$$\mathbf{F}_{12} = k \frac{q_1 q_2}{R^2} \hat{\mathbf{R}}_{12} \tag{1}$$

where k is Coulomb's constant, R is the distance between the charges, and  $\hat{\mathbf{R}}_{12}$  is the unit vector pointing from  $q_1$  to  $q_2$ . The force exerted by  $q_2$  on  $q_1$  is equal and opposite:

$$\mathbf{F}_{12} = -\mathbf{F}_{21} \tag{2}$$

Coulomb's constant k is defined as:

$$k = \frac{1}{4\pi\epsilon_0} \tag{3}$$

where  $\epsilon_0$  is the permittivity of free space.

## 2 2019 Redefinition of SI Base Units

Since 2019, the SI base units are defined by setting the numerical values of seven defining constants. This includes the elementary charge, e. As a result, the vacuum permittivity  $\epsilon_0$  is no longer a defined constant but is a measured value with an associated uncertainty. This is a change from the pre-2019 definition where  $\epsilon_0$  was an exact value.

The redefinition emphasizes the fundamental constants of nature. For example, the gravitational force  $F_G$  and the electric force  $F_E$  can be compared:

$$F_G = G \frac{m_p m_e}{R^2} \approx 10^{-47} N \tag{4}$$

$$F_E = \frac{1}{4\pi\epsilon_0} \frac{e^2}{R^2} \approx 10^{-7} N \tag{5}$$