

3.3 Thin Film Deposition by PVD

3.3.1 Overview

PVD = Physical Vapor Deposition

Typical vacuum process: the materials will be deposited on the substrate via


→ **Evaporation** (e-beam / thermal) or **Sputtering** (using a sputtering target)

Principle:

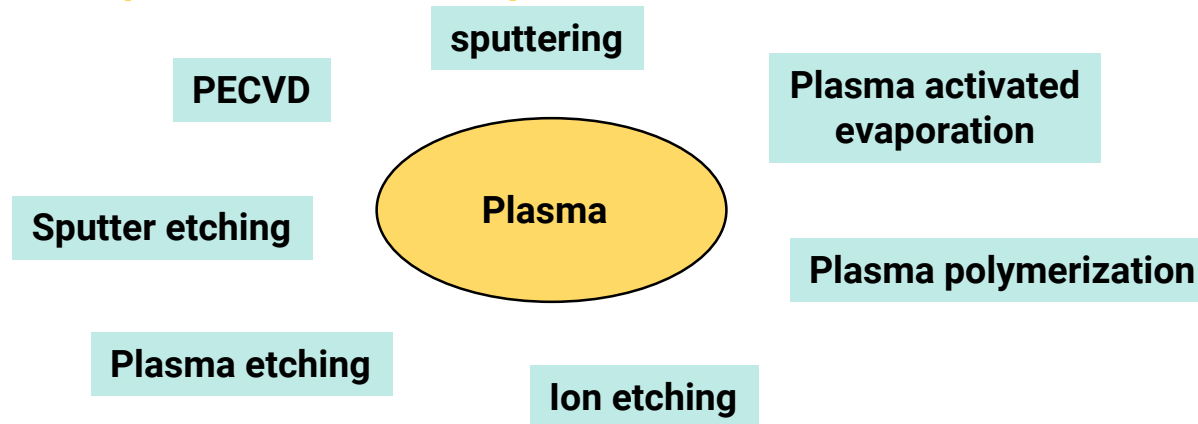
- The material will be transferred in a molecular or atomic state and transported to the substrate.
- On the surface of the substrate the **condensation** takes place and therewith the deposition of the film.

Main field of application: Generation of conductive metal films

Process flow

- 
- Surface preparation (cleaning, e.g. sputter etching)
 - Deposition process
 - Post-treatment (annealing)

3.3.2 Thin Film Deposition: role of the plasma



kinetic energy

The plasma provides charged particles important for:
sputtering (plasma sputtering)
sputter etching (Al, TiN, MoSi₂...)

radicals

Generation of species with a high chemical reactivity (radicals, radical ions) important for:
reactive etching (Cu, TiN, SiO₂, Si₃N₄...)
plasma enhanced chemical vapor deposition

Thin Film Deposition: What is a plasma?

Plasma

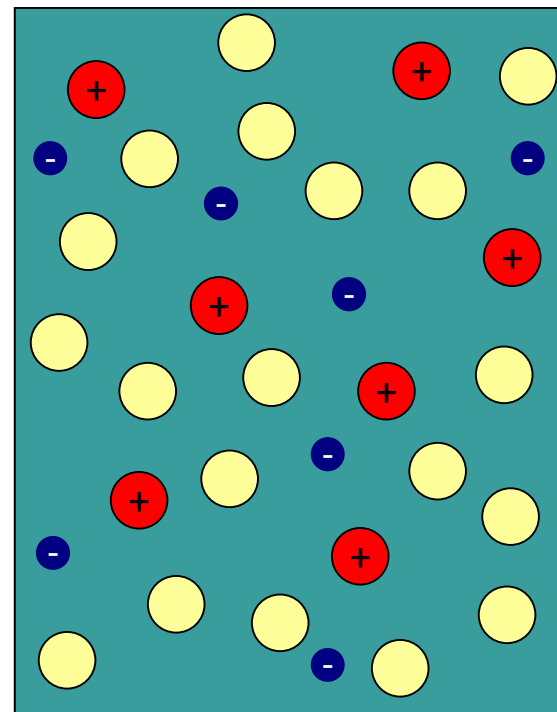
- electrical conductive gas
- mixture of free electrons, ions and atoms or molecules (formation of radicals* is also possible)

Appearance:

- from a gas
- caused by ionization (generation of electrons and ions from atoms and molecules)
 - collision ionization (collision of fast particles)
 - photo ionization (absorption of photons)
 - high temperatures ($\times 1000$ K)

→ required energy of ionization:

3.9 ... 24.6 eV



* radical: uncharged rest of a splitted molecule

Thin Film Deposition: What is a plasma?

Properties of low pressure plasmas

- External charge neutrality
 - Electron density: $10^9 \dots 10^{12} \text{ cm}^{-3}$, degree of ionization $10^{-6} \dots 10^{-4}$
electron energy: 1 ... 20 eV
↓
„temperature“ of the electrons is nearly 30 ... 1000 times higher than the average temperature of the molecules
 - The excited atoms and molecules convert in the initial state after a defined time. During the process a characteristic radiation will be emitted.
- Glow of the plasma (optical emission)

Applications in the microelectronics technology

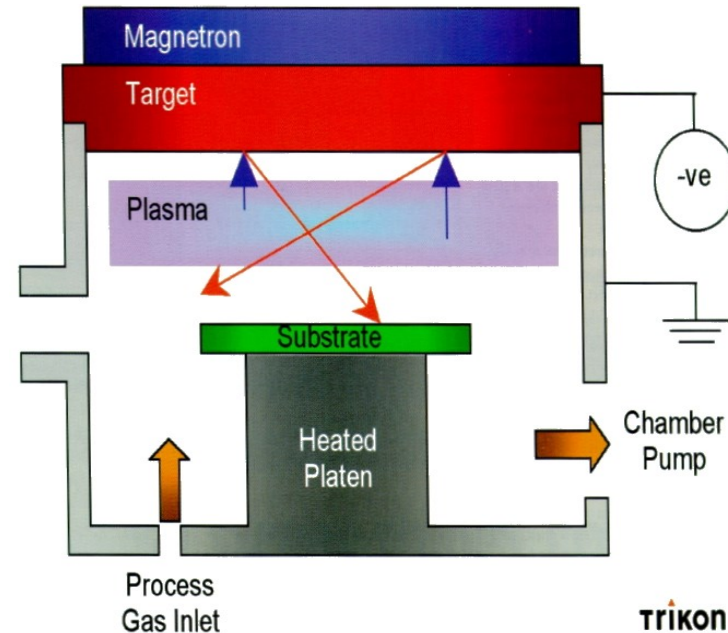
- CVD, etching, ion implantation, sputtering
arc discharge- and low pressure- plasmas
(CVD, etching 10 ... 200 Pa; sputtering 0,1 ... 1 Pa; ion implantation 10^{-5} Pa)

3.3.3 Sputtering Process

Principle:

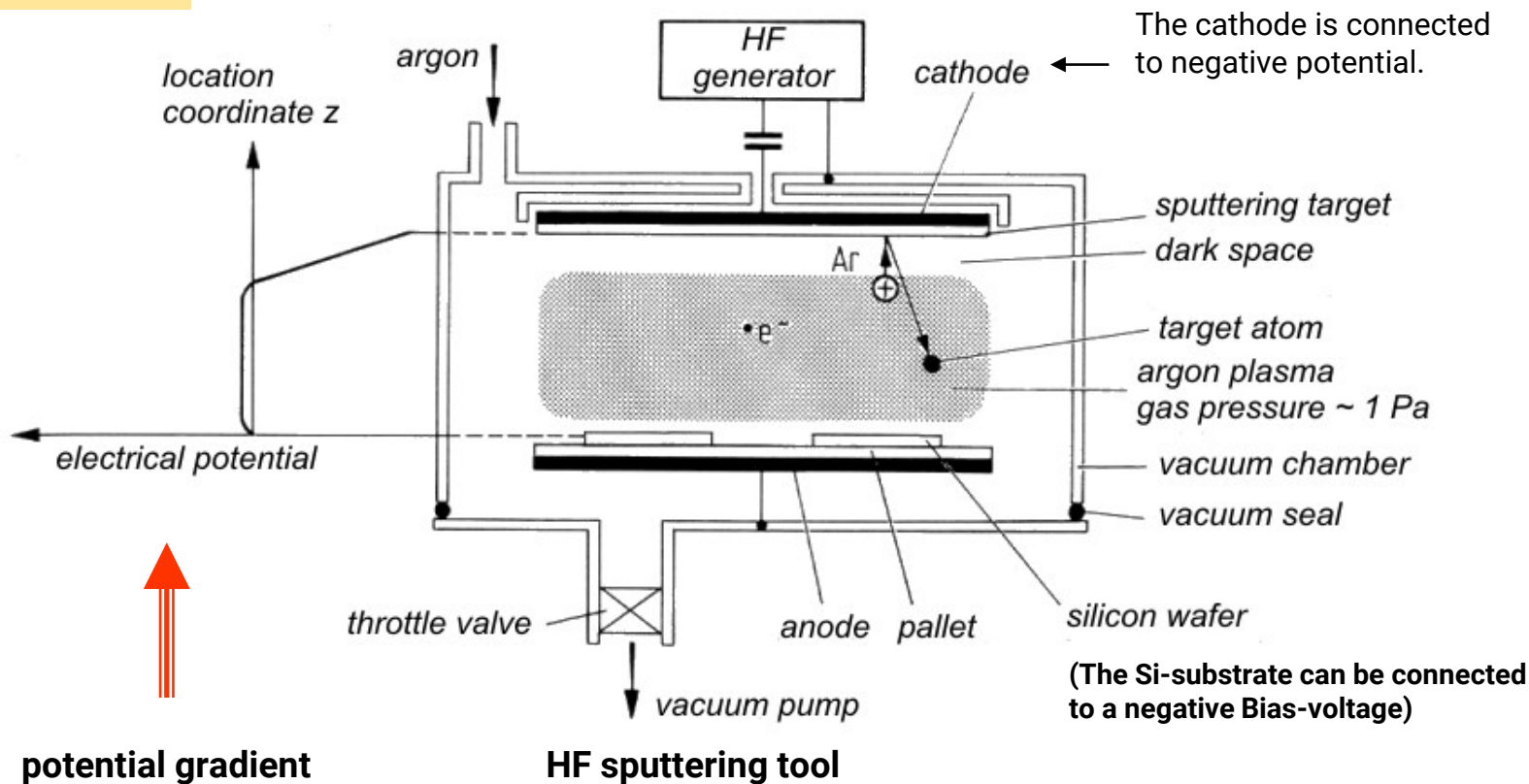
The target consists of the material, which should be deposited on the wafer.

- Area of ionization will not be separated from the area of deposition !
- The pressure will be defined by the plasma.
- Ballistic transport of the target atoms to the wafer (few collisions on the way to the substrate)
- Main field of application today in the microelectronic technology and technology of MEMS
reason: productivity and purity of the targets e.g.: 99.9995 % [5N5]



TRIKON

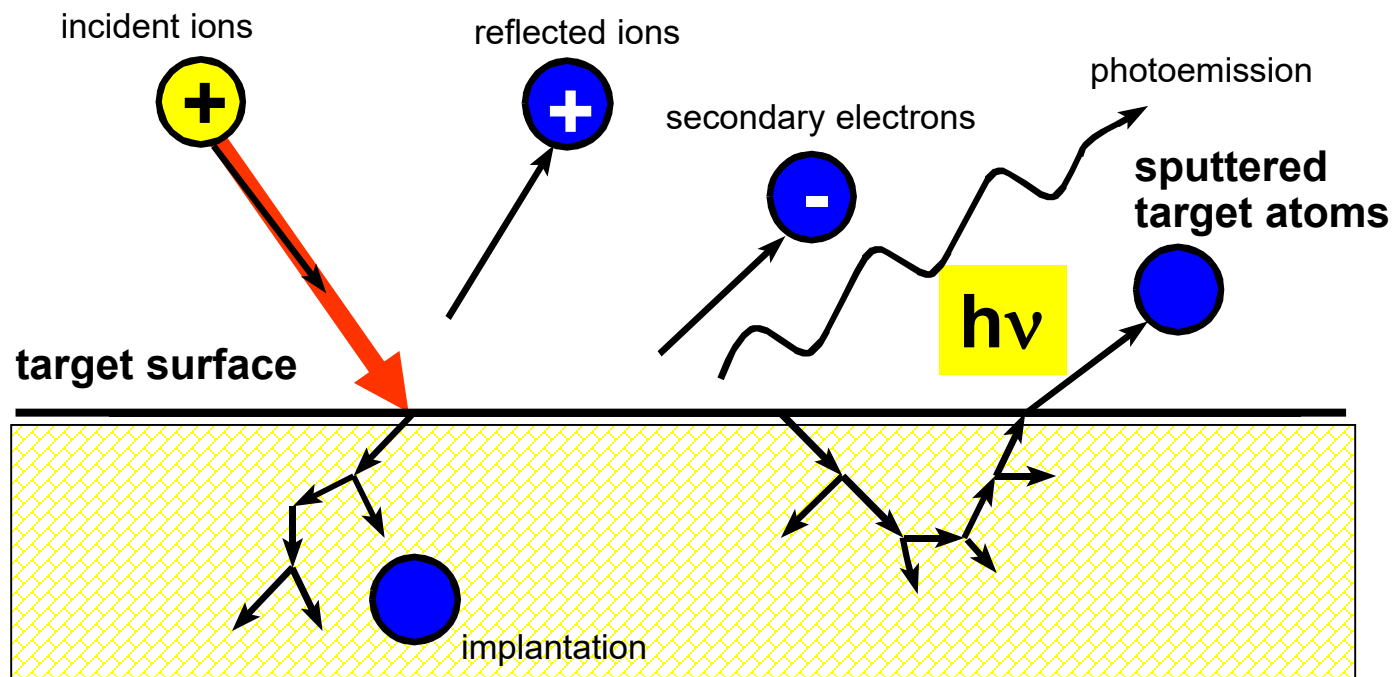
Sputtering



/Widmann/

- DC or RF mode
- dark spaces
- pressure range: 1 ... 10 Pa, voltage: $x \cdot 1000 \text{ V}$

Elementary processes of sputtering



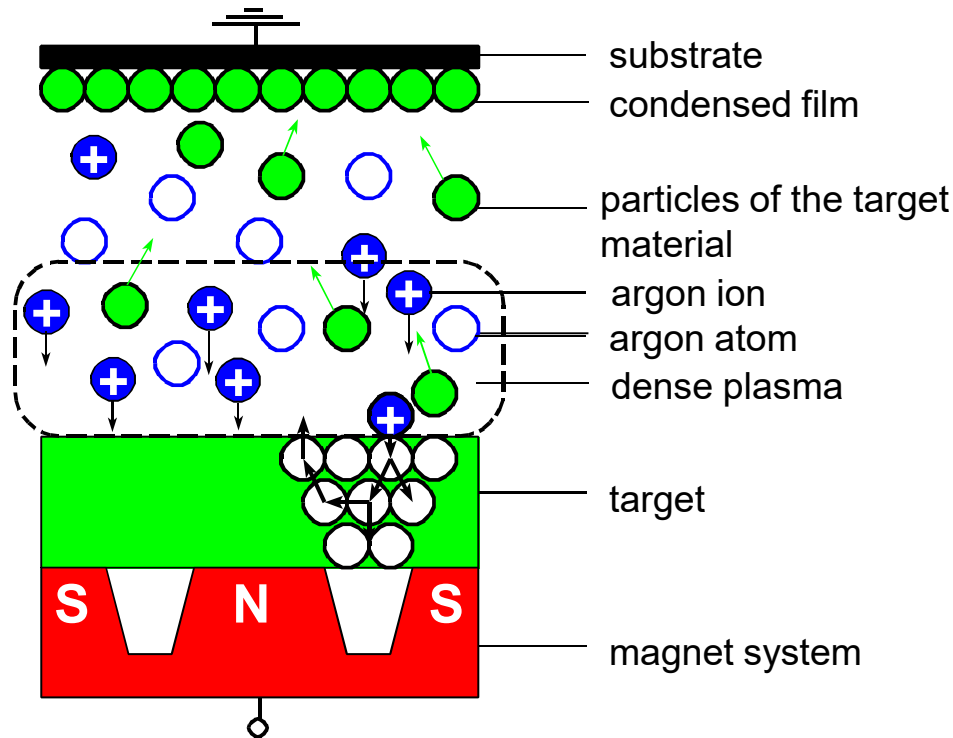
Interaction of ions with the surface

reflection secondary electrons change of stoichiometry
lattice defects radiation damages ion implantation

Sputtering: **Generation of a collision cascade**

It differs, if the last collision hits an atom in the bulk or at the surface of the target.
Only in the latter case an emission takes place - efficiency 5% to 25 %!

Sputtering: DC magnetron sputtering

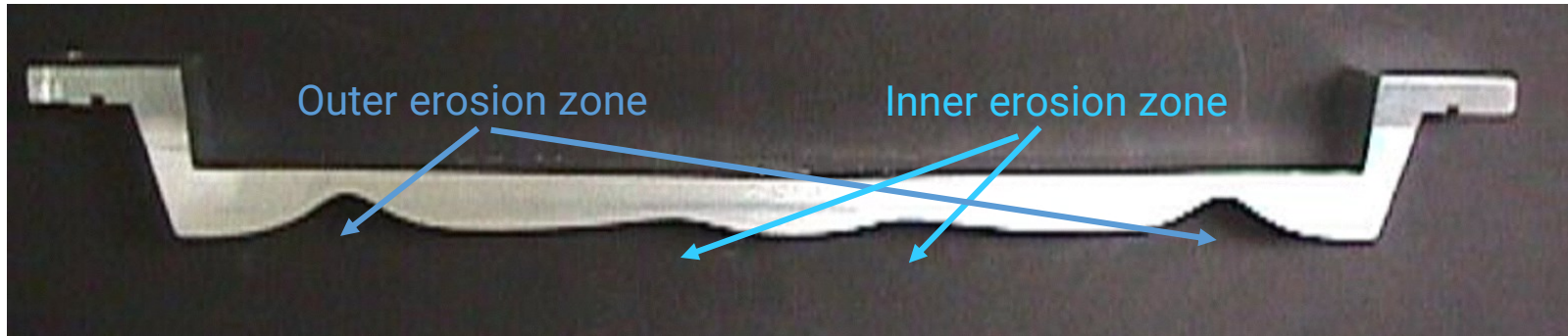


- Plasma (Ar^+ , e^-) generated by glow discharge
- the magnetic field elongates the trajectory of the electrons
- DC:
for conducting target materials
 - average kinetic energy approx. 2-10 eV
 - insulator would be charged →
- RF:
for insulating target materials
(electrons compensate for positive charges on the target)

Sputtering – cathode sputtering

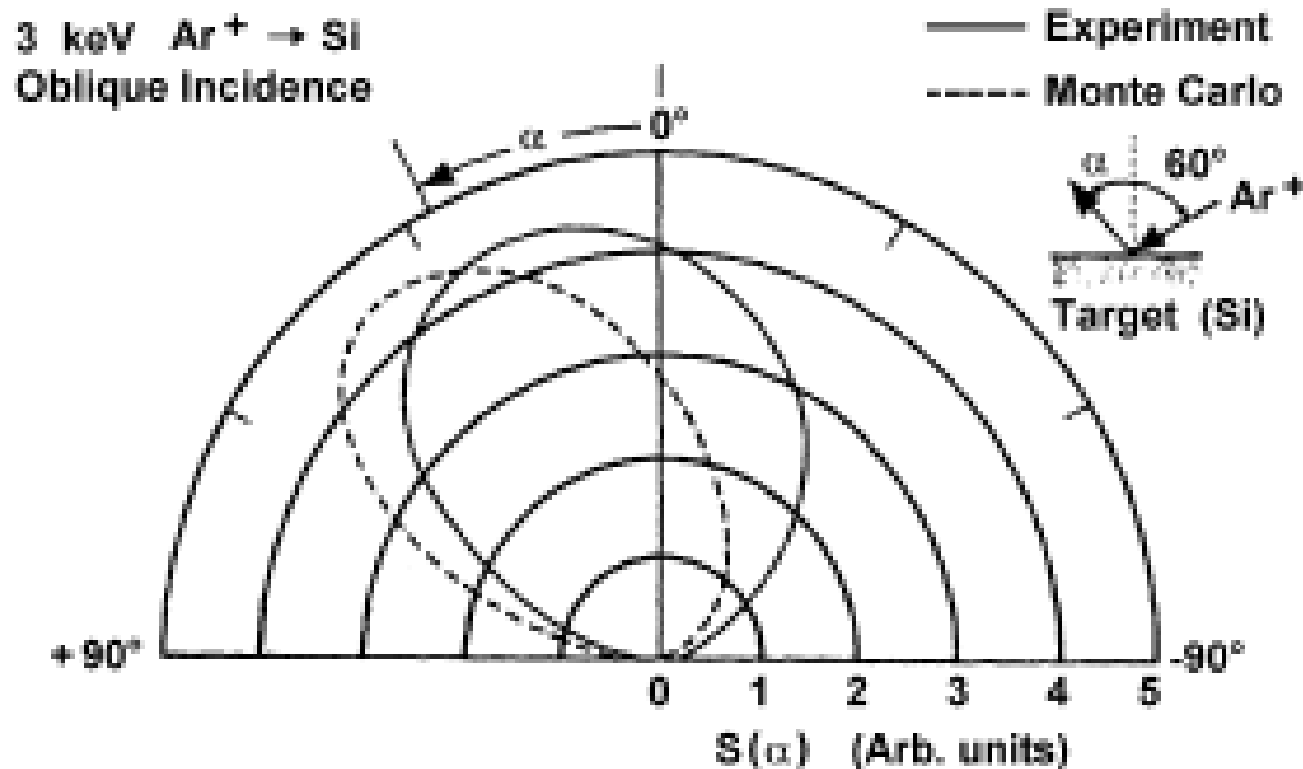
Accelerated positively charged gas ions with an enough amount of kinetic energy collide with the cathode (target); by collision processes atoms at the surface will be ejected; these atoms leave the surface with preferential direction (cosine distribution); the energy of the atoms decrease by interaction with the plasma from initial approx. 10 eV to 1 to 2 eV (compare with e-beam evaporation: approx. 0.1 eV).

Erosion trench on a magnetron target



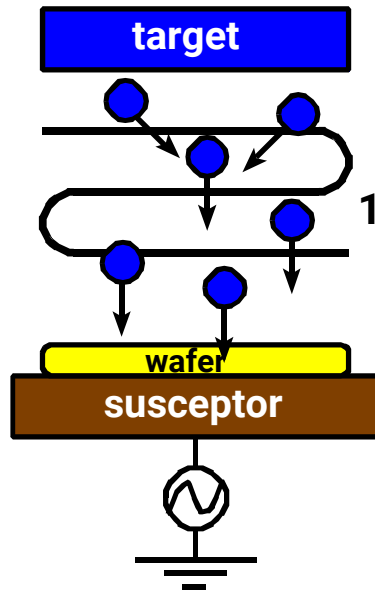
Target, diameter approx. 400 mm

The angular distribution of sputtered atoms



Angular distributions of sputtered Si atoms for 3 keV, Ar^+ ion bombardment at an incident angle of 60°

Thin Film Deposition: Sputtering – advanced techniques

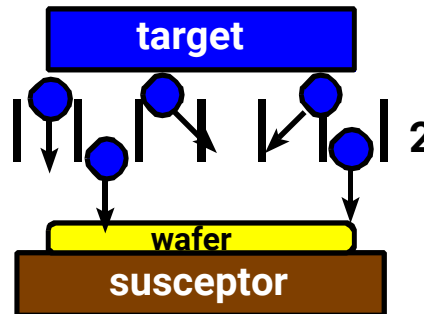


1 water-cooled RF-electrode 1,985 MHz

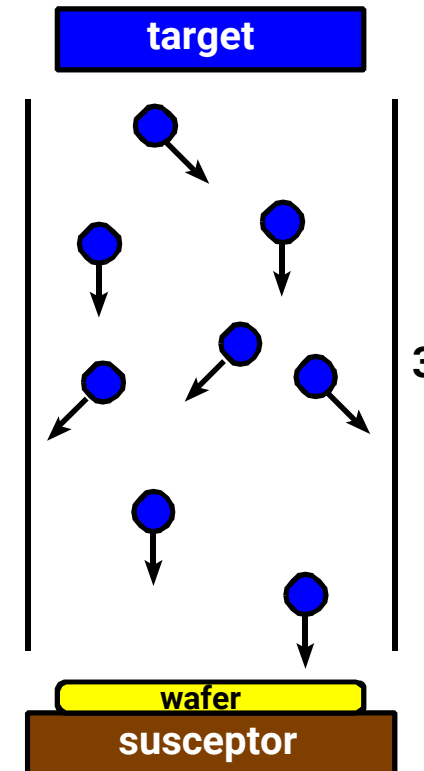
2 collimator

3 chamber wall

ionized metal deposition



collimated sputtering



long throw sputtering

Modern principles to coat contact and via holes with a high aspect ratio
(e.g. TiN, TaN or WN barriers for the copper metallization)

Only target particles with a small angle to the substrate normal reach the surface !

The goal is to coat high aspect ratio patterns more conformally by more directional sputtering with narrow arrival angle distribution.

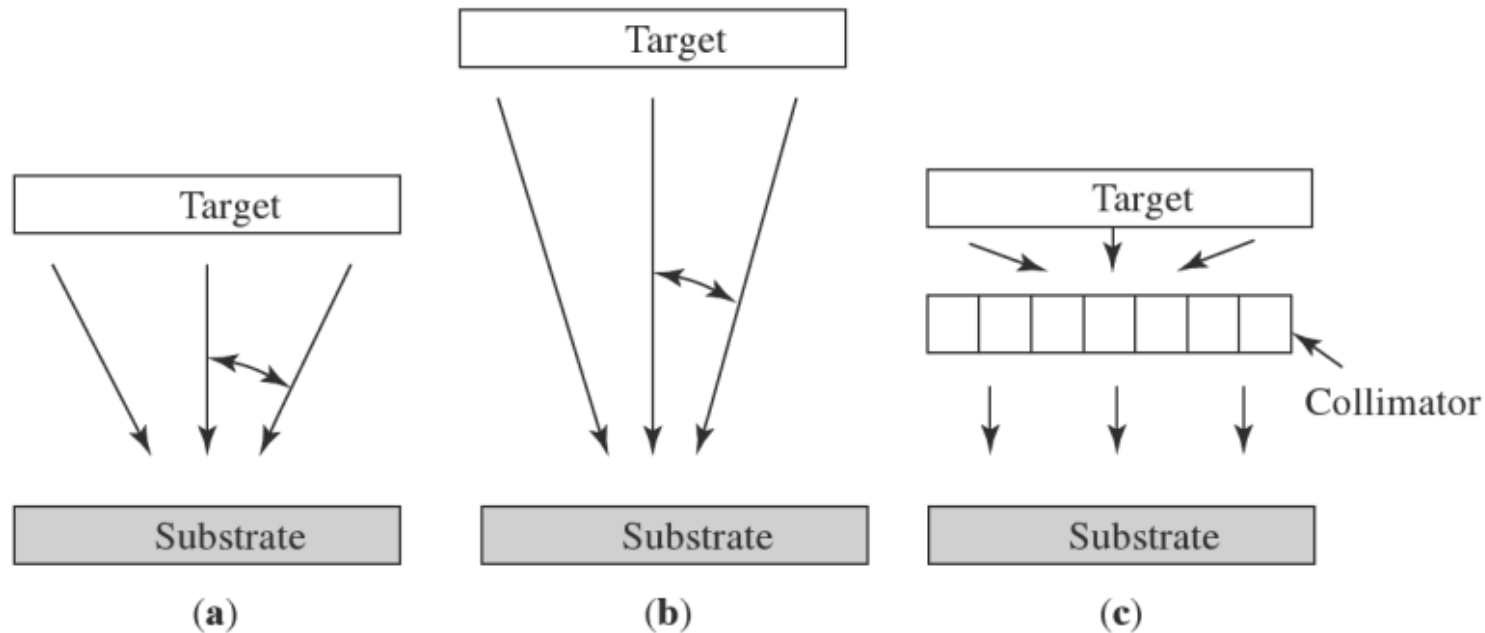


Figure 2.27. (a) Standard sputtering; (b) long-throw sputtering; (c) sputtering through a collimator [1].

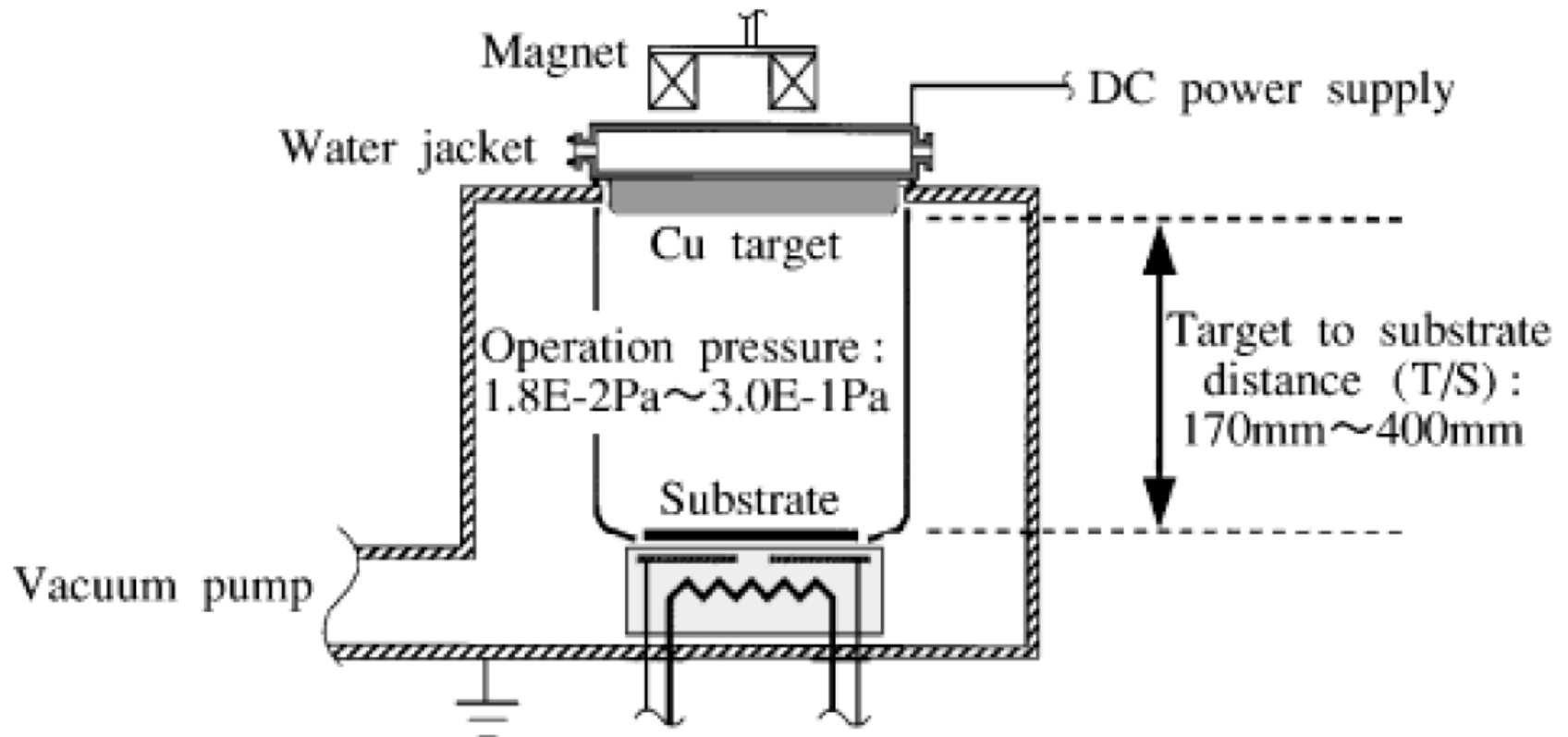
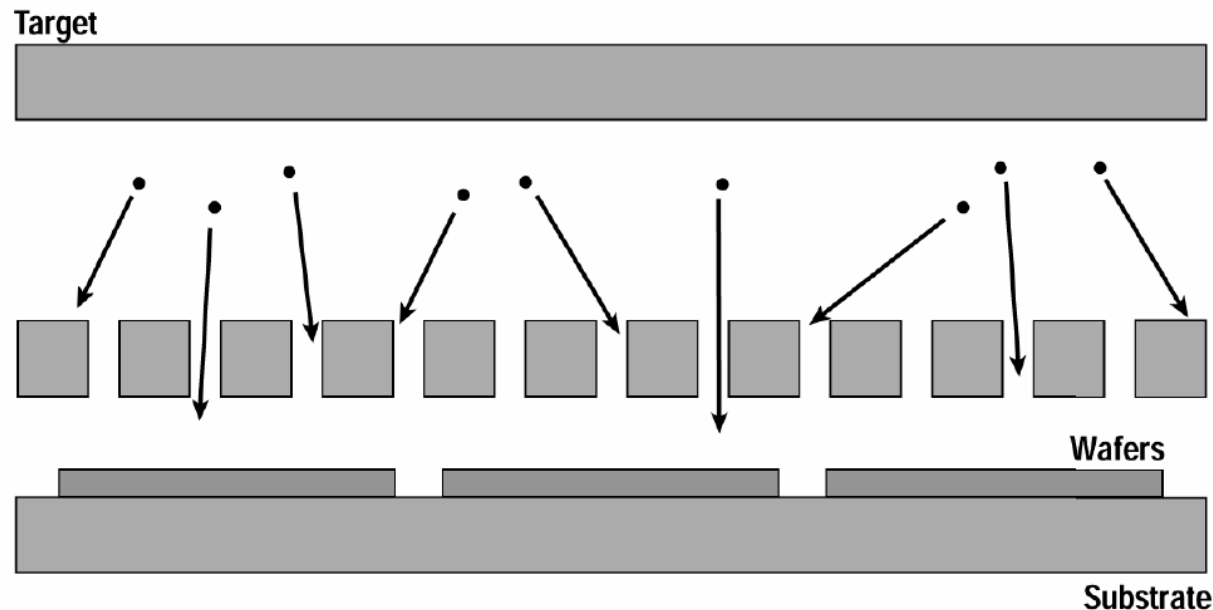


Fig. 1 Schematic diagram of low pressure long throw sputtering system.

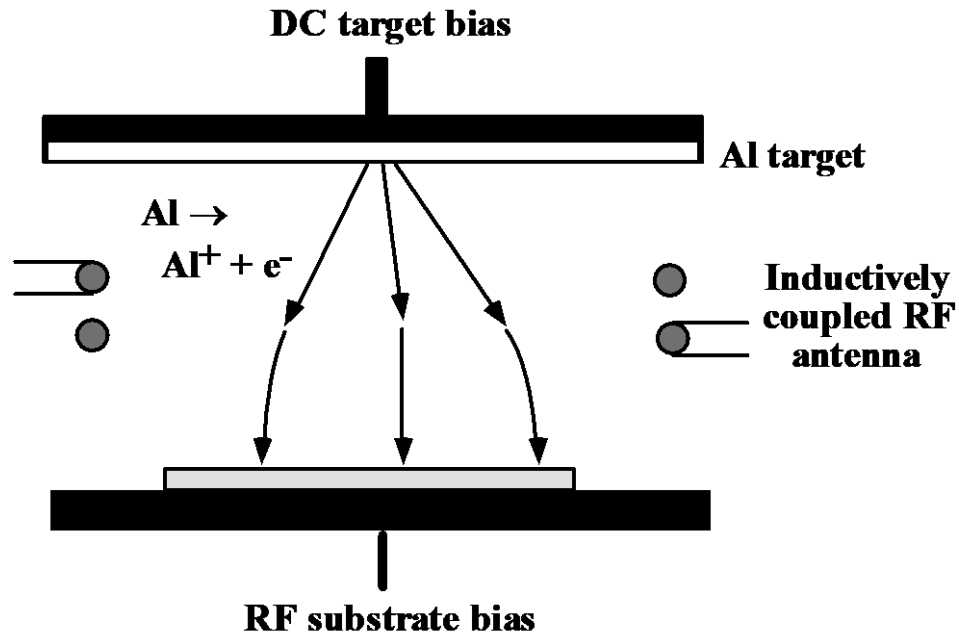
T. Saito, Materials Transactions, Vol. 43, No.7 (2002) pp. 1599-1604

Collimated sputtering



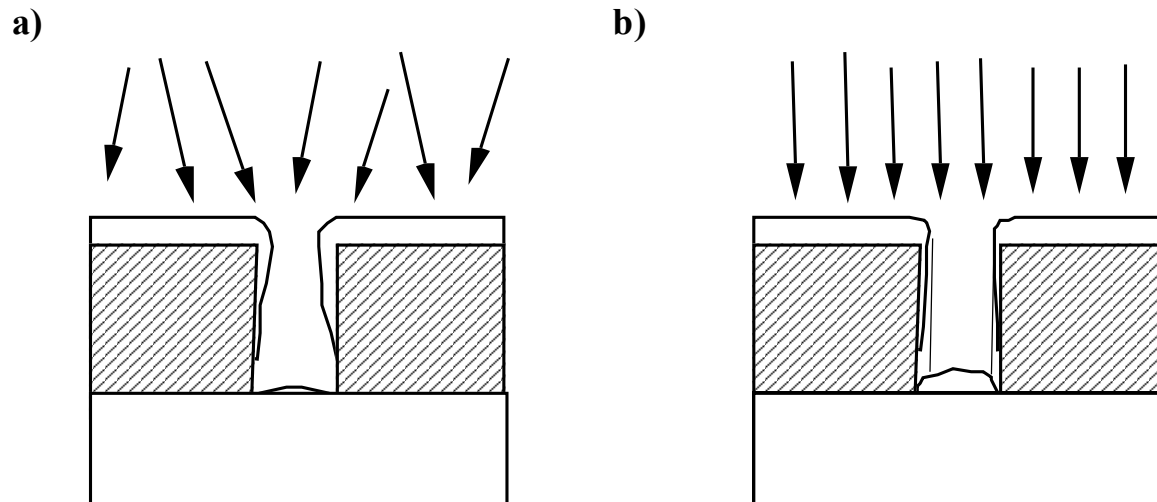
- Insert a plate with high-aspect-ratio holes.
- Sputter at low pressure, mean path is long enough that few collisions occur between collimator and wafer.
- Species with velocities nearly perpendicular to wafer surface pass through the holes.
- Reduce deposition rate considerably (most sputtered atoms cannot reach the substrate).

Ionized sputter deposition



- The depositing atoms themselves are ionized.
- An RF coil around the plasma induces collisions in the plasma, creating the ions (50-85% ionized).
- Most sputtered atoms can reach the substrate (better solution than a collimator).
- Provides a narrow distribution of arrival angles, which may be useful when covering or coating high AR patterns.

- (a) Regular sputter deposition.
- (b) Sputter deposition, by using long throw configuration, a collimator, or ionized sputter deposition.



Reactive sputtering

From a metallic target

Deposition of a compound, which derives from target material and reactive gas

TiN (Ti in Ar/N₂)

TaN (Ta in Ar/N₂)

Reactive gases:

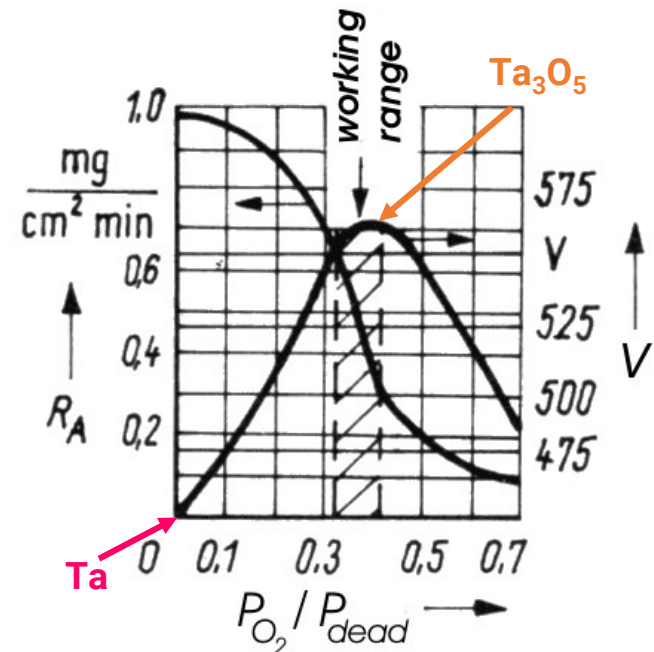
O₂ (oxide), **N₂ or NH₃** (nitride), **O₂/N₂** (oxynitride)
C₂H₂ or CH₄ (carbide), **SiH₄** (silicide)

- The reactive gas is consumed during film deposition.
- Compensation using a higher flow rate.
- Control of the reactive gas is realized using a selected emission line of the sputtered target atoms.
- This procedure is complex → a special adaptation to the used target-gas system is necessary.

From a compound target

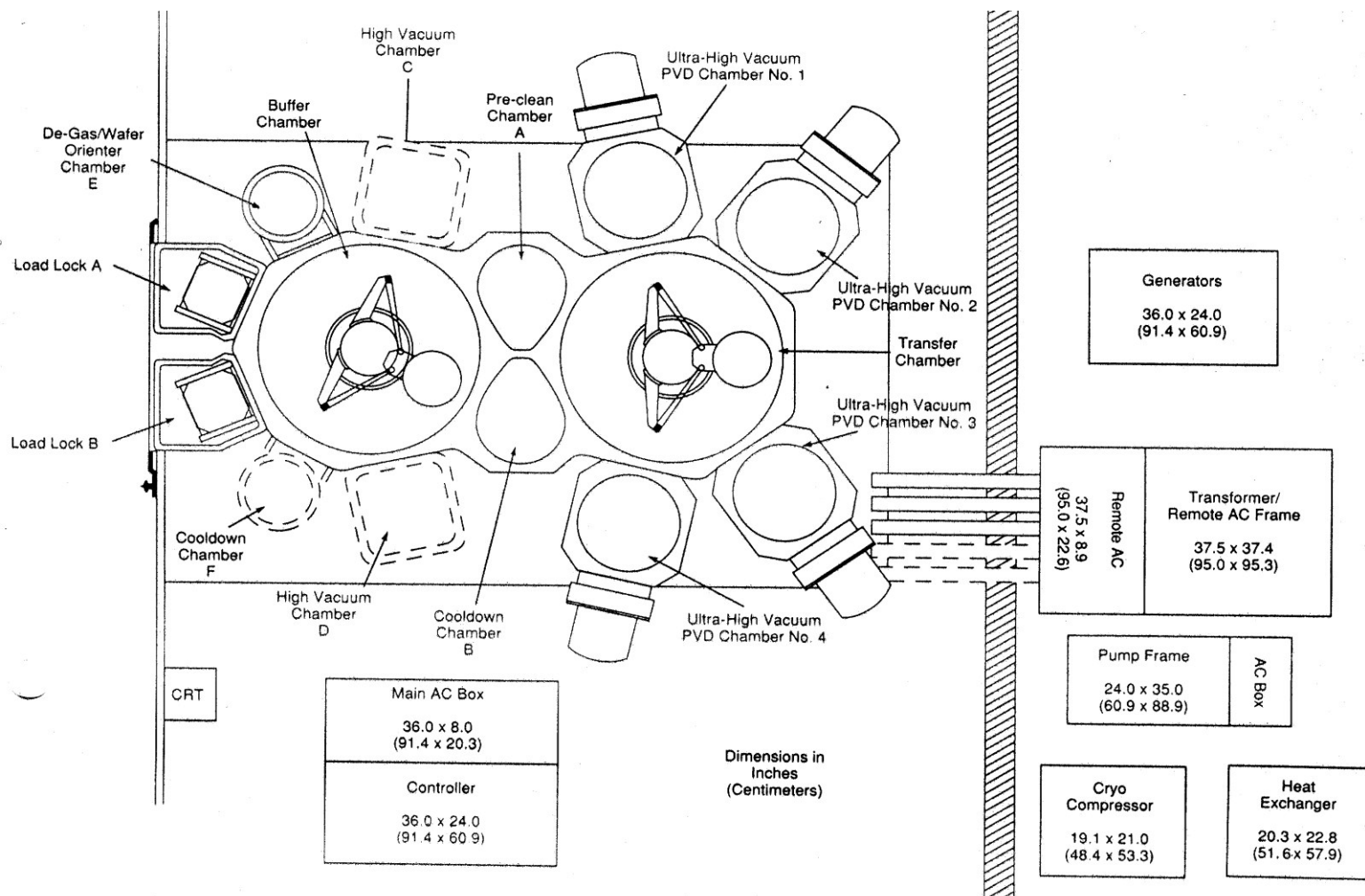
Prevention of the decomposition of the compound during sputtering: stoichiometric film deposition

SiO₂ (SiO₂ in Ar/O₂)



Deposition rate, V_{cathode} as a function
of $p_{\text{O}_2}/p_{\text{tot}}$ ($p_{\text{tot}} = 0.3 \text{ Pa}$; $P = 5 \text{ kW}$)

Sputtering Systems – Single wafer systems



Endura 5500: cluster tool.

Applied Materials, USA



Endura 5500: cluster tool

Applied Materials, USA

Selected application examples for PVD

Microelectronic production

Al	→	Interconnect lines
W	→	Interconnect lines (M0/M1)
Cu	→	seed layer for damascene structures (afterwards filling with electroplating)
TiN	→	barrier, seed layer (liner) for W-CVD, ARC (antireflection coating)
Ta, TaN/Ta	→	barrier against Cu diffusion / liner
Ti, Co, Ni, Pt	→	for silicide contacts
TiW	→	barriers (especially for the packaging)