

Table of Polyatomic Ions

1−	2−	1+ and 3−
Acetate: $\text{CH}_3\text{COO}^{1-}$ <i>and</i> $\text{C}_2\text{H}_3\text{O}_2^{1-}$	Oxalate: $\text{C}_2\text{O}_4^{2-}$ Carbonate: CO_3^{2-}	Ammonium: NH_4^{1+}
Bicarbonate: HCO_3^{1-}		
Hypochlorite: ClO^{1-} Chlorite: ClO_2^{1-} Chlorate: ClO_3^{1-} Perchlorate: ClO_4^{1-}	Chromate: CrO_4^{2-} Dichromate: $\text{Cr}_2\text{O}_7^{2-}$	
Cyanide: CN^{1-}		
Hydroxide: OH^{1-}		
Iodate: IO_3^{1-}		Arsenate: AsO_4^{3-}
Nitrite: NO_2^{1-} Nitrate: NO_3^{1-}	Sulfite: SO_3^{2-} Sulfate: SO_4^{2-}	Phosphite: PO_3^{3-} Phosphate: PO_4^{3-}
Permanganate: MnO_4^{1-}		
Thiocyanate: SCN^{1-}	Thiosulfate: $\text{S}_2\text{O}_3^{2-}$	