

Green Chemistry

- Definition
- principles
- ❖ Goal & need
- Efficiency Parameters



design of chemical products and processes that reduce or eliminate the use or generation of hazardous substances

Focus on Chemical Synthesis Process Chemistry and Chemical Engineering

- **❖** Alternative Reaction Condition
- Incorporation of catalyst
- **❖** Alternative pathway
- **❖Synthesis of biodegrabale material**



- ❖Green Chemistry was Coined by Paul T Anastas
- ❖Green chemistry also called sustainable chemistry
- ❖ An area of chemistry and chemical engineering focused on the designing of products and processes that minimize the use and generation of hazardous substances.



Paul T Anastas



Waste Green Chemistry Materials Aims at Reducing **Hazard** Risk **Energy**



Principles Of Green Chemistry

4.Safer Chemicals

3. Less
Hazardous
Synthesis

2.High Atom Economy

1.Prevention of Waste

5.Design for Energy Efficiency

6.Safer Solvents

7.Renewable Feedstock

8.Reduce Derivatization

9.Use of Catalyst

10.Degradable Products

11.Pollution Prevention

12. Analytical Methods



Goals of Green Chemistry

- **❖** To reduce adverse Environmental Impact
- **❖** To prefer renewable materials
- to Reduce use of Toxic materials
- **❖** To prefer hazard free processes
- ❖ To reduce byproduct formation
- To reduce use of Toxic Solvents
- **❖** To improve Energy Efficiency
- **❖** To develop processes for toxicity monitoring
- **❖** To prefer and synthesize biodegradable materials

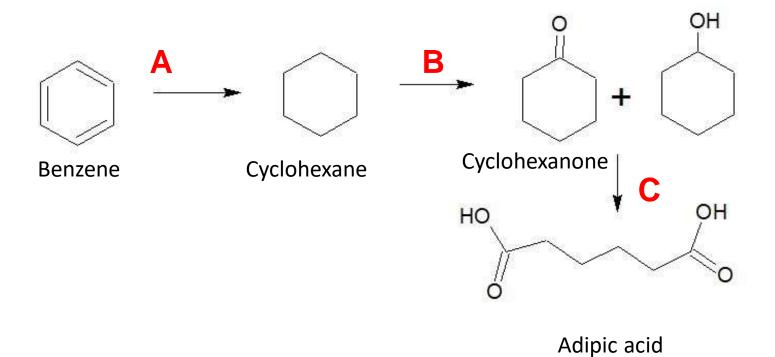


Efficiency Parameters

- Atom Economy
- Yield of reaction
- Environmental load factor
- Mass intensity
- Conversion
- Reaction Selectivity



Chemical Route For Adipic Acid





Drawbacks Of Chemical Route Of Adipic Acid

- ❖ Benzene is Carcinogenic Staring Material
- High Pressure and high Temp requirement
- Less Atom Economy



Greener Route For Adipic Acid

L:E.Coli M: E.Coli K: H₂/Pt



Chemical Route For Polycarbonate



Greener Route For Polycarbonate



Advantages Of Greener Route Of Polycarbonate

- Komiya et al reported synthesis of polycarbonate (Asahi chemical Japan)
- No need of Solvent (Solid phase Synthesis)
- **❖** Polymerization takes place in Molten state
- Use of DCM is Eliminated



Industrial Applications Of Green Chemistry

- DMC is used as Electrolyte in Ion Battery
- Selective palladium (Pd(OAc)2)- catalyzed cyclocarbonylation reactions were carried out for the synthesis of lactones from allyl phenol derivatives in DMC

Reference: https://doi.org/10.1016/j.cogsc.2017.03.012 (Sang-Hyun Pyoa, Ji Hoon Parkb, Tae-Sun Changb, Rajni Hatti-Kaula)



Use Of DMC

- DMC chemistry has been extended to include acid catalyzed reactions of different aliphatic alcohols and phenols
- DMC has shown a good extraction performance for triglycerides
- DMC as Methylating agent in place of DMS



Polyurethane Via Green Approach

- Chemical route involve use of phosgene
- ❖ In greener route Phosgene is replace by CO₂

$$R-NH_2 + COCI_2 \longrightarrow R-N=C=O + 2HCI$$

$$R-NH_2 + CO_2 \longrightarrow R-N=C=O + H_2O$$