Problem 1)

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 \text{Im}_{[70]} = \text{M} = \text{UnitConvert} \Big[ \text{ sodium } \text{ ELEMENT} \Big[ \text{ atomic mass} \Big], \text{ "Kilograms"} \Big]; \text{ StringForm}["M = ``", M]   \lambda = \text{Quantity}[589, \text{ "nanometers"}];   k = \frac{2\pi}{\lambda}; \text{ StringForm}["k = ``", \text{UnitConvert}[k, \text{"inverse nm"}] // N]   p = \hbar \text{ k; StringForm}["p = ``", \text{UnitConvert}[p, \text{"Kg m/s"}] // N]   v = \frac{p}{M}; \text{ StringForm}["v = ``", \text{UnitConvert}[v, \text{"cm/s"}]]   \text{Out}_{[70]} = \text{M} = 3.81754100 \times 10^{-26} \text{ kg}   \text{Out}_{[72]} = \text{k} = 0.0106675 \text{ /nm}   \text{Out}_{[73]} = \text{p} = 1.12497 \times 10^{-27} \text{ kg m/s}   \text{Out}_{[74]} = \text{v} = 2.94684318 \text{ cm/s}   \text{Im}_{[81]} = \lambda' = \frac{2\pi \text{ c}}{\frac{2\pi \text{ c}}{\lambda} - \text{kv}}; \text{ StringForm}["Doppler \text{ shifted frequency } \lambda' = ``. \text{ So, } \Delta\lambda = ``", \lambda', \lambda' - \lambda]   \text{Out}_{[81]} = \text{Doppler shifted frequency } \lambda' = 589.00000000578964075 \text{ nm} . \text{ So, } \Delta\lambda = 5.78964075 \times 10^{-8} \text{ nm}
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Problem 2)

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\label{eq:total_state} \begin{array}{lll} & & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & &
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