

Trace metals, climate, and ocean iron fertilization (OIF)

Ben Twining

17 September 2018

Elements of greatest biological relevance

the stuff of life!

1A		2A		Elements of greatest biological relevance																0						
1 H 1.00794		3 Li 6.941	4 Be 9.009																	2 He 4.0026						
				<i>the stuff of life!</i>																						
11 Na 22.98976	12 Mg 24.304			3B	4B	5B	6B	7B	8B		1B	2B	3A	4A	5A	6A	7A	10 Ne 20.179								
19 K 39.0983	20 Ca 40.078	21 Sc 44.9559	22 Ti 47.88	23 V 50.9415	24 Cr 51.9961	25 Mn 54.9380	26 Fe 55.845	27 Co 58.9332	28 Ni 58.69	29 Cu 63.546	30 Zn 65.38	31 Ga 69.723	32 Ge 72.64	33 As 74.9216	34 Se 78.96	35 Br 79.904	36 Kr 83.80									
37 Rb 85.4678	38 Sr 87.62	39 Y 88.9058	40 Zr 91.224	41 Nb 92.9063	42 Mo 95.94	43 Tc (98)	44 Ru 101.07	45 Rh 102.9055	46 Pd 106.42	47 Ag 107.8682	48 Cd 112.411	49 In 114.818	50 Sn 118.710	51 Sb 121.757	52 Te 127.60	53 I 126.9055	54 Xe 131.29									
55 Cs 132.9054	56 Ba 137.327	57 La 138.905	72 Hf 178.49	73 Ta 180.948	74 W 183.84	75 Re 186.207	76 Os 190.23	77 Ir 192.22	78 Pt 195.084	79 Au 196.967	80 Hg 200.59	81 Tl 204.37	82 Pb 207.2	83 Bi 208.98	84 Po (209)	85 At (210)	86 Rn (222)									
87 Fr (223)	88 Ra (226)	89 Ac (227)	104 Rf (261)	105 Ha (262)	106	107	108	109																		
									Lanthanides																	
									Actinides																	
									58 Ce 140.12	59 Pr 140.908	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.925	66 Dy 162.50	67 Ho 164.930	68 Er 167.259	69 Tm 168.933	70 Yb 173.054	71 Lu 174.967				
									90 Th 232.038	91 Pa 231.036	92 U 238.029	93 Np (237)	94 Pu 244.064	95 Am 243.061	96 Cm 247.070	97 Bk 247.070	98 Cf 251.08	99 Es 252.083	100 Fm 257.10	101 Md 258.10	102 No 259.10	103 Lr 262.10				

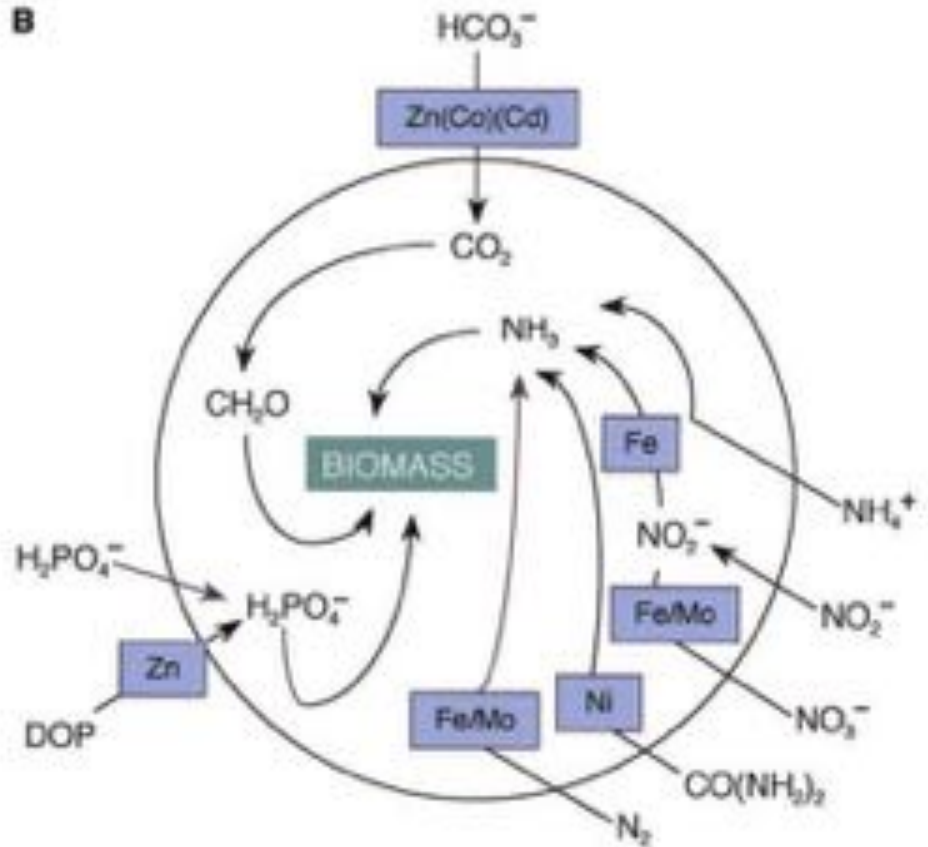
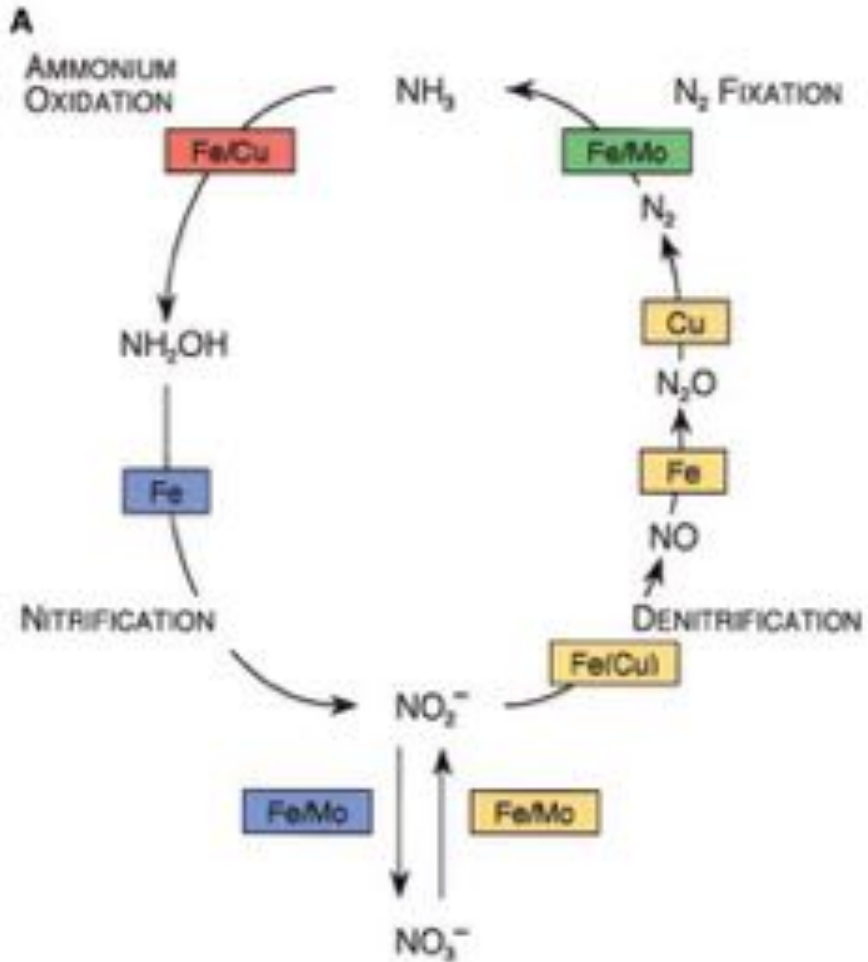
Major building blocks of biomolecules

Metals and other trace components of biomolecules

Important ions

Significant bio-toxins

Uses of metals in cells



(Morel and Price 2003)

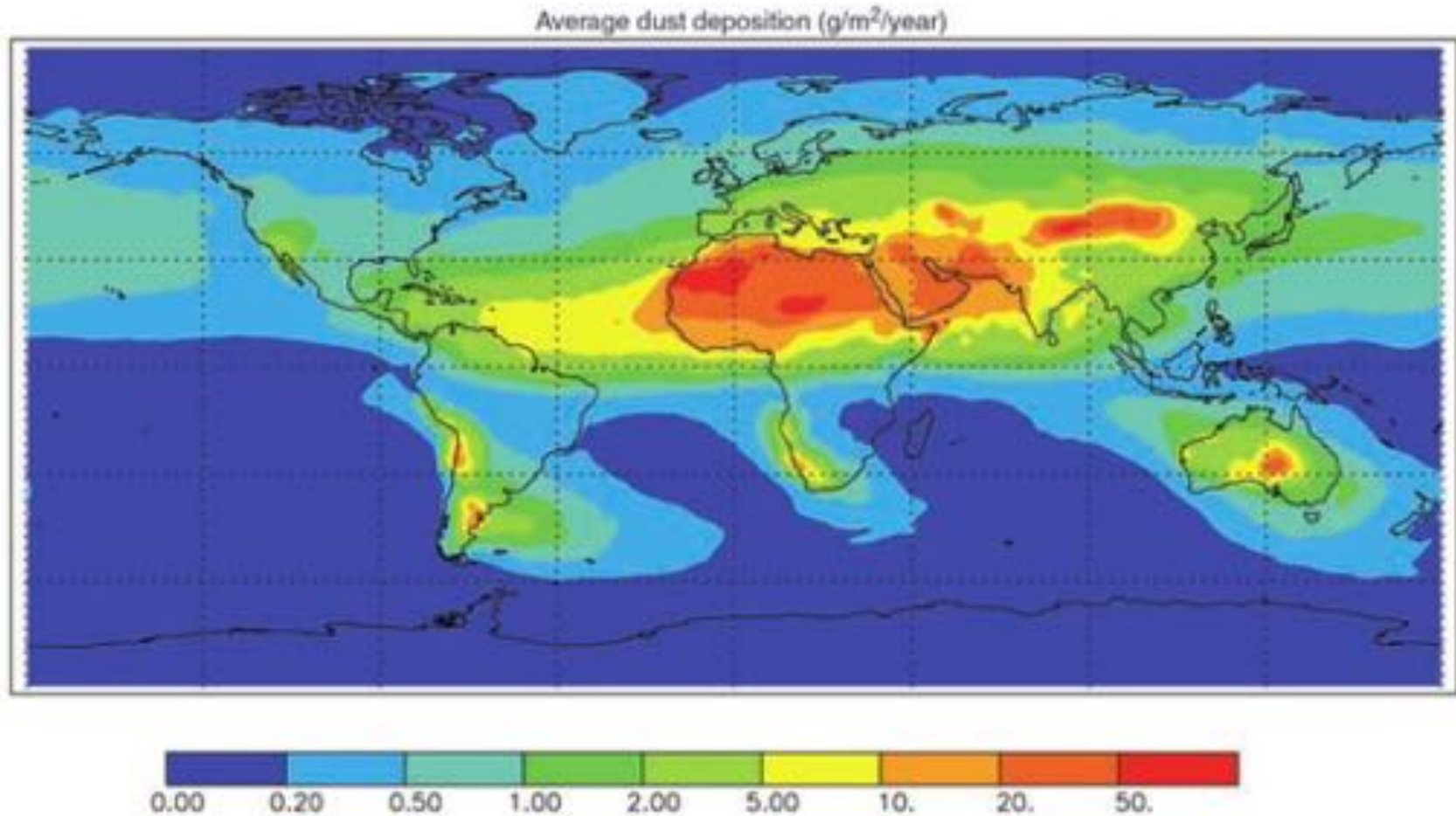
Importance of iron



Borrelia burgdorferi



How does iron get into the ocean?



Jickells et al. 2005

William Putman, NASA/Goddard

Ice rafted debris (dirt) in icebergs



HNLC Regions, Grazing and Iron



John Martin (1935-1993)

Colby '62

High nutrient,
low chlorophyll
HNLC regions

surface nitrate

Sea-surface nitrate [mmol N m^{-3}]





Claire Patterson



1922-1995

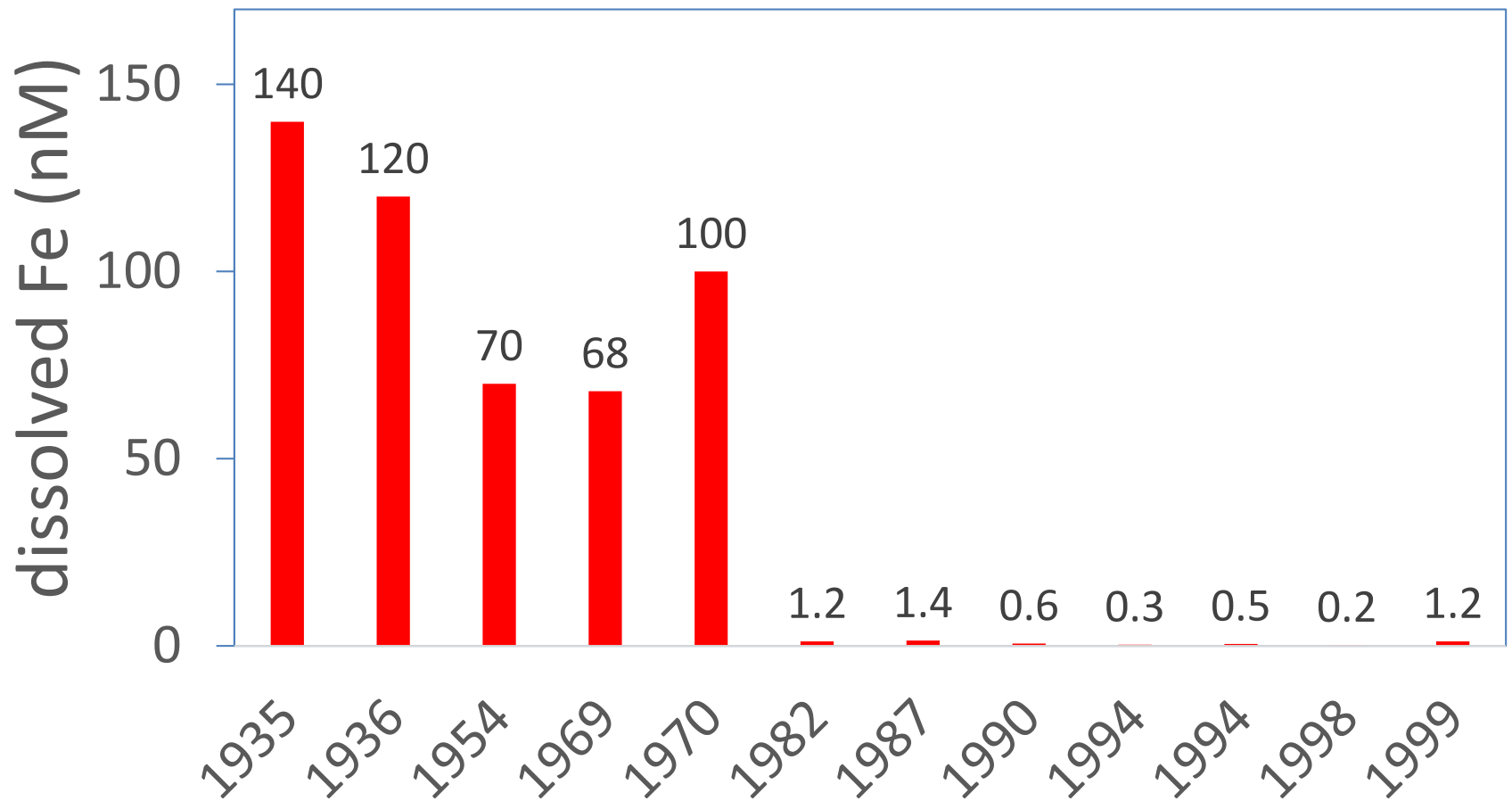




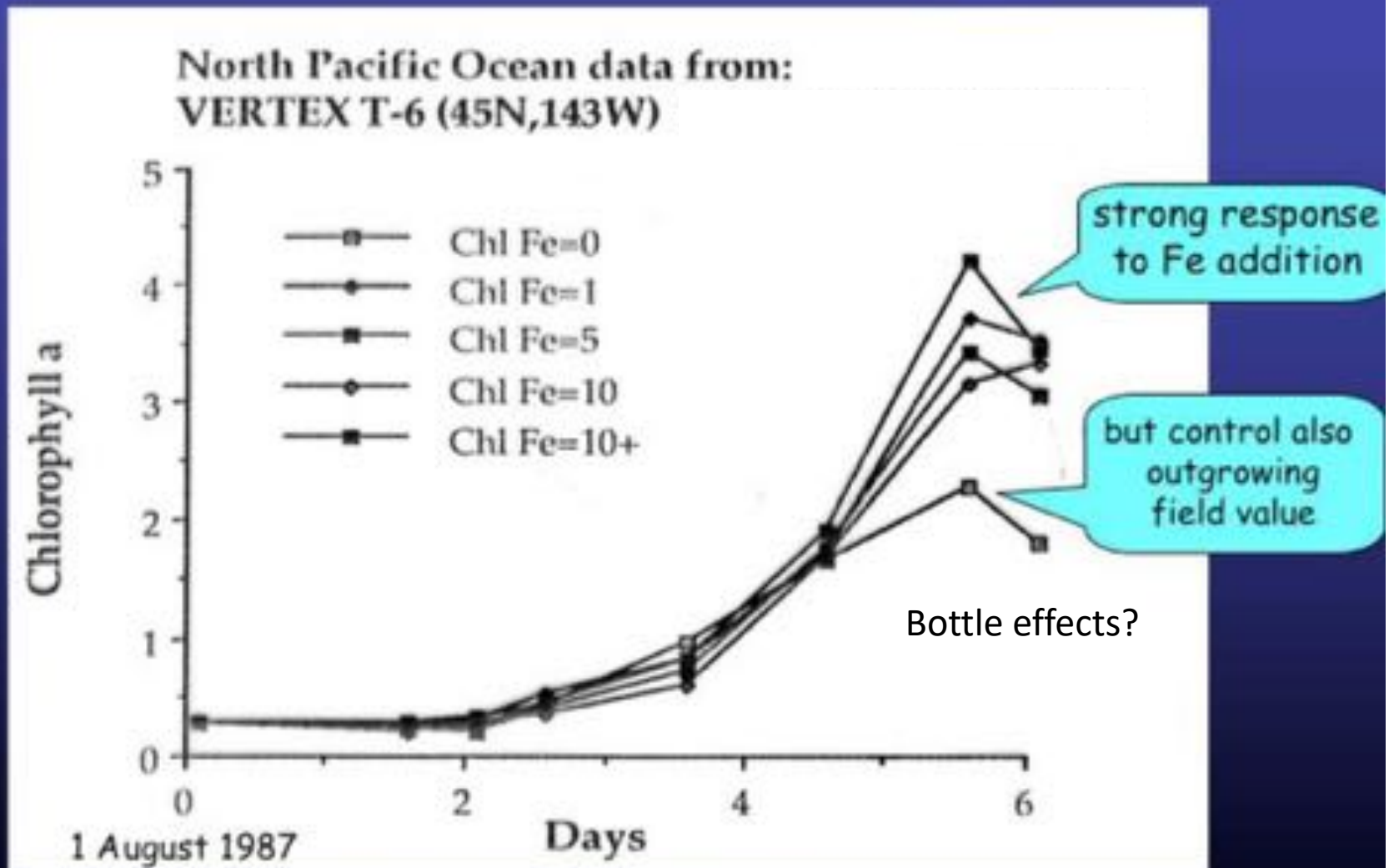




Ocean iron concentrations through time...



Subarctic North Pacific, August 1987



....as shown in Paris by John Martin in September 1987

Oscar Schofield - Rutgers

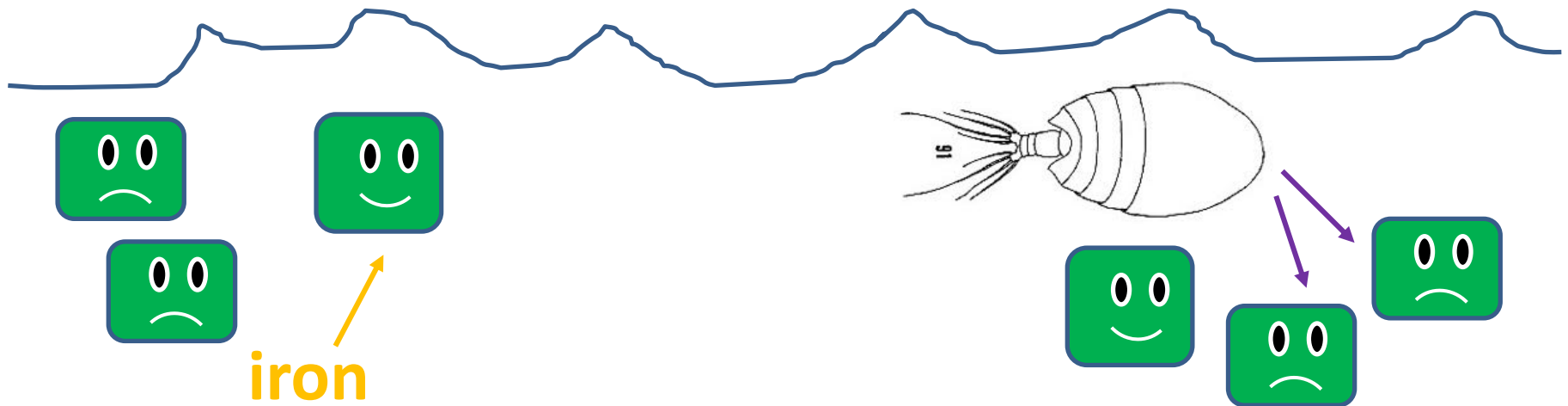
Martin and Fitzwater (1988) *Nature*, 331, 341-343.

*Why are nutrients not depleted in
HNLC regions?*

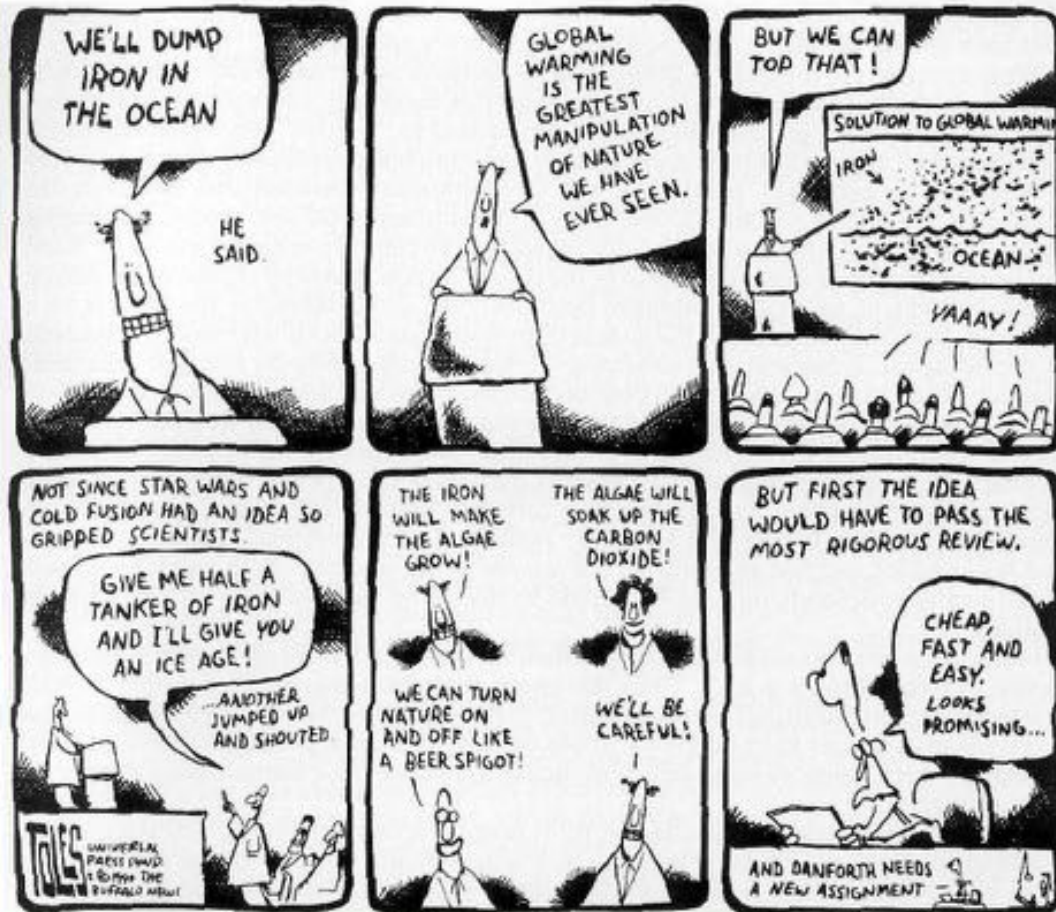
*Iron
hypothesis*

vs.

*Grazing
hypothesis*



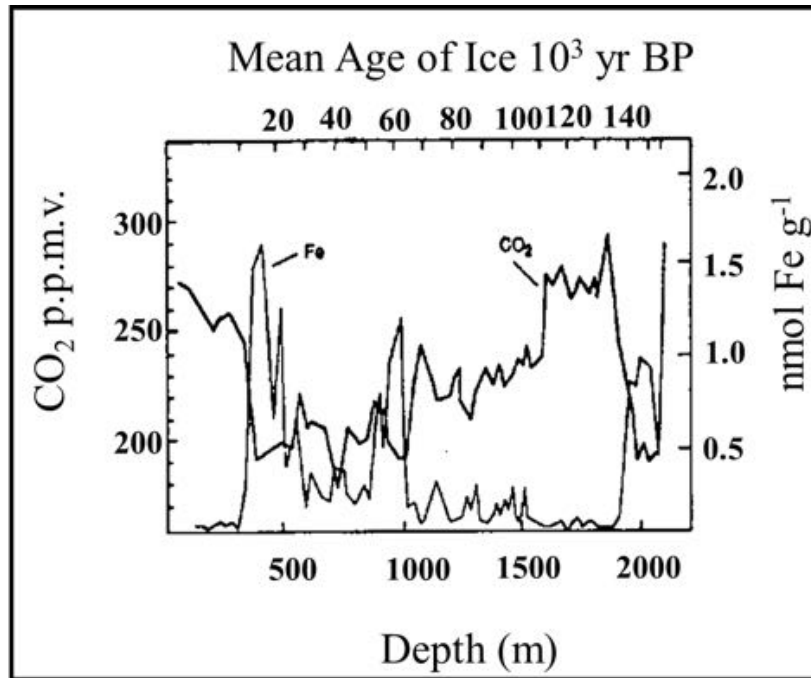
“give me half a tanker of iron, and I will give you the next ice age”



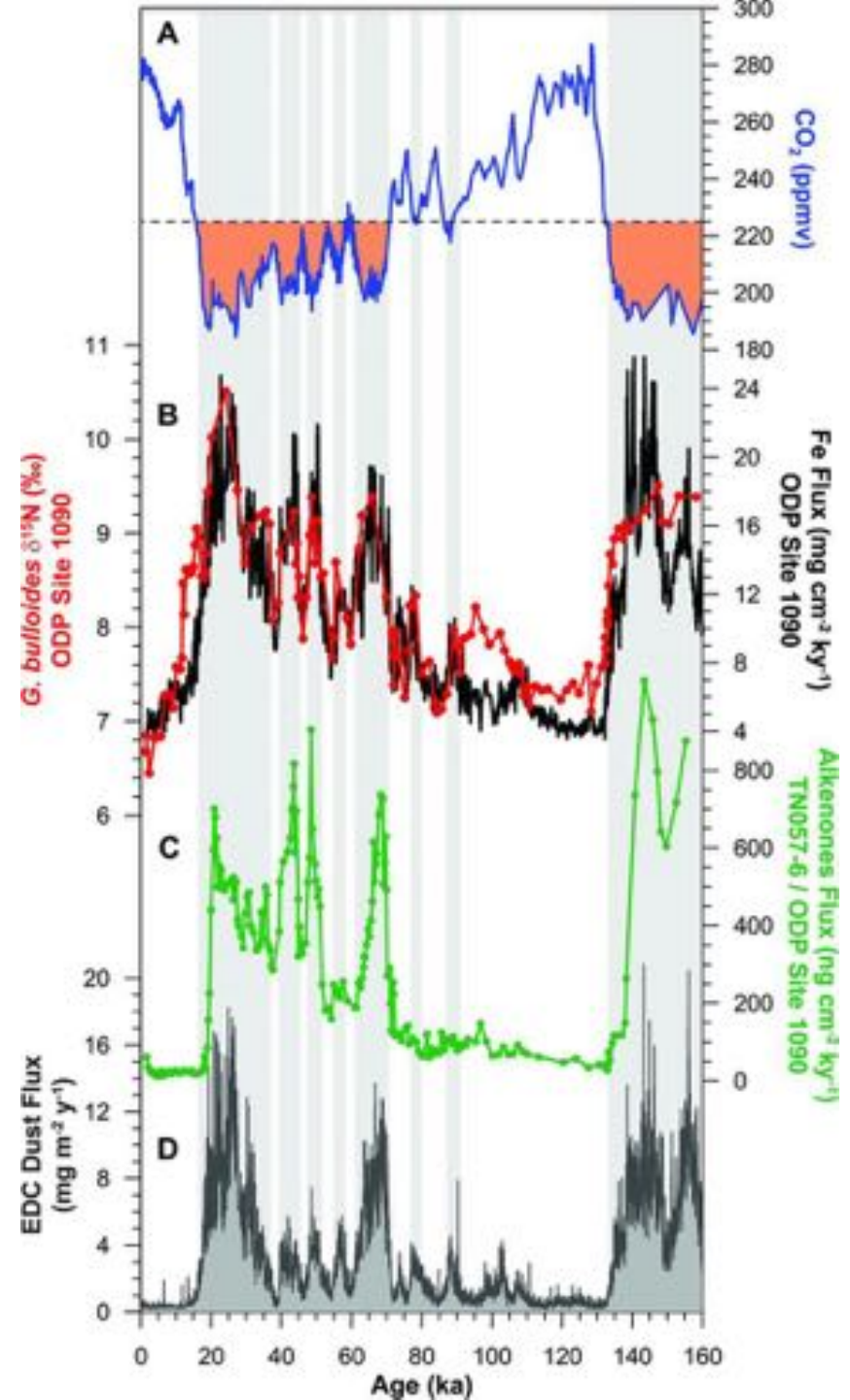
... John Martin
(Colby '62)

Over beer, on the Redfield patio,
Woods Hole Oceanographic
Institution, July 1988

Iron and atmospheric carbon dioxide

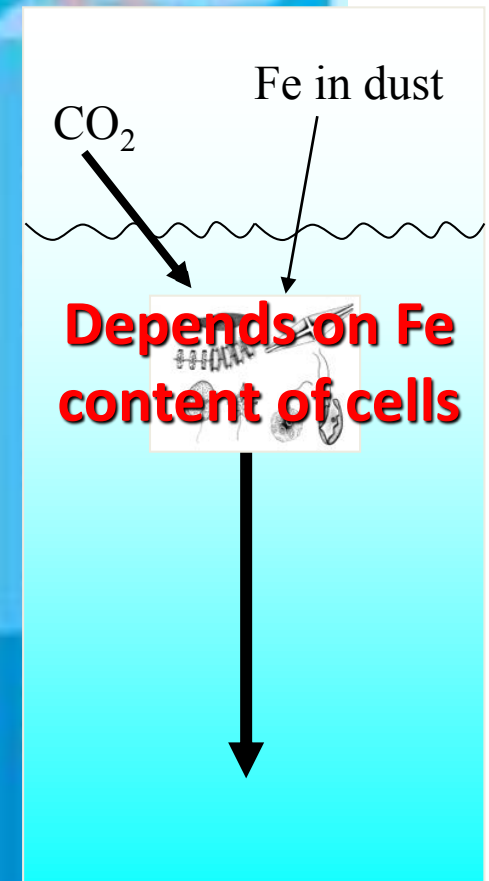
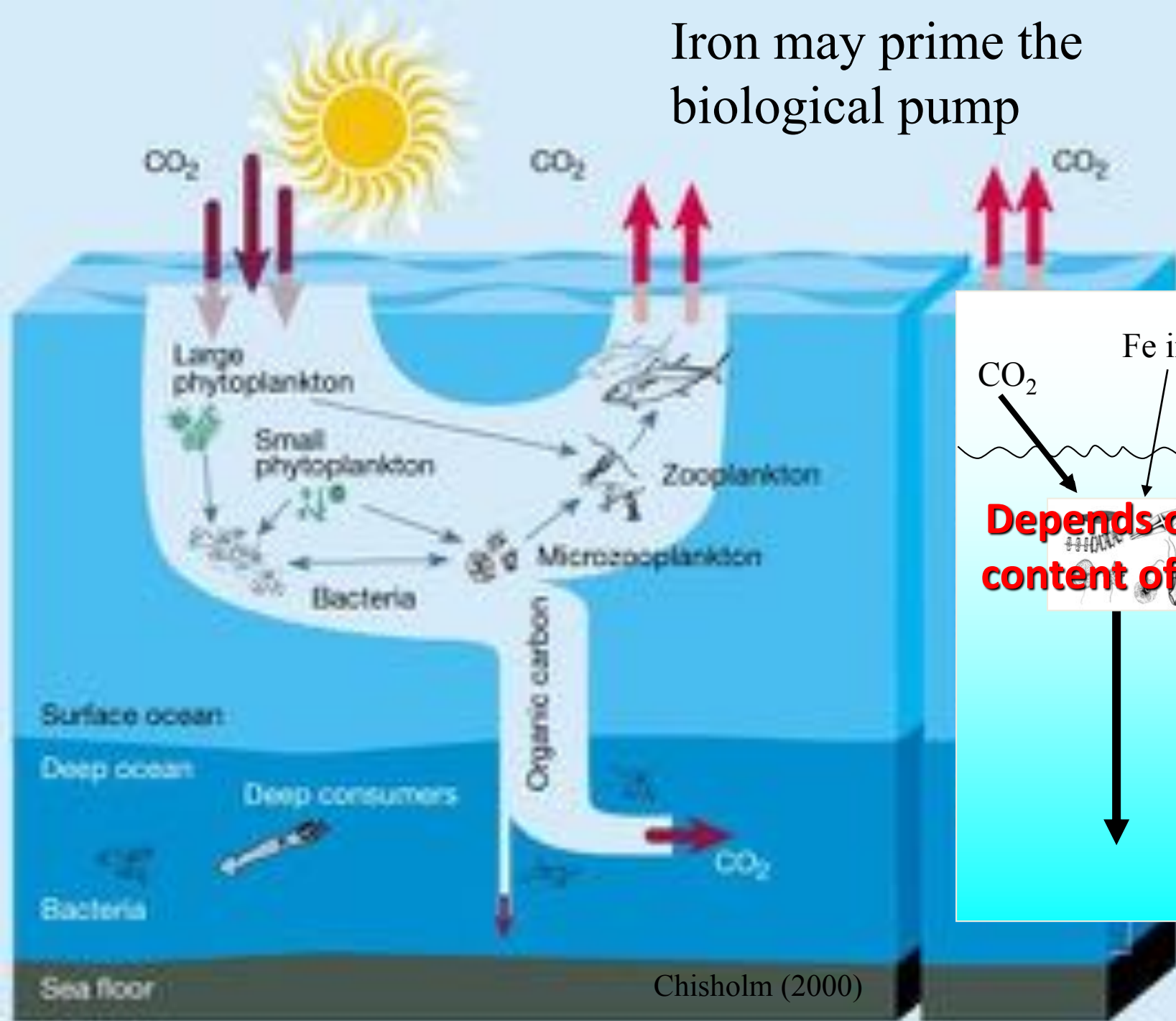


(Martin 1990)

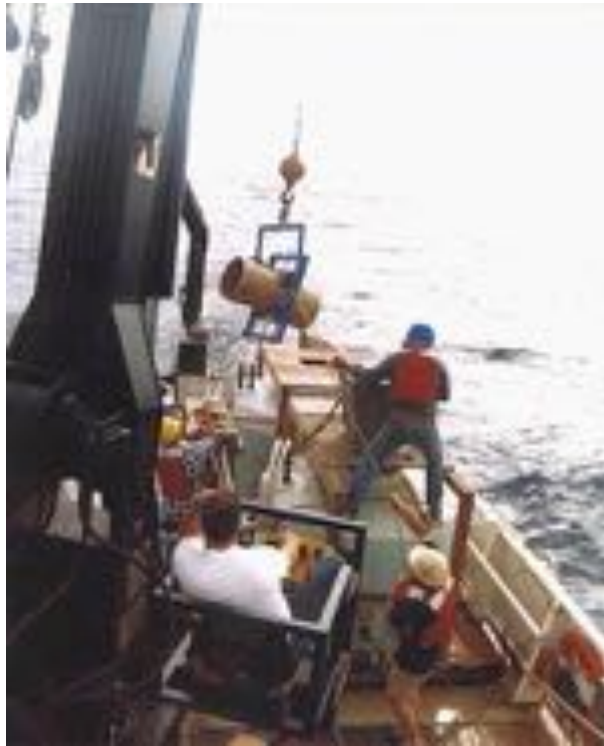


(Martinez-Garcia et al. 2014)

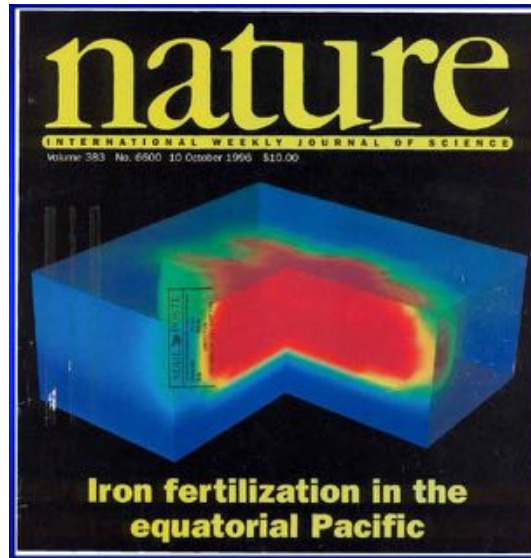
Iron may prime the biological pump



In situ iron addition experiments



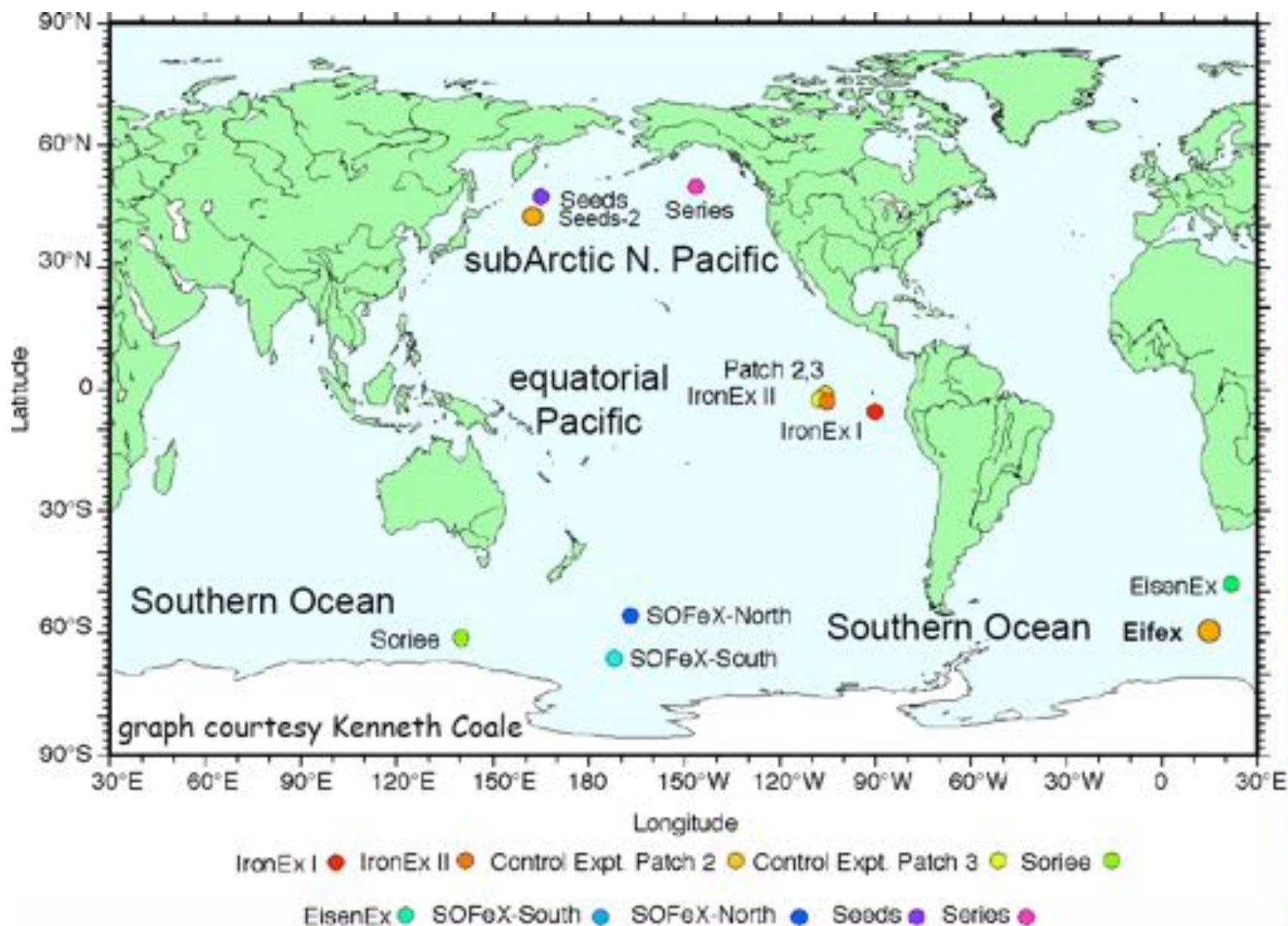
IRONEX, 1993



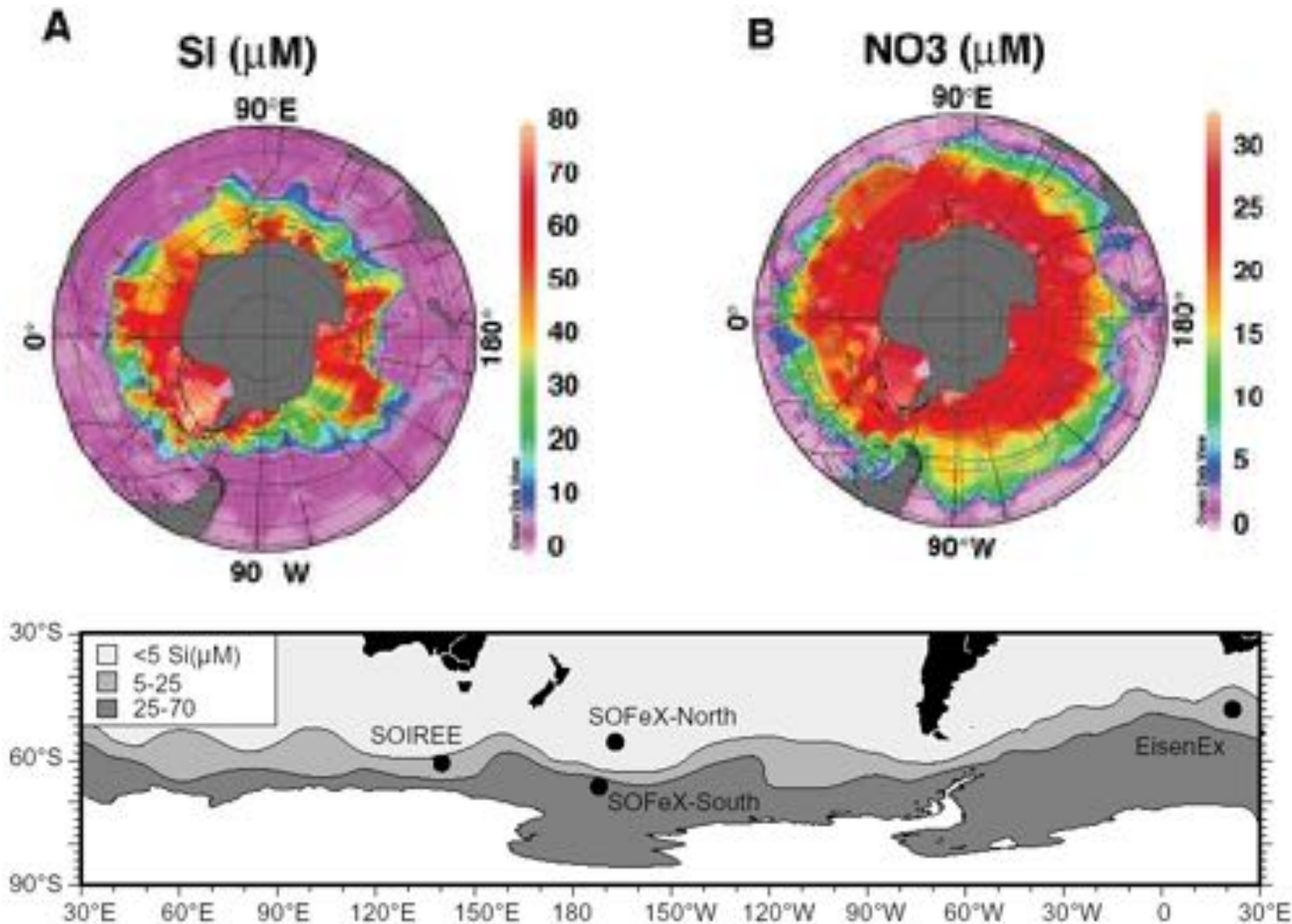
IRONEX II, 1995



SOFEX, 2002

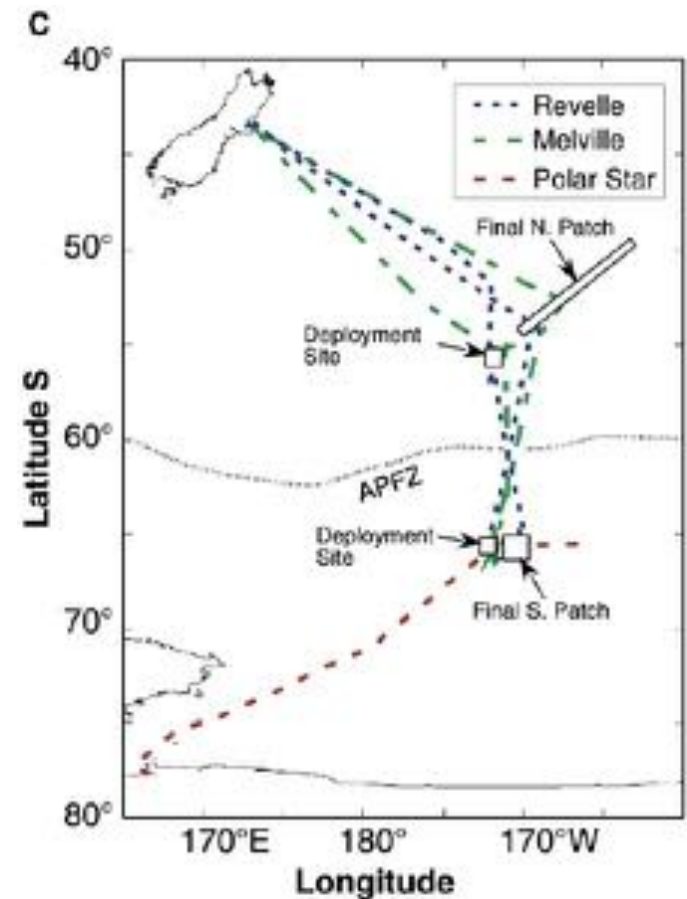


Southern Ocean Iron Experiment



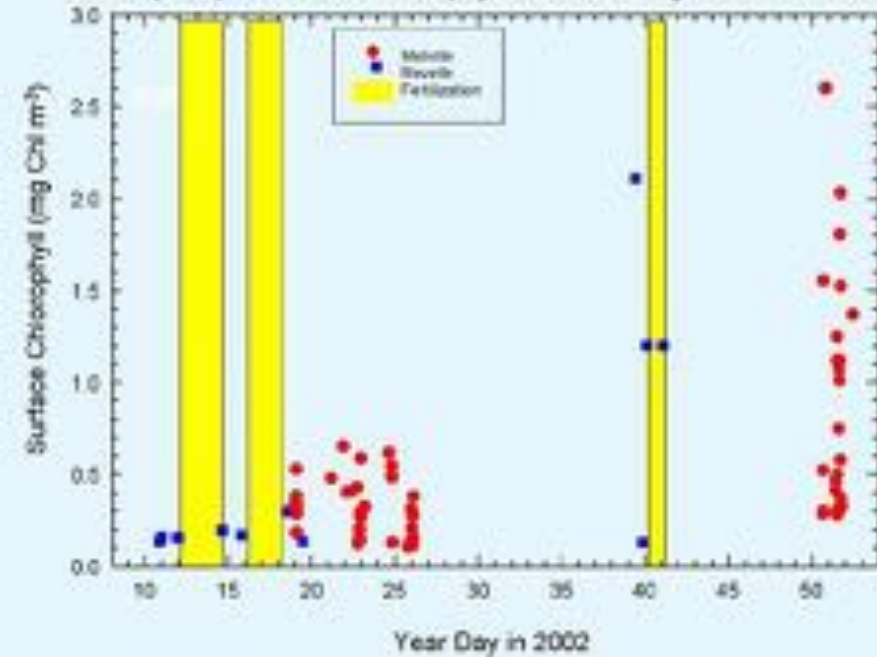
SOFeX cruise plan

- Austral summer: Jan-Feb
- Fertilized two patches of water:
 - North: high N, P; low Si
 - South: high N, P, Si
- Plankton samples collected inside and outside each patch for x-ray fluorescence analysis

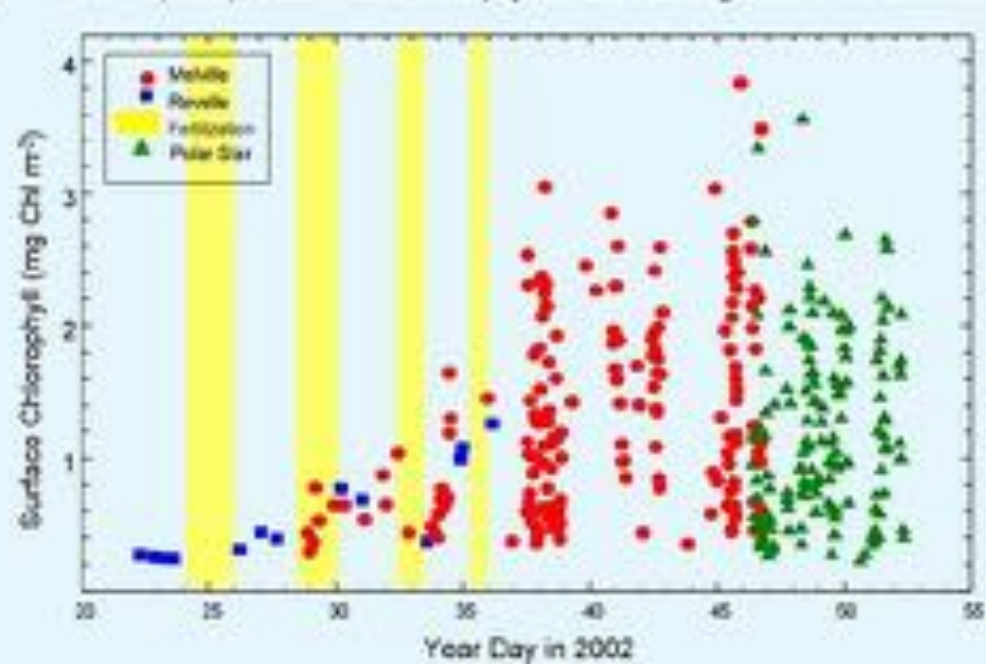


Response of phytoplankton community

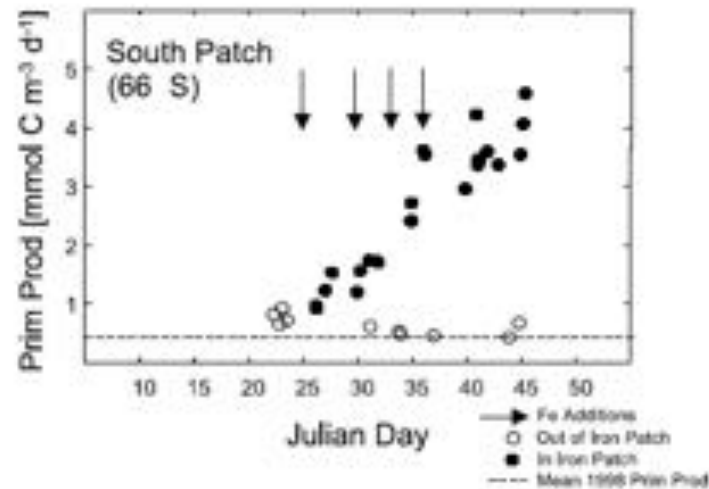
SOFEX (2002) North Patch Chlorophyll increase during and after Fe addition



SOFEX (2002) South Patch Chlorophyll increase during and after Fe addition

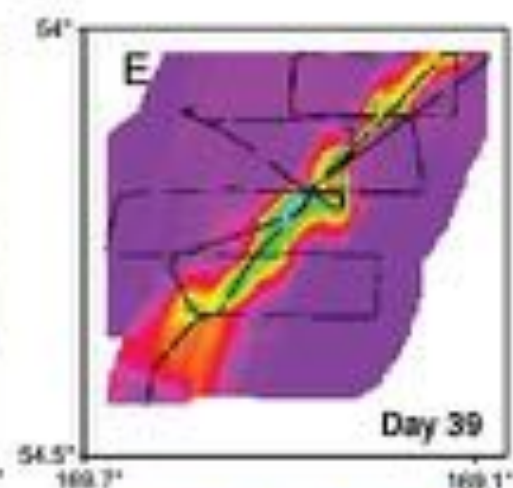
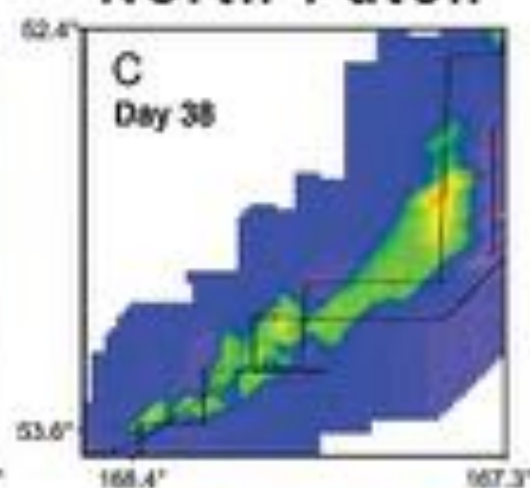
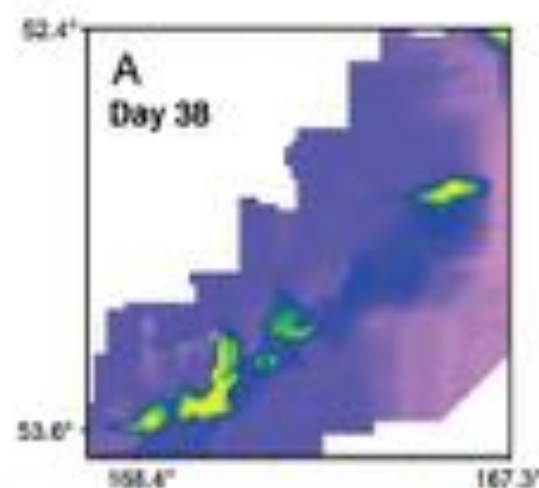


North



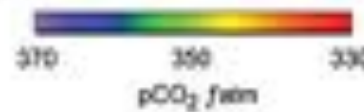
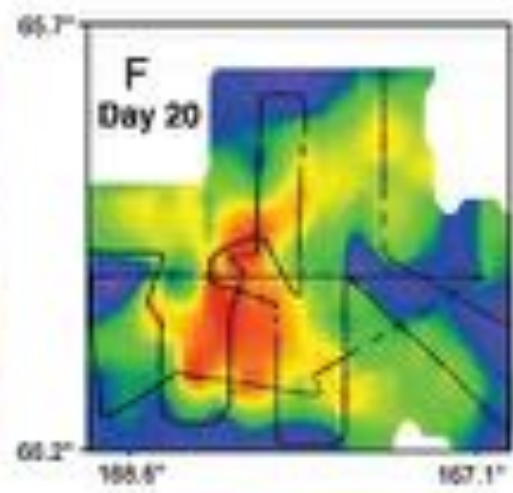
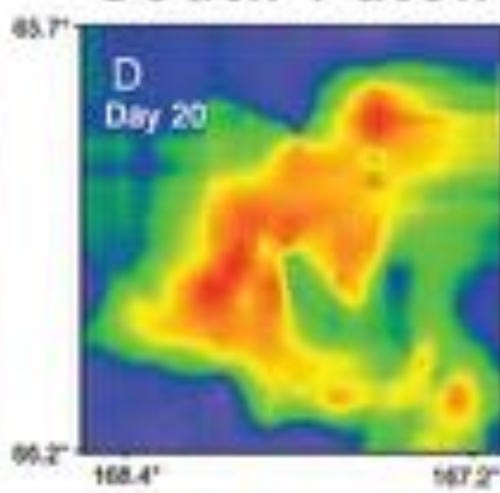
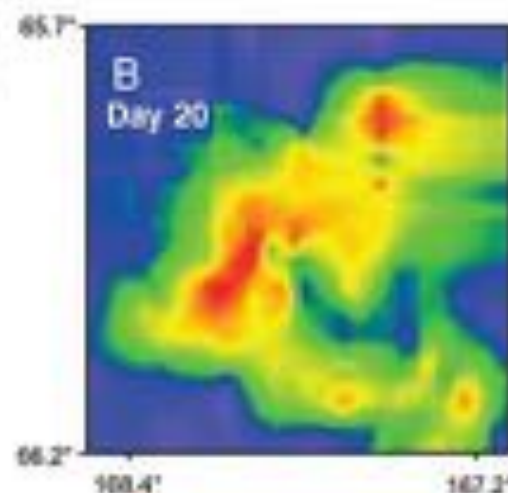
South

North Patch



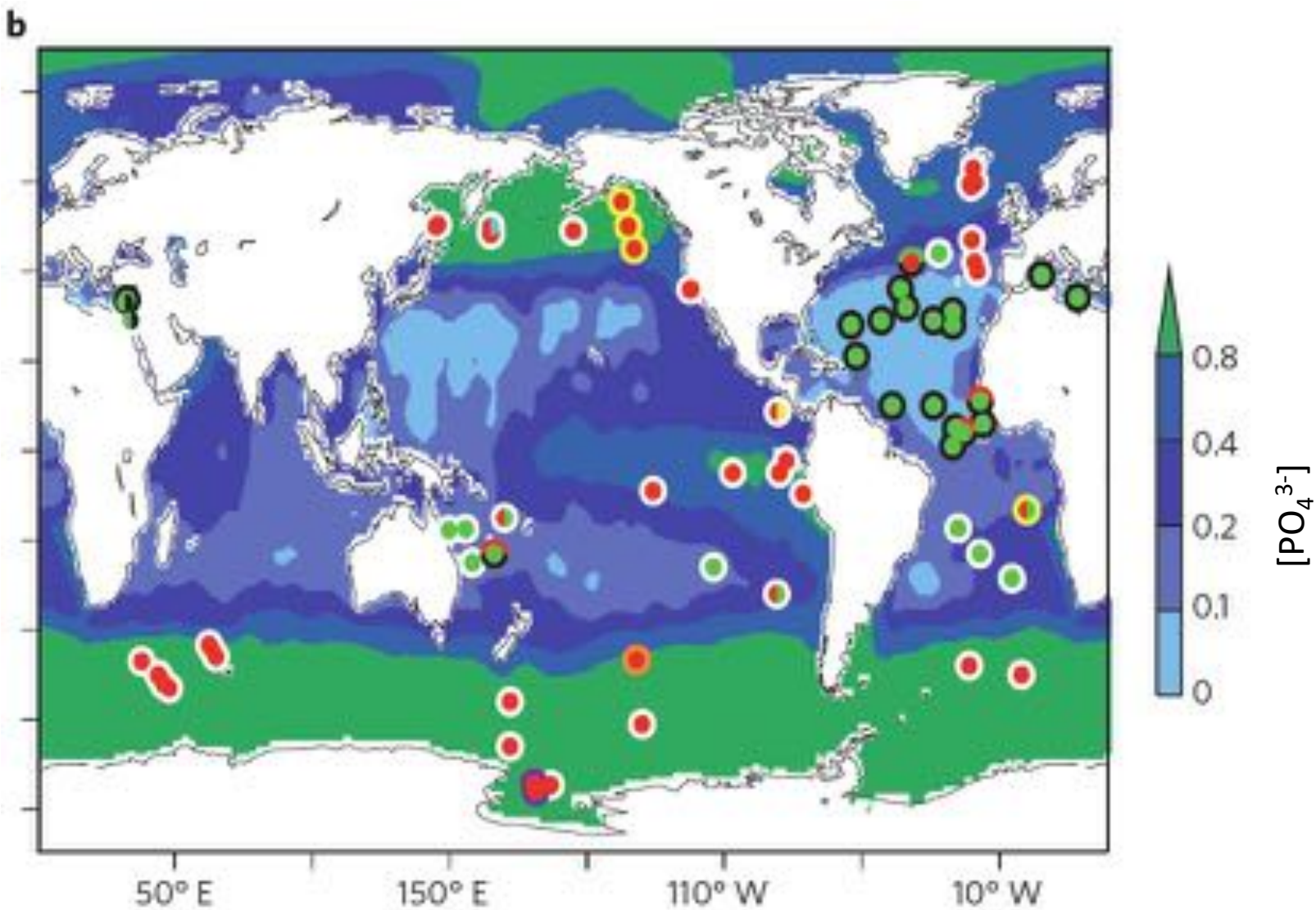
Latitude S

South Patch



Longitude ($^{\circ}\text{W}$)

Metal limitation of primary production



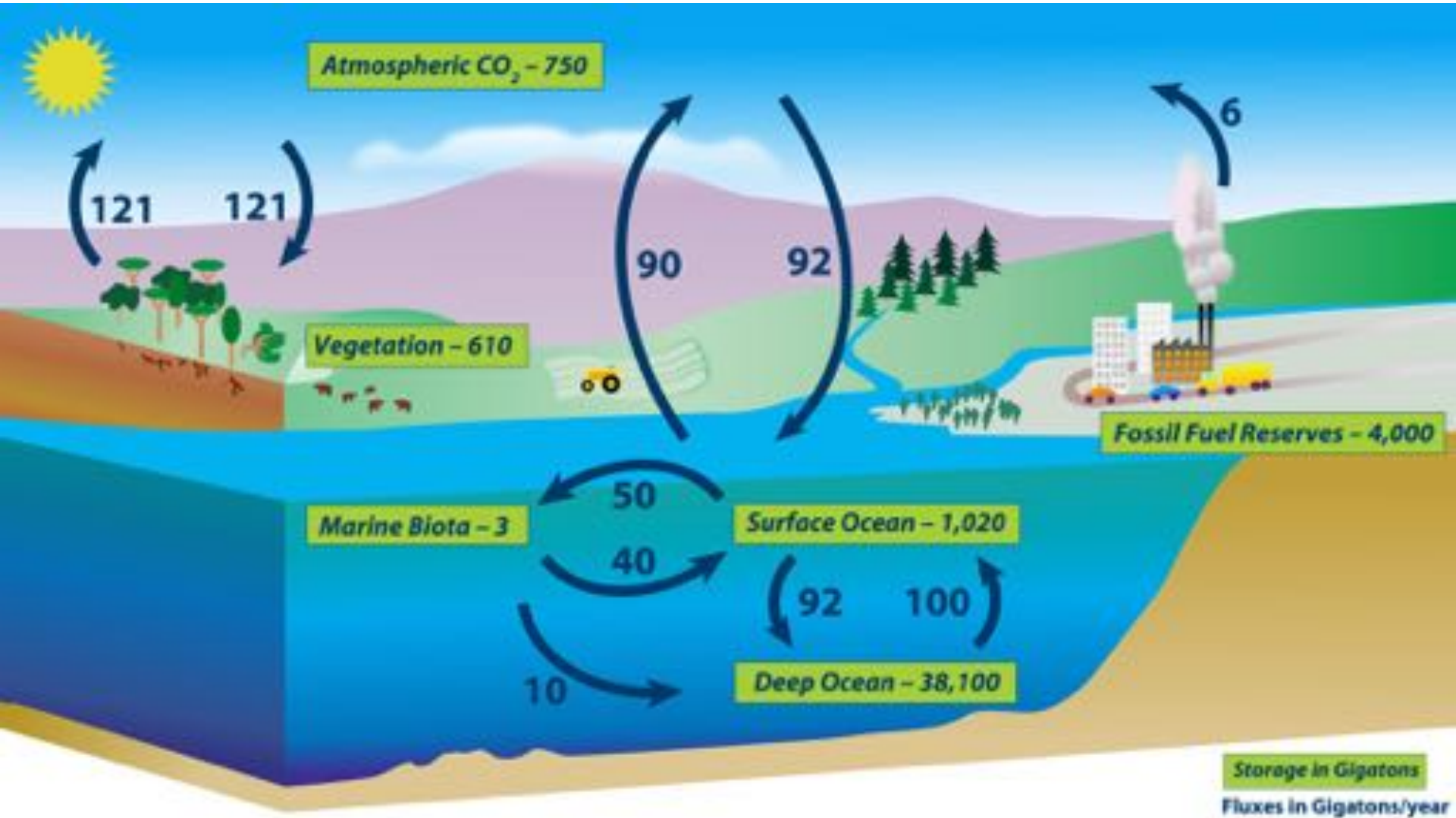
Limiting nutrient: N P Fe Co Si

(Moore et al. 2013 *Nature Geo.*)

Climate geoengineering

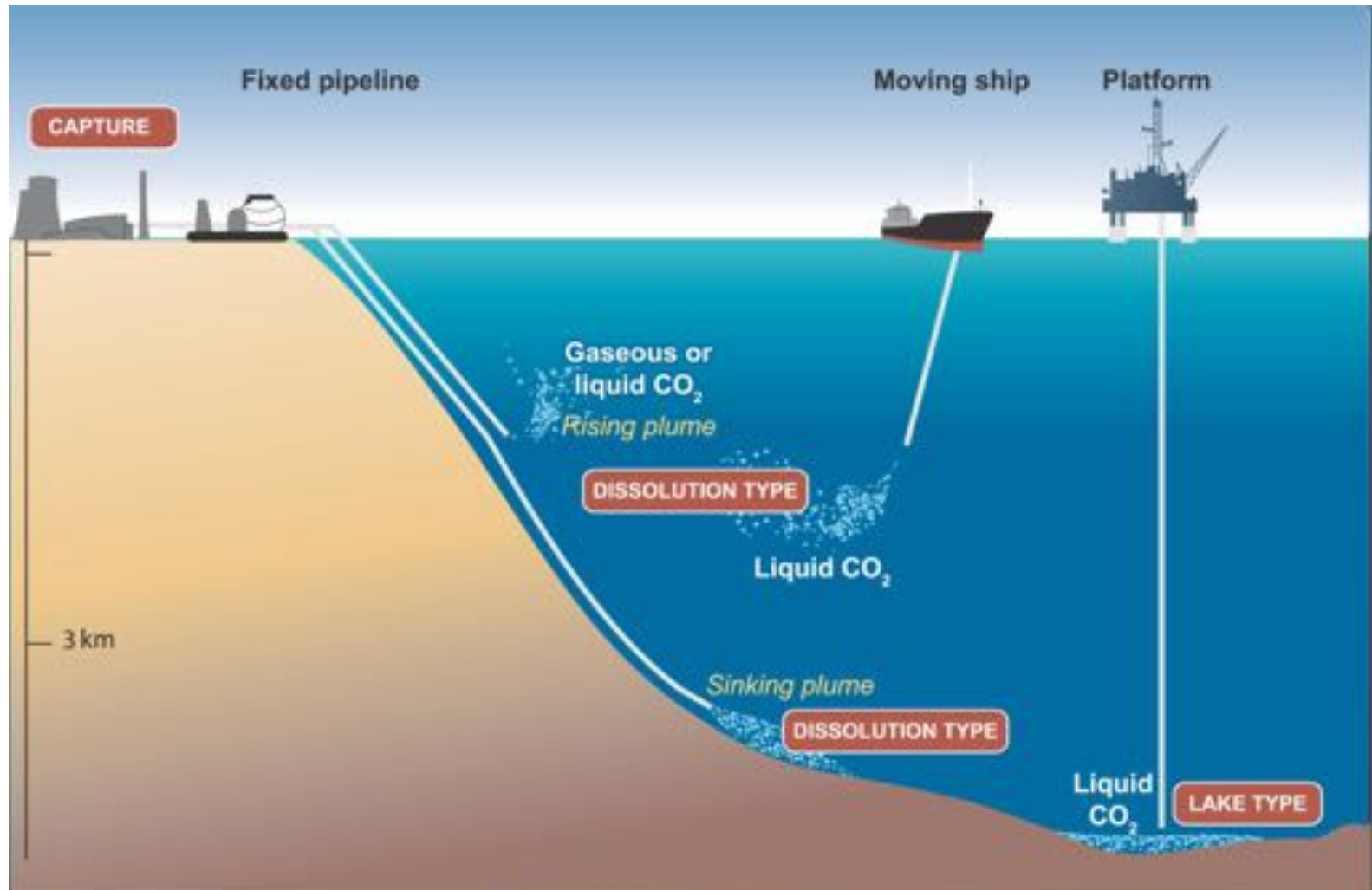
- National Academy of Sciences: “climate intervention”
- Albedo modification
- Carbon sequestration

The Global Carbon Cycle



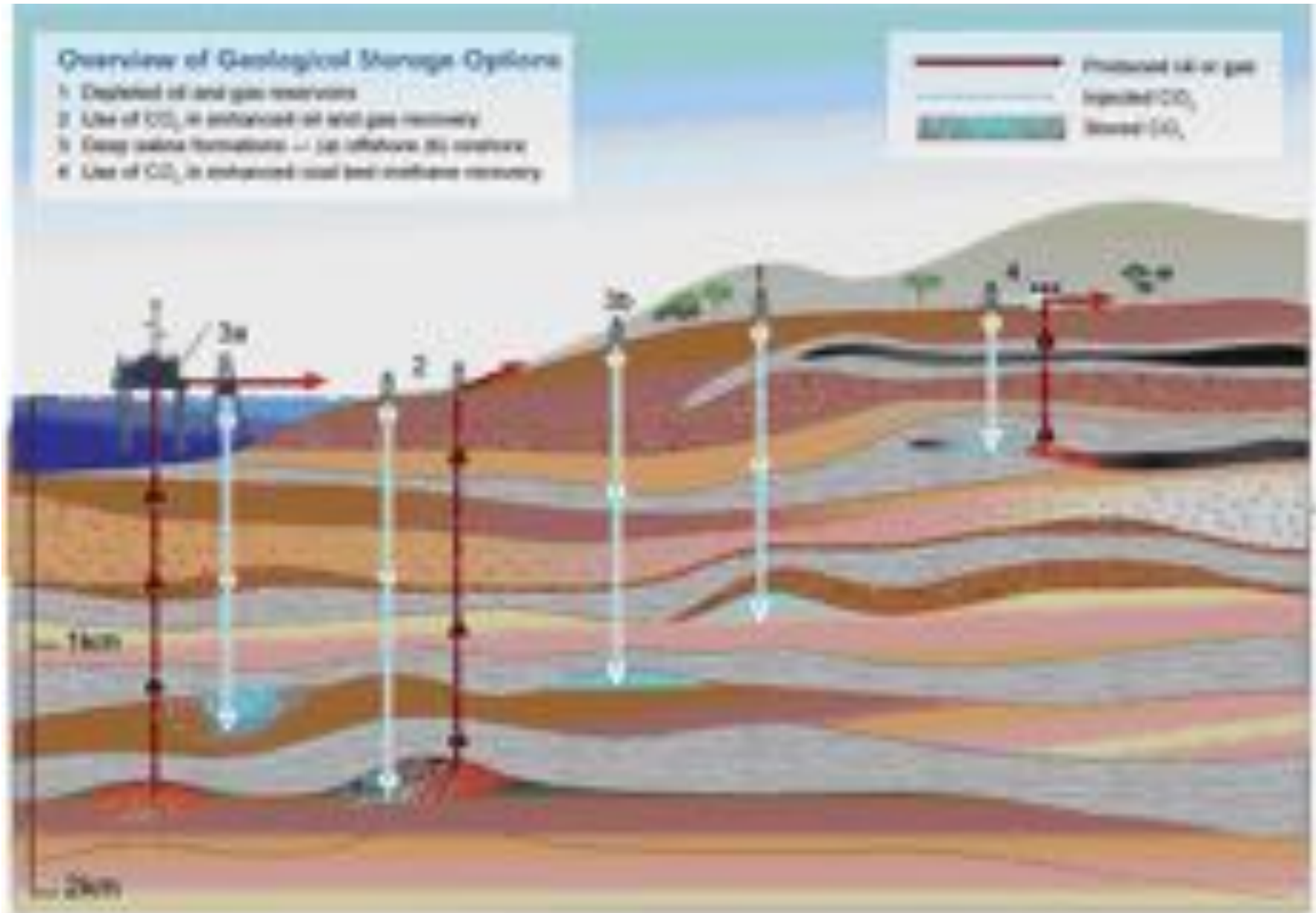
Storage in Gigatons
Fluxes in Gigatons/year

Direct oceanic CO₂ sequestration

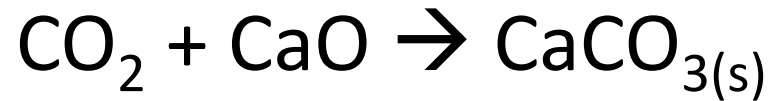
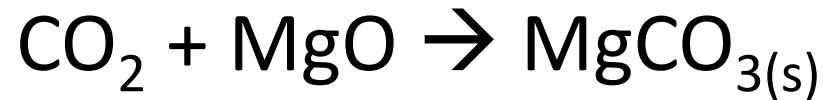


Potential impacts: 1) change in ocean pH at location of input; 2) benthic kills / other ecosystem impacts; 3) eventual release of CO₂ back to atmosphere due to MOC

Geological CO₂ sequestration

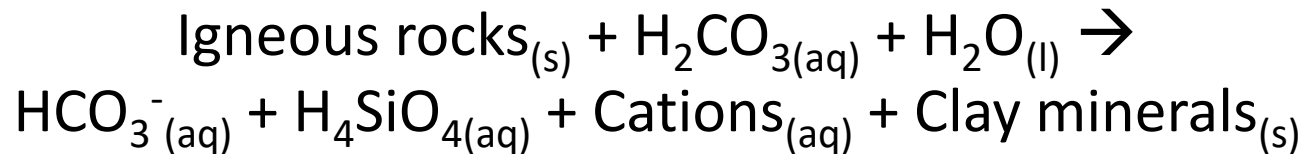


Mineral carbonation



Then store resulting salts on seabed (above calcium carbonate compensation depth)

Enhance chemical weathering



Generate acid from seawater using solar energy and spray on coastal volcanic rocks. Uses CO_2 ; generates HCO_3^-

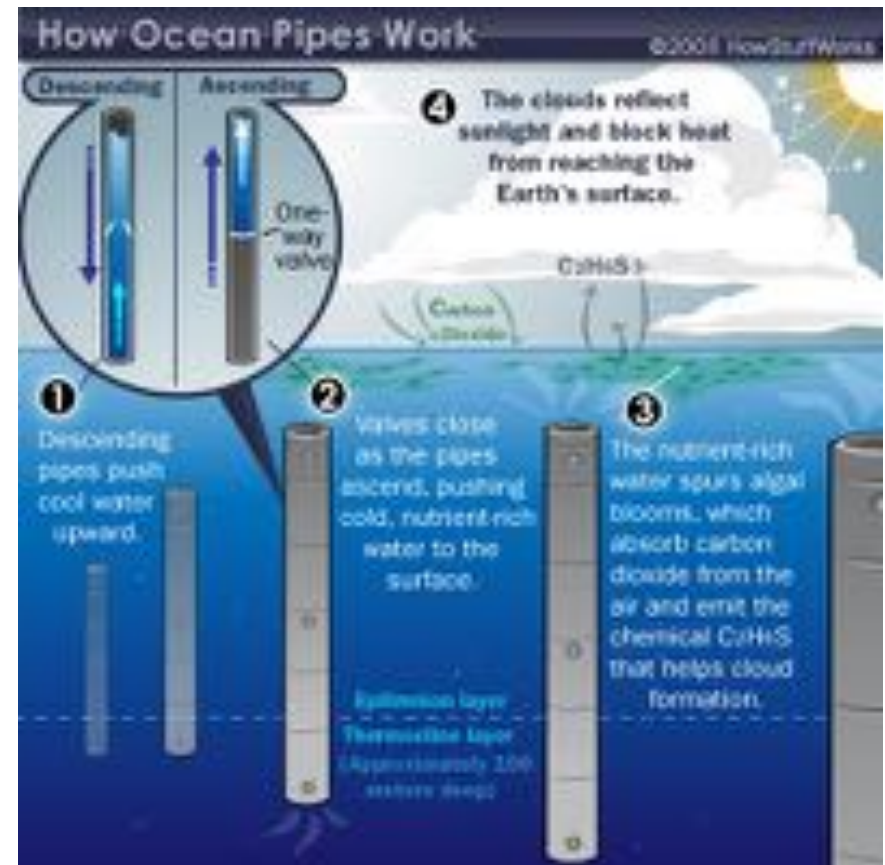
Ocean iron fertilization motivations

- Mitigate atmospheric CO₂ emissions by using the ocean to store carbon
- Increase fisheries productivity

Ocean fertilization

Ocean Nourishment Corporation: fertilization of surface ocean waters with urea and other N compounds

Atmocean, Inc.: “wave energy sequestration technology”



Ocean iron fertilization

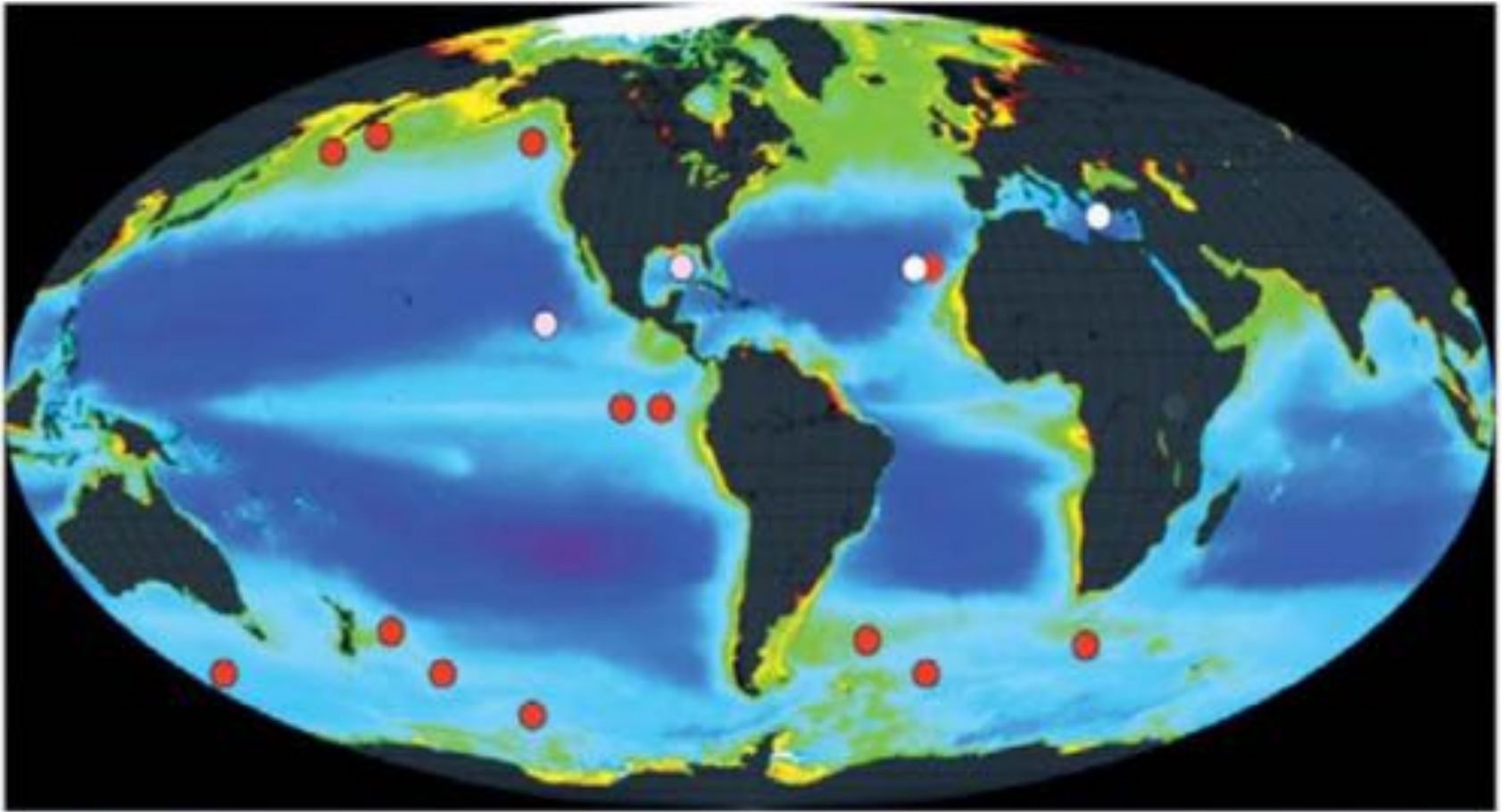
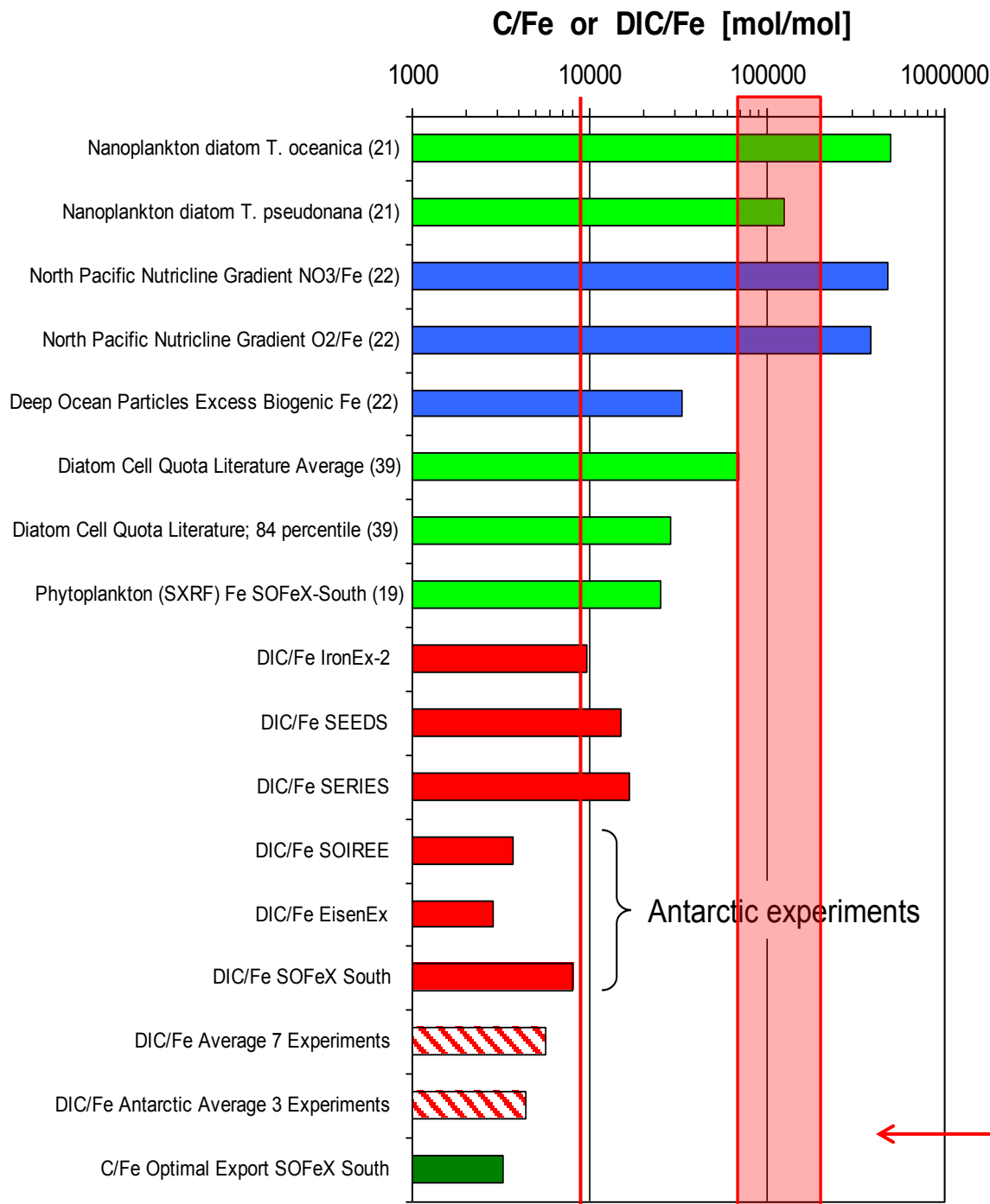
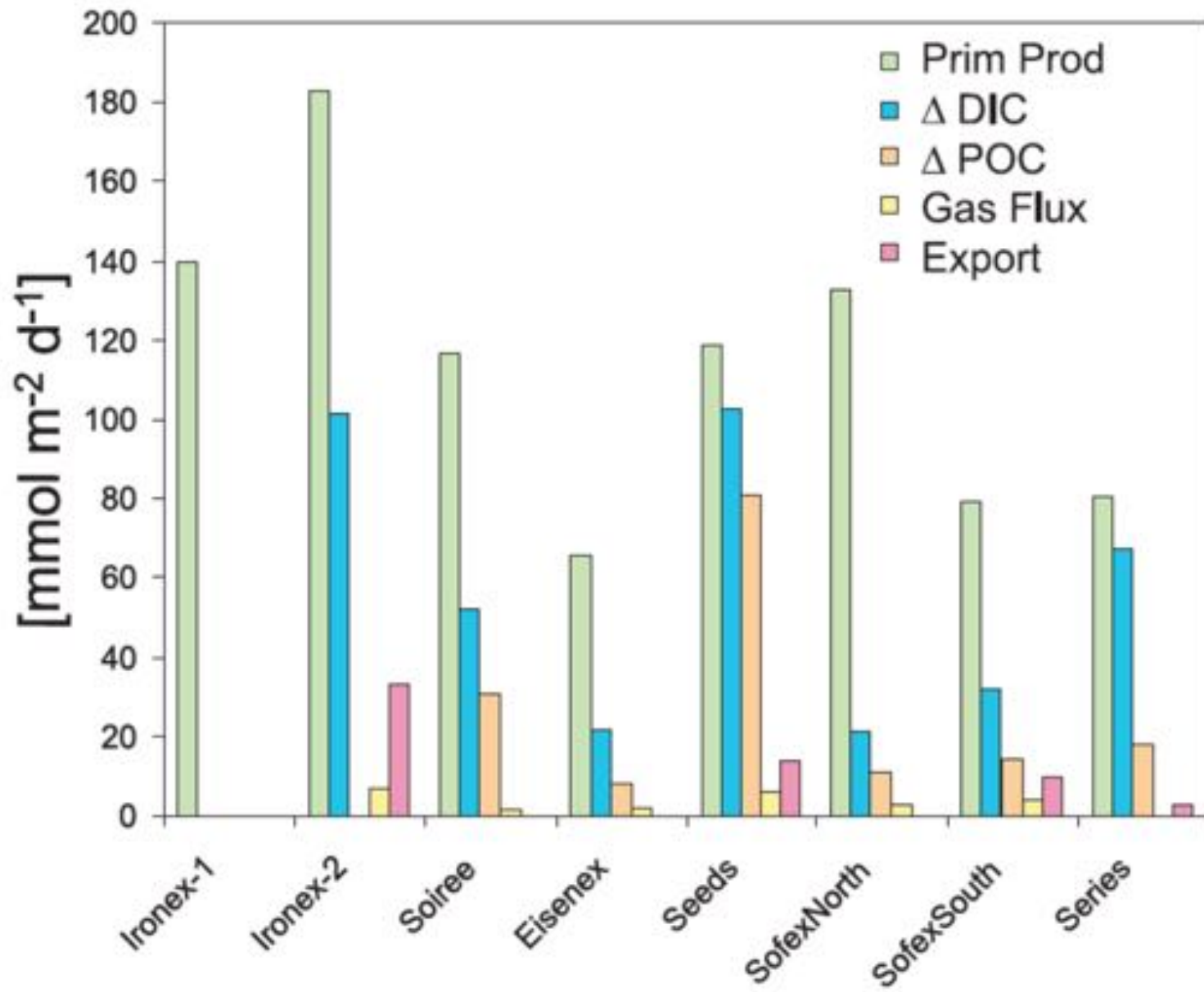


Fig 2. Sites of the 13 iron fertilization experiments (red), two commercial trials using iron (pink) and two phosphate addition studies (white) carried out to date, on map of satellite-based ocean primary production (yellow/green, high; dark blue, low).



Sequestration
efficiency
(C removed per
mole Fe added)

So will it work?



Modeling effects of OIF

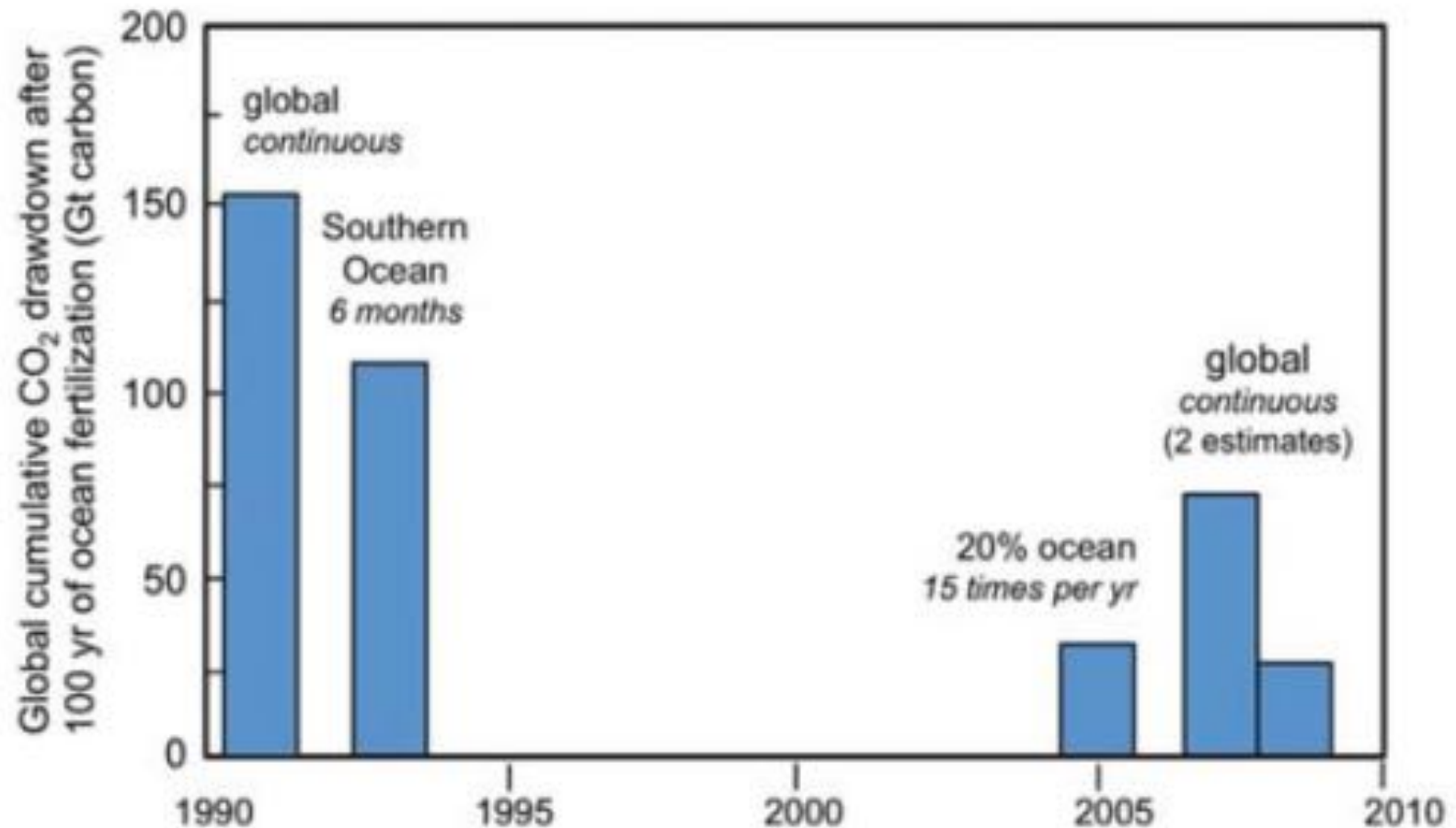
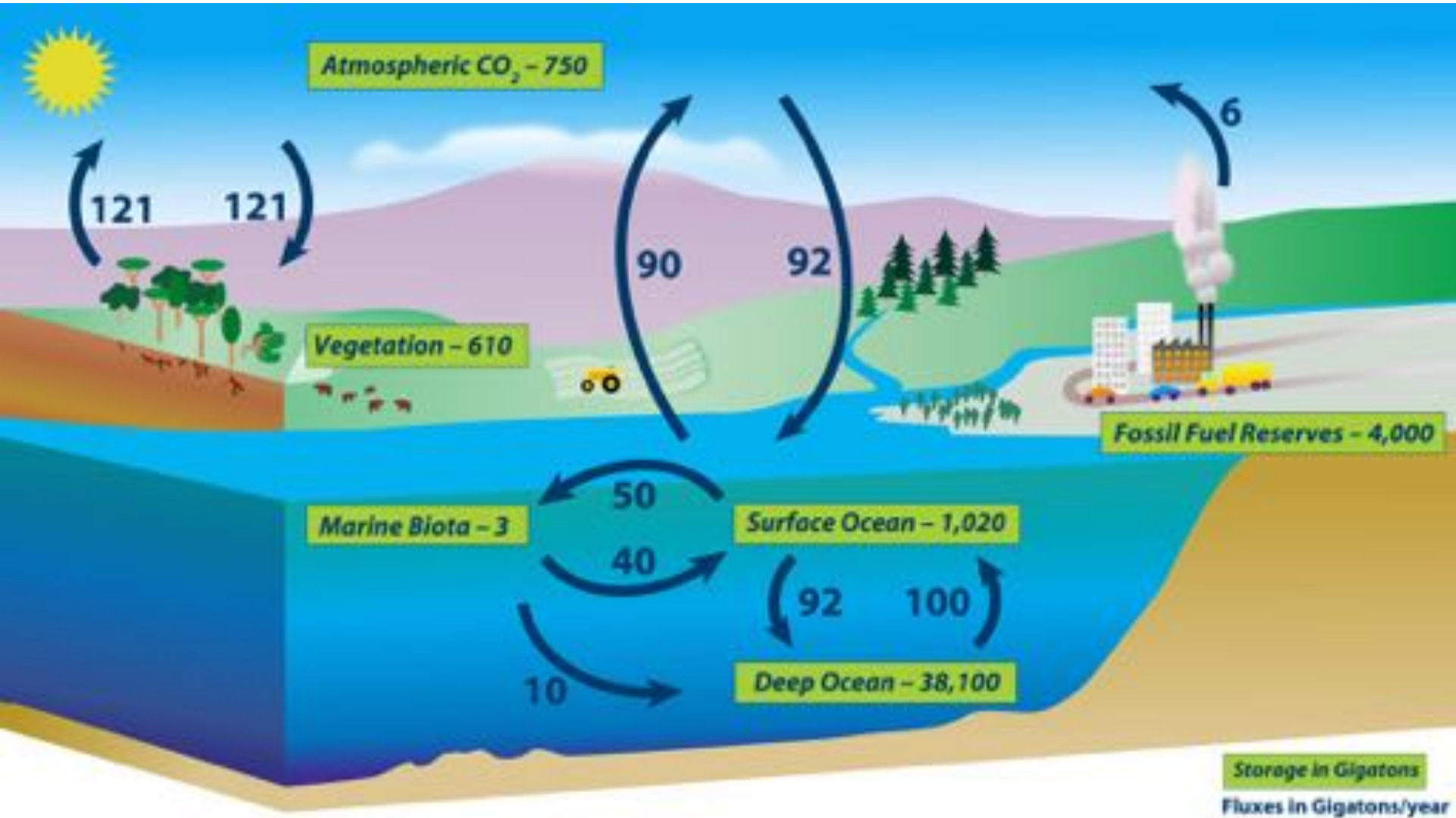


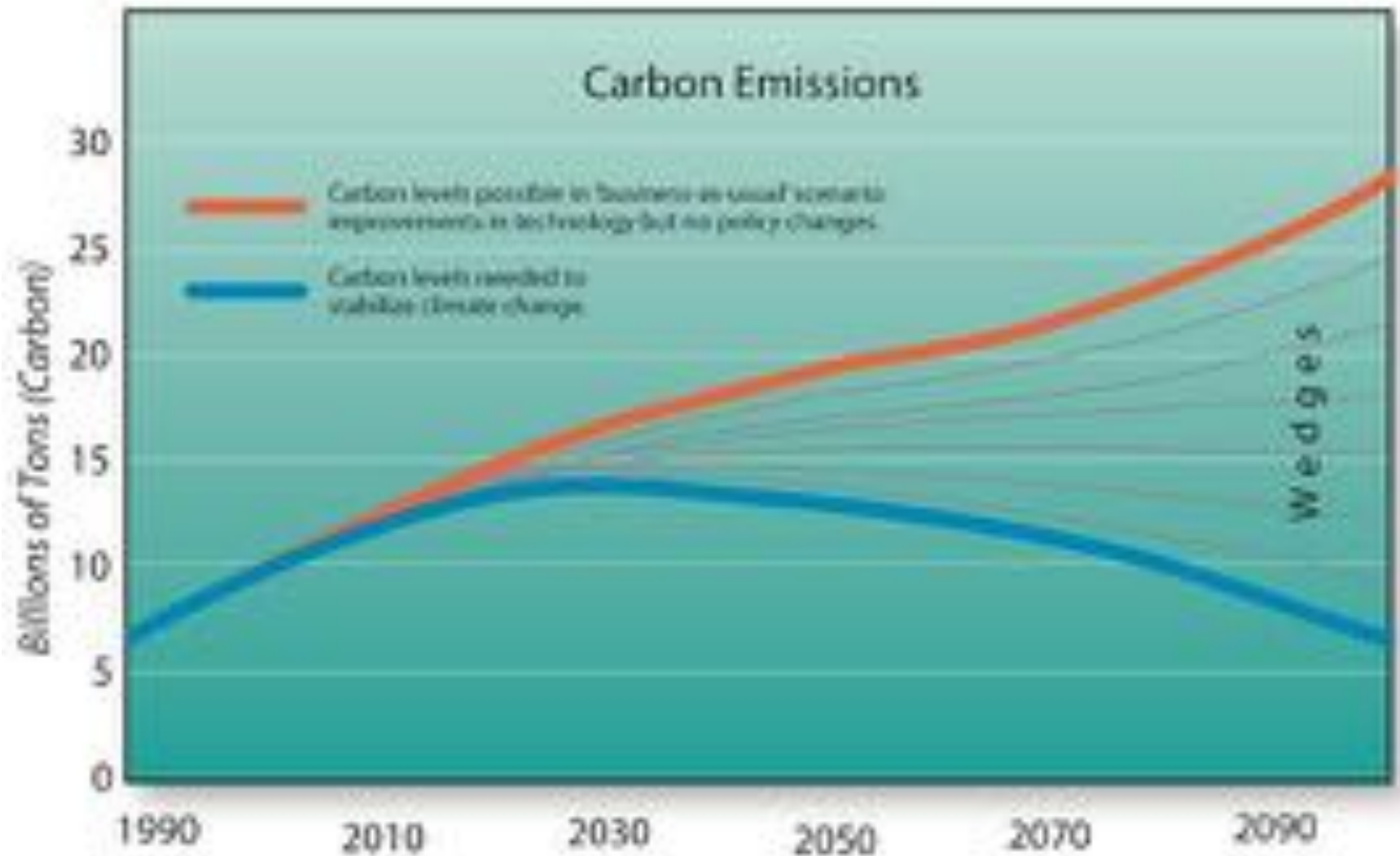
Fig 4. Model-based estimates of the effectiveness of carbon sequestration (cumulative drawdown over 100 yr) for large-scale, iron-based ocean fertilization. Dates relate to year of publication.

The Global Carbon Cycle



Storage in Gigatons
Fluxes in Gigatons/year

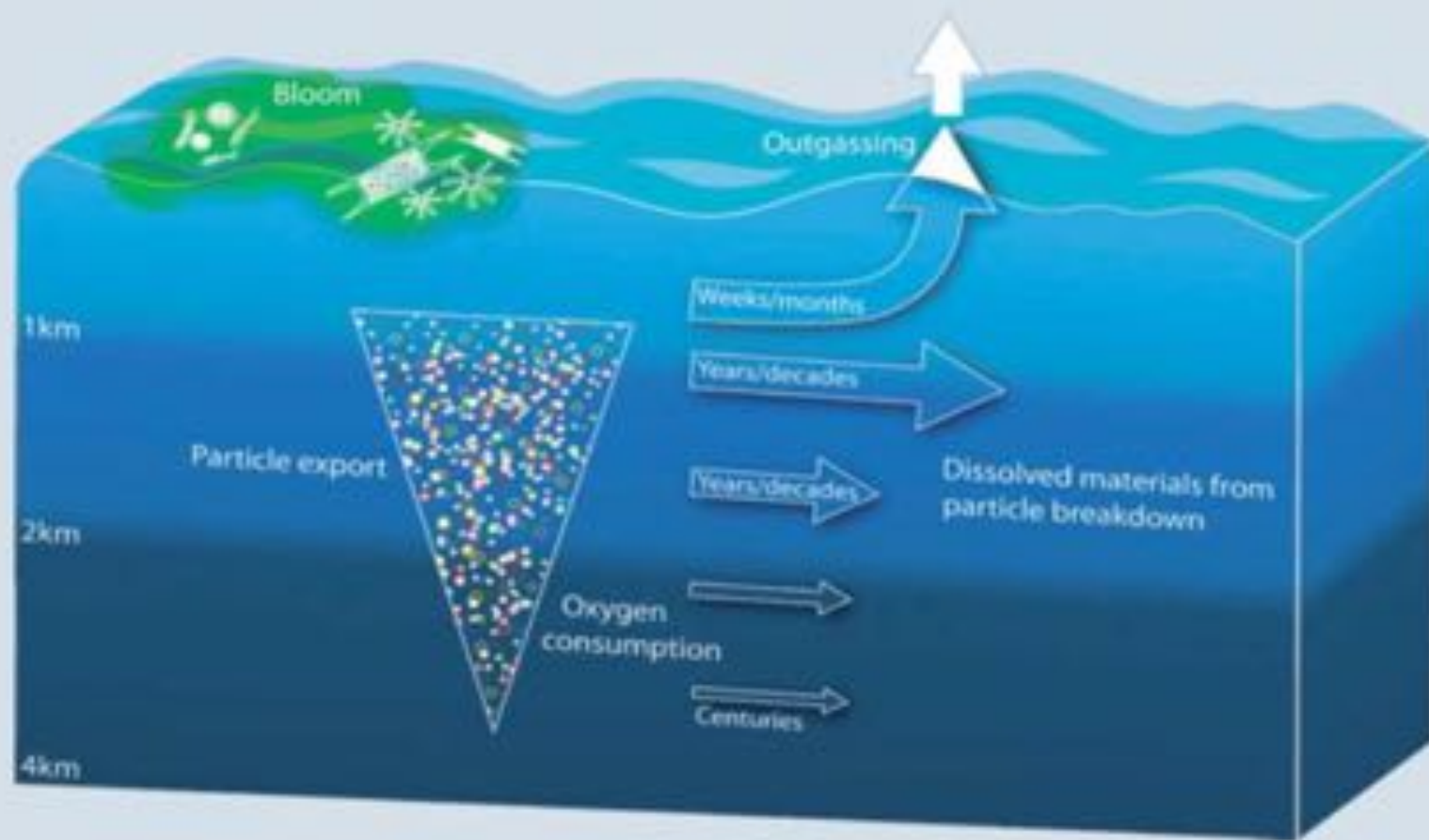
A 'stabilization wedge'?



Potential negative effects

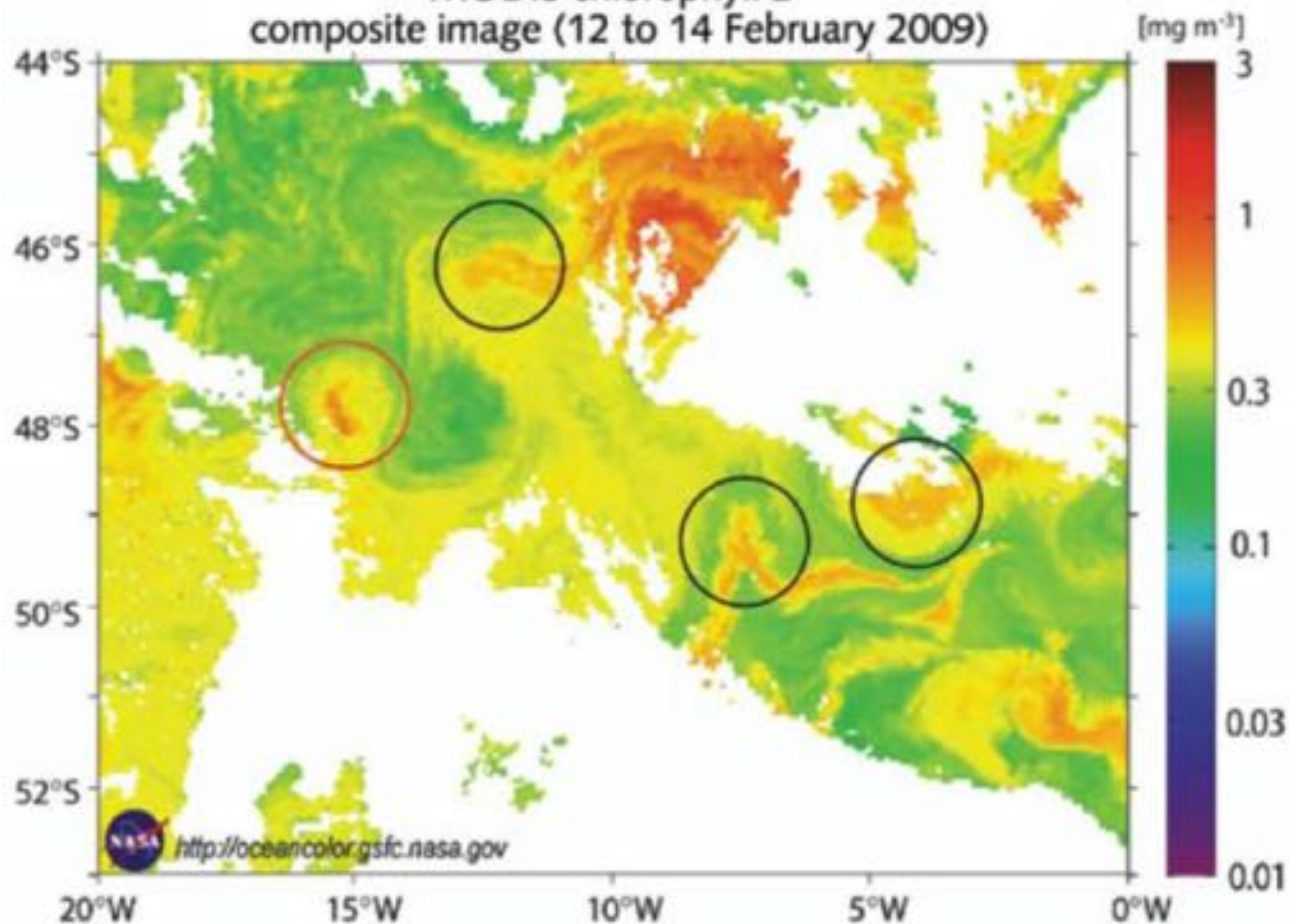
- Removal of oxygen at depth
- Generation of N_2O and methane
- 'nutrient robbing' from other regions
- Changes in food webs
- Difficult to verify C export

The importance of transport and timescales



Vertical and horizontal transport processes over a range of timescales affect the fate of biologically-fixed carbon in the ocean

MODIS chlorophyll a
composite image (12 to 14 February 2009)



Carbon Dioxide Removal proposals...	Albedo Modification proposals...
... address the cause of human-induced climate change (high atmospheric GHG concentrations).	...do not address cause of human-induced climate change (high atmospheric GHG concentrations).
...do not introduce novel global risks.	... introduce novel global risks.
...are currently expensive (or comparable to the cost of emission reduction).	...are inexpensive to deploy (relative to cost of emissions reduction).
...may produce only modest climate effects within decades.	...can produce substantial climate effects within years.
...raise fewer and less difficult issues with respect to global governance.	...raise difficult issues with respect to global governance.
...will be judged largely on questions related to cost.	...will be judged largely on questions related to risk.
...may be implemented incrementally with limited effects as society becomes more serious about reducing GHG concentrations or slowing their growth.	...could be implemented suddenly, with large-scale impacts before enough research is available to understand their risks relative to inaction.
...require cooperation by major carbon emitters to have a significant effect.	...could be done unilaterally.
...for likely future emissions scenarios, abrupt termination would have limited consequences	...for likely future emissions scenarios, abrupt termination would produce significant consequences

Recommendation 1: Efforts to address climate change should continue to focus most heavily on mitigating greenhouse gas emissions in combination with adapting to the impacts of climate change because these approaches do not present poorly defined and poorly quantified risks and are at a greater state of technological readiness.

United States Patent [14]

Markles, Jr.

US000433173A

[11] Patent Number: 5,433,173

[21] Date of Patent: Jul. 18, 1995

[54] METHOD OF IMPROVING PRODUCTION OF SEAFOOD

[76] Inventor: Michael Markles, Jr., 1818 Greer Ln., Alexandria, Va. 22307

[21] Appl. No.: 084,574

[22] Filed: Apr. 28, 1994

[51] Int. Cl.⁷ A01K 41/00

[52] U.S. Cl. 189/209

[56] Field of Search 189/180, 181, 208, 209, 189/11.04, 182, 183, 143, 47/14 B, 14 A2, 14 B2

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"Testing the First Hypothesis in Formation of the Equatorial Pacific Ocean", J. H. Martin et al., Nature vol. 370, pp. 133-139 (Sep. 8, 1994).

Primary Examiner—Todd E. Marlette
Attorney, Agent, or Firm—Davis & Thomas

[57] ABSTRACT

A method of improved production of seafood comprising (I) moving the water at the surface of the ocean in order to decrease the nutrients that are missing, (II) applying a fertilizer that contains the missing nutrients to fertilize the surface of the ocean, and (III) harvesting the increased production of seafood that results from the fertilization.

14 Claims, No Drawings

US Patent Application for **PROCESS AND METHOD OF SUSTAINABLE IMPROVEMENT OF SEAFOOD PRODUCTION IN OCEAN WATERS (#20170360065)**

PROCESS AND METHOD OF SUSTAINABLE IMPROVEMENT OF SEAFOOD PRODUCTION IN OCEAN WATERS

Dec 8, 2015

Disclosed is a method and process for manifesting sustainable improvement in fisheries productivity in Ocean waters. This method and process includes (1) selecting a location of the Ocean that is considered both High Nutrient Low Chlorophyll (HNLC), (2) that the location is within proximity to fisheries feeding grounds or migratory routes or within areas that are considered to be fish feeding areas, (3) within this location, a



Ocean-fertilization project off Canada sparks furore

Bid to boost salmon stocks relied on hotly debated science and dubious carbon credits.

BY JEFF TOLLEFSON

When a chartered fishing boat strewed 100 tonnes of iron sulphate into the ocean off western Canada last July, the goal was to supercharge the marine ecosystem. The iron was meant to fertilize plankton, boost salmon populations and sequester carbon. Whether the ocean responded as hoped is not clear, but the project has touched off an explosion on land, angering scientists, embarrassing a village of indigenous people and enraging opponents of geoengineering.

The first reports about the project, which appeared in British newspaper *The Guardian* on 15 October, presented it as a rogue geo-engineering scheme — the largest in history — in “blatant violation” of international treaties. Critics suggested that Russ George, a US



Workers on a Haida Salmon Restoration Corporation boat release iron sulphate into the Pacific Ocean.

October 25, 2012 *Nature*