Lecture 1 - 5th September 2024

density: $\rho = \frac{m}{V} \rightarrow water density: 997 kg/m^3$

Significant Digits

- · all nonzero cligits are significant!
- for quantities less than one, any zero preceding the first non-zero digit is not significant.

 4 eg: 0.00237 has 3 sig figs.
- · in quantities greater than one, any zero following the last non-zero digit is not significant, unless there is a decimal point.

4 eg: 200 has only one sig fig, while 200.00 has four.

Stoichiometry:

Example: what is $7.2 \frac{Btu}{lbm^{\circ}F}$ expressed in units of $Jg^{-1}K^{-1}$? 4 given: $15 = 9.4782 \cdot 10^{-4}$ Btu, 18g = 2.20462 lbm, $18 = 1.8^{\circ}F$

15 2.20462 lbm°F · 9.4782·10⁻⁴ Btu · 2.20462 lbm · 1kg · 10009 · 1K

= 30.1 3K

Example: you are interested in calculating the annual reduction in CO2 emissions by replacing your car with an all-electric vehicle. Assume that you drive an annual distance 14000km. Your car has a fuel effeciency 28 miles

per gallon. The electric car has an average energy consumption of 157 Wh/km. 4 Data for CO2 emission: 25g CO2/kWh, 2.3 kg CO2/L Gos: 14000 km 6.21371.10-4mi 1000 m 1gal $\frac{1000 L}{264.172 gal} \cdot \frac{2.3 kg co_2}{1 L} = 2705 \frac{kg co_2}{4 eac}$ Electric: 14000 km 157 Wh 259 002 1kWh 1000 g $= 55 \frac{\text{kg } \omega_2}{\text{year}}$ reductions: gas - electric = $2705 - 55 = 2650 \frac{\text{kg } \text{Co}_2}{\text{uear}}$