[This question paper contains 4 printed pages.]

Your Roll No.....

Sr. No. of Question Paper: 1352

Unique Paper Code : 3182011101

Name of the Paper : Bio-organic Chemistry

Name of the Course : B.Sc. (H) Biomedical Science

(NEP-UGCF)

Semester : I

Duration: 3 Hours Maximum Marks: 90

Instructions for Candidates

1. Write your Roll No. on the top immediately on receipt of this question paper.

- 2. Attempt Five questions in all.
- 3. Question No. 1 is compulsory.
- 4. Attempt sub-parts of questions together.
- 5. Give example and illustrations and Chemical structures wherever required.
- 1. (a) Define (Any five)
 - (i) Geometrical isomer
 - (ii) Conditionally essential amino acids
 - (iii) Peptide group

- (iv) Phospholipids
- (v) Racemic modification
- (vi) Conformation

 $(5 \times 2 = 10)$

- (b) Differentiate (Any three)
 - (i) Glycogen and cellulose
 - (ii) Cis and trans fatty acid
 - (iii) Sawhorse and Newman projections
 - (iv) Monosaccharides and Disaccharides $(3\times4=12)$
- (c) Give the structure of the following:
 - (i) α (1 \rightarrow 6) linkage in carbohydrates
 - (ii) PUFA
 - (iii) Zwitter ion
 - (iv) Inosine
 - (v) Tryptophan $(4\times2=8)$
- 2. (a) What are anomers? Giving the structure of anomeric forms of glucose, explain what happens when they interconvert in aqueous solution.
 - (b) Draw a labelled diagram of titration curve for Aspartic acid. Answer the following questions after looking at the curve

- (i) Range in which it has buffering capacity
- (ii) Point at which it is a zwitterionic
- (iii) Equation showing successive ionization of groups due to change in pH of medium
 (6,9)
- 3. Give mechanism and biological importance of the following: (Any three)
 - (i) Aldol condensation
 - (ii) Cannizaro reaction
 - (iii) Michael addition
 - (iv) Baeyer Villiger oxidation (5,5,5)
- 4. (a) Giving the structure of sucrose explain why a solution of sucrose gives negative Benedict's test, however upon boiling with concentrated HCI the solution becomes positive for Benedict's Test.
 - (b) Write all the possible isomers for 2,3 dicholopentanoic acid. Classify them as three or erthro isomer. Specify the stereochemical relationship between these isomers.
 - (c) RNA undergoes hydrolysis under basic condition but DNA doesnot. Justify giving the mechanism involved. (5,5,5)

- 5. (a) Give the structure for various conformation of cyclohexane. Give their stability order and justify your answer.
 - (b) What are triacylglycerols? Why are they preferred for storing energy?
 - (c) Enlist five bioactive compounds and their precursor amino acids. (5,5,5)
- 6. (a) Assigning priority order and assign E/Z configuration to each of the following:

(b) Assigning priority order and assign R/S configuration to each of the following:

COOH

$$CI$$
 H
 CI
 H
 III
 III
 III
 III
 III
 III
 III
 III

(200)