This question paper contains 3 printed pages]

S. No. of Question Paper : 5696

Unique Paper Code : 2173012010

Name of the Paper : DSE: NUCLEAR AND ENVIRONMENTAL

CHEMISTRY

Name of the Course : B.Sc. (Hons.) Chemistry

Semester : IV

Duration: 3 Hours Maximum Marks: 90

(Write your Roll No. on the top immediately on receipt of this question paper.)

Attempt six questions in all.

All questions carry equal marks.

- 1. (a) In what way is the mode of decay of a particular nucleus related to the ratio of neutrons to protons? Explain on the basis of Stability Curve.
 - (b) Calculate the energy released per fusion in MeV per fusion 2D_1 + $^1H_1 \longrightarrow ^3H_1$. Given the atomic masses are 2D = 2.01410 amu, 1H = 1.007825 amu, 3H = 3.01603 amu.
 - (c) The half-life period of ³²P is 8 days. What percentage of the original radioactive material will be present after 40 days? 5,5,5

P.T.O.

- 2. Explain the following:
 - (a) Disposal of nuclear waste and its management
 - (b) Different theories on the stability of nucleus.
 - (c) Nuclear fission and fusion reactions.

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- 3. (a) Explain the construction and working principle of Geiger Muller Counter.
 - (b) What are fast breeder reactors? How are they different from conventional thermal reactors? Describe the main components of a fast breeder reactor.
 - (c) How are isotopes separated for use in nuclear reactors? Explain using at least three different methods.

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- 4. (a) What are air pollutants and how are they classified? Elaborate on the major sources contributing to air pollution.
 - (b) Discuss the major regions of Earth's atmosphere. Describe the characteristics of each region.
 - (c) Explain different air pollution control methods. 5,5,5
- 5. (a) Discuss the chemistry and environmental consequences of photochemical smog. How is it formed and what are its main components?
 - (b) What is ozone depletion? Explain the mechanisms involved, highlighting the role of chlorofluorocarbons (CFCs) and other ozone-depleting substances.

- (c) Describe the greenhouse effect and the role of greenhouse gases in global warming. Explain the chemical nature and environmental impact of major greenhouse gases.

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- 6. (a) What is water pollution? What are its impacts on the hydrological ecosystem?
 - (b) What are the key physicochemical parameters used to assess water quality? Explain the significance of each parameter.
 - (c) Explain the working principles and applications of two different water purification methods.

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- 7. (a) Explain the sources and causes of water pollution.
 - (b) Describe the various stages of the hydrological cycle and explain how each stage contributes to the continuous movement and transformation of water within the Earth's system.
 - (c) Explain the effluent treatment process. How is the effluent from different industries treated?

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- 8. Write short notes on any three of the following: 5,5,5
 - (a) α , β and γ
 - (b) Chernobyl disaster
 - (c) Nuclear reactors in India
 - (d) Radioactive Series.

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