Question number	Answer	Notes	Marks
12 (a) (i)	neutron numbers correct; particle X numbers correct;  1  n + N  7	1 14 C + X 6	2
(ii)	proton / p;	allow hydrogen, H, H <sup>+</sup> (ion)	1
(iii)	any two from:  MP1.both have same number of protons (and electrons);  MP2.C-12 has <b>fewer</b> {neutrons / nucleons} than C-14;  MP3.C-12 is lighter than C-14;	both C atoms have 6 protons allow RA C-14 has 8 neutrons, C-12 has 6 neutrons allow RA	2
(iv)	mass number is constant; atomic number increases by one;	however expressed, including numerically	2
(b)	working seen / appropriate line(s) on graph seen; 5 500 (years)	e.g. line drawn across from 125 Bq allow 5000-6000 (years)	2
(c) (i) (ii)	(due to) background radiation; idea that activity depends on the mass;	allow 'fair test' idea ignore 'to have the same activity'	1

Question number	Answer	Notes	Marks
(d)	{activity / amount of C-14} too low (to measure);	allow activity is zero allow no longer emits radiation allow 'all C-14 has gone'	2
	quantitative supporting statement;		
	<ul><li>e.g.</li><li>age of bone is much greater than one half-life</li></ul>		
	<ul> <li>activity becomes zero after 35 000 years</li> <li>C-14 decays fully after 35 000 years</li> </ul>	allow any value given greater than 35 000 years allow any value given greater than 35 000 years	

Total for question 12 = 13 marks