

Question number	Answer	Notes	Marks
2 (a)	10 000; N;	allow 9800, 9810, $10^4$ allow "newton(s)" marks are independent	2
(b) (i)	density = mass / volume;	allow abbreviation, e.g. $\rho = m/V$ , $d = m/V$ or rearrangements	1
(ii)	substitution OR rearrangement; evaluation;  e.g. $2300 = 1000/\text{volume}$ $= 0.43 \text{ (m}^3\text{)}$	award if either seen in working   allow 0.4, 0.434, 0.435, 0.4347... condone 0.44	2
(c) (i)	bar chart / bar graph;	condone histogram	1
(ii)	any 1 from: MP1. idea that (density) data is discontinuous; MP2. materials have non-numerical values / are not quantifiable; MP3. material types identified as categories; MP4. idea that a line graph would indicate continuity;	discrete, categoric, non/not continuous	1
(iii)	cork is less dense OR water is denser; cork 25%, $\frac{1}{4}$ as dense OR water four times denser;	accept correct calculation of <u>both</u> densities for 2 marks	2

**Total 9 marks**

Question number	Answer	Notes	Marks									
3 (a)	<p>any one line correct for one mark; all three lines correct for two marks;;</p> <div><div>state of matter</div><div>particles</div><div><div>solid</div><div>close together, moving about and can slide past one another</div></div><div><div>liquid</div><div>far apart, moving quickly and at random</div></div><div><div>gas</div><div>close together, vibrating about fixed positions</div></div></div>	more than one line to a box does not score	2									
(b) (i)	<p>18; 192;</p> <table><tr><td></td><td>Temperature in °C</td><td>Temperature in kelvin</td></tr><tr><td>room temperature</td><td>18</td><td>291</td></tr><tr><td>triple point of ethyne</td><td>-81</td><td>192</td></tr></table>		Temperature in °C	Temperature in kelvin	room temperature	18	291	triple point of ethyne	-81	192		2
	Temperature in °C	Temperature in kelvin										
room temperature	18	291										
triple point of ethyne	-81	192										
(ii)	decreases / OWTTE;	ignore “molecules slow down”	1									
(iii)	remains constant / no change / nothing;		1									

Total 6 marks