

Question number	Answer	Accept	Reject	Marks
4 (a) (i)	Anomaly clearly identified (20.44 mm);			1
(ii)	Averaging seen $/162.7 \div 8$ $/142.26 \div 7$; Anomaly excluded/ $\div 7$ seen ; Final answer rounded to 2 decimal places; e.g.: 20.32 (mm)	Ignore sig figs in working Allow full marks for correct answer, no working, i.e.: 20.32 (mm) = 3 marks If no working accept these other bald answers: 20.3228.. etc (mm) = 2 marks 20.34 (mm) = 2 marks 20.3375.. (mm) = 1 mark 20.33 (mm) = 1 mark		3

Question number	Answer	Accept	Reject	Marks
4 (b)	<p>Any two of:</p> <p>Yes / No (no mark)</p> <p>MP1 Good way of measuring small values / Measures a larger value;</p> <p>MP2 Taking a larger measurement might reduce (%) errors;</p> <p>MP3 Not actually measuring what is required (a particular coin);</p> <p>MP4 Possible to make a maths error e.g. when dividing / counting /rounding;</p> <p>MP5 Not all coins are necessarily the same / idea of anomalous coin / bent / worn;</p>	<p>Accept reverse arguments</p> <p>Ignore comments about human error</p> <p>Ignore reference to caliper precision</p> <p>Ignore comments about gaps</p>		2

Question number	Answer	Accept	Reject	Marks
4 (c)	<p>Any three of:</p> <p>MP1 Measure/find <u>mass</u>;</p> <p>MP2 Using a named instrument - e.g. (top pan) balance, scale(s);</p> <p>MP3 A sensible experimental precaution: e.g. Repeat readings / measure mass of several of coins and divide/ check balance zero;</p> <p>MP4 Formula to use (density = mass ÷ volume);</p> <p>MP5 A correct <u>density</u> unit mentioned (e.g. kg/m³);</p>	<p>Ignore information about calculating or finding volume</p> <p>Accept "Weighing" to find <u>mass</u></p> <p>Ignore measuring weight</p> <p>Ignore volume = $\pi r^2 h$</p>		3

Total 9 marks