

Question number	Answer	Notes	Marks
3 (a)	<p>(some) bonds (between particles) are broken or weakened;</p> <p>particles begin to move around or slide over each other;</p>	<p>award either or both marks if clear from diagram accept idea of reduction in intermolecular forces</p> <p>reject MP2 if there is a reference to the speed of particles or KE changing during melting</p>	2
(b) (i)	<p>any two from:</p> <p>MP1. energy is transferred to air/particles (by heating);</p> <p>MP2. idea that particles move faster;</p> <p>MP3. temperature (in K) is (directly) proportional to KE of particles;</p>	<p>accept molecules for particles</p> <p>accept particles gain KE</p>	2
(ii)	<p>particles hit (flask) walls more frequently;</p> <p>particles hit (flask) walls with more force / harder;</p>		2
(iii)	136 (kPa);	allow answers in the range 135-137 (kPa)	1
(iv)	<p>substitution;</p> <p>rearrangement;</p> <p>evaluation;</p> <p>e.g. $136 / 400 = p_2 / 230$ $(p_2 =) (136 \times 230) / 400$ $(p_2 =) 78 \text{ (kPa)}$ </p>	<p>allow ecf from (b)(iii) e.g. 86 (kPa) for (iii) gives 79 (kPa) for (iv) substitution and rearrangement in either order</p> <p>allow 78.2...</p> <p>accept extrapolation back to $T = 230$ for 2 marks accept value between 79 and 77 by this method</p> <p>accept calculation using graph (i.e. extrapolation numerically/algebraically) back to 230 for 2 marks accept value between 79 and 77 by this method</p>	3

Total for question 3 = 10 marks