

**3** This question is about solids, liquids and gases.

(a) Melting is the process that occurs when a solid changes into a liquid.

Using ideas about particles, describe the process of melting.

(2)

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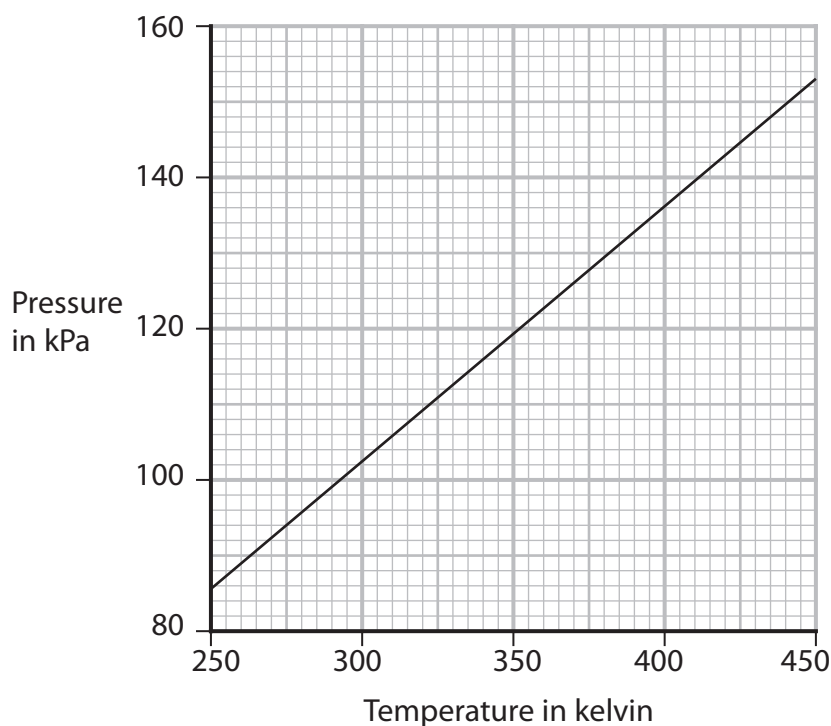
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(b) A sealed metal flask contains air at room temperature.

The flask is then heated.

The graph shows how the pressure of the air increases as its temperature increases.



- (i) Explain how heating the flask increases the temperature of the air inside the flask. (2)

- (ii) Explain why the pressure of the air increases as its temperature increases. (2)

- (iii) Using the graph, determine the pressure of the air when its temperature is 400 K. (1)

pressure = ..... kPa

- (iv) The air in the flask is now cooled to a temperature of 230 K.

Calculate the new pressure of the air.

(3)

pressure = ..... kPa

**(Total for Question 3 = 10 marks)**

