

Question number	Answer	Notes	Marks
12 (a) (i)	neutron numbers correct; particle X numbers correct; <div style="display: flex; align-items: center; justify-content: center; gap: 20px;"> <div style="border: 1px solid black; padding: 5px; text-align: center;">1</div> <div style="text-align: center;"> $n + {}^{14}_{7}\text{N} \longrightarrow {}^{14}_{6}\text{C} + X$ </div> <div style="border: 1px solid black; padding: 5px; text-align: center;">1</div> </div> <div style="display: flex; align-items: center; justify-content: center; gap: 20px; margin-top: 10px;"> <div style="border: 1px solid black; padding: 5px; text-align: center;">0</div> <div></div> <div style="border: 1px solid black; padding: 5px; text-align: center;">1</div> </div>		2
(ii)	proton / p;	allow hydrogen, H, H ⁺ (ion)	1
(iii)	any two from: MP1.both have same number of protons (and electrons); MP2.C-12 has fewer {neutrons / nucleons} than C-14; MP3.C-12 is lighter than C-14;	both C atoms have 6 protons allow RA C-14 has 8 neutrons, C-12 has 6 neutrons allow RA	2
(iv)	mass number is constant; atomic number increases <u>by one</u> ;	however expressed, including numerically	2
(b)	working seen / appropriate line(s) on graph seen; 5 500 (years)	e.g. line drawn across from 125 Bq allow 5000-6000 (years)	2
(c) (i)	(due to) background radiation;		1
(ii)	idea that activity depends on the mass;	allow 'fair test' idea ignore 'to have the same activity'	1