

Question 3.7.3

a.

Let e_T denote the total enzyme concentration: $e_T = e(t) + c(t)$

$$\frac{d}{dt}s(t) = k_{-1}c(t) - k_1s(t)e(t) = k_{-1}c(t) - k_1s(t)(e_T - c(t))$$

$$\frac{d}{dt}c(t) = k_1s(t)e(t) + k_{-2}p(t)e(t) - k_{-1}c(t) - k_2c(t) = k_1s(t)(e_T - c(t)) + k_{-2}p(t)(e_T - c(t)) - k_{-1}c(t) - k_2c(t)$$

$$\frac{d}{dt}p(t) = k_2c(t) - k_{-2}p(t)e(t) = k_2c(t) - k_{-2}p(t)(e_T - c(t))$$

Apply a quasi-steady-state assumption to the complex C, where

$$\frac{d}{dt}c^{qss}(t) = 0, \text{ and the goal is to arrive the description of } c^{qss} = \frac{k_1e_Ts + k_{-2}e_Tp}{k_1s + k_{-2}p + k_{-1} + k_2}$$

$$0 = k_1s(t)[e_T - c^{qss}] + k_{-2}p(t)[e_T - c^{qss}] - k_{-1}c^{qss} - k_2c^{qss}$$

$$\Rightarrow (k_{-1} + k_2)c^{qss} = [k_1s(t) + k_{-2}p(t)][e_T - c^{qss}]$$

$$\Rightarrow (k_{-1} + k_2)c^{qss} + [k_1s(t) + k_{-2}p(t)]c^{qss} = [k_1s(t) + k_{-2}p(t)]e_T$$

$$\Rightarrow [k_{-1} + k_2 + k_1s(t) + k_{-2}p(t)]c^{qss} = [k_1s(t) + k_{-2}p(t)]e_T$$

$$\Rightarrow c^{qss} = \frac{k_1e_Ts(t) + k_{-2}e_Tp(t)}{k_1s(t) + k_{-2}p(t) + k_{-1} + k_2}$$

$$\frac{d}{dt}p(t) = k_2c(t) - k_{-2}p(t)(e_T - c(t))$$

Substitute $c(t)$ with c^{qss} and simplify:

$$\begin{aligned} &= k_2 \frac{k_1e_Ts(t) + k_{-2}e_Tp(t)}{k_1s(t) + k_{-2}p(t) + k_{-1} + k_2} \\ &\quad - k_{-2}p(t) \left(e_T - \frac{k_1e_Ts(t) + k_{-2}e_Tp(t)}{k_1s(t) + k_{-2}p(t) + k_{-1} + k_2} \right) \\ &= \frac{k_1k_2e_Ts(t) + k_2k_{-2}e_Tp(t)}{k_1s(t) + k_{-2}p(t) + k_{-1} + k_2} \\ &\quad - \frac{k_1k_{-2}e_Ts(t)p(t) + k_{-2}^2e_T[p(t)]^2 + k_{-1}k_{-2}e_Tp(t) + k_2k_{-2}e_Tp(t) - k_1k_{-2}e_Ts(t)p(t) - k_{-2}^2e_T[p(t)]^2}{k_1s(t) + k_{-2}p(t) + k_{-1} + k_2} \\ &= \frac{k_1k_2e_Ts(t) - k_{-1}k_{-2}e_Tp(t)}{k_1s(t) + k_{-2}p(t) + k_{-1} + k_2}. \end{aligned}$$

Done.

b.

Assuming the complex C is in quasi-steady-state with respect to S and P,

$$\begin{aligned}
\text{Net rates } S \rightarrow P &= \frac{V_f \frac{s}{K_S} - V_r \frac{p}{K_P}}{1 + \frac{s}{K_S} + \frac{p}{K_P}} \\
&= \frac{k_1 k_2 e_T s(t) - k_{-1} k_{-2} e_T p(t)}{k_1 s(t) + k_{-2} p(t) + k_{-1} + k_2} \\
&\quad \text{divide by } k_{-1} + k_2 \Rightarrow \frac{\frac{k_1 k_2 e_T s(t) - k_{-1} k_{-2} e_T p(t)}{k_{-1} + k_2}}{\frac{k_1 s(t) + k_{-2} p(t) + k_{-1} + k_2}{k_{-1} + k_2}} \\
&= \frac{\frac{k_1 k_2 e_T s}{k_{-1} + k_2} + \frac{k_{-1} k_{-2} e_T p}{k_{-1} + k_2}}{\frac{k_1}{k_{-1} + k_2} s + \frac{k_{-2}}{k_{-1} + k_2} p + \frac{k_{-1} + k_2}{k_{-1} + k_2}} \\
&\quad \text{Substitute } \frac{k_{-1} + k_2}{k_1} \text{ with } K_S, \frac{k_{-1} + k_2}{k_{-2}} \text{ with } K_P \\
&\Rightarrow \frac{\frac{k_2 e_T s}{K_S} + \frac{k_{-1} e_T p}{K_P}}{\frac{s}{K_S} + \frac{p}{K_P} + 1} \\
&\quad \text{Substitute } k_2 e^T \text{ with } V_f, k_{-1} e^T \text{ with } V_r \\
&\Rightarrow \frac{V_f \frac{s}{K_S} + V_r \frac{p}{K_P}}{\frac{s}{K_S} + \frac{p}{K_P} + 1}
\end{aligned}$$

$$V_f = k_2 e^T, V_r = k_{-1} e^T, K_S = \frac{k_{-1} + k_2}{k_1}, K_P = \frac{k_{-1} + k_2}{k_{-2}}$$

Done.

c.

$$v = \frac{d}{dt} p(t) = \frac{k_1 k_2 e_T s(t) - k_{-1} k_{-2} e_T p(t)}{k_1 s(t) + k_{-2} p(t) + k_{-1} + k_2}$$

when $k_{-2} = 0$

$$\begin{aligned}
&= \frac{k_1 k_2 e_T s(t)}{k_1 s(t) + k_{-1} + k_2} \\
&= \frac{k_1 k_2 e_T s(t)}{k_1 \left(s(t) + \frac{k_{-1} + k_2}{k_1} \right)}
\end{aligned}$$

Define *Michaelis* constant $K_M = \frac{k_{-1} + k_2}{k_1}$, $V_{\max} = k_2 e_T$

$$\Rightarrow \frac{V_{\max} s(t)}{s(t) + K_M}, \text{ which is the form of the irreversible Michaelis-Menten rate law}$$

Done.