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Answer 1.1) If XCl■ then X should be a metal with valency 2.

Sulphate Formula: XSO■

Hydroxide Formula: X(OH)■

Answer 1.2) XN is nitride. X is a valency 3 metal as nitrogen has valency 3.

Therefore, Sulphate's formula would be X■(SO■)■ and Hydroxide's formula would be X(OH)■

Answer 1.3) Valency of nitrogen in:

1) NO: +2

2) N■O: +1

3) NO■: +4

Answer 2.1) Three symbols whose first letter is the symbol for them.

(1) Boron (B)

(2) Carbon (C)

(3) Hydrogen (H)

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Answer 2.2) The eight metals showing variable valency:

1) Iron (Fe) : +2, +3

2) Copper (Cu) : +1, +2

3) Mercury (Hg) : +1, +2

4) Tin (Sn) : +2, +4

5) Lead (Pb) : +2, +4

6) Gold (Au) : +1, +3

7) Chromium (Cr) : +2, +3, +6

8) Manganese (Mn) : +2, +3, +4, +6, +7

Answer 2.3) Examples of chemical equations:

(a) One product : $2H_2 + O_2 \rightarrow 2H_2O$

(b) Two product : $CaCO_3 \rightarrow CaO + CO_2$

(c) Three Products : $4HNO_3 \rightarrow 2H_2O + 4NO_2 + O_2$

(d) four Products : $2KClO_3 \rightarrow 2KCl + 2O_2 + O_2 + \Delta$

Answer 2.4) Symbols in chemical equation mean:

(i) $\uparrow$ - Gas evolved

(ii) $\downarrow$ - precipitate formed

(iii) (s) - solid state

(iv) (l) - liquid state

(v) (g) - gaseous state

(vi) (aq) - aqueous solution