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Answer 1A) If XCl2 then X should be a metal with valency 2

Therefore

(1) Sulphate formula : XSO4

(2) Hydroxide formula : X(OH)2

Answer 1B) XN2 is nitride X is a valency 3 metal as nitrogen has valency 3

Therefore Sulphate's formula would be X2(SO4)3 and Hydroxide formula would be X(OH)3

Answer 1C) valency of Nitrogen in

1) NO : +2

2) N2O : +1

3) NO2 : +4

Answer 2B) The eight metals showing variable valency

1) Iron (Fe) : +2,+3

2) copper (Cu) :+1,+2

3) Mercury (Hg) : +1,+2

4) Tin (Sn) : +2,+4

5) Lead (Pb) : +2,+4

6) Gold (Au) : +1,+3

7) Chromium (Cr) : +2,+3,+4,+6

8) Manganese (Mn) : +2,+3,+4,+6,+7

Answer 2C) Examples of Chemical Equations :

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(a)

* One product: 2H2 + O2 \(\rightleftharpoons\) 2H2O

(b)

* Two products: CaCO3 \(\rightarrow\) CaO + CO2 + \(\Delta\)

(c)

* Three products: 4HNO3 \(\rightarrow\) 2H2O + 4NO2 + O2

(d)

* Four products: 2KClO3 \(\rightarrow\) 2KCl + 3O2 + \(\Delta\)

Answer: 2D