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****Answer 1A)**** If XCl_2 then X should be a metal with valency 2

Therefore

(1) Sulphate formula : XSO_4

(2) Hydroxide formula : X(OH)_2

****Answer 1B)**** XN is nitride X is a valency 3 metal as nitrogen has valency 3

Therefore Sulphate's formula would be $\text{X}_2(\text{SO}_4)_3$ and Hydroxide formula would be X(OH)_3

****Answer 1C)**** valency of Nitrogen in

1) NO : +2

2) N_2O : +1

3) NO_2 : +4

****Answer 2B)**** The eight metals showing variable valency

1) Iron (Fe) : +2,+3,+6

2) Copper (Cu) : +1,+2

3) Mercury (Hg) : +1,+2

4) Tin (Sn) : +2,+4

5) Bismuth (Pb):+2,+4

6) Gold (Au):+1,+3

7) Chromium (Cr):+2,+3,+6

8) Manganese (Mn):+2,+3,+4,+6,+7

****Answer 2C)**** Examples of Chemical Equations:

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(a) One product: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

(b) Two Products: $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2 + ______$

(c) Three Products: $4\text{HNO}_3 \rightarrow 2\text{H}_2\text{O} + 4\text{NO}_2 + \text{O}_2$

(d) Four Products: $2\text{KClO}_3 \rightarrow 2\text{KCl} + 2\text{O}_2 + \text{O}_2 + \Delta$