

Conc : (g/dl)	Flow Time of Polymer -Solvent system (t) sec	Flow Time of Solvent t (t ₀) sec	$\eta_r = \frac{t}{t_0}$	$\eta_{sp} = \eta_r - 1$	Reduced Viscosity, $\eta_{red} = \frac{\eta_{sp}}{C}$ (dl/g)	$\ln \eta_r$	Inherent Viscosity, $\eta_{inh} = \frac{\ln \eta_r}{C}$ (dl/g)
0.02	107.16	102.99	1.040	0.04	2	0.039	1.95
0.04	109.72	102.99	1.065	0.065	1.625	0.062	1.55
0.06	113.55	102.99	1.102	0.102	1.7	0.097	1.61
0.08	116.76	102.99	1.133	0.133	1.662	0.124	1.55
0.1	121.18	102.99	1.176	0.176	1.76	0.162	1.62

Calculation:
(Include formula)

Graph

From the above graph, $[\eta] = \frac{151}{\text{polyvinylacetate}}$ ml/g

For (polymer- solvent system) acetone $K = 41.5 \times 10^{-3}$ ml/g $\alpha = 0.62$

Substituting these values in the formula, $[\eta] = KM^\alpha$

We get, (show calculation):

$$151 = 41.5 \times 10^{-3} \times (M)^{0.62}$$

$$151 = 415 \times 10^{-4} \times (M)^{0.62}$$

$$0.363 \times 10^4 = (M)^{0.62}$$

$$3630 = (M)^{0.62}$$

$$\ln(3630) = \ln(M) \times 0.62$$

$$8.19 = \ln(M) \times 0.62$$

$$\ln(M) = \frac{8.19}{0.62}$$

$$\ln(M) = 13.20$$

$$M = 5,40,364 \text{ g/mol}$$

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Assignment:

1. Name any 5 compounding agents with an example and their functions in polymers.
2. List important characteristics of an ideal polymer.
3. Write a note on doped conducting polymers.

Result/ Conclusion:

The viscosity average molecular weight of the (polymer) ' M_v ' = 5,40,364 g/mol

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inherent / reduced viscosity
vs
concentration.

SCALE:- X axis:- $\Delta \text{cm} = 0.02$
Y axis:- $\Delta \text{cm} = 20 \text{ units}$
□ reduced viscosity
○ inherent viscosity



