

Unit 3:- Biomass Energy



Syllabus...Unit 3

- Energy from Biomass, Thermo-Chemical, Bio-Chemical Conversion to fuels. Bio-gas and its applications

Books ...

- Gilbert M. Masters, *Renewable and Efficient Electrical Power Systems*, Wiley - IEEE Press, August 2004.
- Godfrey Boyle, *Renewable Energy*, Third edition, Oxford University Press, 2012.
- Chetan Singh Solanki, *Solar Photovoltaics-Fundamentals, Technologies and Applications*, PHI Third Edition, 2015.

Supplementary Reading:

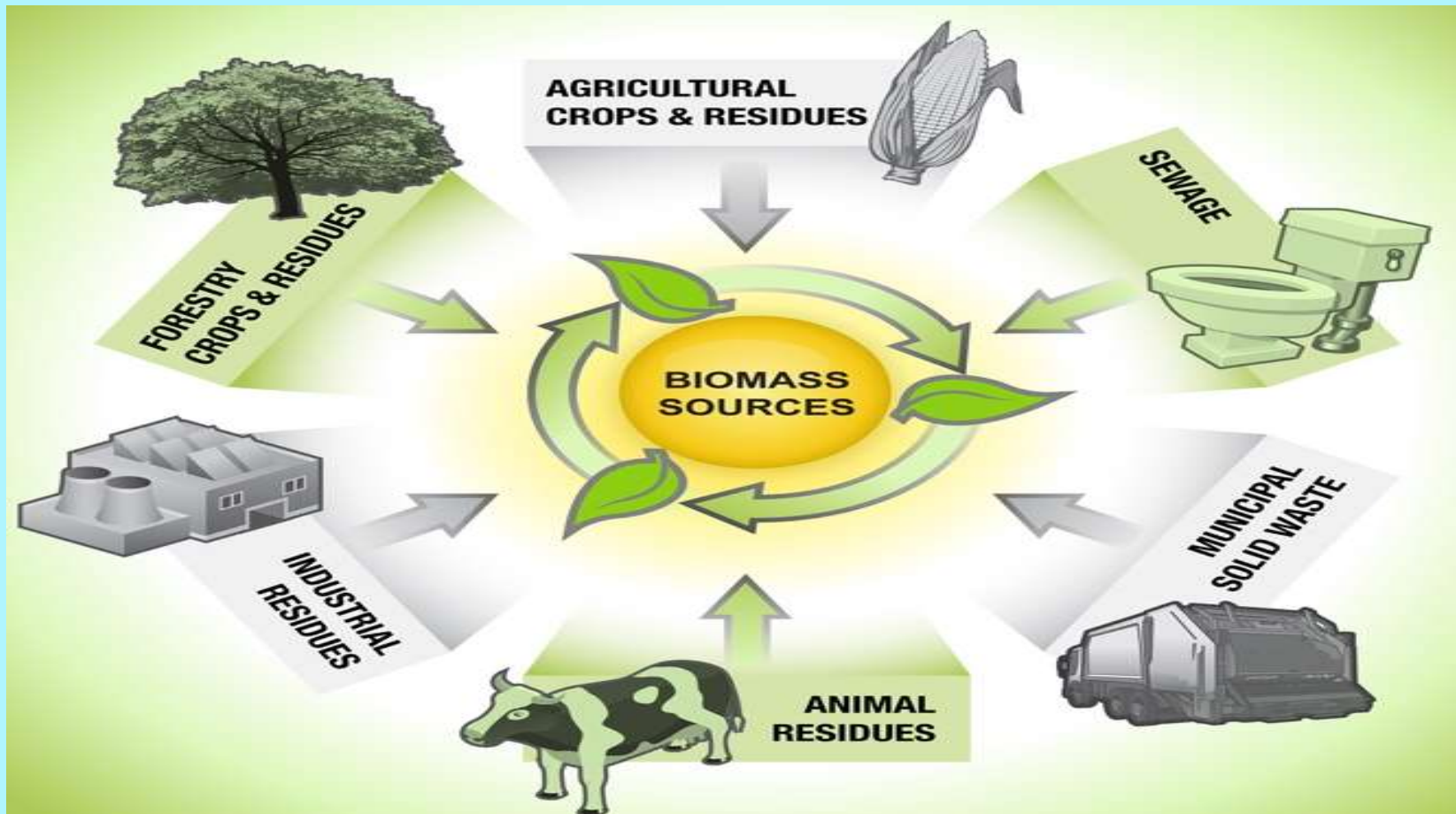
- D.P.Kothari, K.C.Singal, Rakesh Rajan, *Renewable Energy Sources and Emerging Technologies*, PHI Second Edition, 2011.

Lecture 1

- What is Biomass?
- Biomass Algae
- Biomass as Solar Energy Photosynthesis..
- Carbon Neutral
- Bio mass Cycle
- Why Biomass?
- World's Statistics
- Biomass Energy Conversion
- Biomass Direct Combustion
- Biomass Direct Combustion Plant
- Direct Combustion Sources
- Thermo-Chemical Conversion

Bio-mass

- Biomass is a organic matter such as Wood, Straw, Crops, **Algae**, Sewage Sludge, Animal Waste and/ or other Biological waste.

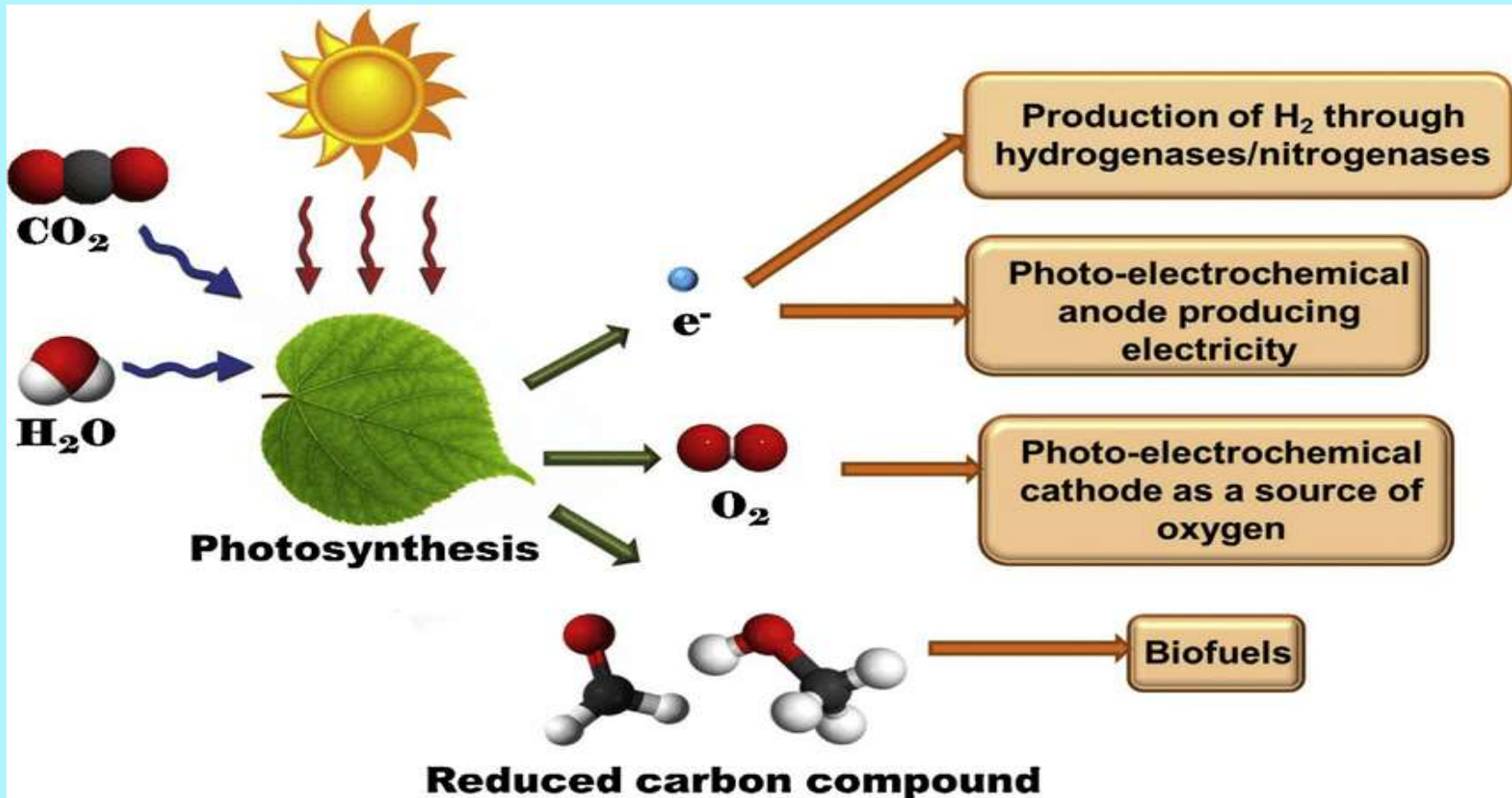


Biomass Algae



- Biomass **Algae?** **Algae**; (singular **alga**) is an informal term for a large, diverse group of photosynthetic eukaryotic organisms that are not necessarily closely related, and is thus polyphyletic. **Algae** are simple plants that can range from the microscopic (microalgae), to large seaweeds such as giant kelp more than one hundred feet in length.

Biomass as Solar Energy Photosynthesis..



- Biomass can be viewed as a formed of stored solar energy.
- Sun's energy is captured and stored in the bio material. How?
 - ✓ By Photosynthesis

What is Photosynthesis ?

Steps in Photosynthesis

Splitting of water molecule into H_2 & O_2 under influence of chlorophyll.
“Light Reaction”

Hydrogen is transferred to CO_2 to form Starch or Sugar.

Necessary Conditions

Light



CO_2 Concentration

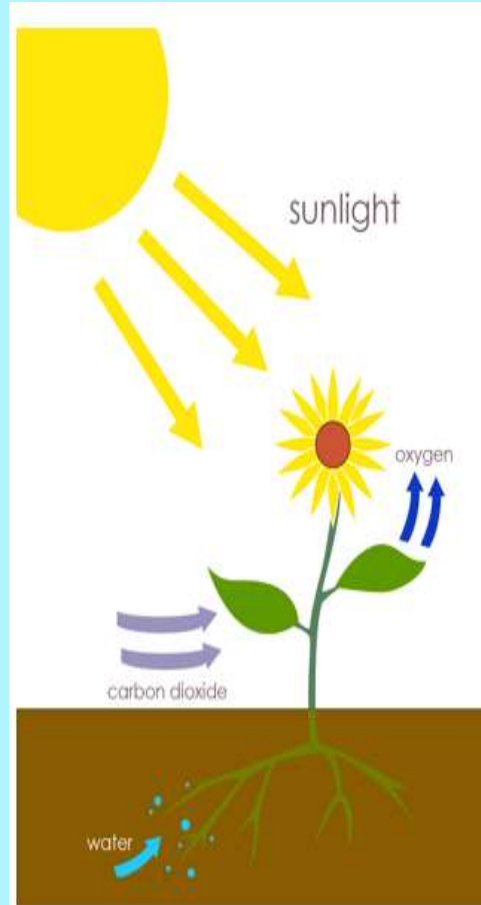


Temperature

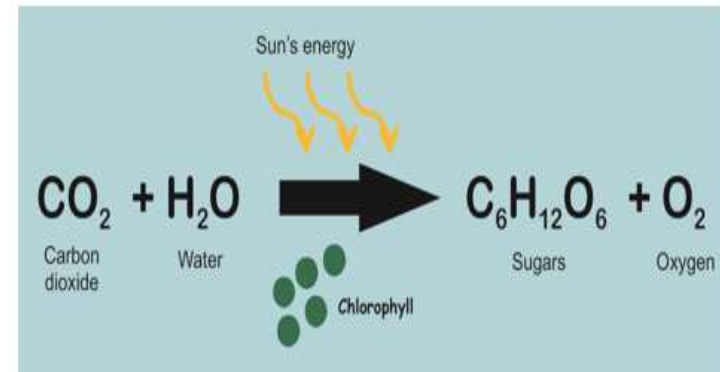


What is Photosynthesis ?

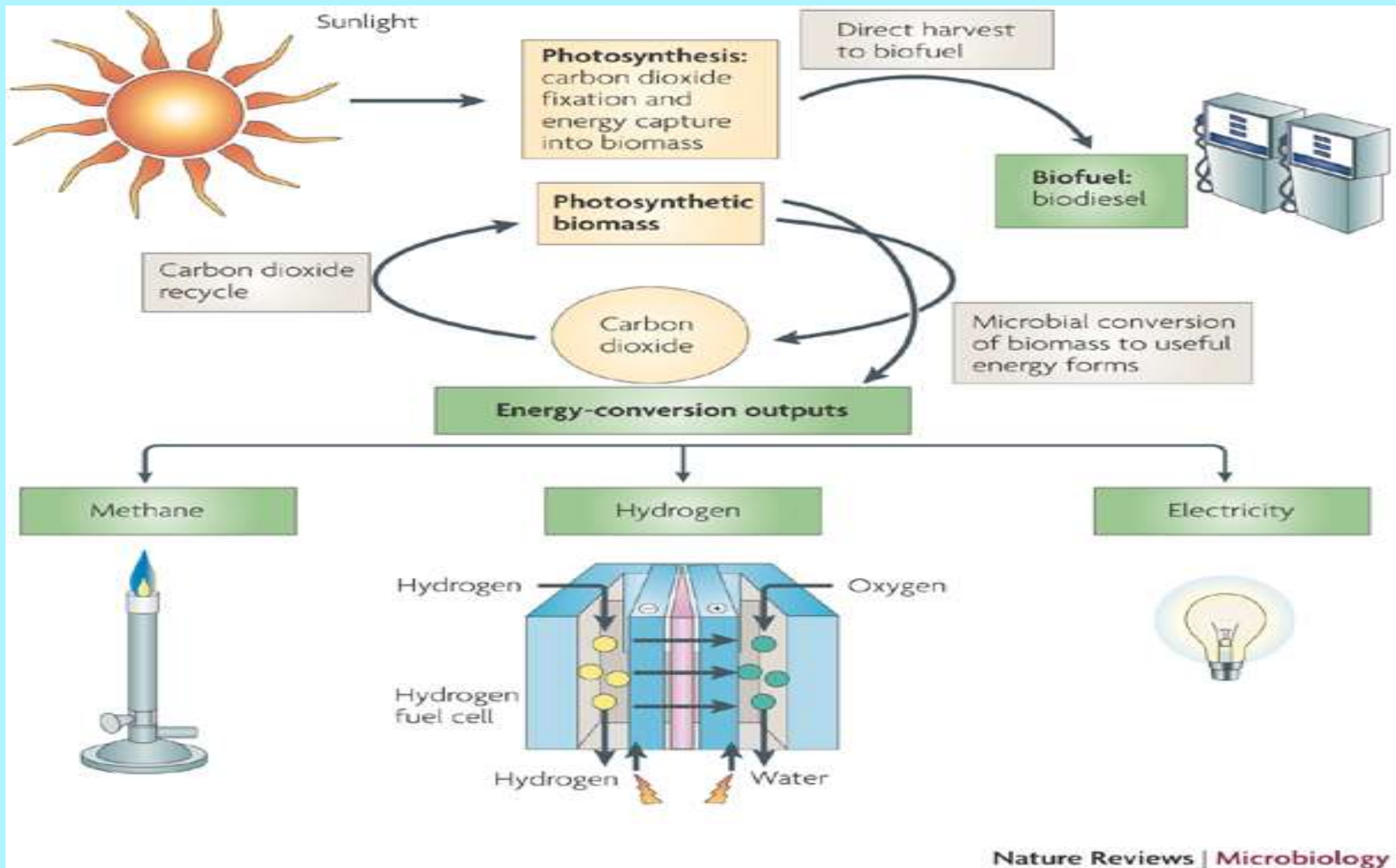
Photosynthesis is a chemical process that converts carbon dioxide in to organic compounds, especially sugar, using energy from sunlight.



Photosynthesis converts CO_2 and H_2O into plants and trees; into biomass.



What is Photosynthesis ?

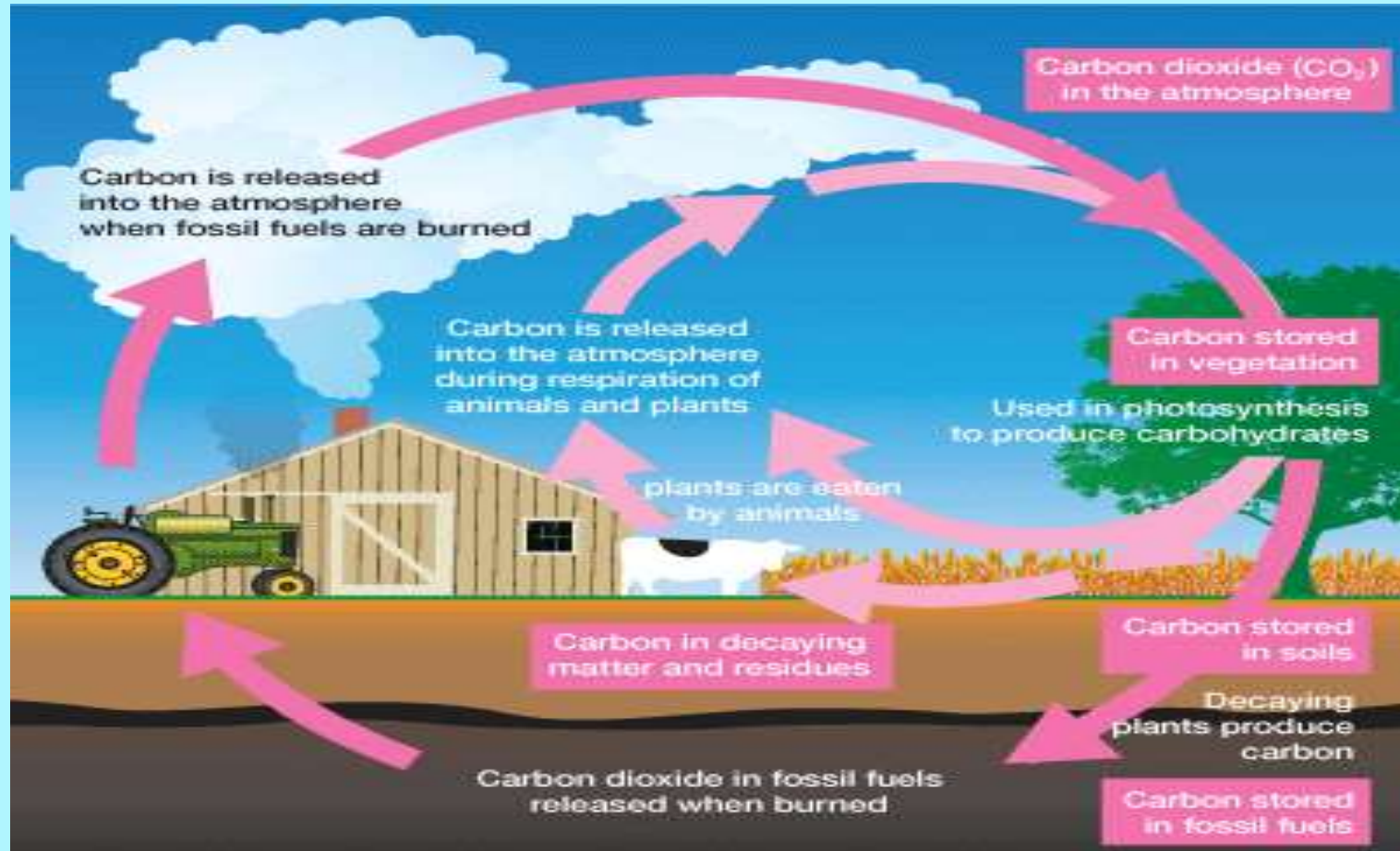


Bio Energy and Carbon Neutral

Bio energy is the energy derived from bio mass. The carbon dioxide released during burning of biomass is largely balanced by absorption / capture of carbon dioxide during its growth in photosynthesis. Hence it is considered as '**Carbon Neutral**'



Bio-mass Cycle



Why Biomass ?

UNDERSTANDING BIOMASS

What Is It?

A renewable low carbon fuel available throughout

A sustainable fuel that can deliver a significant reduction in net carbon emissions when compared with fossil fuels

Biomass includes organic matter like grass, leaves, wood, wood chips, rice husk, peanut shells, sugarcane fiber, sewage etc

Fuels are sourced from wood, including pellets, chips and logs

What Can It Do?

20% Average rate of return provided

30%-50% The amount you can save on fuel bills

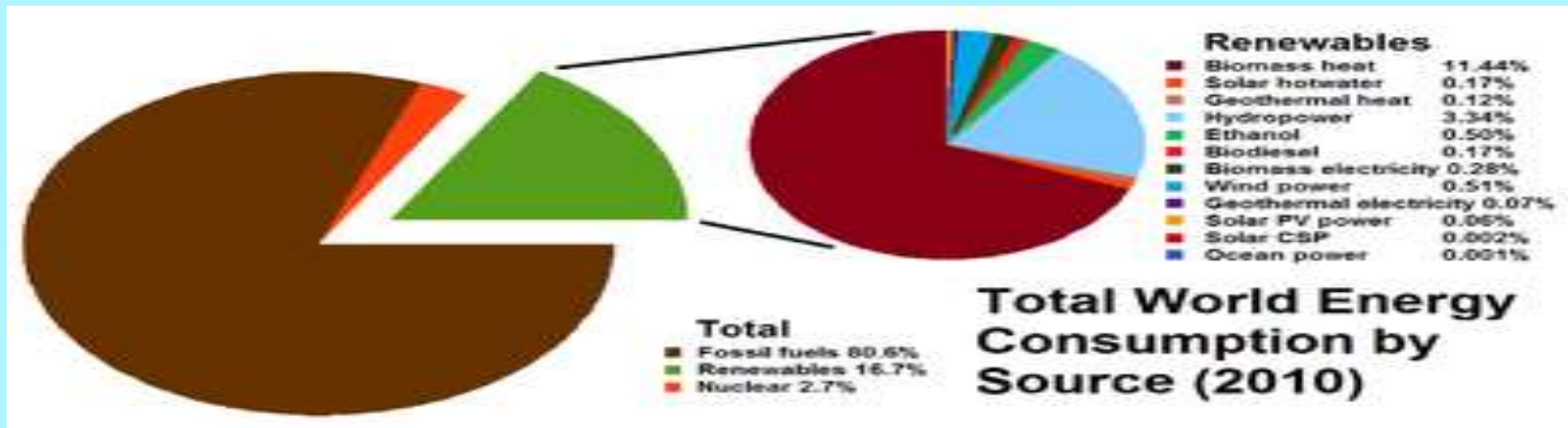
Beat oil, gas and electricity prices

Provide all your home or business' heating needs

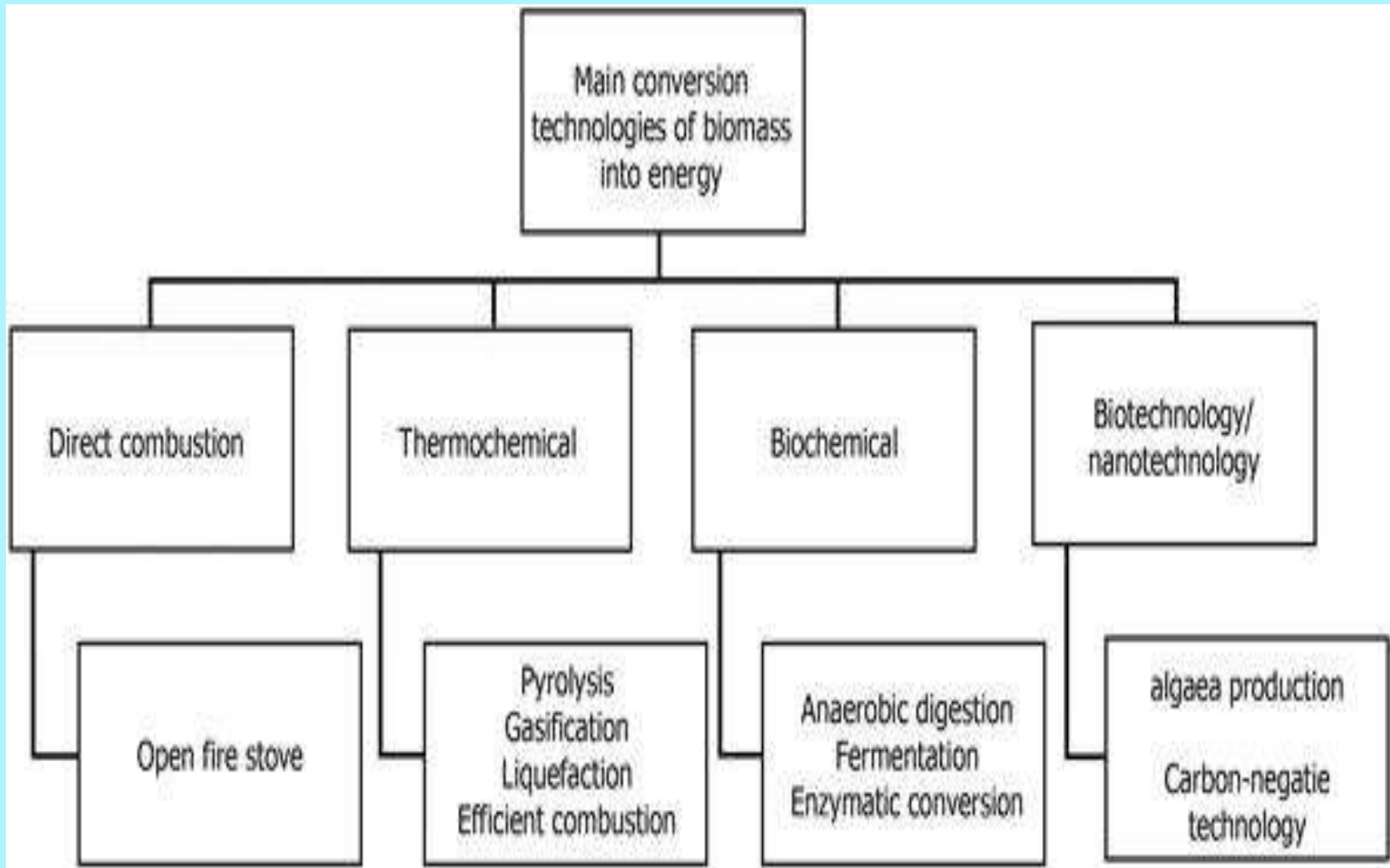
5-7 years Average payback time

World's Statistics.....

- Bio mass energy provides 13% of primary energy in the world.
- More than 75% of World's renewable energy.
- Up to 2050 bio energy could contribute 25% to 33% of global Energy.
- World production of Biomass is estimated 146 Billion Metric Tons per year.
- Mostly a wild plant growth is expected.



Biomass Energy Conversion.....



Biomass Energy Conversion.....

Biomass to Energy Conversion Pathways

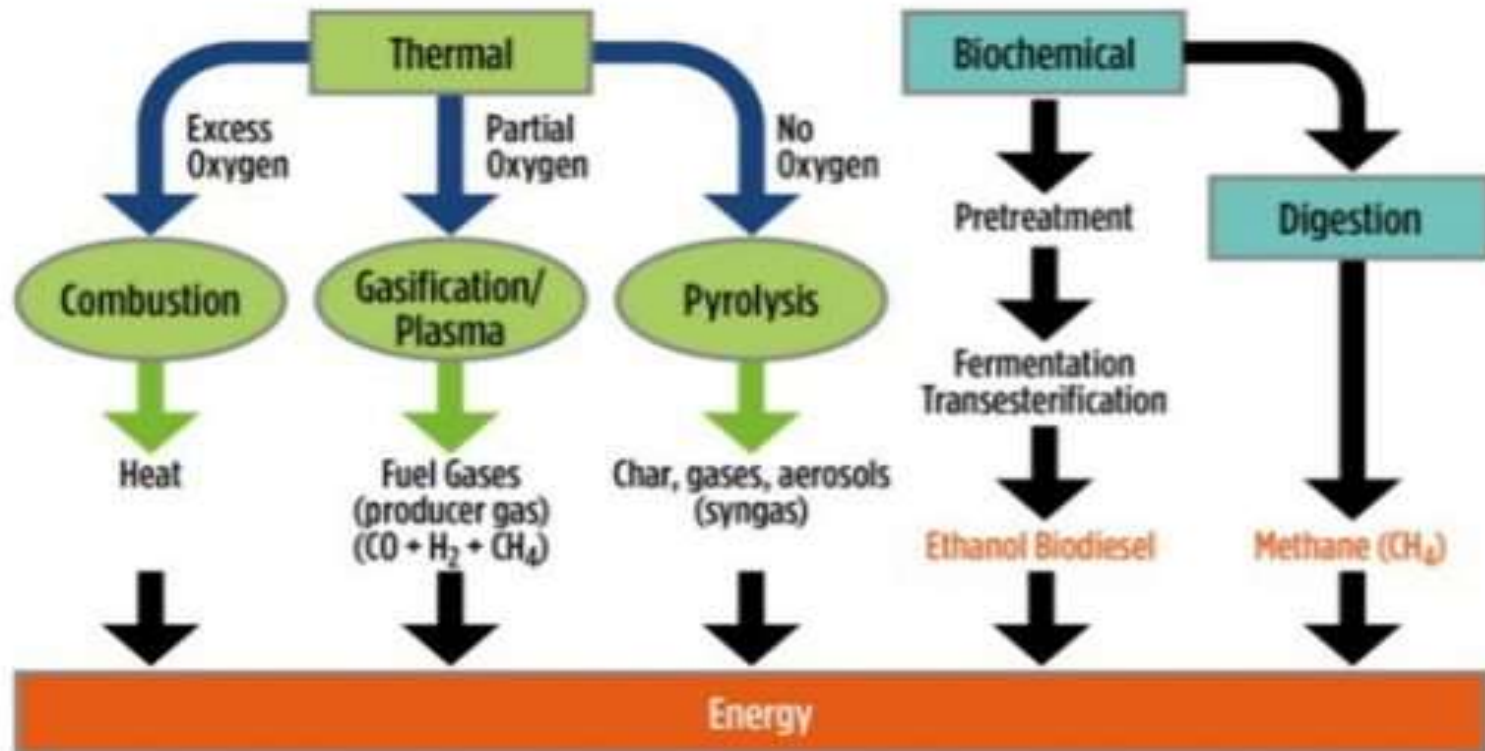


Illustration by NREL

Biomass Direct Combustion.....

The direct combustion of biomass in presence of oxygen/air to produce heat and by products is called direct combustion.

The complete combustion of biomass into ash is called incineration.

This heat energy in the product gases or in the form of steam can be used for various applications like space heating or cooling, power generation, process heating in industries or any other application.

However, if biomass energy by combustion is used as co generation with conventional fuels, the utilization of biomass energy makes it an attractive proposition.

Biomass Direct Combustion.....

Wood is mainly just carbon, hydrogen, and oxygen: $[\text{CH}_2\text{O}]_x$

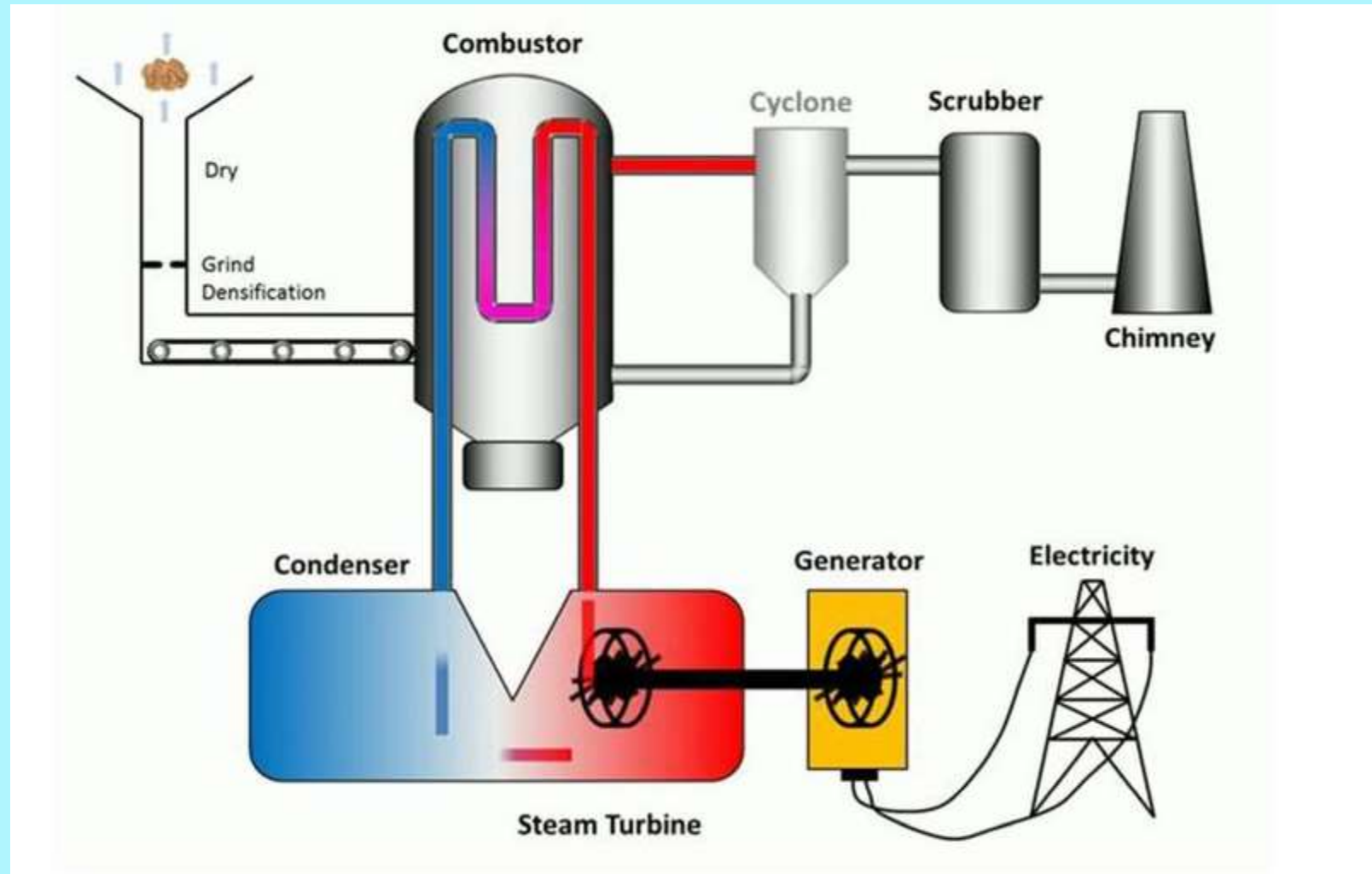
Combustion: $\text{CH}_2\text{O} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{heat}$

Why doesn't wood emit only CO_2 and H_2O when it is burned?



Answer: Incomplete combustion – unavoidably, some of the wood carbon is not completely combusted into CO_2 .

Biomass Direct Combustion Plant.....



Biomass Energy Conversion.....

Direct Combustion

- It includes small scale boilers to large scale MW generation plants
- Uses solid biomass
- Like.....coconut shells, rice husks, wood waste, bagasse, De Oiled-Bran(DOB) and oil seed cakes



Biomass Energy Conversion.....

Direct Combustion

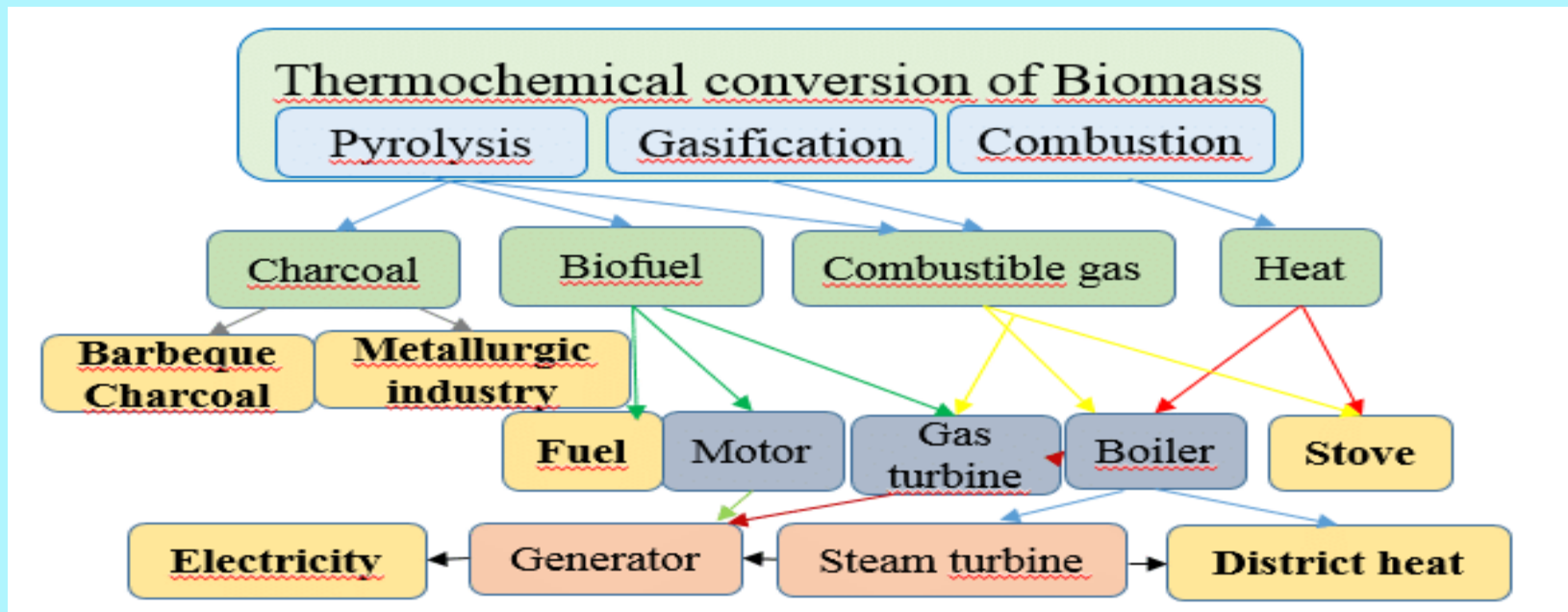
- Biomass of low bulk density are processed to *Pellets or Briquettes*



Biomass Energy Conversion.....

Thermo-Chemical Conversion

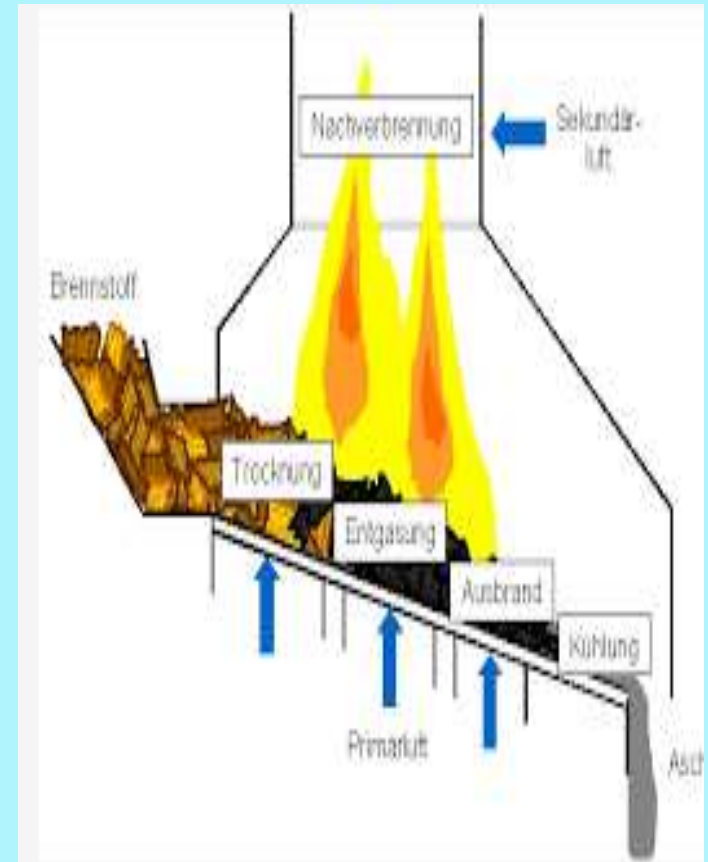
- In this reaction organic biomass is converted in to more valuable and convenient form of products as gaseous and liquid fuels, residue and Bi-products



Biomass Energy Conversion.....

Direct Combustion

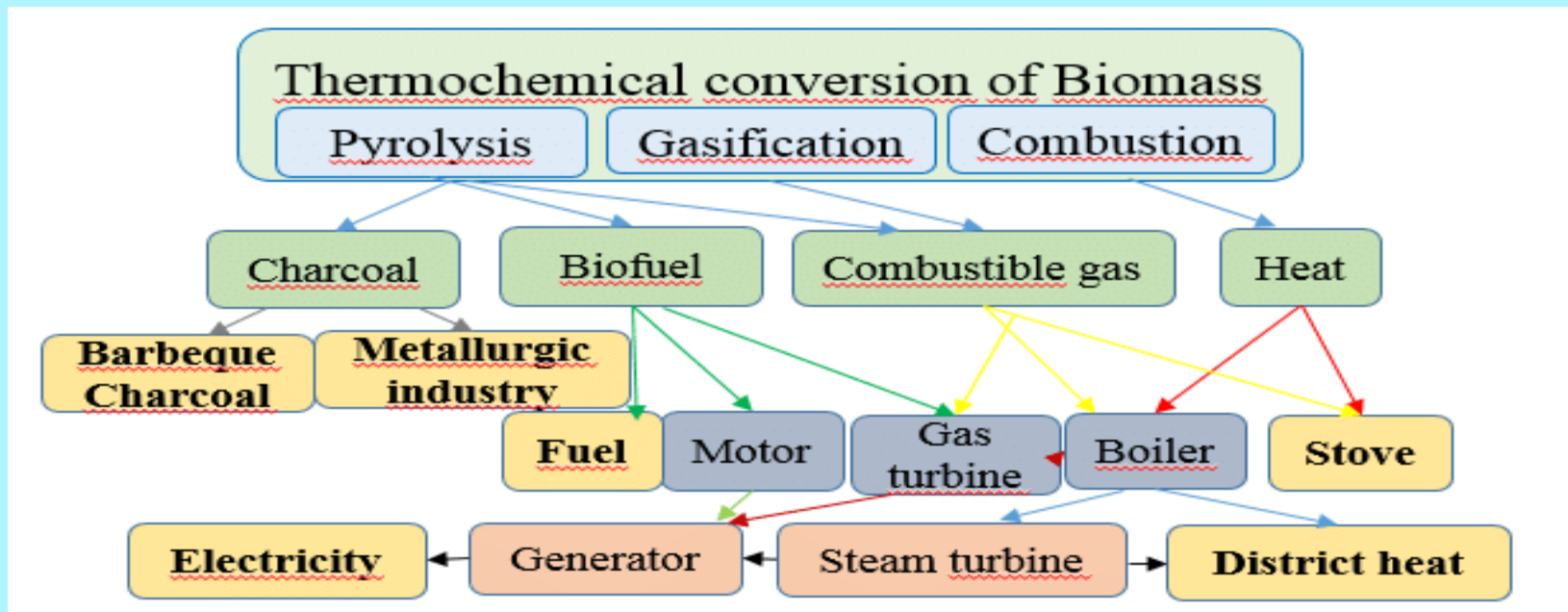
Combustion: the process by which flammable materials are burned in the presence of air or oxygen to release heat. It is the simplest method by which biomass can be used for energy. In its rudimentary form, combustion is used for space heating (i.e. a fire for warmth) but can also be used to heat steam for electricity generation.



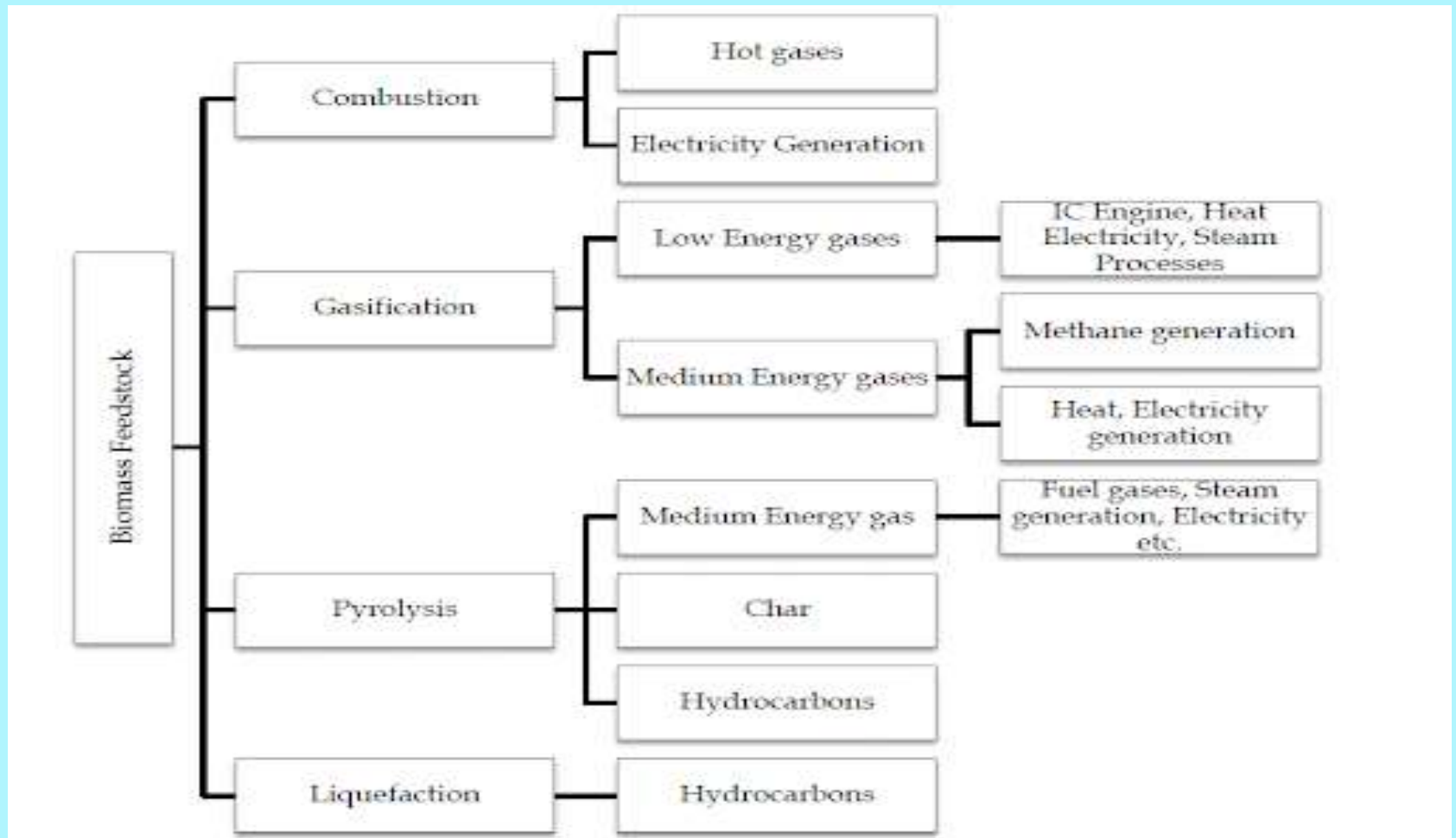
Biomass Energy Conversion.....

Thermo-Chemical Conversion

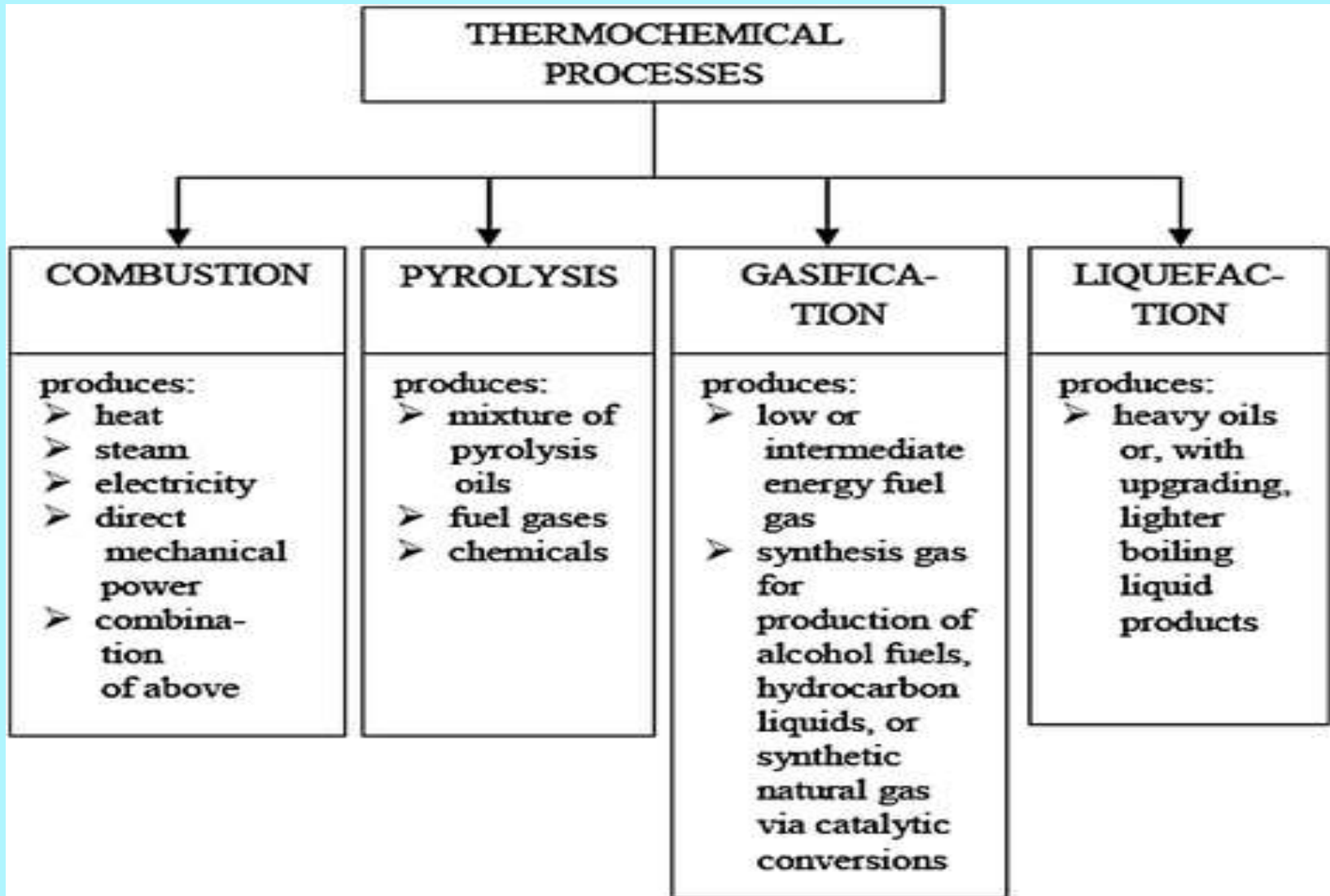
- In this reaction organic biomass is converted in to more valuable and convenient form of products as gaseous and liquid fuels, residue and Bi-products



Thermochemical Conversion.....



Thermochemical Conversion.....



Gasification.....

Definition : *Gasification is a process that converts organic or fossil fuel based material into syngas and carbon dioxide.*

It is the conversion of biomass into a combustible gas mixture referred to as Producer Gas ($\text{CO} + \text{H}_2 + \text{CH}_4$) or Syngas. The gasification process uses heat, pressure and partial combustion to create syngas, which can then be used in place of natural gas.

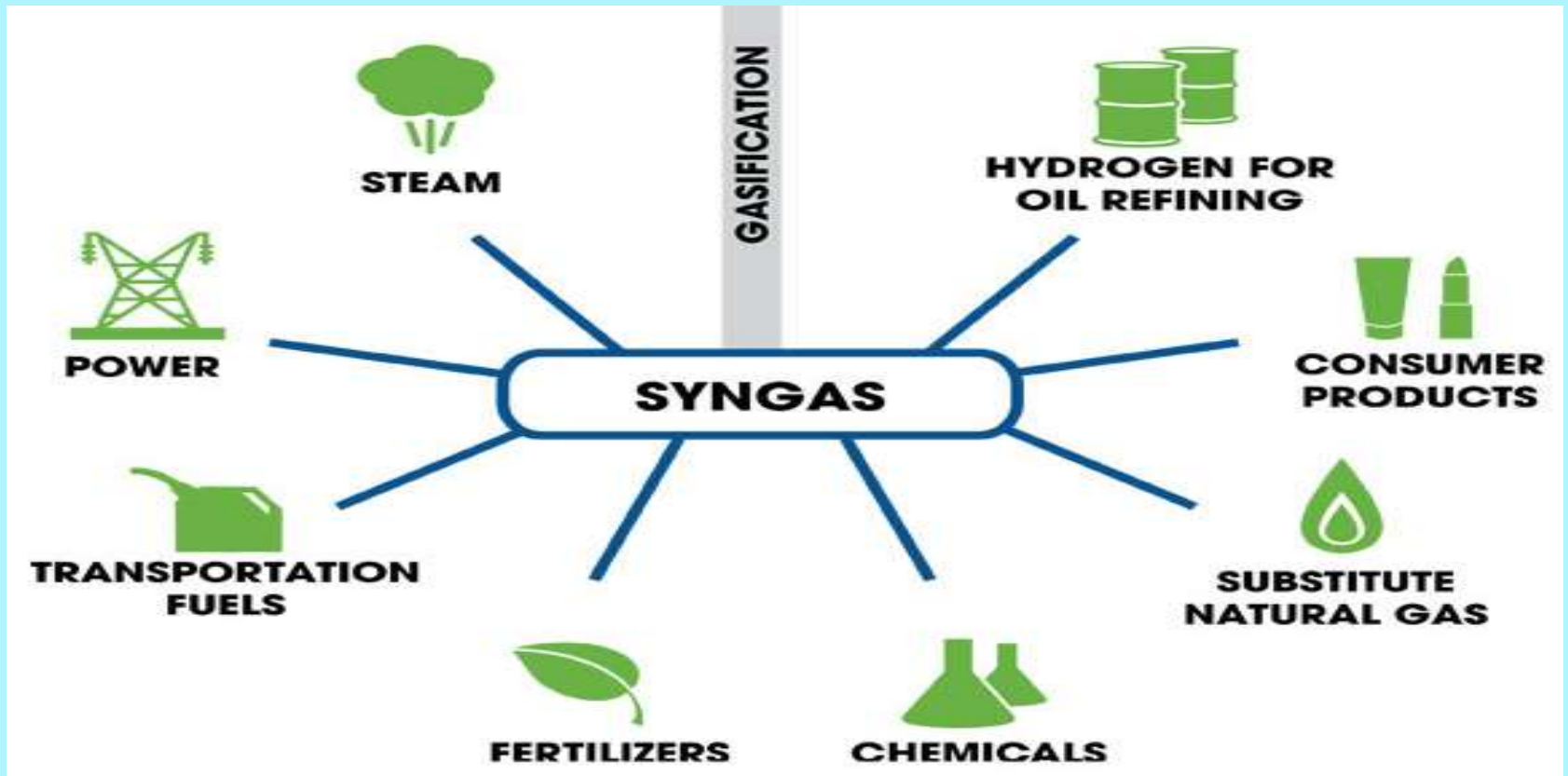
Combustion	vs.	Gasification
$\text{C} + \text{O} \rightarrow \text{CO}_2$		$\text{C} + \frac{1}{2} \text{O}_2 \rightarrow \text{CO}$
		$\text{C} + \text{CO}_2 \rightarrow 2\text{CO}$
		$\text{C} + \text{H}_2\text{O} \rightarrow \text{CO} + \text{H}_2$
A flame is present		Syngas is produced

Gasification.....

- Gasification does not involve combustion, but instead uses little or no oxygen or air in a closed reactor to convert carbon-based materials directly into a synthesis gas, or syngas.
- The gasification process breaks these carbon-containing materials down to the molecular level, so impurities like nitrogen, sulphur, and mercury can be easily removed and sold as valuable industrial commodities.



Gasification.....



Gasification is a technology that creates synthesis gas which in turn can be used to produce electricity and other valuable products, such as chemicals, fuels, and fertilizers.

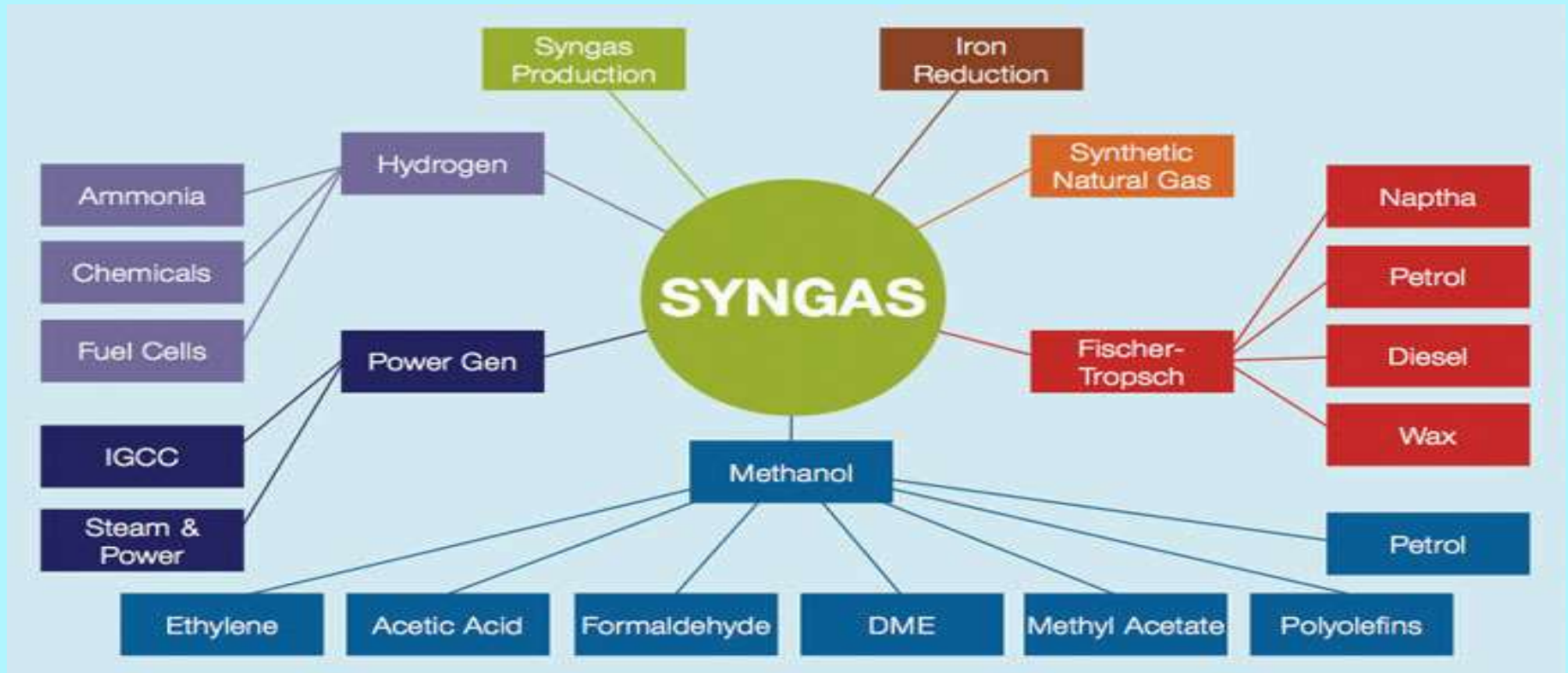
Syngas...

Syngas, or synthesis gas, is a fuel gas mixture consisting primarily of **hydrogen**, **carbon monoxide**, and very often some **carbon dioxide**. The name comes from its use as intermediates in creating synthetic **natural gas** (SNG) and for producing **ammonia** or **methanol**.

Historically, syngas has been used as a replacement for gasoline, when gasoline supply has been limited; for example, wood gas was used to power cars in Europe during WWII however, Syngas has less than half the energy density of natural gas.



Syngas...



Syngas is the primary product of the gasification plants, marketable products obtained from syngas include chemicals (45%), liquid fuels (28%), gaseous fuels (8%), and electric power (19%)^[2]

Gasification.....

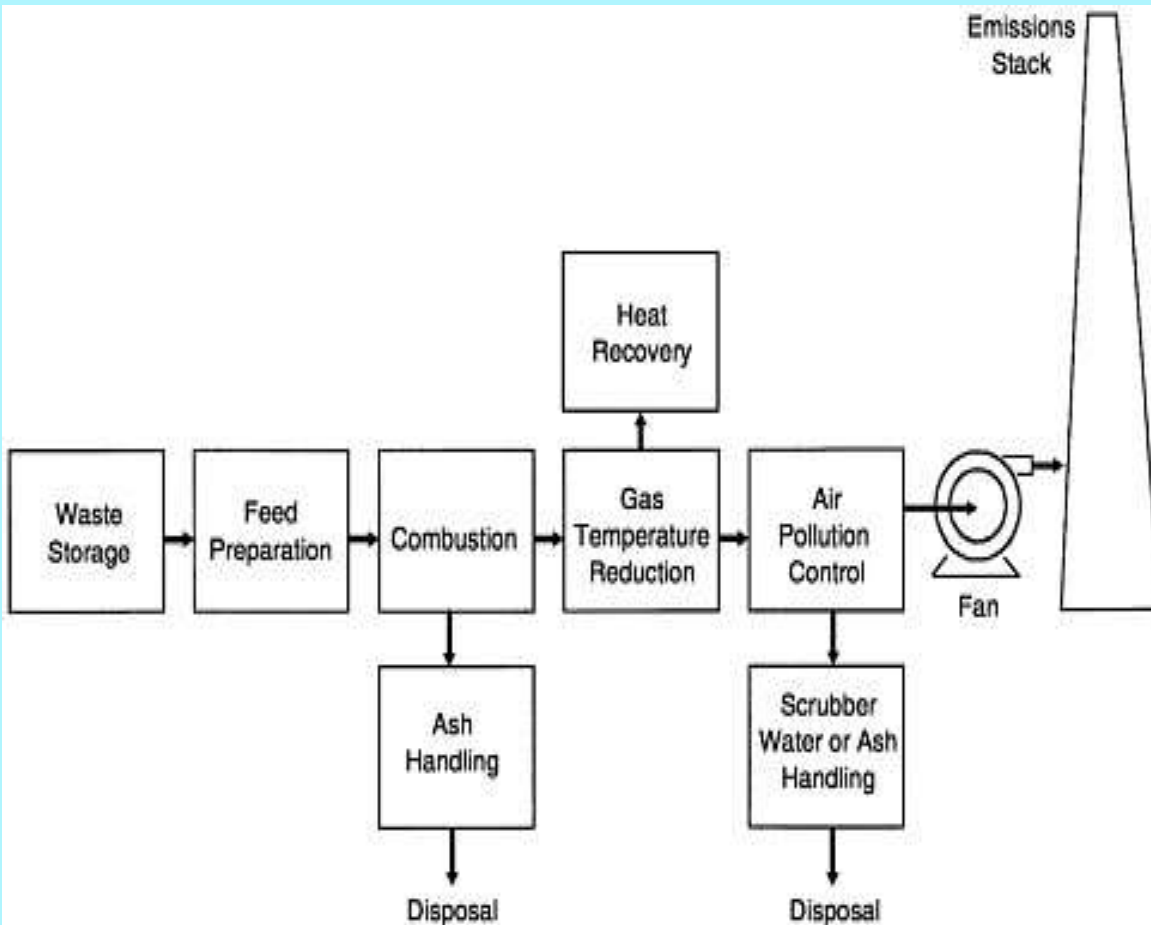
- Gasification has been reliably used on a commercial scale for more than 75 years in the refining, fertilizer, and chemical industries, and for more than 35 years in the electric power industry.
- Gasification produces electricity with significantly reduced environmental impacts compared to conventional technologies. Compared to the old coal-burning plants, gasification can capture carbon dioxide much more efficiently and at a lower cost.
- There are concerns over gasification of waste such as the undermining of zero waste programs. Such as Incineration and landfilling problems.

Landfilling.....



In Gasification Municipal solid waste is converted in to Syngas,
It is eliminating the need for landfilling.

Incineration.....



Gasification also eliminates the need for incineration

Gasification Vs Incineration

- Gasification is not incineration.
- Incineration is the burning of fuels in an oxygen-rich environment, where the waste material combusts and produces heat and carbon dioxide, along with a variety of other pollutants.
- Gasification is the conversion of wastes into their simplest molecules.
- The process works by having carbon monoxide, hydrogen and methane form a syngas which then can be used for generating electricity or producing valuable products that can be cleaned of pollutants before being used, unlike incineration.
- The oxygen-deficient atmosphere in a gasifier does not provide the surroundings needed for dioxins and furans to form or reform.
- When the syngas is primarily used as a fuel for making heat, it can be cleaned as necessary before combustion which cannot occur in incineration.

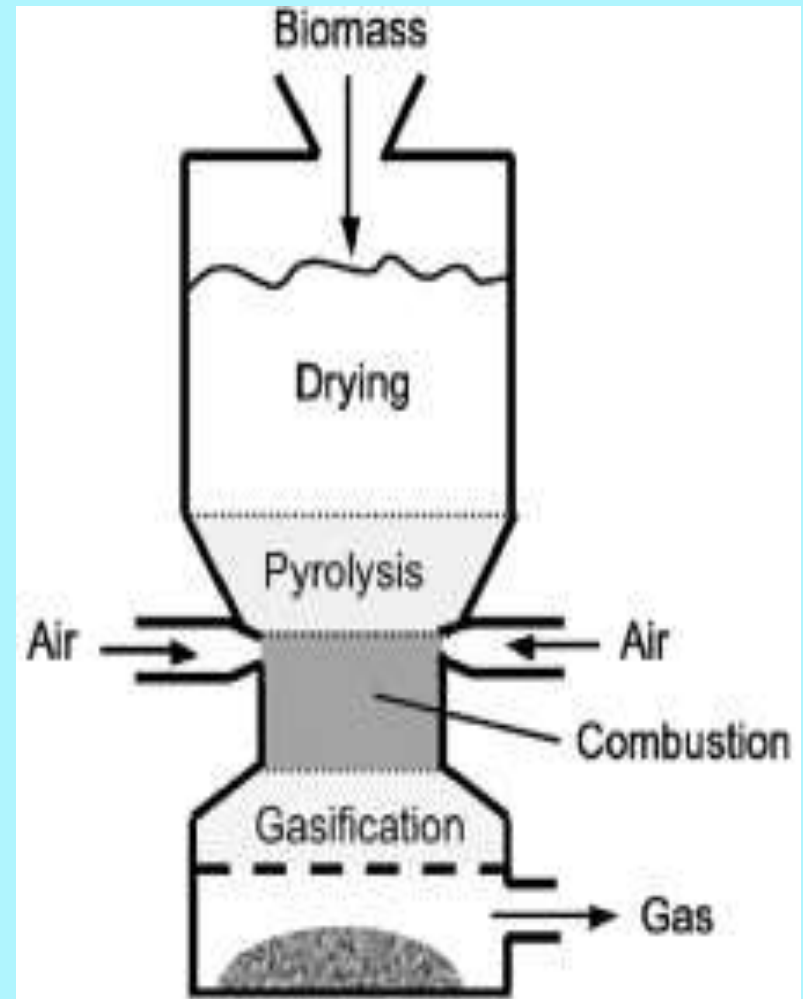
Gasification in Thermal Power Plant.....

A coal gasification power plant : First, the coal gases are fired in a gas turbine to generate electricity. The hot exhaust of the gas turbine, and some of the heat generated in the gasification process, are then used to generate steam. This dual source of electric power, called a "*combined cycle*". The fuel efficiency of a coal gasification power plant in this type of combined cycle can be boosted to 50 percent or more.



Gasification Chamber.....

- A conversion efficiency of gasification is 60-70%
- The produced gas has low calorific value 1000-1200 Kcal/Nm³
- When used in DG set as dual fuel it can save 65-85% fuel.
- Gasification occurs at 1000 °C in Gasifier.
- Process carried out by supply of less air (partial combustion) less than stoichiometric requirements.



Gasification Reactions.....

A gasification system consists of four main stages:

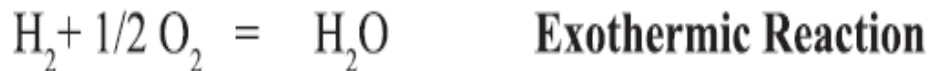
- ✓ Feeding of feedstock
- ✓ Gasifier reactions where gasification takes place
- ✓ Cleaning of resultant gas
- ✓ Utilization of cleaned gas.

Biomass gasifier (Figure 1) is a thermo chemical converter / reactor where various physical and chemical reactions take place. Biomass is passed through various zones – Drying/ Distillation Zone, Pyrolysis Zone, Combustion Zone & Reduction Zone. When the biomass is passed through all the above zones, it gets converted into a high quality combustible gas called Producer Gas.

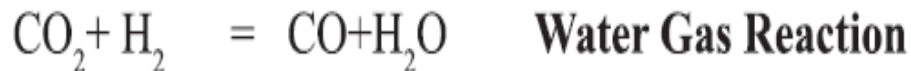


Gasification Reactions.....

The following reactions take place in the biomass gasification.



When these gases pass through bed of biomass converted charcoal in the reduction zone, the following main reactions take place:

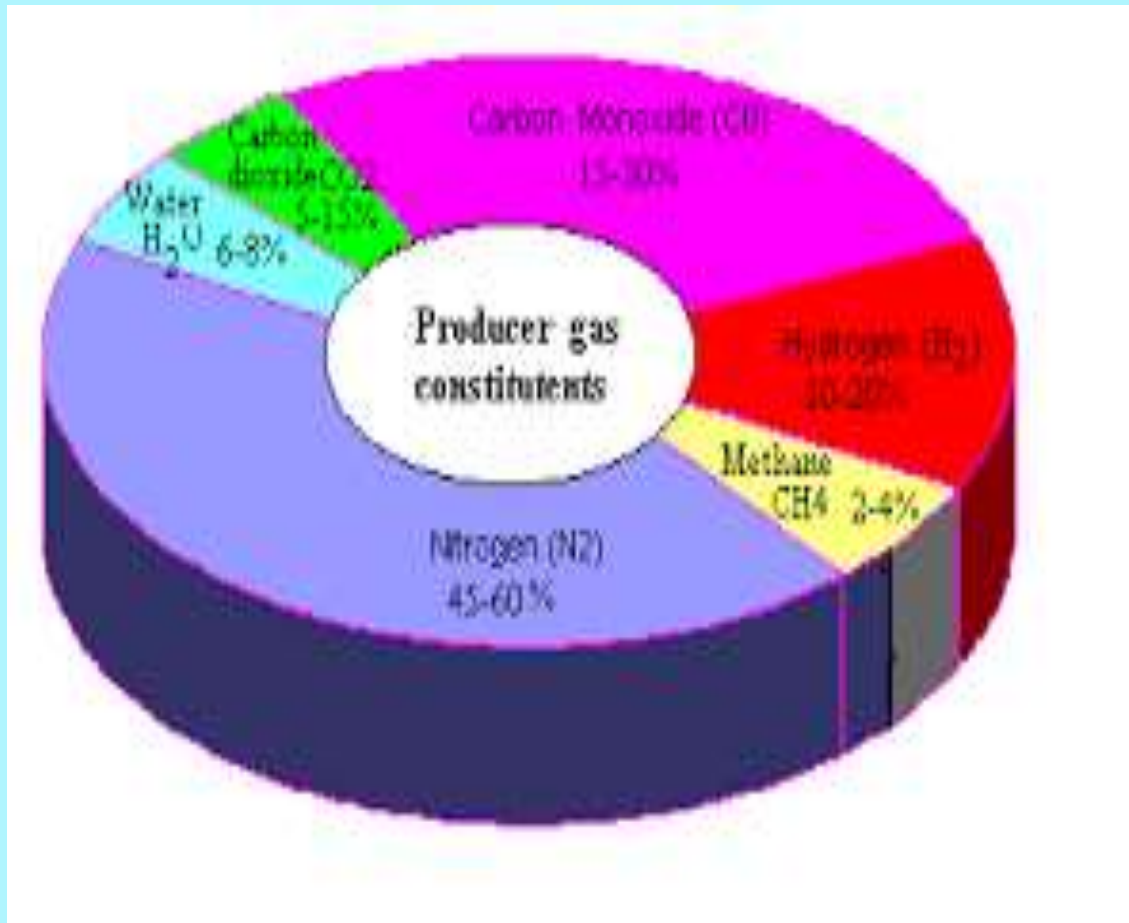


Syngas and Producer gas.....

Syngas is a mixture of Carbon Monoxide and Hydrogen which is the product of steam or oxygen gasification of organic material such as biomass. ... If the gasification product contains significant amount of non-combustible **gases** such as nitrogen and carbon dioxide, the term used for such mixtures is '**Producer Gas**'.

In the USA, producer gas is a generic term referring to wood gas, town gas, synthesis gas, syngas or raw gas.

Producer Gases



$$\text{CO} = 19 \pm 3 \%$$

$$\text{H}_2 = 18 \pm 2 \%$$

$$\text{N}_2 = 50 \pm 2 \%$$

$$\text{CH}_4 = 03 \pm 1 \%$$

$$\text{CO}_2 = 10 \pm 3 \%$$

Gasifier Efficiency

Average conversion efficiency of a gasifier is given by the following formula:

$$\eta_{\text{Gas}} = \frac{\text{Calorific value of gas/kg of fuel}}{\text{Avg. calorific value of 1 kg of fuel}}$$

Spark ignition engines running on producer gas on an average produces 0.55-0.75 kWh of energy from 1 kg of biomass. Compression ignition (diesel) engines cannot run completely on producer gas. Thus to produce 1 kWh of energy they consume 1 kg of biomass and 0.07 litres of diesel. Consequently they effect 80-85% diesel saving.

Example.....

Example:

1 kg of wood produces 2.5 m³ of gas with average calorific value of 1000 kcal/Nm³. Average calorific value of wood (dry) is 3500 kcal/kg.

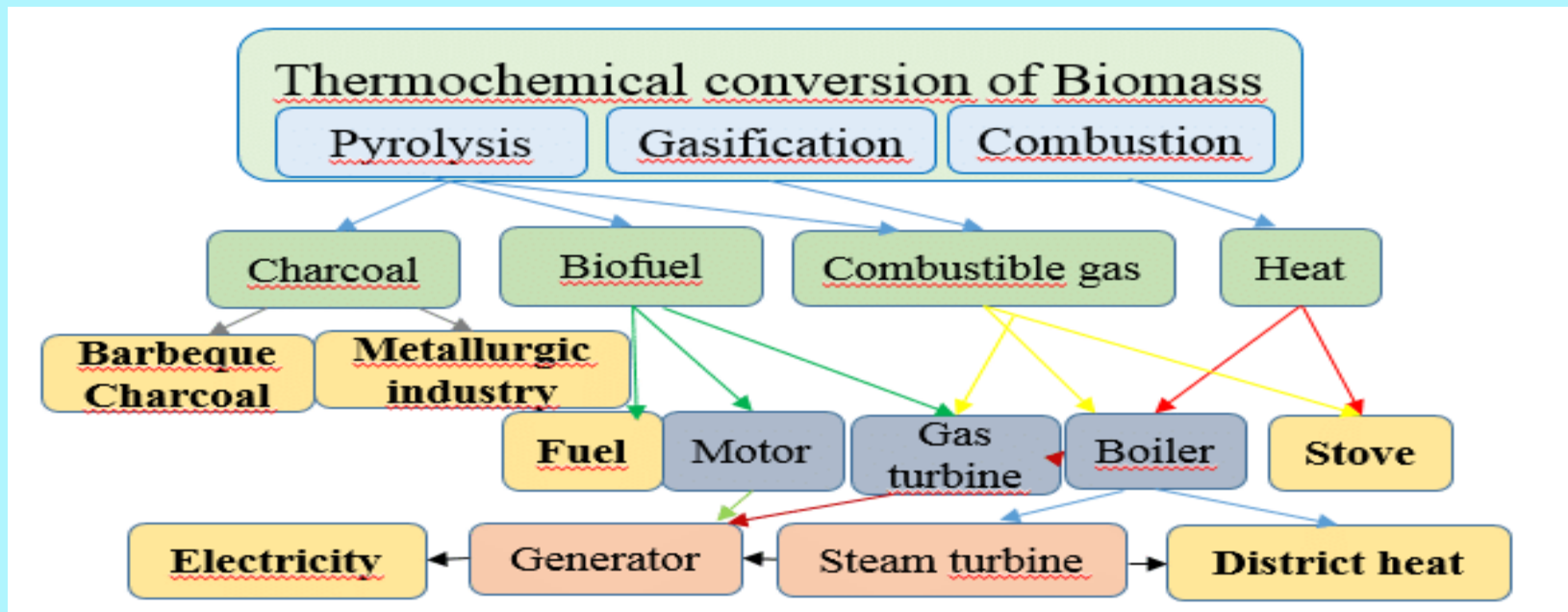
Hence

$$\eta_{\text{Gas}} = \frac{2.5 \text{ (m}^3\text{/kg of wood)} \times 1000 \text{ kcal/m}^3}{3500 \text{ kcal/kg of wood}} = 71\%$$

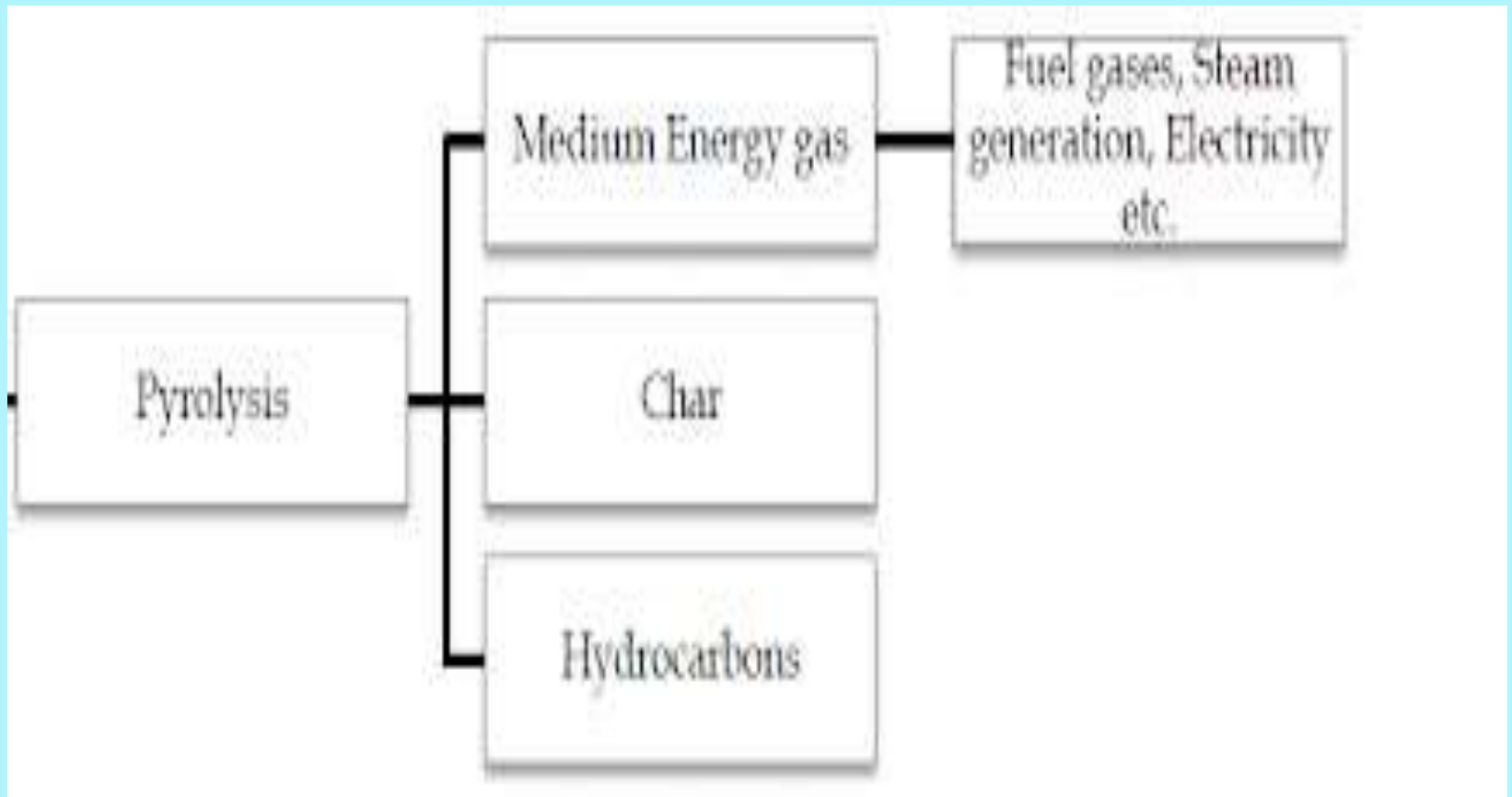
Biomass Energy Conversion.....

Thermo-Chemical Conversion

- In this reaction organic biomass is converted in to more valuable and convenient form of products as gaseous and liquid fuels, residue and Bi-products



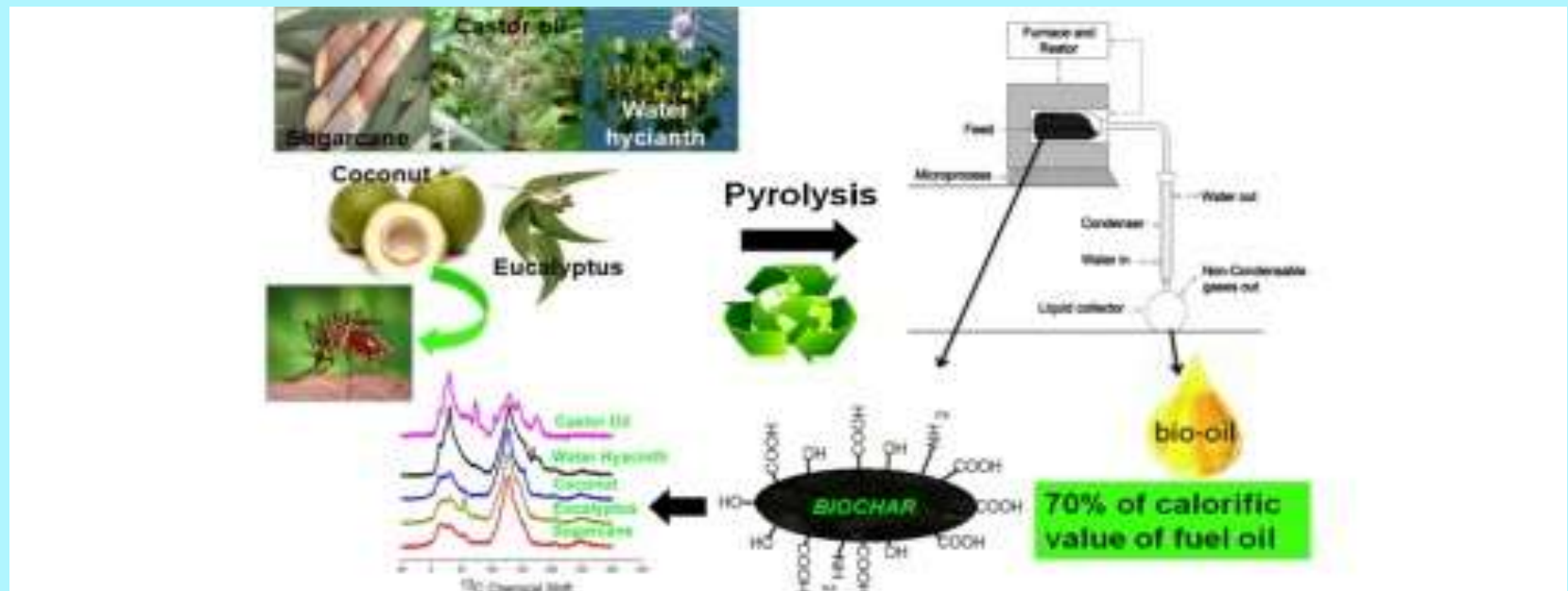
PYROLYSIS.....



PYROLYSIS.....

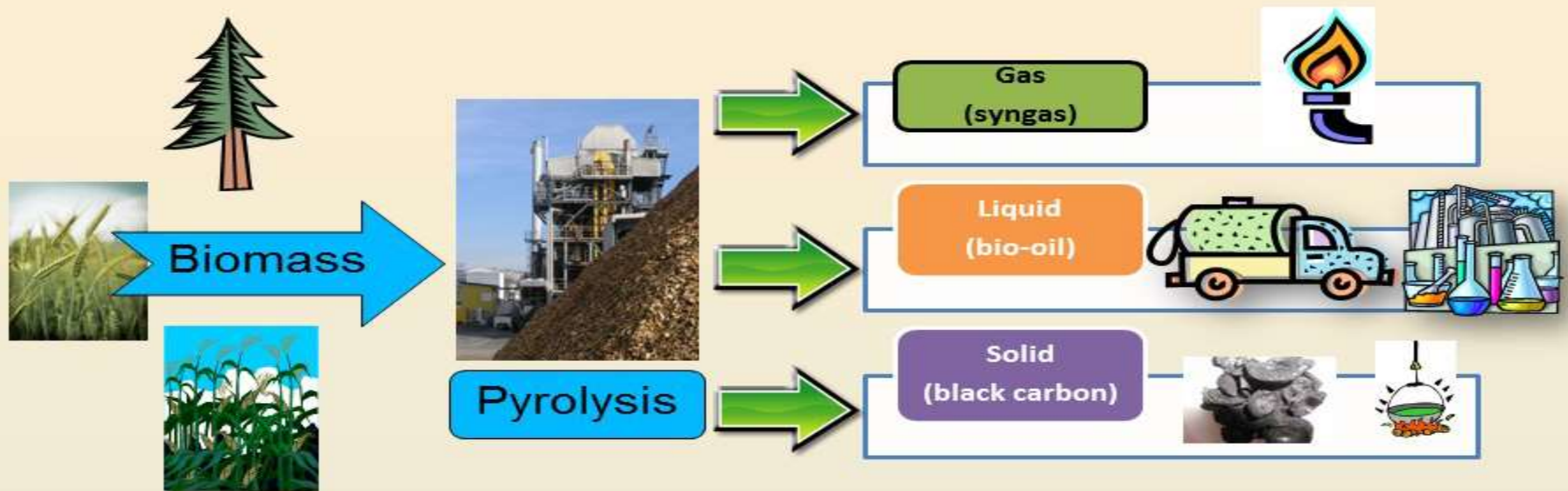
Definition : *Pyrolysis is the process of heating organic material at high temperatures in the absence of oxygen.*

It is the heating of biomass in a closed vessel at temperatures in the range 500°C-900°C in absence of O₂/air or with steam. It produces solid, liquid and gases.



PYROLYSIS.....

Pyrolysis is the process of heating organic material at high temperatures in the absence of oxygen. Since no oxygen is present, the organic material does not combust. Instead, the chemical compounds (i.e. cellulose, hemicellulose and lignin) that make up the material decompose into combustible gases and charcoal.



Building Blocks → Tear apart and reorganize → Form new compounds and chemicals

PYROLYSIS Products.....

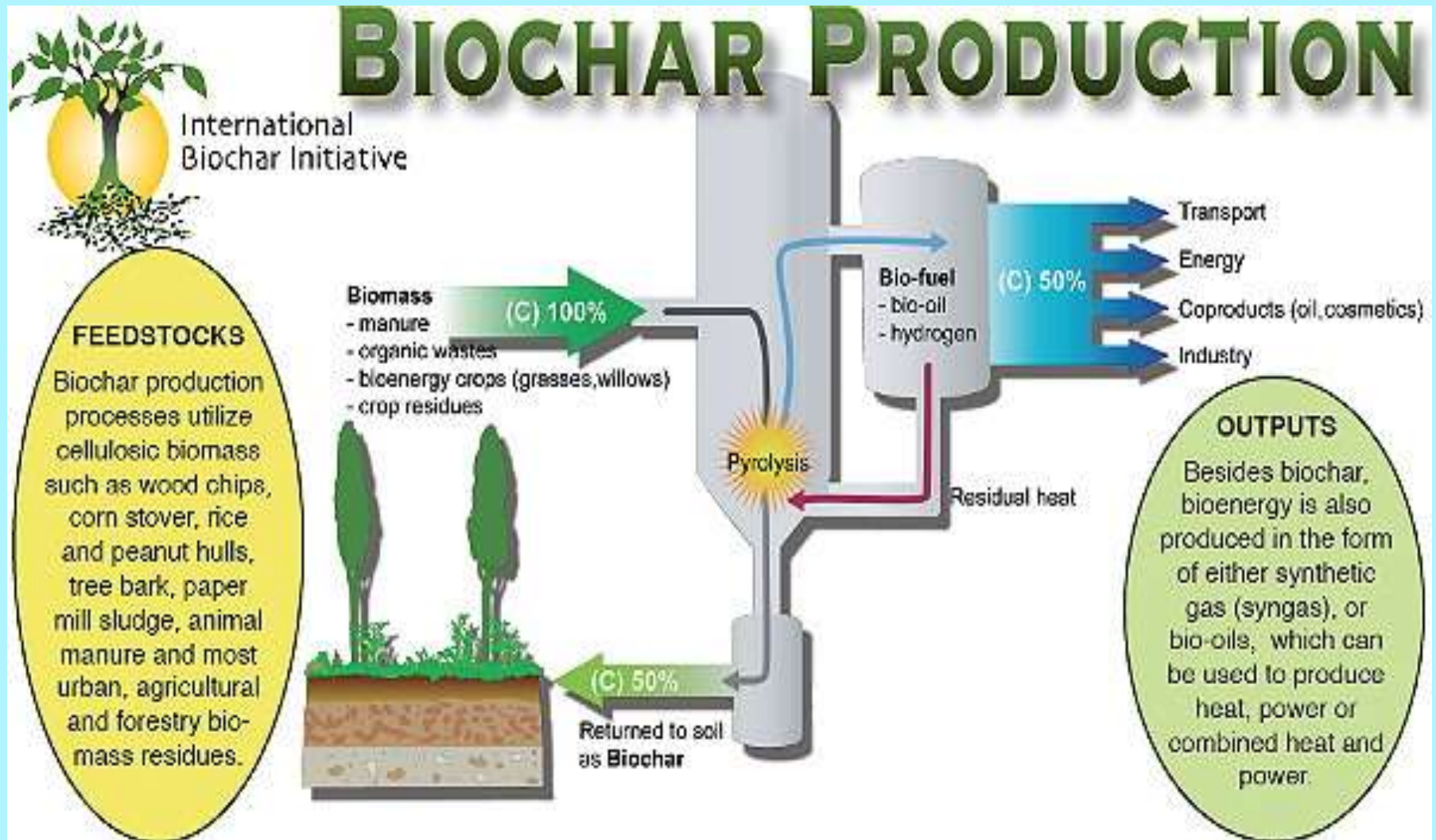
Pyrolysis of biomass produces three products - a liquid, a gas and a solid:

Bio-oil: comes from the combustible gases of the process, which are condensed, into a liquid. Bio-oil can be used as a low grade diesel oil.

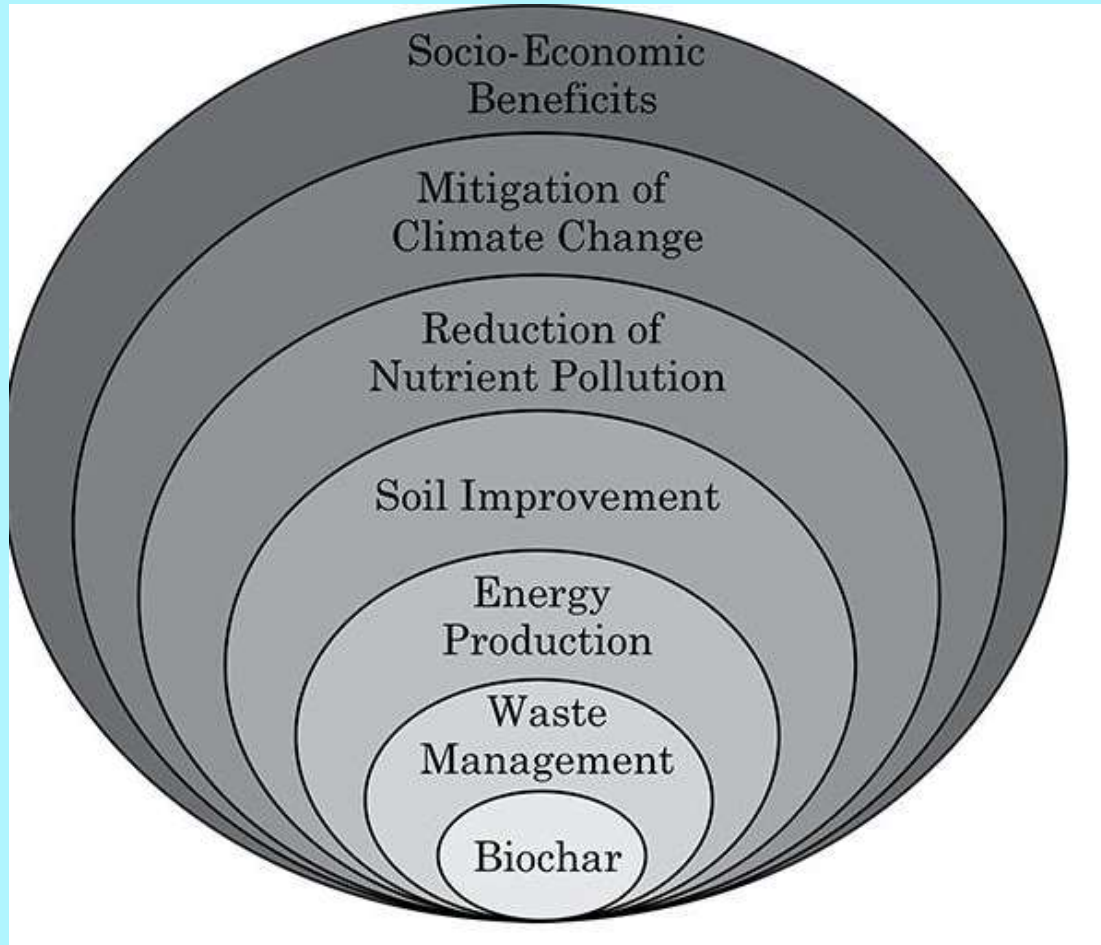
Bio-char: is the solid material that is created in the process. Bio-char is being promoted for its potential to improve soil properties and fertility as well as sequester carbon.

Syngas: are the permanent gases (CO_2 , CO , H_2 , light hydrocarbons) that remain after the pyrolysis process is complete. Syngas can be used in place of natural gas or converted with catalysts to ethanol.

Bio-Char.....

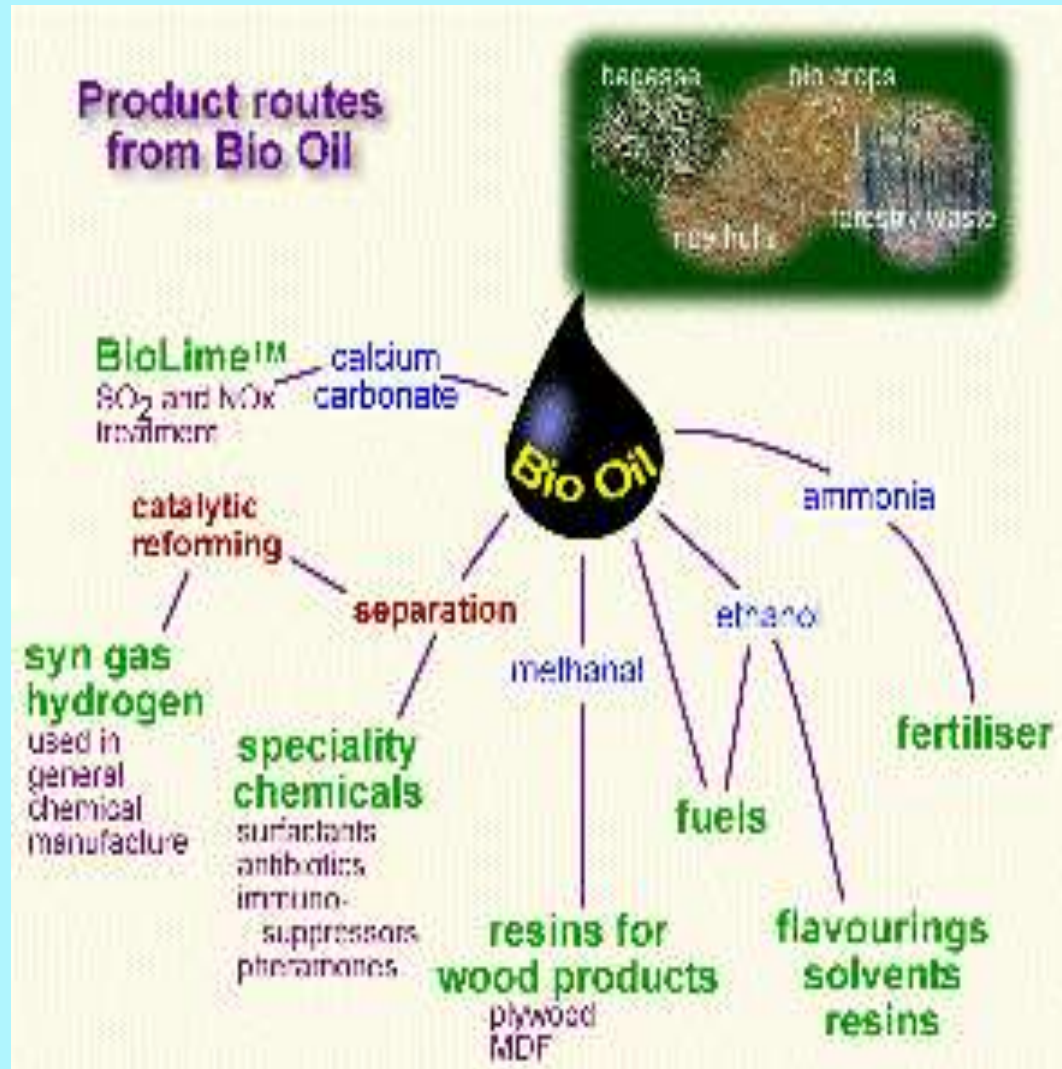


Bio-Char.....



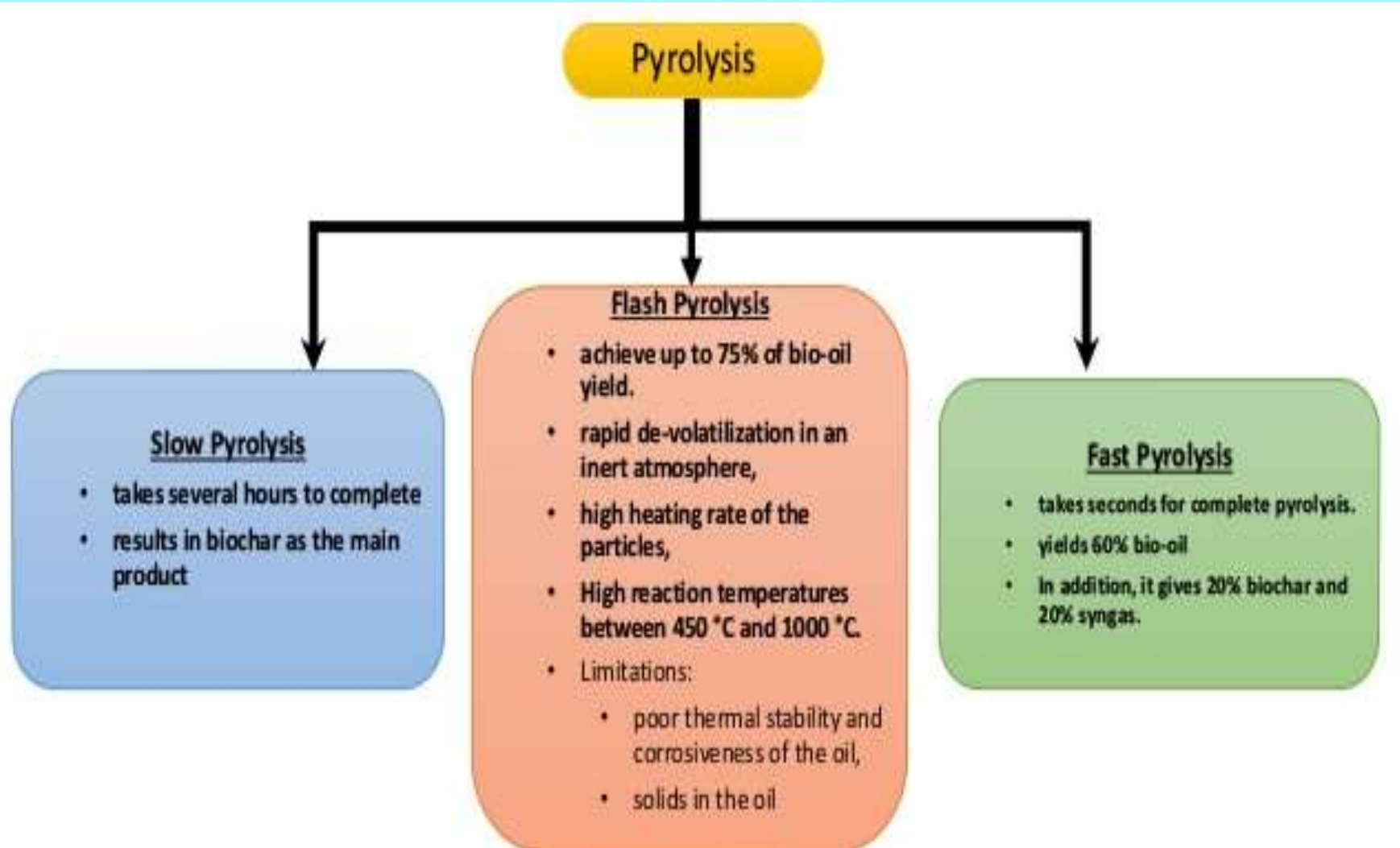
- As a solid Fuel in boiler
- Used further for the gasification process to obtain hydrogen rich gas
- Used directly as activated carbons
- Useful as a sorbent for air pollution control as well as for wastewater treatment.
- Serve as catalysts and catalyst supports

Bio-Oil.....



- As a combustion fuel
- Used for power generation
- Can be used for production of chemicals and resins
- Can be used as a transportation fuel
- Suitable blend with diesel oil may be used as diesel engine fuels
- Easily stored and transported

PYROLYSIS Types.....



PYROLYSIS Types.....

Process	Conditions	Products		
		Liquid	Char	Gas
Fast pyrolysis	Moderate temperature, short residence time	75% Water 25 %	12%	13%
Slow pyrolysis (carbonization)	Low temperature, long residence time	30% Water 75 %	35%	35%
High temperature pyrolysis	High temperature	5% Tars 5 %	10%	85%

PYROLYSIS Types.....

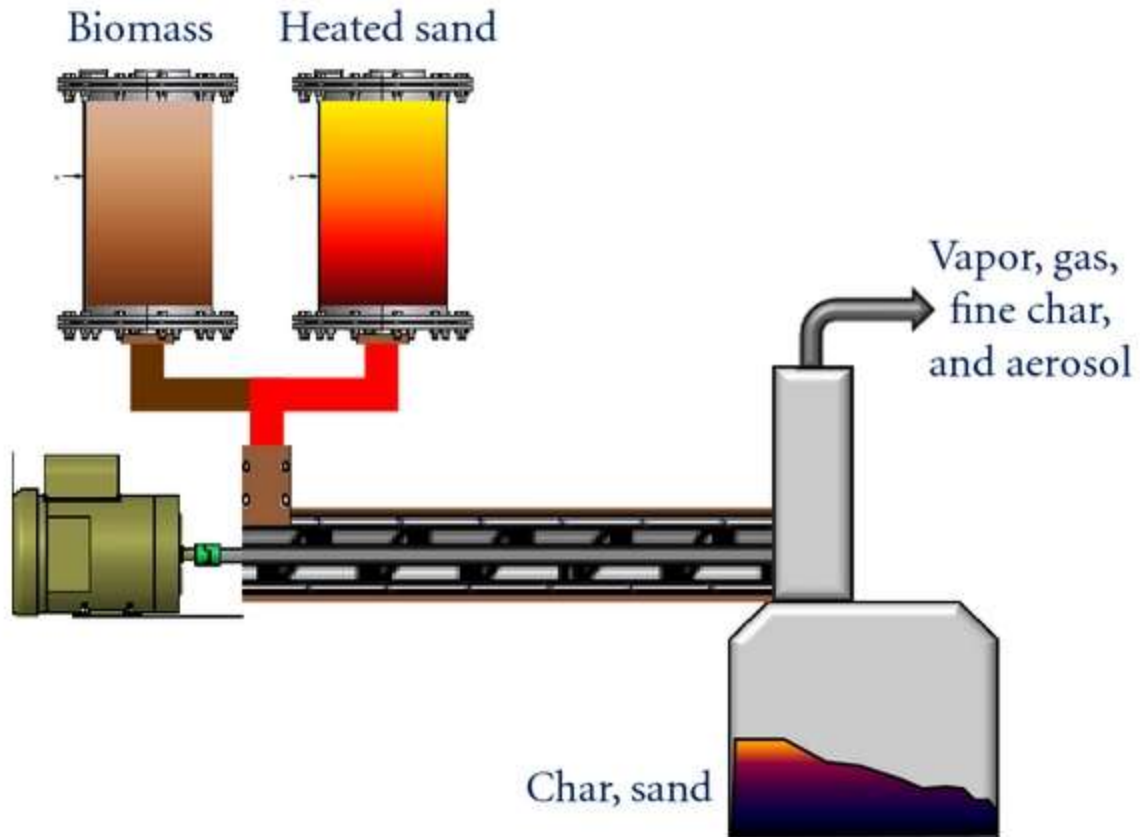
- Fast Pyrolysis optimizes bio-oil production by increasing the rate of pyrolysis temperature to 1000°C/second.
- This process yields approximately 60-70% bio-oil, 15-25% bio-char, and 10-15% syngas.
- In contrast, Slow Pyrolysis uses slower heating rates and bio-char is the major end product.
- In both process, the system is self-sustaining by harnessing the syngas to drive the reaction.

PYROLYSIS Types.....

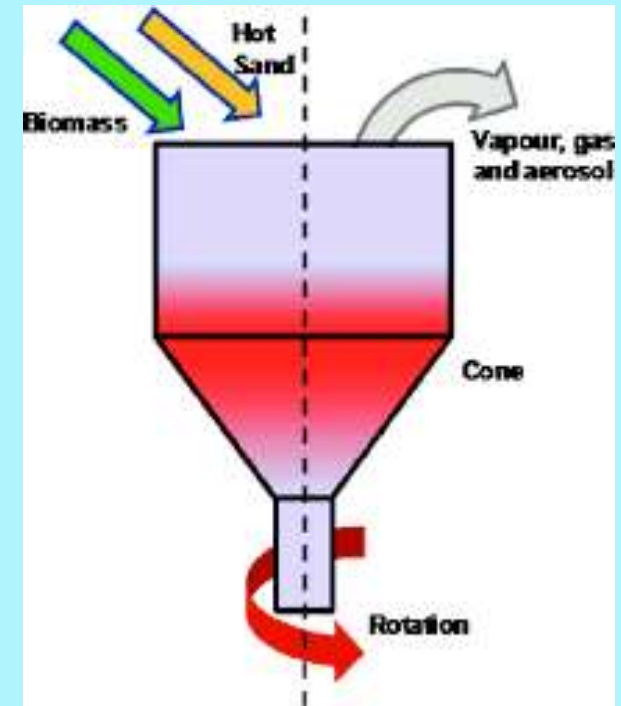
Classification of pyrolysis methods with differences in temperature, residence time, heating rate, and major products. The major products are listed in decreasing importance.

Method	Temperature (°C)	Residence Time	Heating rate (°C/s)	Major products
Conventional/slow pyrolysis	Med-high 400-500	Long 5-30 min	Low 10	Gases Char Bio-oil (tar)
Fast pyrolysis	Med-high 400-650	Short 0.5-2 s	High 100	Bio-oil (thinner) Gases Char
Ultra-fast/flash pyrolysis	High 700-1000	Very short < 0.5 s	Very high >500	Gases Bio-oil

PYROLYSIS Reactor ...



Screw Pyrolysis Reactor



Rotating Cone Reactor

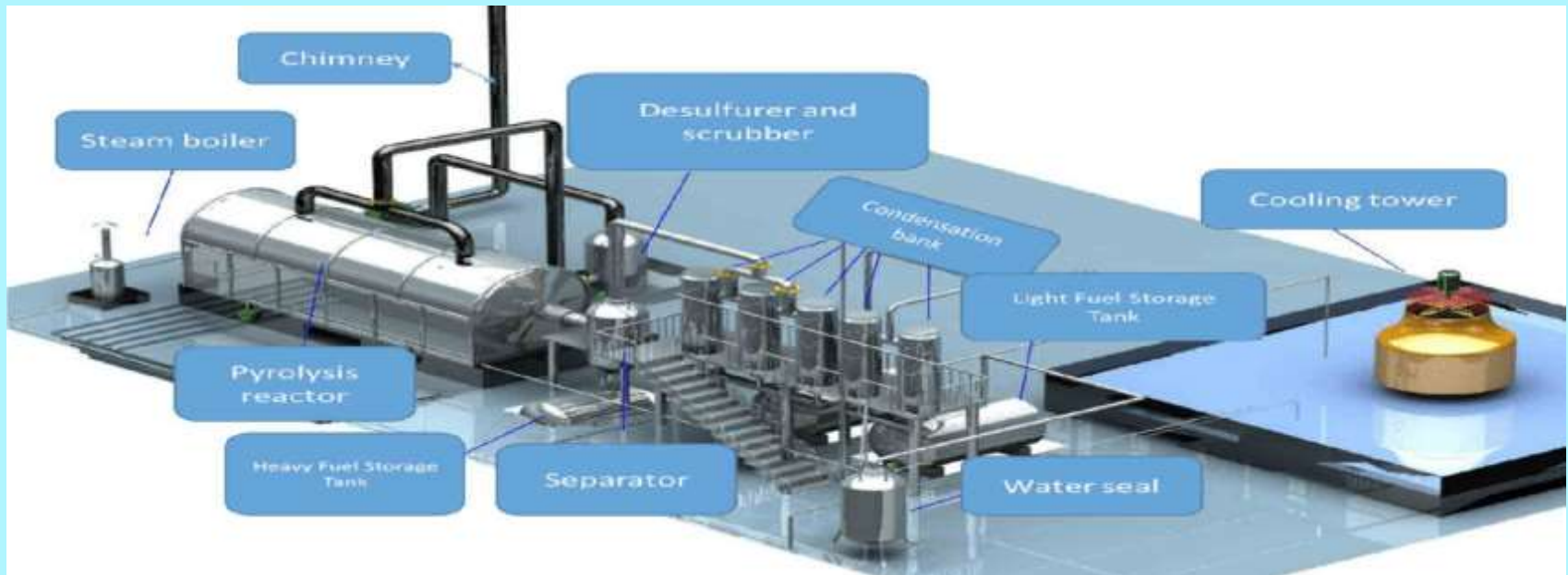
Pyrolysis Advantages

Advantage...

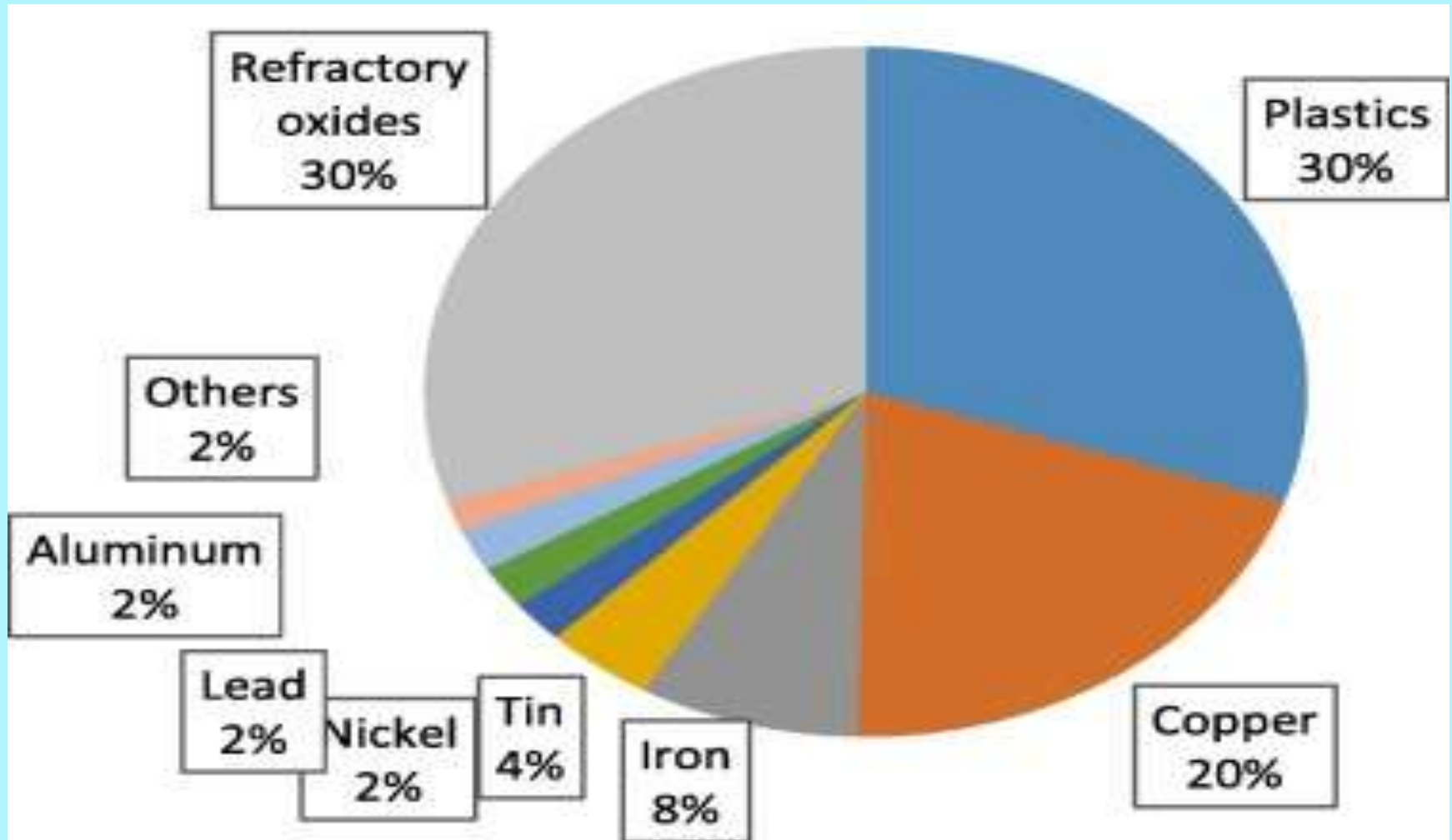
- Pyrolysis presents an opportunity to manufacture low-carbon liquid fuels and decrease the emissions of the industry.
- Pyrolysis can be carbon negative if the bio-char produced is buried into the soil and used for crop enhancement instead of combustion
- The fuels produced are considered second generation fuels since the feed stocks are from renewable sources, like waste and biomass. As a result, they are less carbon-intensive than fossil fuels
- These fuels are also compatible with our existing transportation infrastructure which eliminates the need to introduce costly new infrastructure

Pyrolysis Limitations

- Pyrolysis does have higher costs associated with the machinery and heating.
- It is dependent on a supply of cheap biomass.
- The quality of the bio-oil is also low grade and cannot be used in all applications where fossil fuels are used.



Waste Management Sector.....



Potential of pyrolysis processes in the waste management sector

Comparison.....

	Combustion	Gasification	Pyrolysis
Oxidizing Agent	Greater than stoichiometric supply of oxygen*	Less than stoichiometric oxygen* or steam as the oxidizing agent	Absence of oxygen or steam
Typical Temperature Range with Biomass Fuels	800°C to 1200°C (1450°F to 2200°F)	800°C to 1200°C (1450°F to 2200°F)	350°C to 600°C (660°F to 1100°F)
Principle Products	Heat	Heat and Combustible gas	Heat, Combustible liquid and Combustible gas
Principle Components of Gas	CO ₂ and H ₂ O	CO and H ₂	CO and H ₂

Biomass Energy next

Biomass to Energy Conversion Pathways

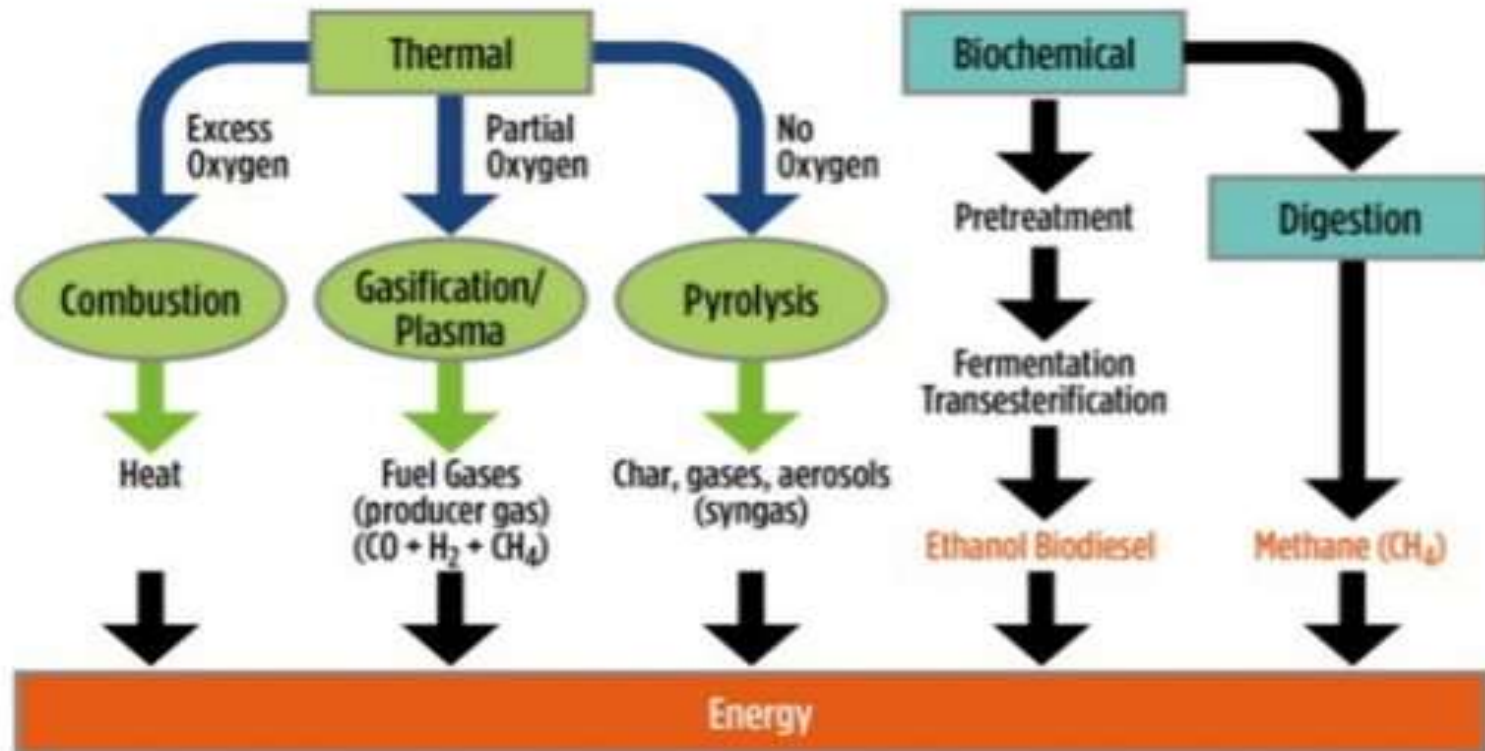
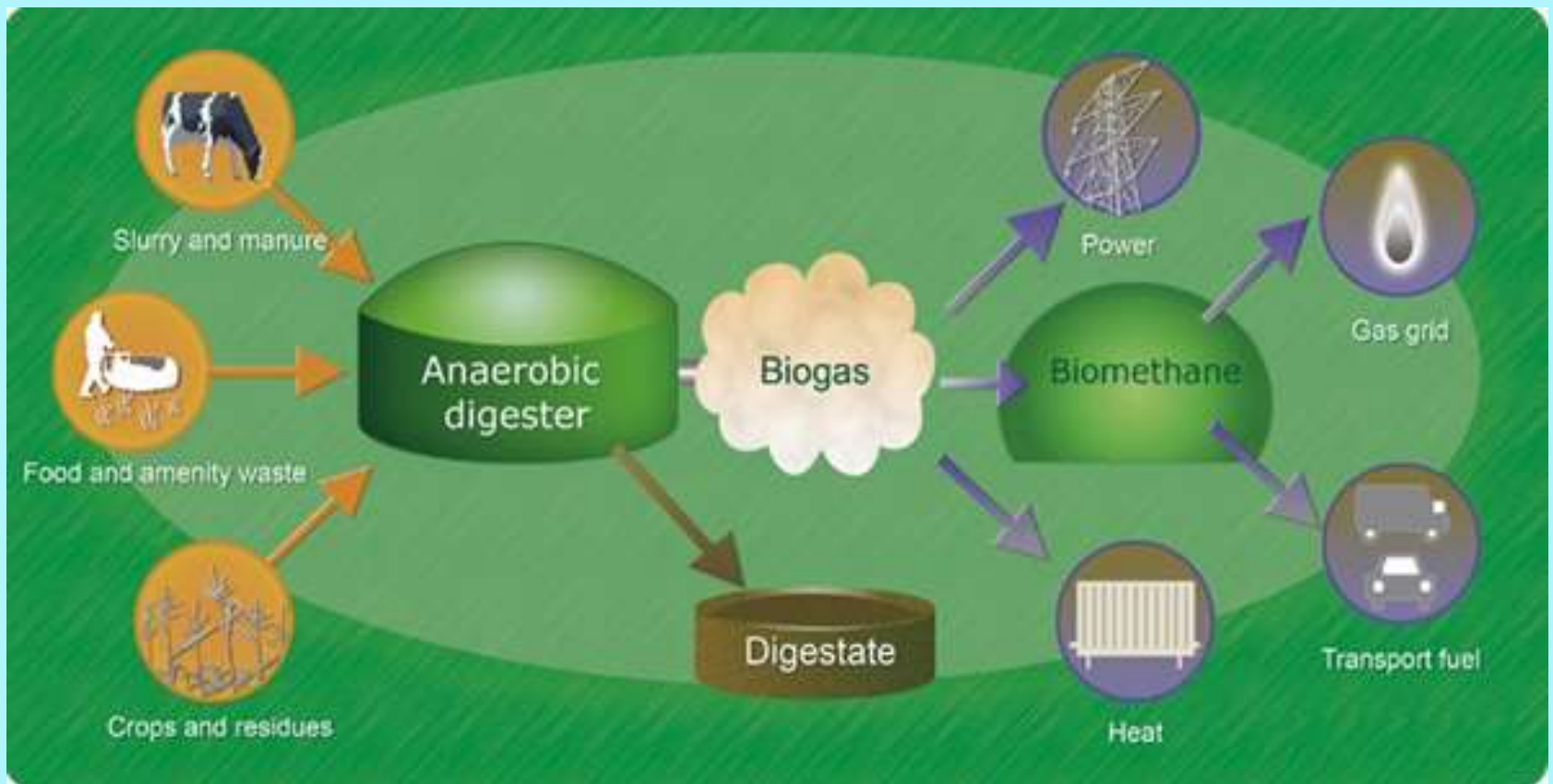


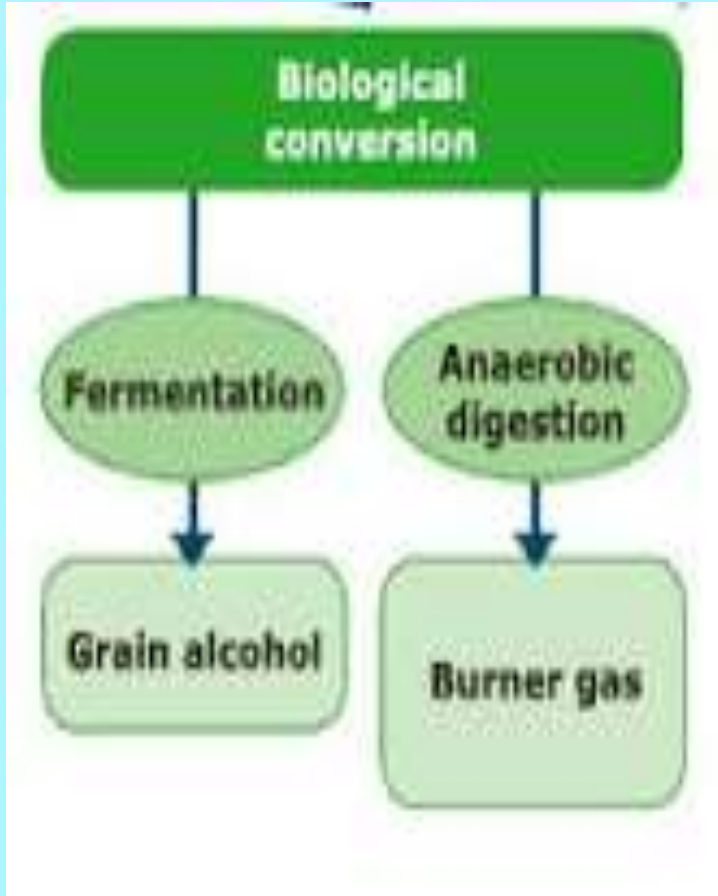
Illustration by NREL

Biochemical conversion of Biomass

Biochemical conversion of **biomass** involves use of bacteria, microorganisms and enzymes to Breakdown **biomass** into gaseous or liquid fuels, such as biogas or bioethanol.



Biochemical Energy Transform

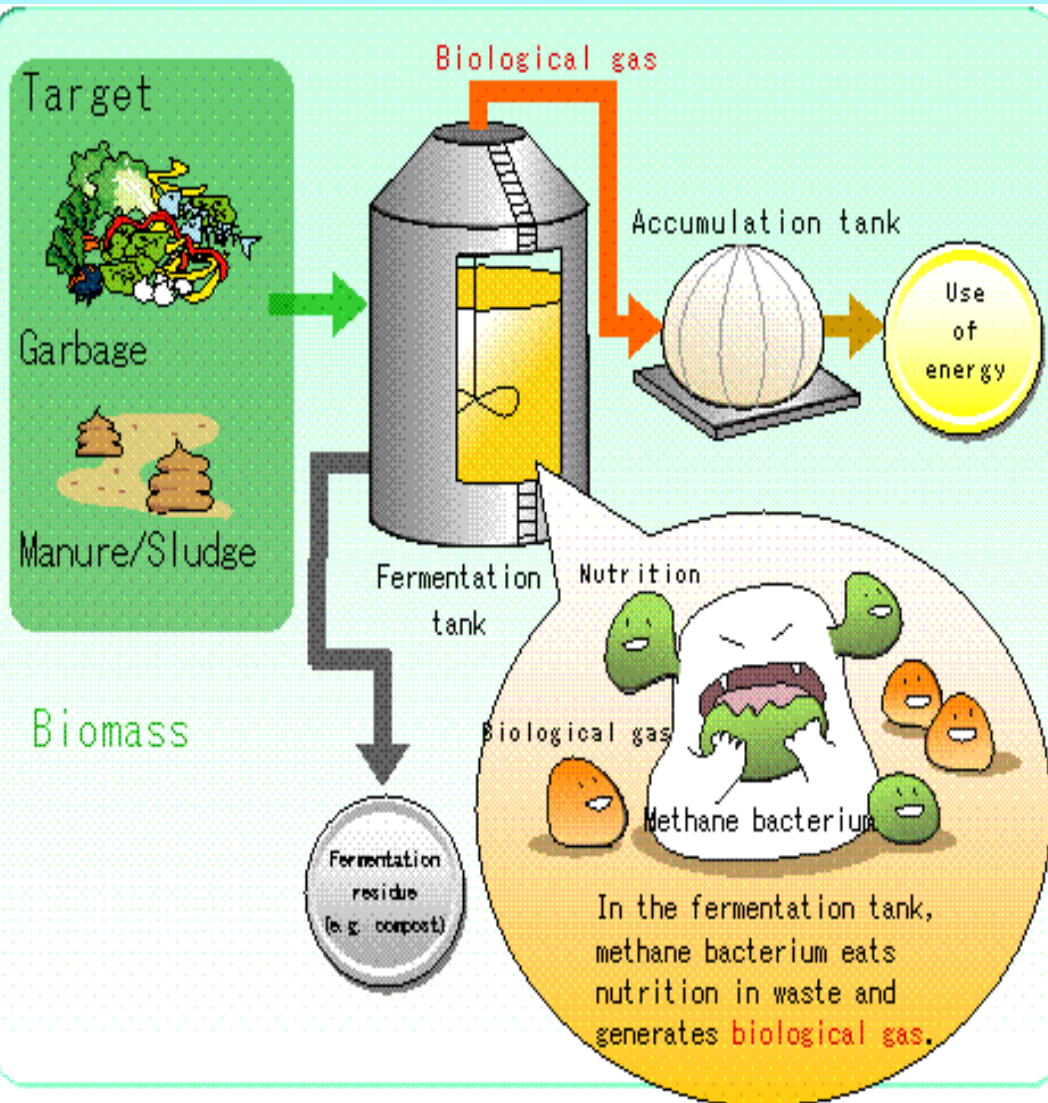


In Biochemical processes the bacteria and micro organisms are used to transform the raw biomass into useful energy like *Methane and Ethane gas*.

Following organic treatments are given to the biomass:

- 1) Fermentation of biomass (Aerobic digestion)
- 2) Anaerobic digestion of biomass

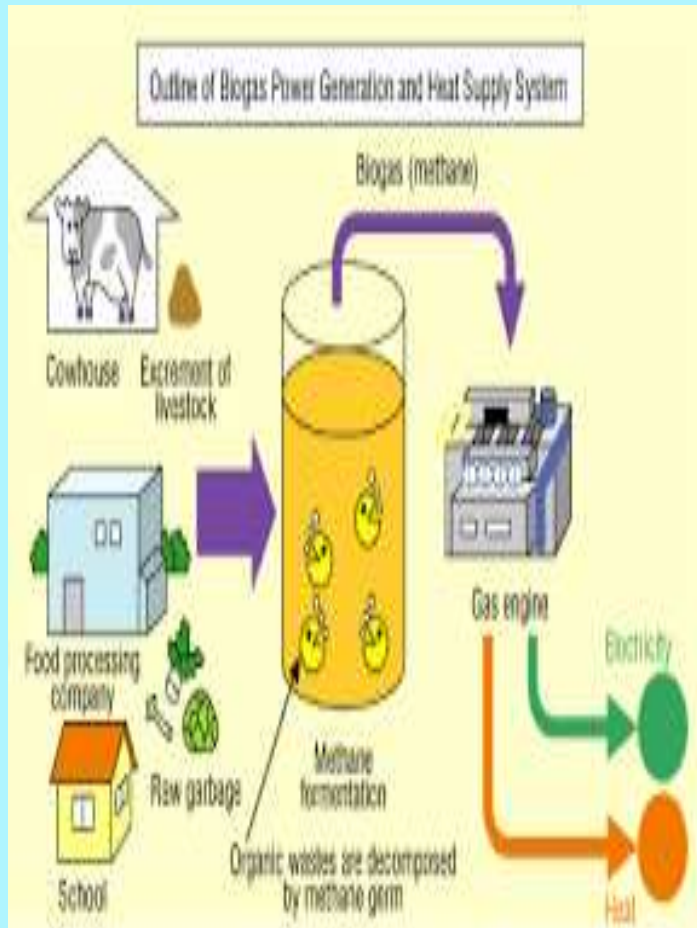
Fermentation.....



Fermentation is a process of decomposition of complex molecules of organic compound under the influence of micro-organism(ferment) such as yeast, bacteria, enzymes etc.

The example of fermentation process is the conversion of grains and sugar crops into ethanol and CO₂ in presence of yeast.

Fermentation.....



- The conversion of a plant's glucose (or carbohydrate) into an alcohol or acid.
- Yeast or bacteria are added to the biomass material, which feed on the sugars to produce ethanol (an alcohol) and carbon dioxide.
- The ethanol is distilled and dehydrated to obtain a higher concentration of alcohol to achieve the required purity for the use as automotive fuel.
- The solid residue from the fermentation process can be used as cattle-feed and in the case of sugar cane can be used as a fuel for boilers

Anaerobic Digestion.....

Definition:

Bio digestion or Anaerobic Digestion (AD) is a biological process that occurs when organic matter is decomposed by bacteria in the absence of oxygen to create biogas.

It is a conversion of decaying wet biomass and animal waste into biogas through decomposition process by the action of anaerobic bacteria.

The most useful biomass for production of biogas are animal and human waste, plant residue and other organic waste material with high moisture content.



Types of Anaerobic Digestion.....

There are two basic anaerobic digestion processes, which take place over different temperature ranges:

Mesophilic digestion:

The most commonly used process for anaerobic digestion. It takes place between 20°C and 40°C and can take a month or two to complete, produces less biogas and requires additional sanitation.

Thermophilic digestion:

Less common and more expensive technology. Typically takes place from 50-65°C, it requires additional energy to maintain the higher operating temperature. It produces more biogas in a shorter period faster but the bacteria are more sensitive.

Stages of Anaerobic Digestion.....

Anaerobic digestion has a defined process flow that consists of four distinct phases.

1.Pre-Treatment: In pre-treatment, wastes may be processed, separated, or mixed to ensure that they will decompose in the digester.

2.Digestion: During digestion, waste products are broken down by bacteria and biogas is produced and collected.

3.Biogas Processing: Biogas produced is either combusted or upgraded and then used to displace fossil fuels. During upgrading, scrubbers, membranes or other means are used to remove impurities and carbon dioxide (CO₂) from biogas.

Stages of Anaerobic Digestion.....

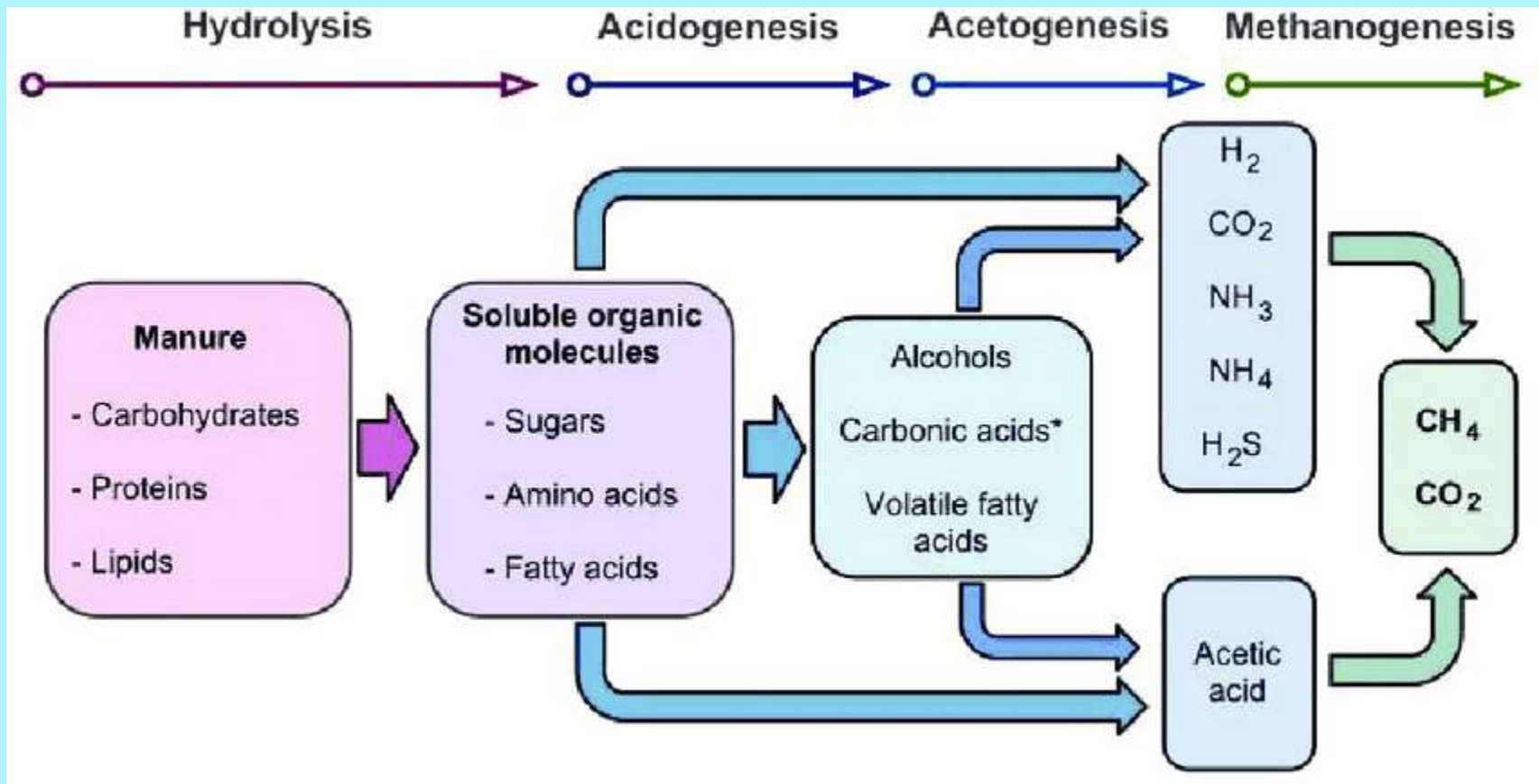
4.Reuse or disposal of solid digested waste: Digested waste has a high nutrient content and can be used as fertilizer so long as it is free of pathogens or toxins, or it can be composted to further enhance nutrient content.

Anaerobic digestion systems are often referred to as “anaerobic digesters”, “biodigesters” or “biogas recovery systems”.

Four stages can also define as

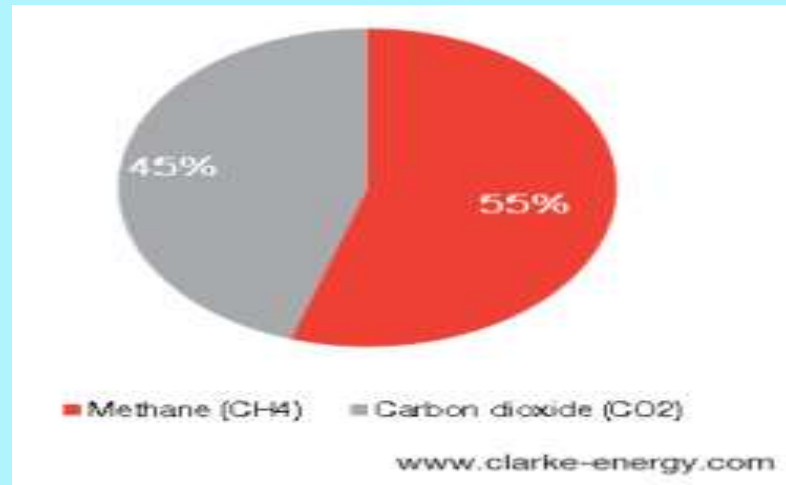
1. Hydrolysis
2. Acidogenesis
3. Acetogenesis
4. Methanogenesis

Stages of Anaerobic Digestion ...



Bio Gas How Useful?

- Bio Methane is composed mainly with Methane and carbon dioxide.
- Bio Methane can completely replace Natural gas.
- Bio Gas produced using cow dung is called as Gobor Gas
- Gobar gas comprises of 60% Methane and 40% Carbon dioxide.

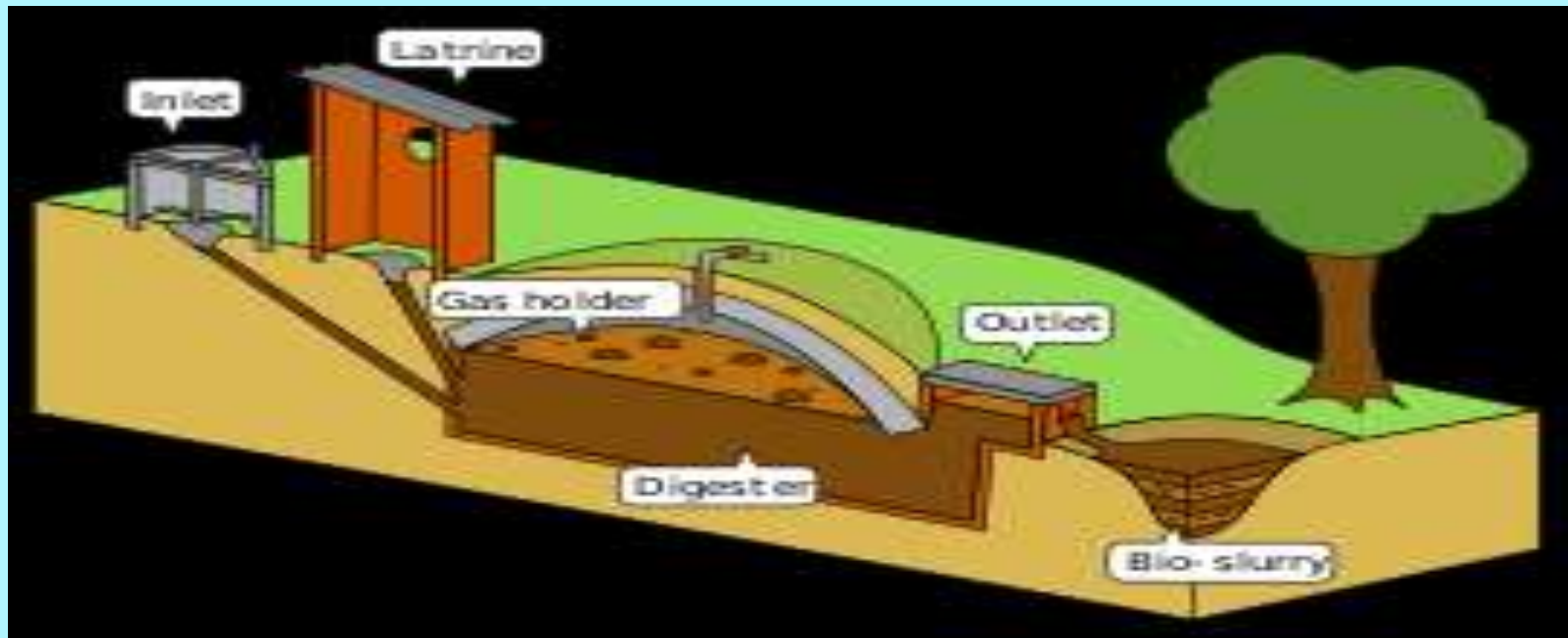


Bio Gas statistics

25 kg Fresh Dung		
Direct Burning		Biogas
5 kg dry dung	Product	1 cu m biogas
10460 kCal	Gross energy	4713 kCal
10%	Device efficiency	55%
1046 kCal	Useful energy	2592 kCal
None	Manure	10kg air dried manure

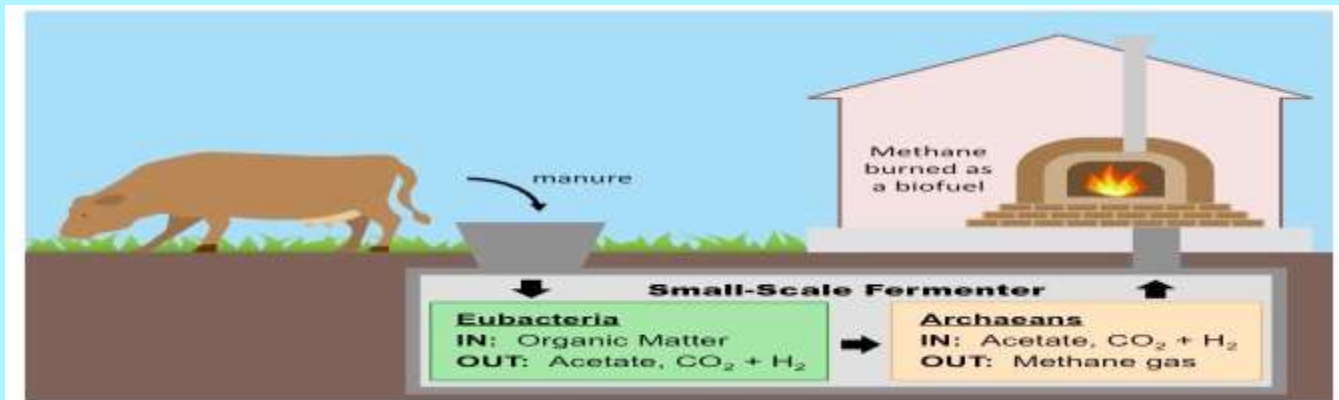
Biogas Generation.....

- Biogas contains 55-65% methane, 30-40% CO₂, and the remainders are impurities like H₂S, H₂, N₂ gases.
- Cattle dung can produce 0.037 m³ of biogas per kg of cow dung. The calorific value of gas is 21000 to 23000 kJ/kg or about 38000 kJ/m³ of gas.



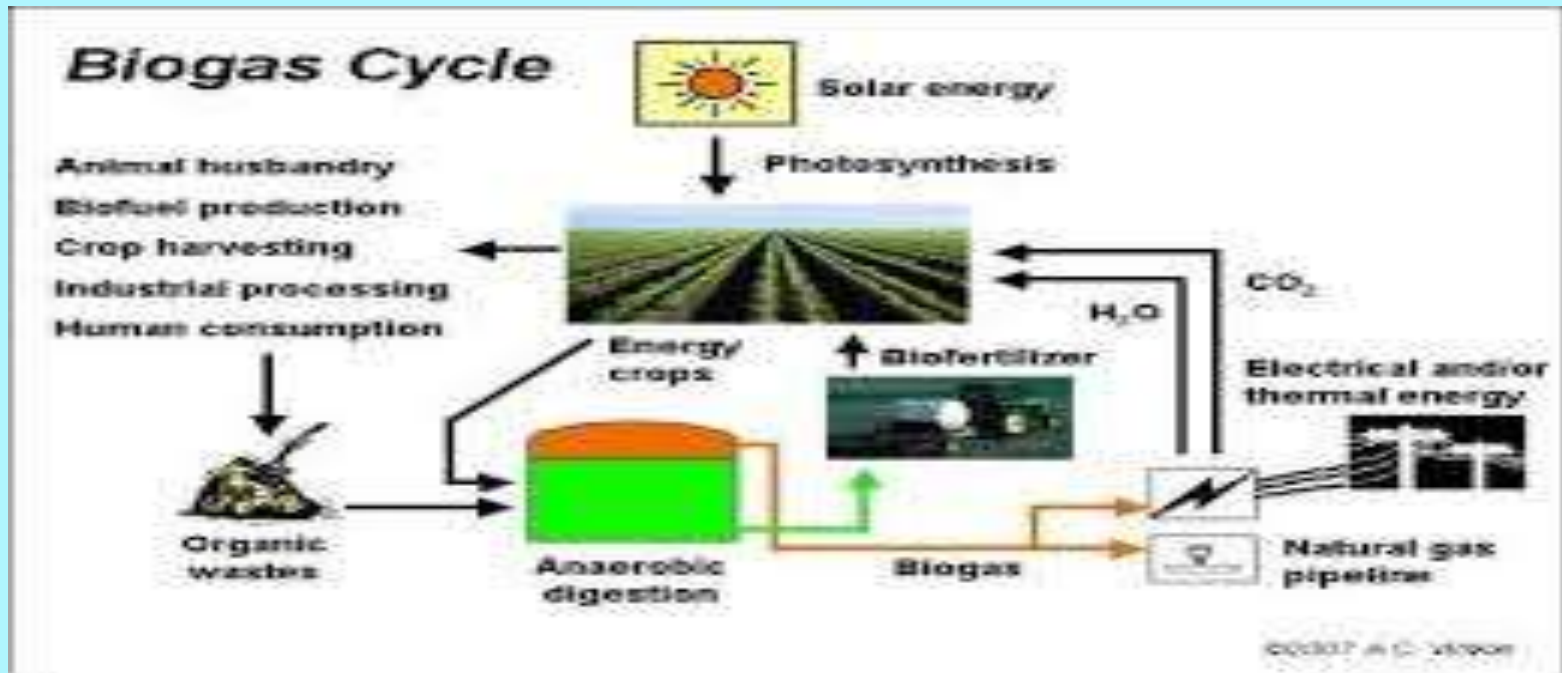
Biogas Generation.....

- The material from which biogas is produced retains its value as fertilizer or as animal feed which can be used after certain processing.
- Biogas can be produced by digestion pyrolysis or hydro gasification. Digestion is a biological process that occurs in absence of O_2 and in presence of anaerobic organisms at atmospheric pressure and temperatures of $35^{\circ}C$ - $70^{\circ}C$. The container in which the digestion takes place is called digester.



Biogas Generation.....

- When organic matter undergoes fermentation, the anaerobic bacteria extracts oxygen by decomposing the biomass at low temperatures up to 65°C in the presence of moisture. 80-95% of the gas so produced is called biogas.



Principle of Biogas Production.....

Biogas production takes place in three stages:

1) Hydrolysis: In this stage, matters with heavy molecular weight are disintegrated into lower molecular weight. This process takes place by hydrolytic bacteria.

2) Acid Formation: In this stage, organic matters are converted into acetates and H_2 . This conversion takes place by acetogenes. Then H_2 and C are converted into acetate by acetogenes.

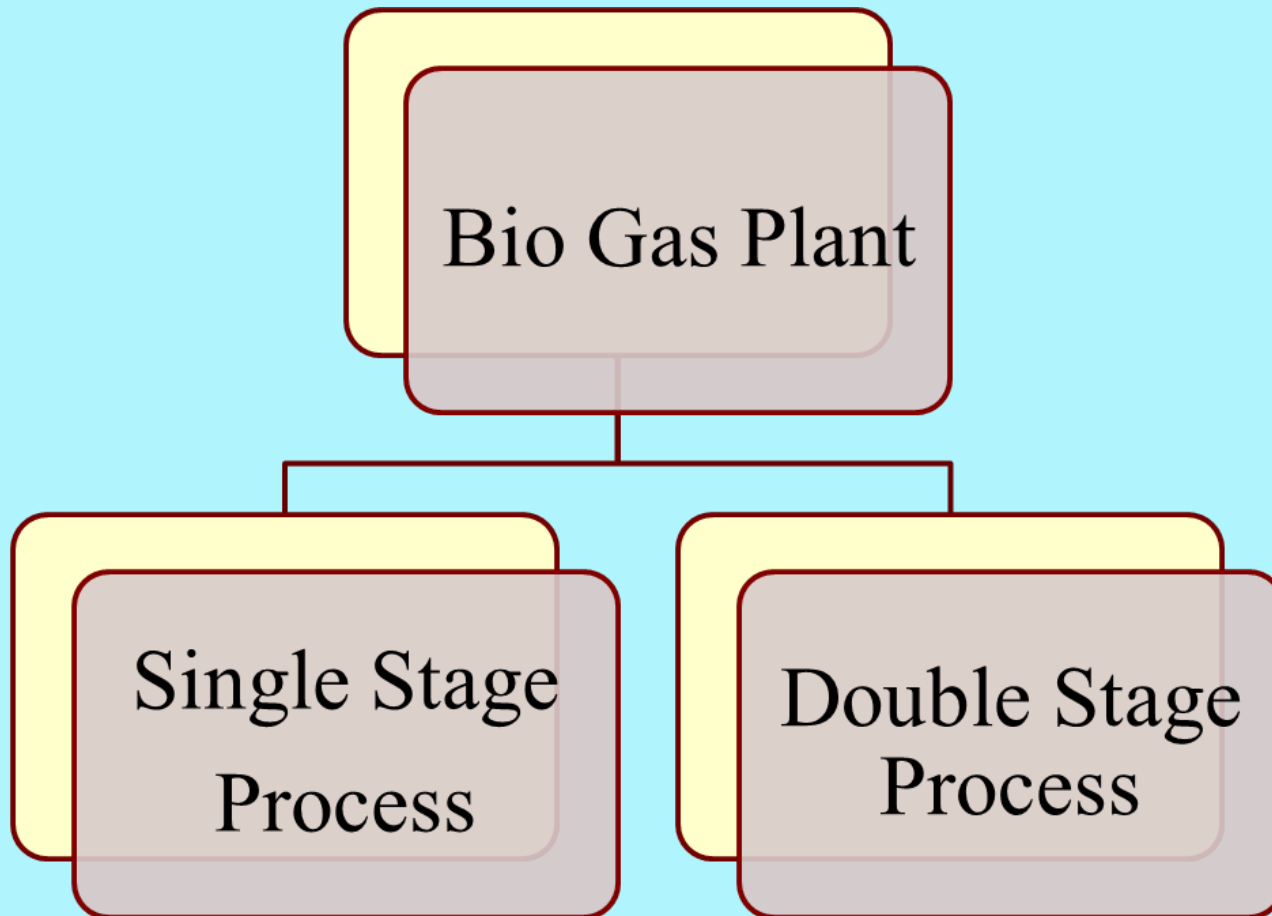
3) Methane Gas Formation: In this stage, acetates and simple CO_2 are converted into CH_4 . This is carried out by methanogenes.

Factors Affecting Biogas Production.....

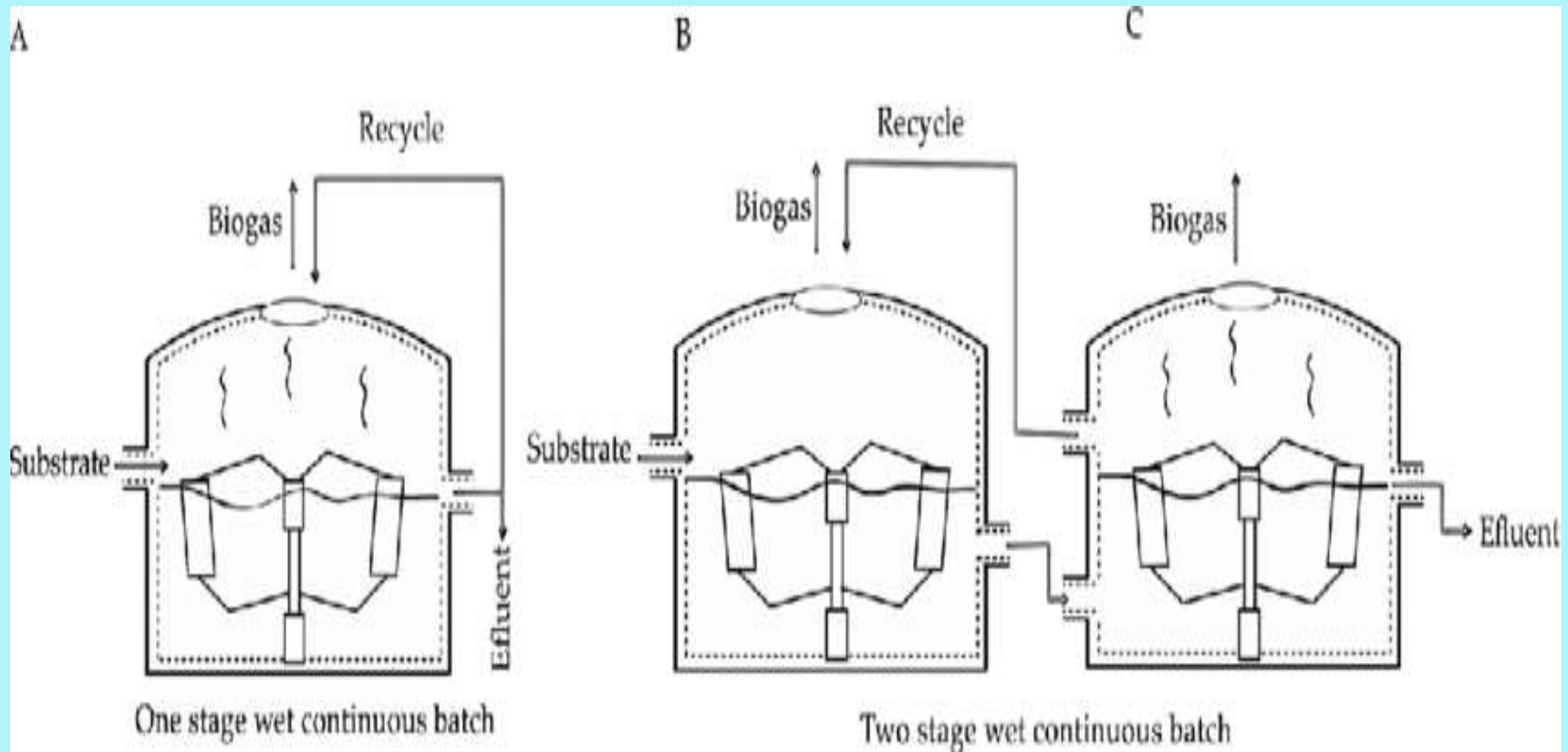
The rate of production of biogas depends on the following factors:

- 1) Temperature & Pressure
- 2) Solid concentration & Loading rate
- 3) Retention period
- 4) pH value
- 5) Nutrients composition
- 6) Toxic substances
- 7) Digester size & shape
- 8) Stirring agitation of the content of digestion

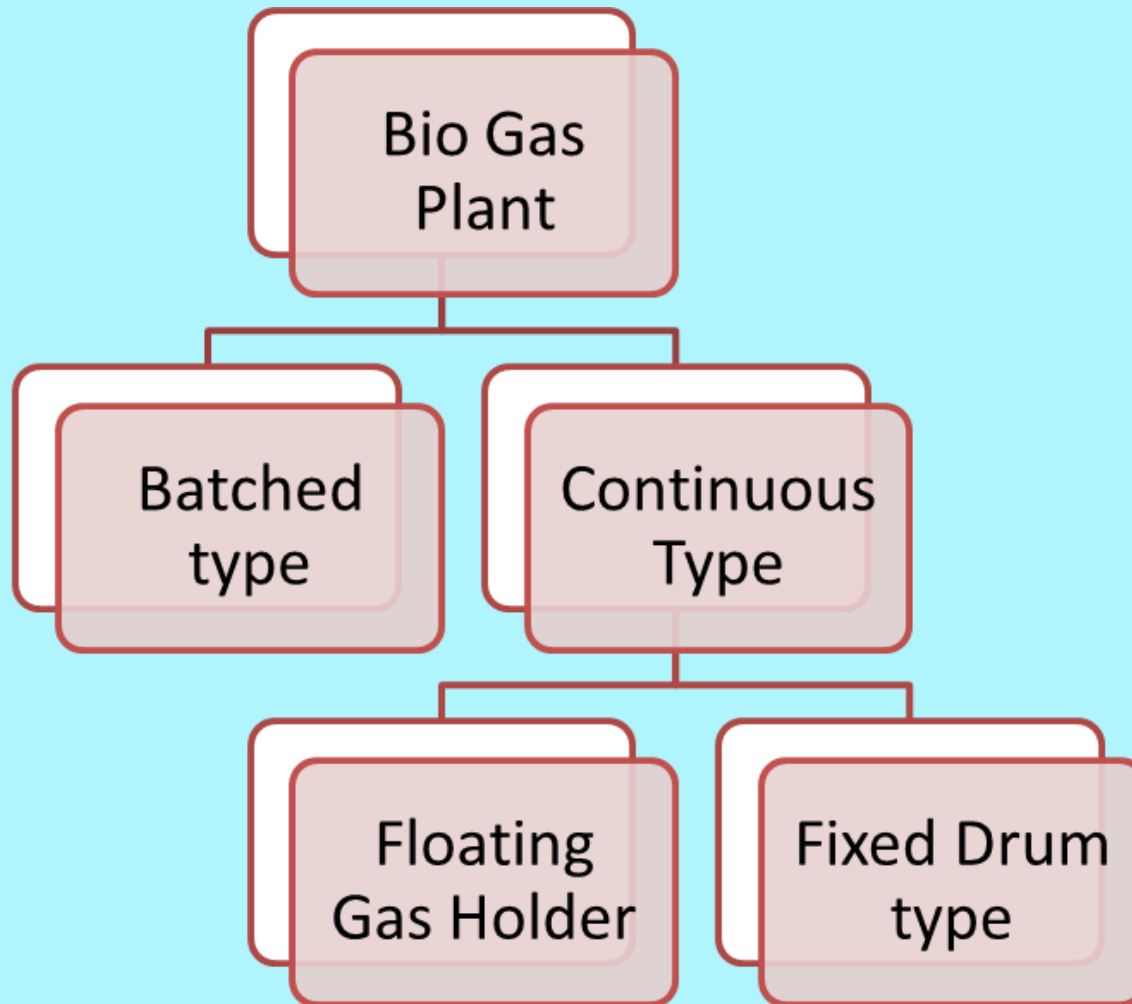
Biogas Plant Classification.....



Biogas Plant Classification.....



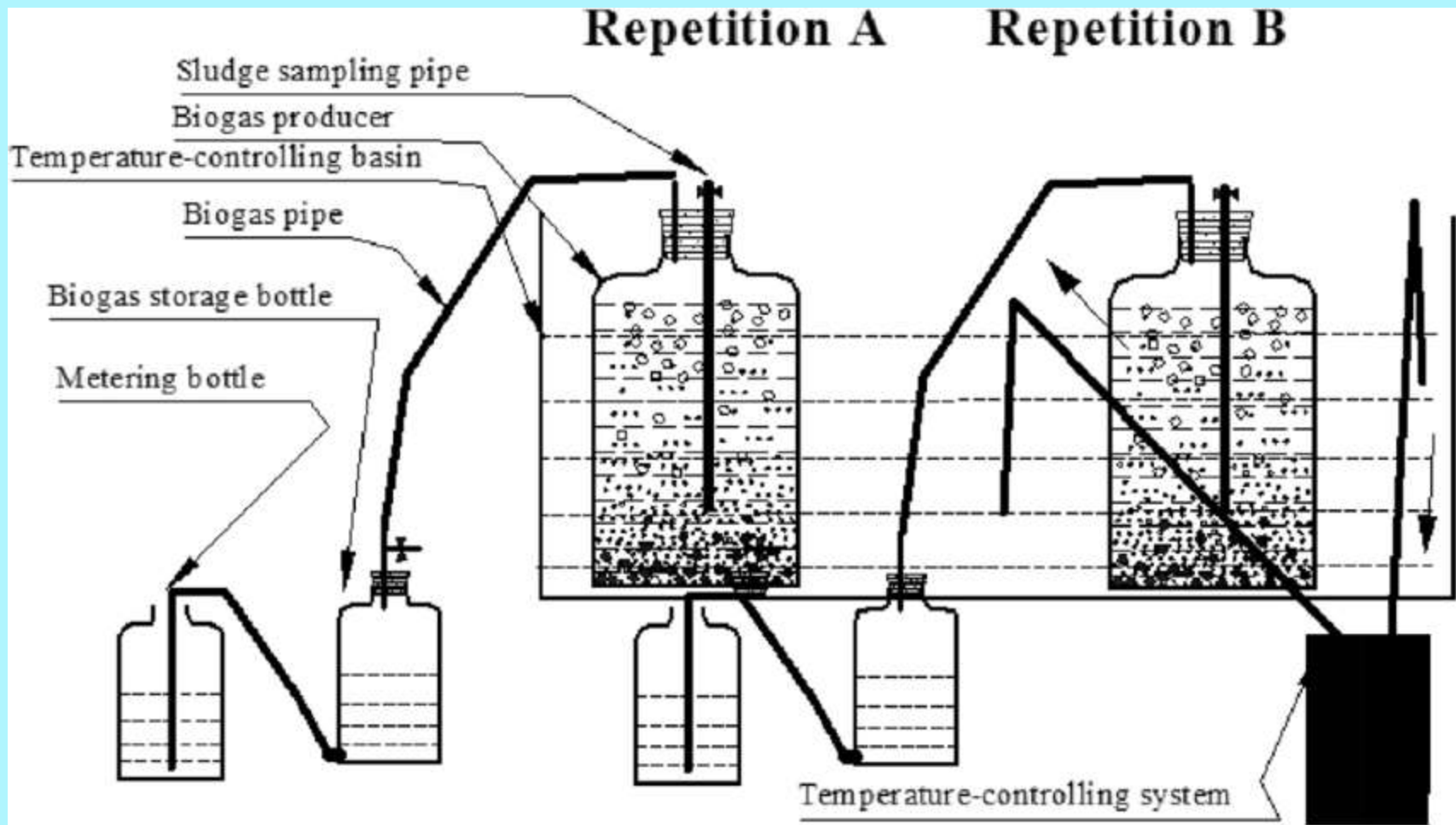
Biogas Plant Classification.....



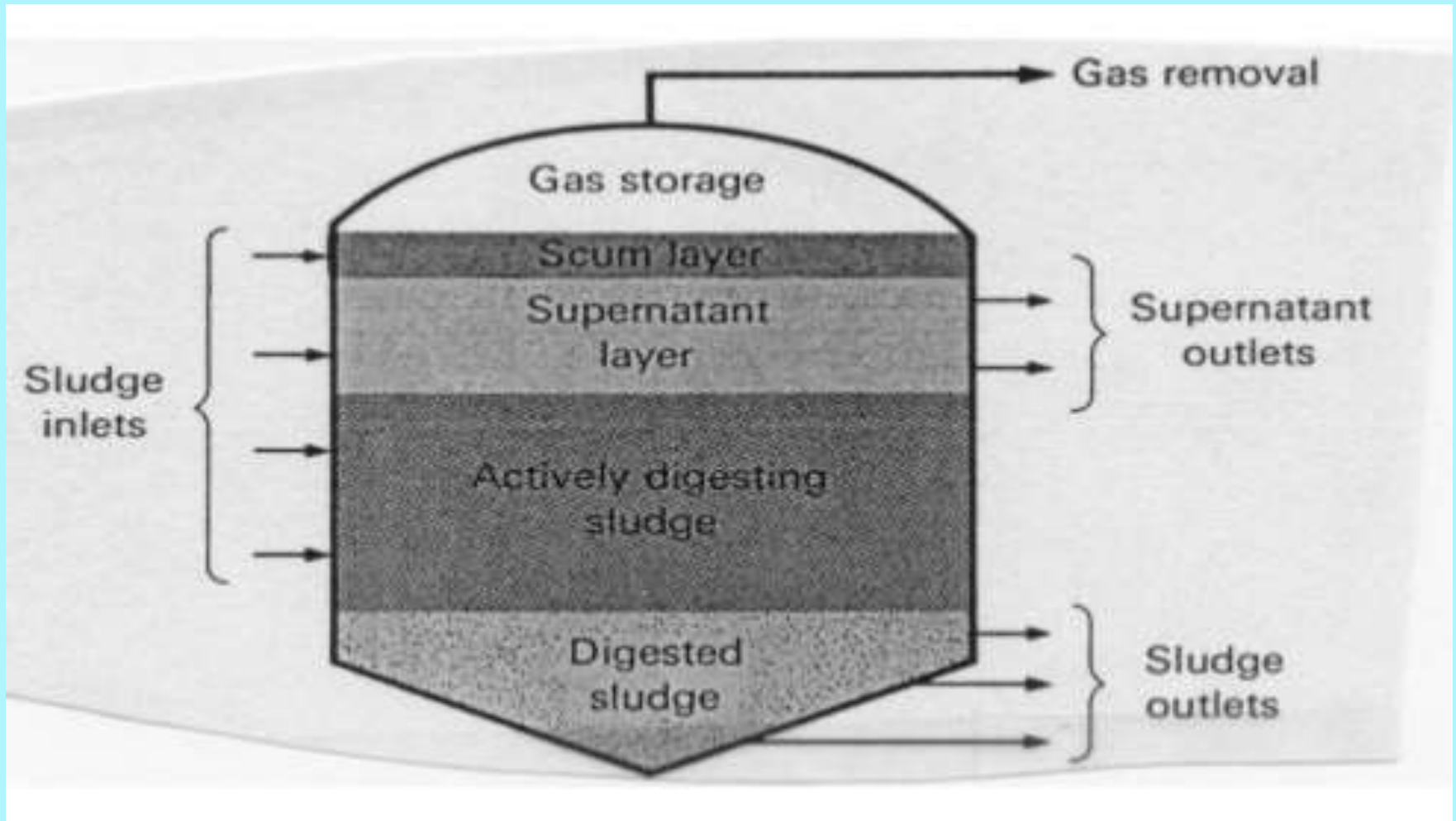
Batched Type Plant

- Batch type biogas plants are appropriate where daily supplies of raw waste materials are difficult to be obtained.
- A batch loaded digester is filled to capacity sealed and given sufficient retention time in the digester.
- After completion of the digestion, the residue is emptied and filled again.
- Gas production is uneven because bacterial digestion starts slowly, peaks and then tapers off with growing consumption of volatile solids.
- This difficulty can overcome by having multiple digesters so that at least one is always in operation.
- This problem can also minimize by connecting batch loaded digester in series and fed at different times so that adequate biogas is available for daily use.

Batched Type Plant



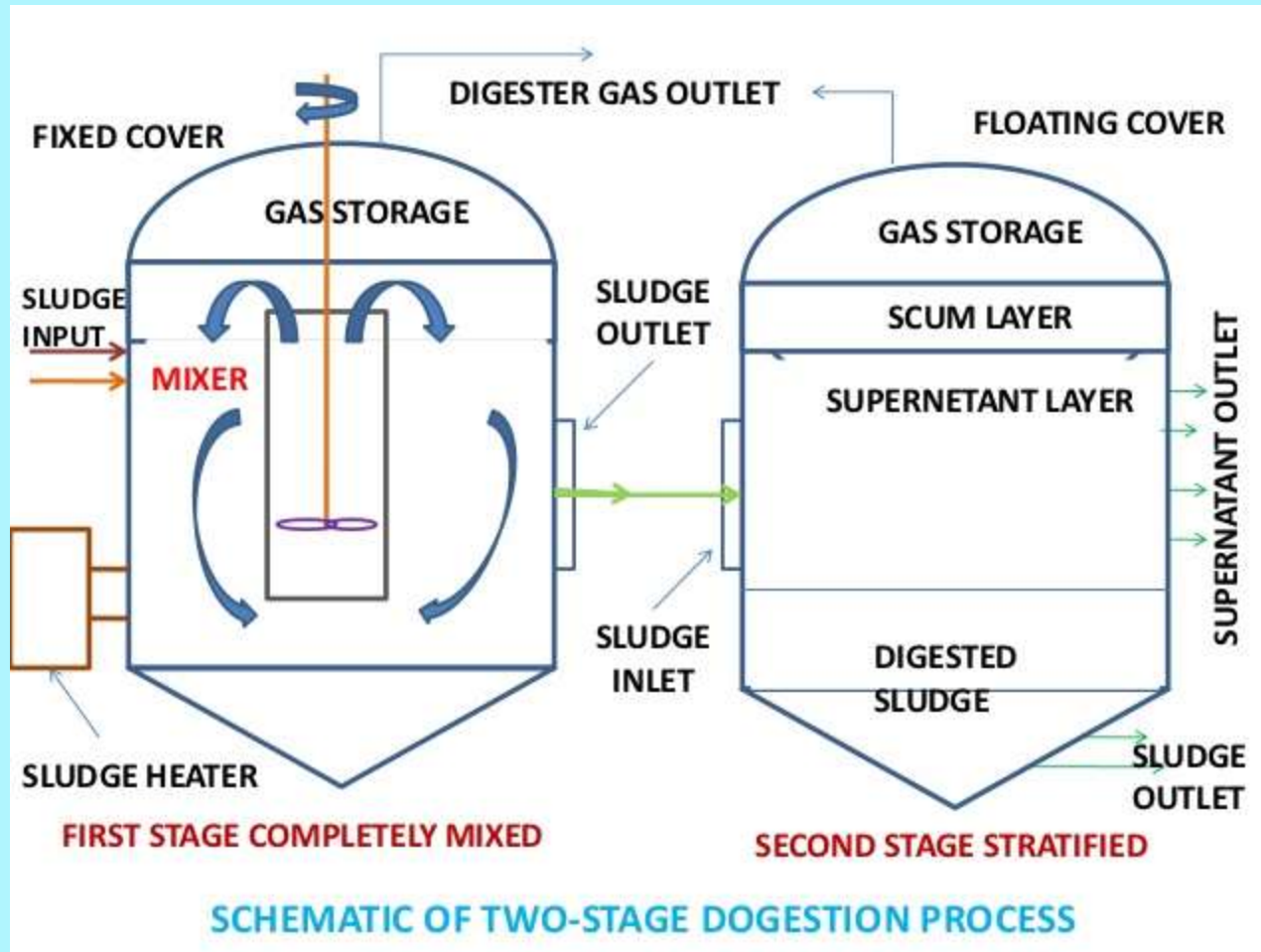
Single Stage Continuous Plant



Continuous Type Plant

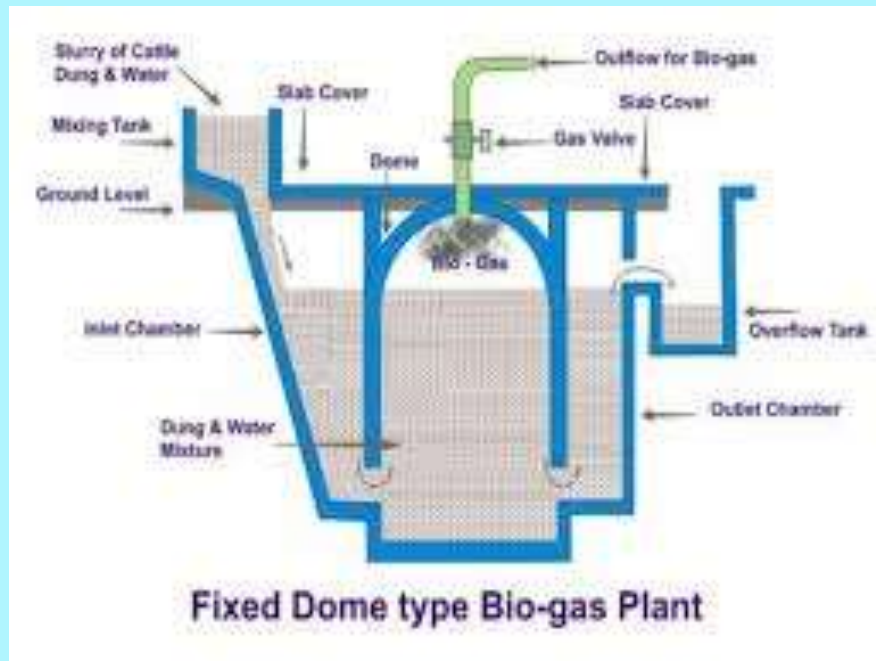
- In continuous type biogas plant, the supply of the gas is continuous and the digester is fed with biomass regularly.
- Continuous biogas plants may be single stage, double stage or multiple stage.
- Digestion of waste materials in a single chamber or digester is called single stage process, in two chambers or digesters is called multi stage process.
- In double stage process, acidogenic and methanogenic stage are physically separated into two chambers. Thus, the first stage of acid production is carried out in a separate chamber and only diluted acids are fed into the second chamber where biomethanation takes place.
- In single stage ,acidogenic and methanogenic stage are carried out in the same chamber without barrier.

Two Stage Continuous Plant



Features of Continuous Plant.....

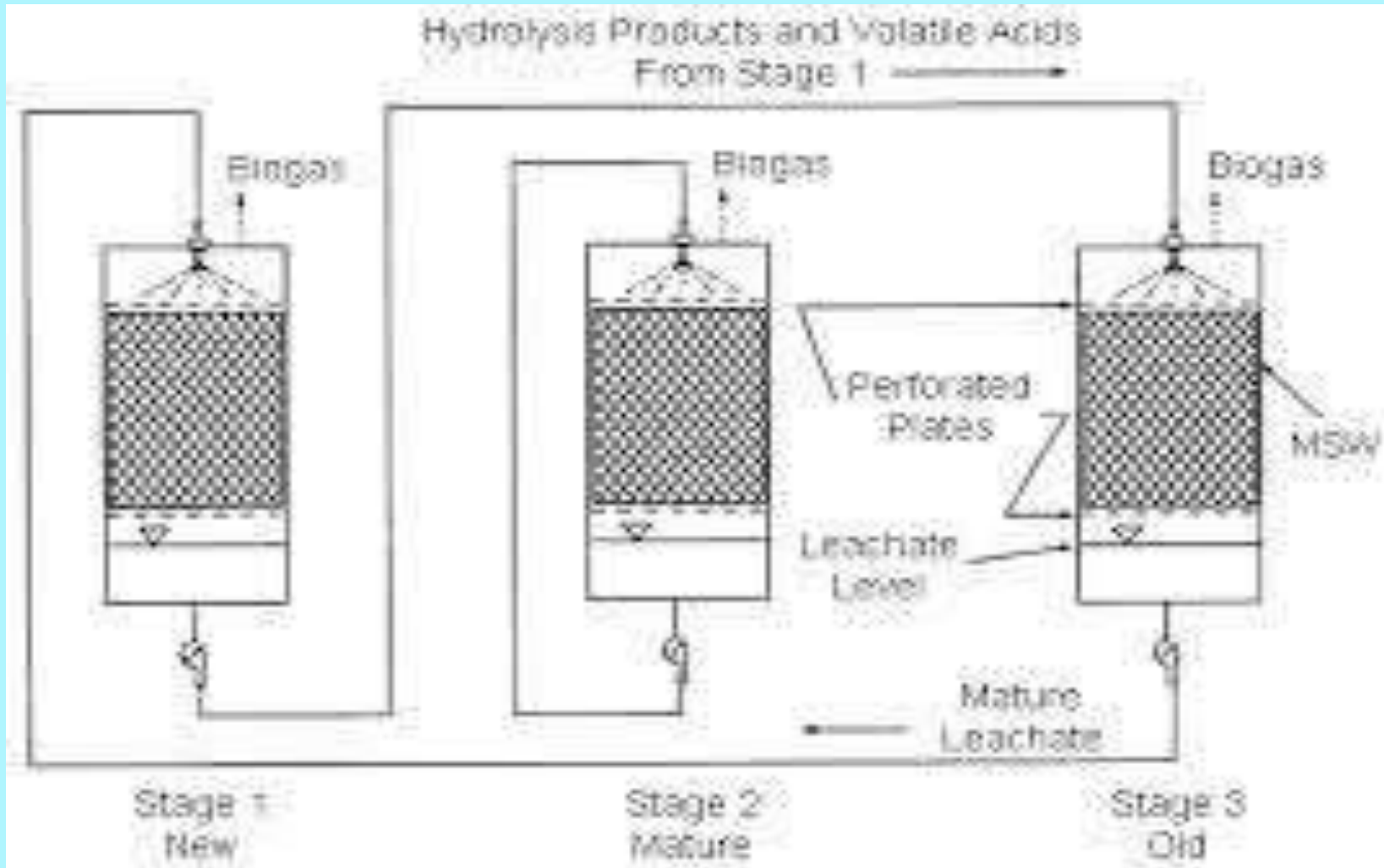
1. Plant will produce gas continuously
2. Plant requires small digestion chambers.
3. Plant needs smaller period of digestion.
4. It has less problem compared to batch type
5. It is easier in operation.



Features of Batch Plant.....

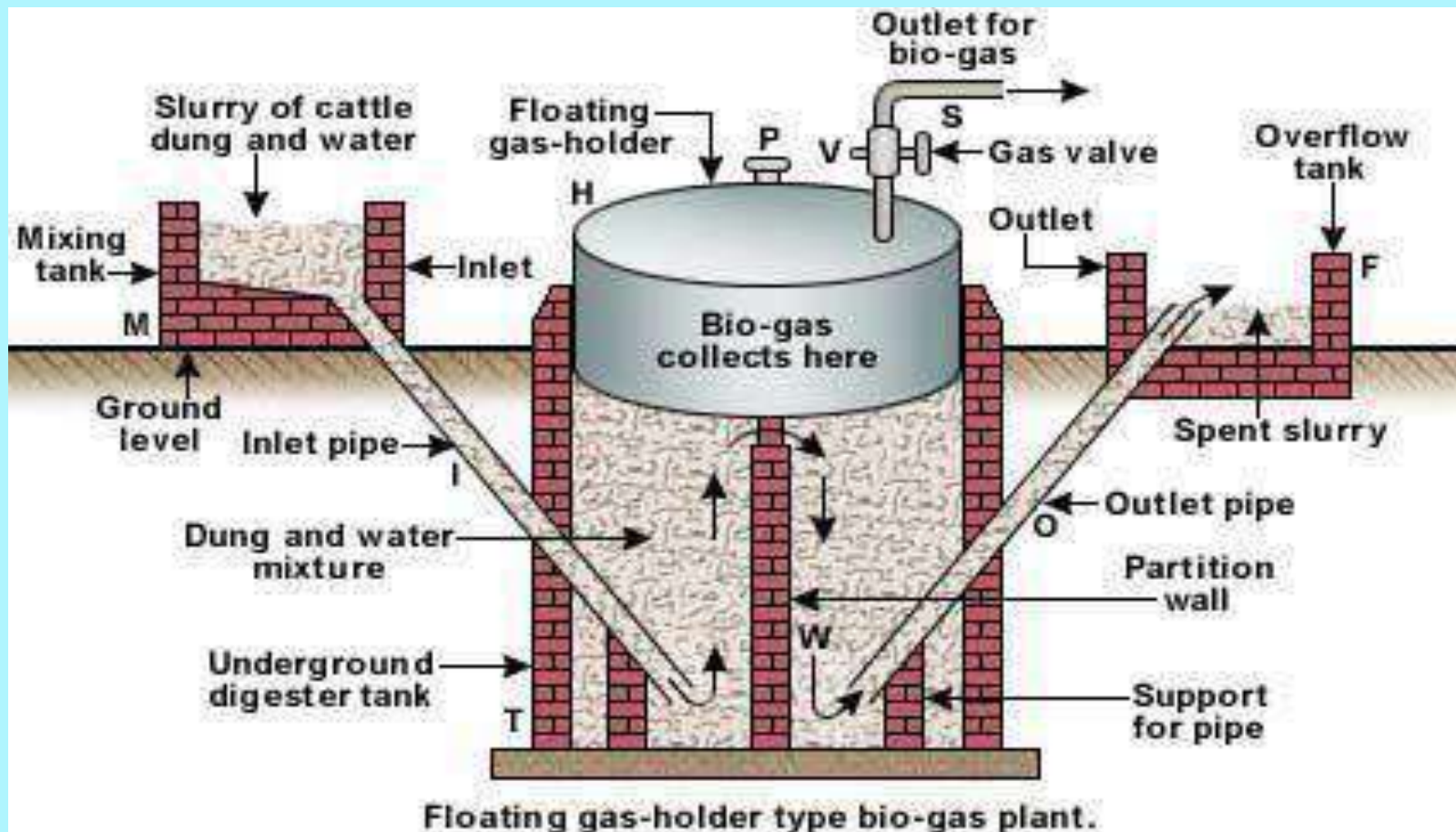
1. Gas production in batch type is uneven.
2. Batch type plants may have several digesters for continuous supply of gas.
3. It depends upon clearing of digester
4. Several digesters occupy more space.
5. Batch plants are good for long fibrous materials
6. This type of plants require large volume of digester, therefore, initial cost becomes high.
7. This plant needs addition of fermented slurry to start the digestion process.
8. The Plant is expensive and has comparatively more problems

Features of Batch Plant.....



Floating Gas Holder Plant.....

In a floating gas holder, gas holder is separated from the digester. Also known as KVIC plant



Floating Gas Holder Plant.....

Floating-drum Plants

1. Mixing tank with inlet pipe & Digester.
2. Overflow on outlet pipe.
3. Gasholder with braces for breaking up surface scum.
4. Gas outlet with main cock.
5. Gas drum guide structure.
6. Difference in level = gas pressure in cm WC.
7. Accumulation of thick sludge.

Floating-drum plants (Figure) consist of a digester and a moving gasholder. The gasholder floats either direct on the fermentation slurry or in a water jacket of its own. The gas collects in the gas drum, which thereby rises. If gas is drawn off, it falls again. The gas drum is prevented from tilting by a guide frame.

Floating Gas Holder Plant.....

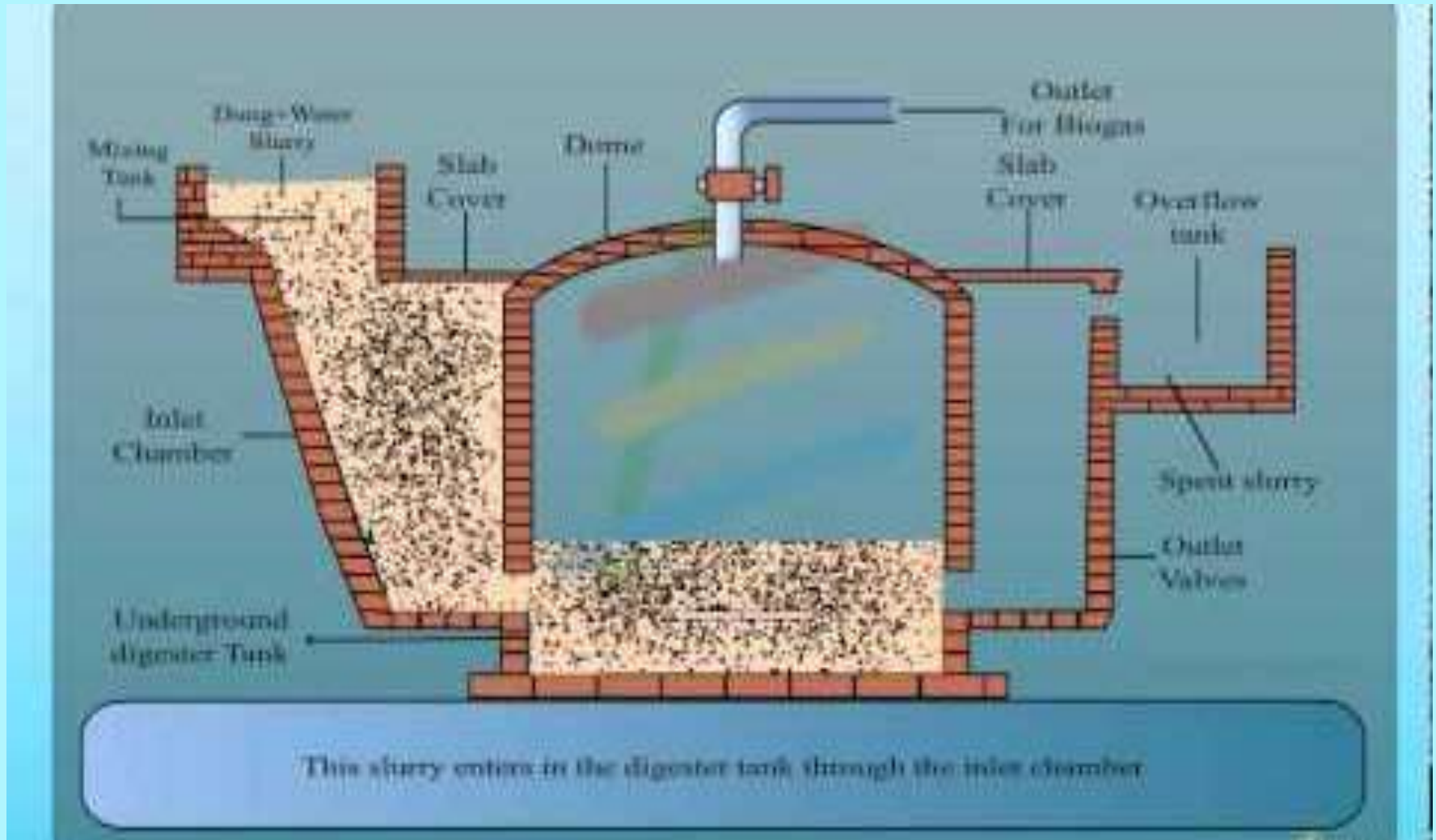
Advantages: Simple, easily understood operation, constant gas pressure, volume of stored gas visible directly, few mistakes in construction.

Disadvantages: High construction cost of floating-drum, many steel parts liable to corrosion, resulting in short life.

In spite of these disadvantages, floating-drum plants are always to be recommended. Water-jacket plants are universally applicable and especially easy to maintain. The drum won't stick, even if the substrate has a high solids content.

Works up to 15 years; in tropical coastal regions about five years for the drum, needs regular maintenance costs

Fixed Dome Type Plant.....



Fixed Dome Type Plant.....

In fixed dome digester, the gas holder and the digester are combined. **Also known as Chinese Plant**

The fixed dome is best suited for batch process especially when daily feeding is adopted in small quantities. The plant is generally built below the ground level and is suitable for cooler regions.

Advantages of Fixed-dome plants

Low construction cost, no moving parts, no rusting steel parts, hence long life (20 years or more), underground construction, affording protection from winter cold and saving space,

Disadvantages: Plants often not gastight (porosity and cracks), gas pressure fluctuates substantially and is often very high, low digester temperatures.

These plants can be recommended only where construction can be supervised by experienced biogas technicians

Variations in Drum Type ...

- There are mainly two variation in floating type plant. One with *water seal* and other *without water seal*.
- Water sealing makes the plant completely anaerobic.
- Cylindrical shape of the digester is preferred because cylinder has no corners and so that there will be no chances of cracks due to faulty construction. This shape also needs smaller surface area per unit volume, which reduces heat losses also.
- Moreover the scum formation may be reduced by rotating gas holder in the cylindrical digester.

Advantages of biogas...

- Cost of equipment's used for making biogas is low and equipment's used are very simple.
- Biogas can be used for lighting, running the engines, farm's machine and cooking gas in the kitchen.
- Biogas is the best medium for cooking food.
- Organic feed stocks used in the plants are easily available at all places.
- Biogas plant gives efficiency as much as 60%.
- Distribution of gas has no problems of any gas leakage and fire.
- Waste product obtained from digester is best quality of fertilizer and gives best yields.

Disadvantages of biogas...

- Biogas produced from biogas plant has to be used at near by places only.
- It can't be transported over long distances.
- Biogas can't be filled in the bottles.
- Biogas plant requires more area.
- It can't be established in urban area where availability of land is limited.

Applications of biogas...

- Biogas is used as cooking fuel.
- Biogas is mental light gas burner for lighting purpose.
- Biogas is used for water heating.
- It is used as fuel in I.C. Engine.
- It is used as fuel to run agricultural machineries.
- It is used to run diesel engine generator set to produce electricity.
- Heat of biogas is utilized in the dryer for drying the agricultural products.
- Heat of biogas is used to heat ammonia of refrigerating plant.
- It is used for running pumps for irrigation purpose.
- Methane and carbon dioxide are used as raw chemical feed stock to manufacture various chemicals.

Thank You