



The Dharwar Gold: Current understanding and future Research

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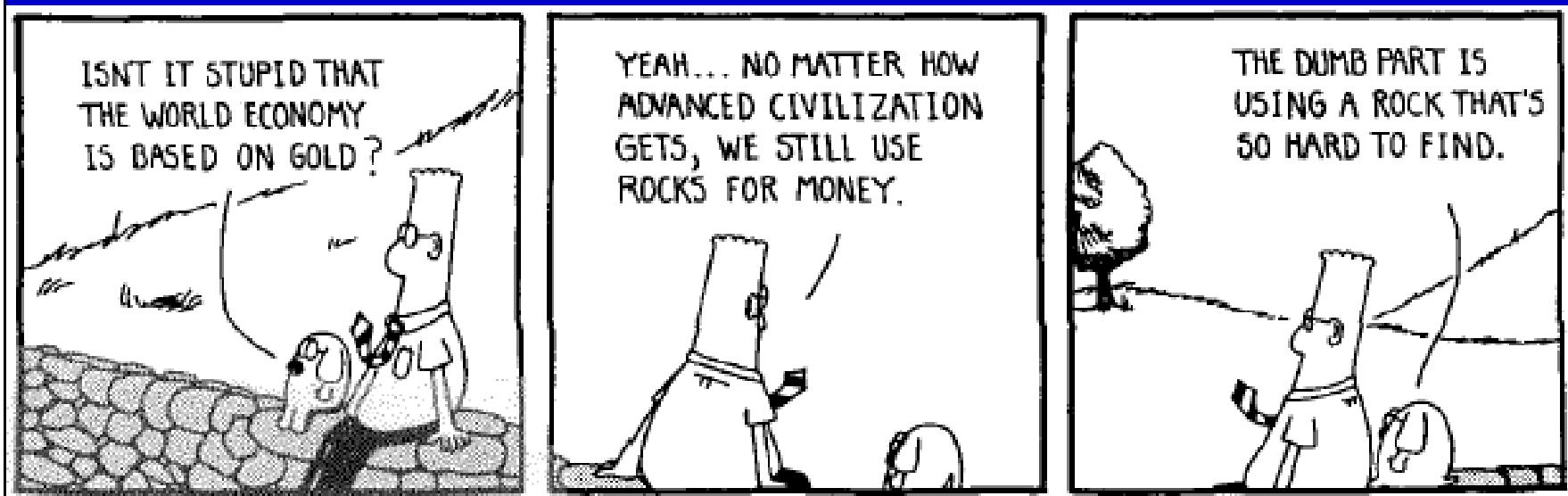
Genesis of metallic ore deposits



How and WHEN do/did ore deposits form? Philosophical
Why do they occur **where** they are? Exploration & Mining

Ore deposits are crustal segments, with minable positive anomalies of one or more metals by efficient/slow earth processes

Geochemically abundant metals (> 100 ppm; Fe, Al and Cr) and scarce metals (< 100 ppm; many in Periodic table)

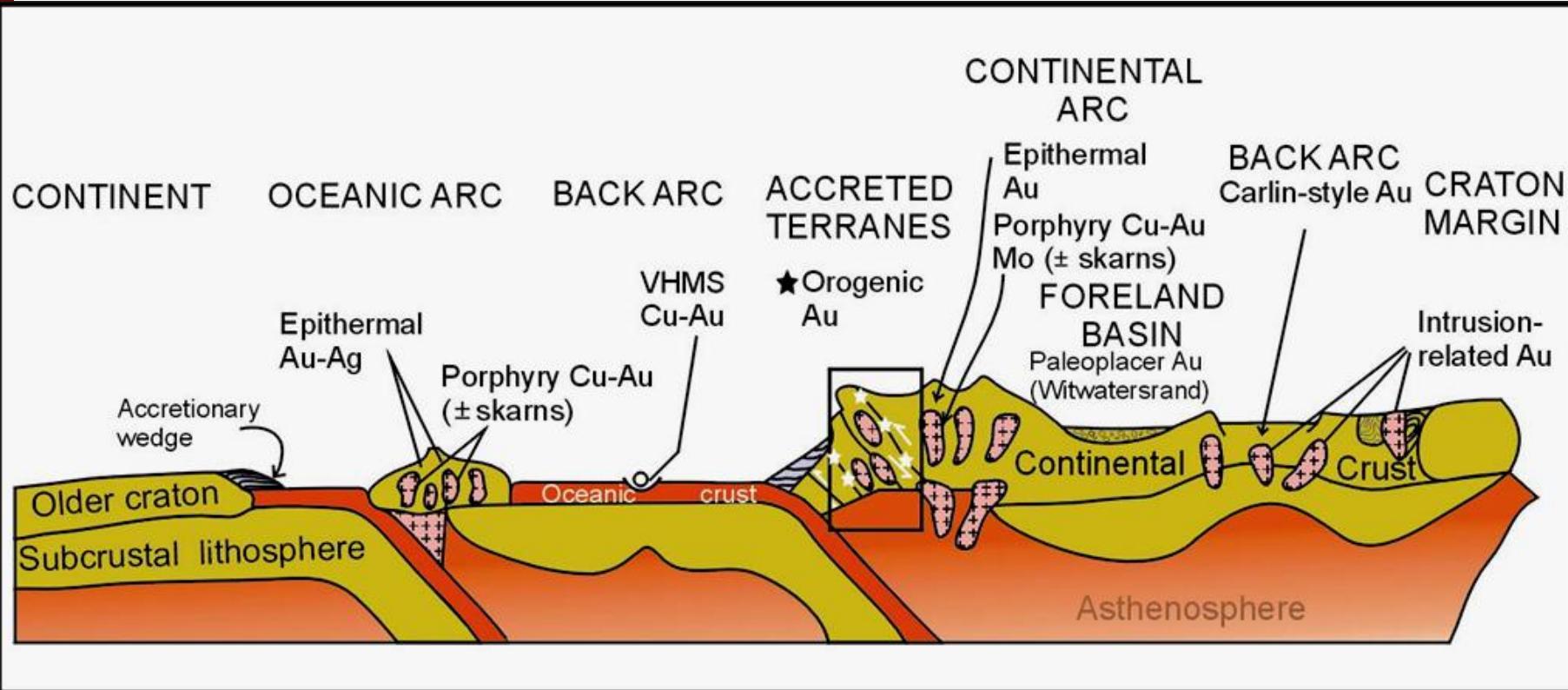


Ore-forming elements at a glance: how much to enrich?



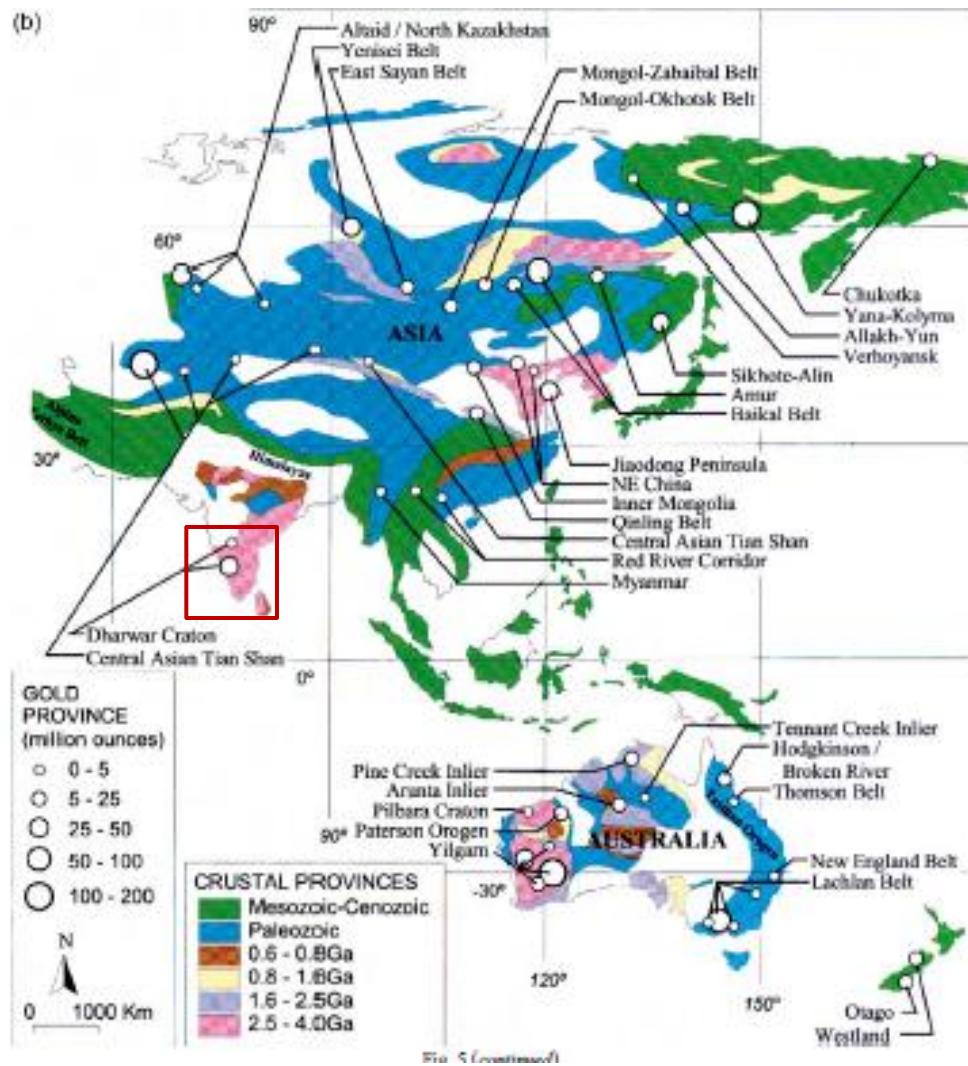
Element	Clarke Value in ppm (X)	Cut-off grade in ppm (Y)	Enrichment Factor (Y/X)
Al	81,300	300,000	4
Fe	50,000	280,000	6
Zr	165*	14,000	85
P	1065	100,000	95
V	135	14,000	104
Cu	55*	10,000	160
Nb	20	3,400	170
Ni	75*	1,500	188
Mn	950	350,000	350
Pt	0.01	4	400
U	1.8	0.1	500
Zn	70*	40,000	600
Mo	1.5	1,300	870
W	1.5	1,400	935
Au	0.003	3	1,000
Pb	13	40,000	3,000
Cr	100	300,000	3,000
Sn	2	10,000	5,000
Ag	0.07	500	7,150
Hg	0.08	3,000	37,500
Sb	0.2	10,000	50,000

Orogenic Au deposits: tectonic setting



Along convergent margins → terrain accretion or collision related to subduction late in the metamorphic-magmatic evolution of the orogen (Groves et al 2003, *Econ Geol*)

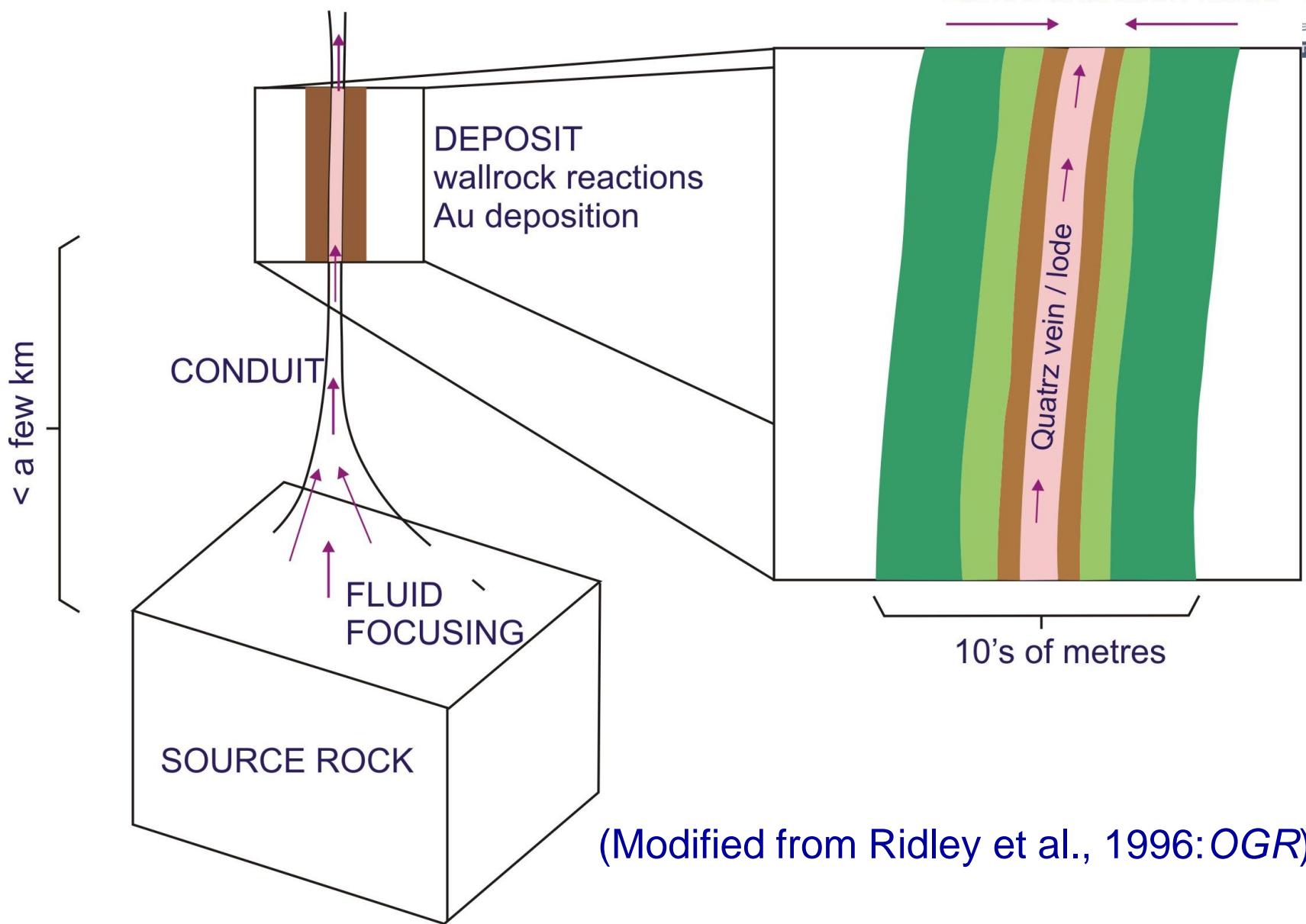
Orogenic Au deposits: Asia-Pacific



(Goldfarb et al, 2001, OGR)

CS-EB meeting, Bangalore, November 28-29, 2015

Orogenic Au deposits: key features



Unresolved questions ?



- Relative timing of alteration/mineralization wrt greenstone volcanism/metamorphism and granitoid intrusion
- Precise architecture of the hydrothermal systems, particularly the relationship between first- and higher-order structures (splays)
- Sources of (i) ore fluid (low salinity $H_2O+NaCl+AuHS_2^-+CO_2\pm CH_4$), (ii) gold, (iii) sulfur

Metamorphic? Magmatic ? Both ???

- Specific depositional mechanisms for gold, particularly for high-grade deposits

Today's discussion



- Greenstone metamorphism: Uti, Kolar, Jonnagiri
- Geochronology (mostly of granitoids), Monazite dating
- Alteration geochemistry/ore mineralogy: Huttı, Jonnagiri
- Fluid inclusions: Kolar, Ramagiri, Huttı, Hira-Buddini, Ajjanahalli, Jonnagiri, Gadag
- C- and oxygen isotope Ajjanahalli/GR Halli/Gadag
- Sulfur isotope: Most camps
- Geochemistry of hydrothermal minerals (tourmaline, scheelite, pyrite and arsenopyrite)
- Synthesis
- Future research directions

Study areas

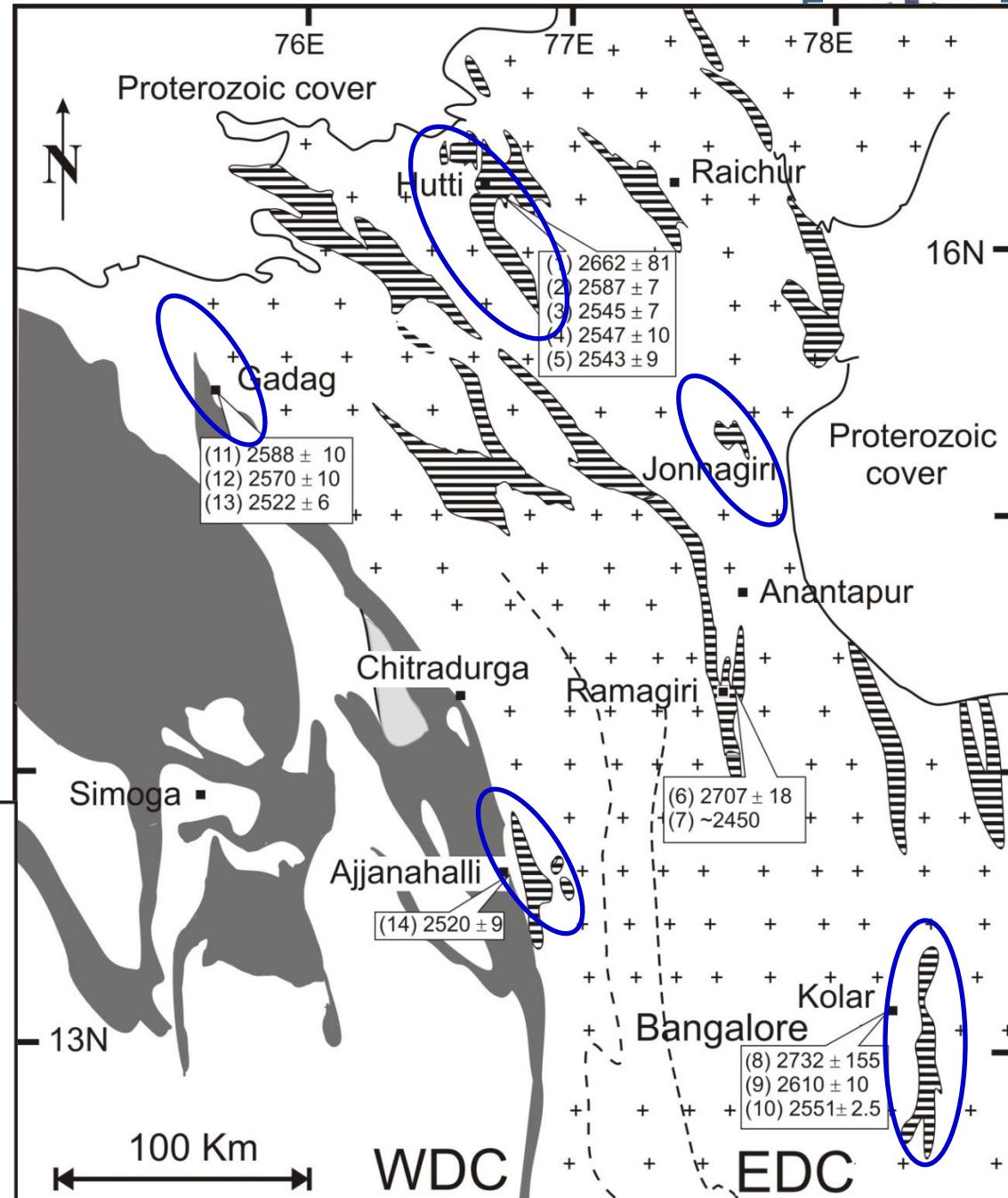
- Age of greenstone metamorphism ?
- Age of volcanism

Hutti: 2662 ± 81 Ma

Kolar: 2732 ± 155 Ma

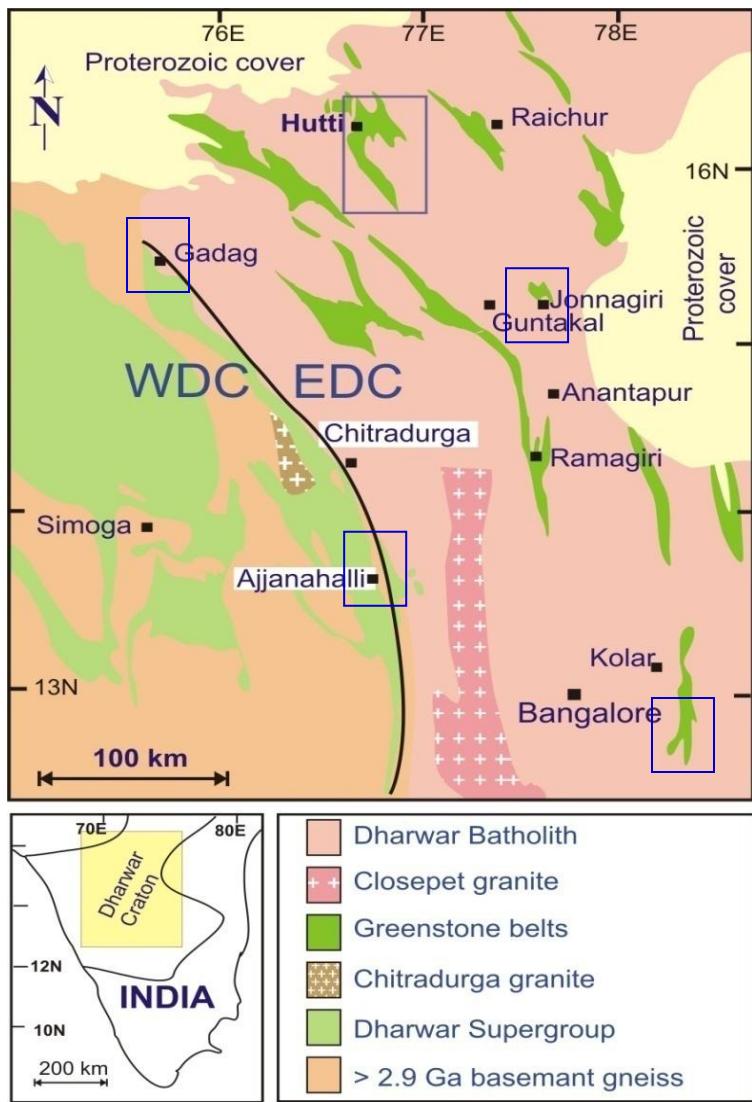
(Anand and Balakrishnan,
2010, JAES)

(Balakrishnan et al., 1990,
CMP)

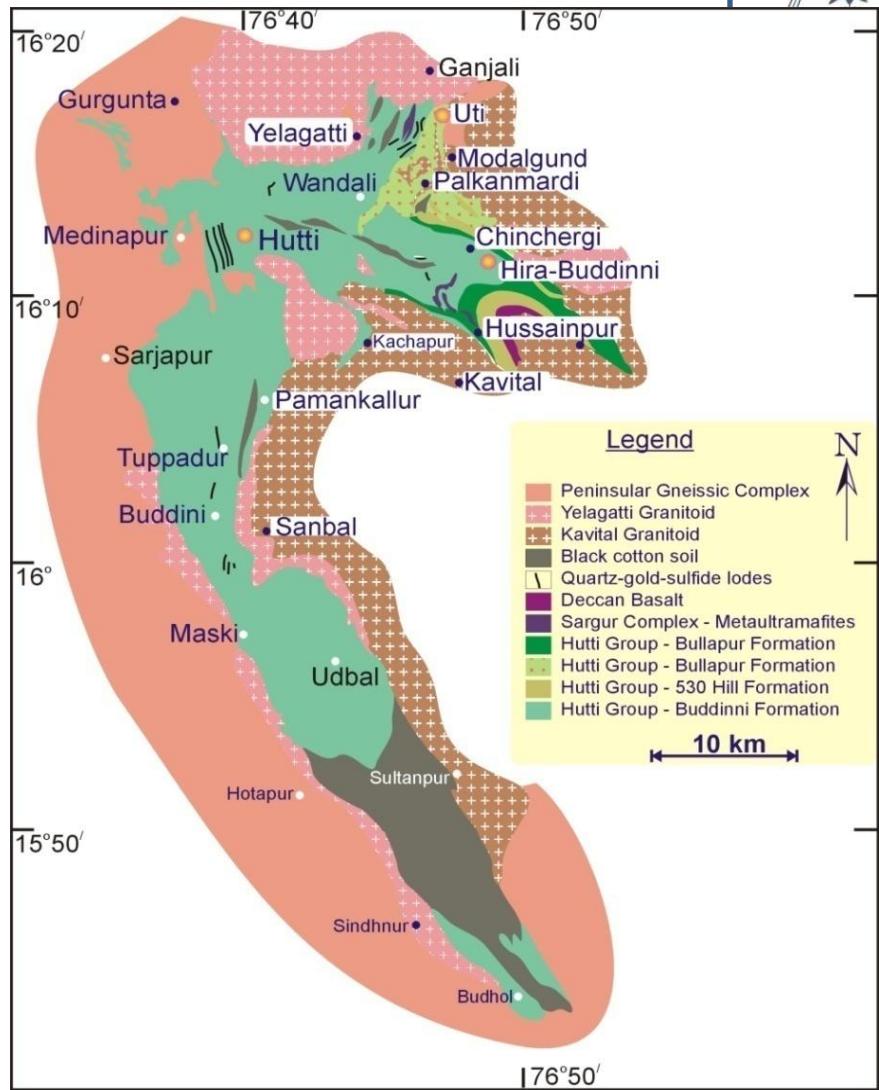


(Modified after Chadwick et al., 2000, PR)

Study areas (contd.)

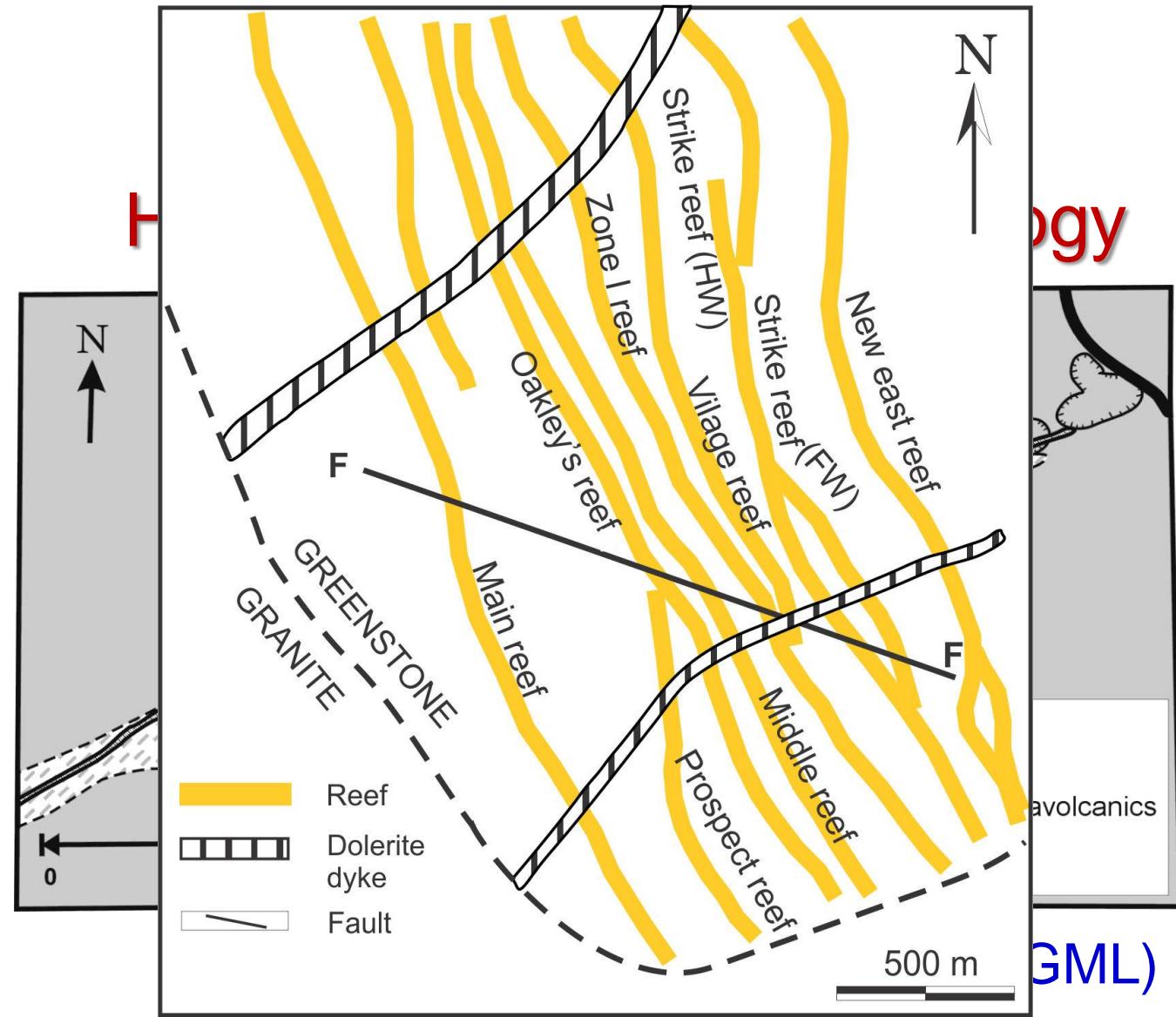


(Chadwick et al, 2000, PR)



(Srikantia 1995, GSIM)

Hutti: auriferous reefs



Curtis and Radhakrishna (1995)

Hutti: world class deposit (>100 t Au)



Uti open pit



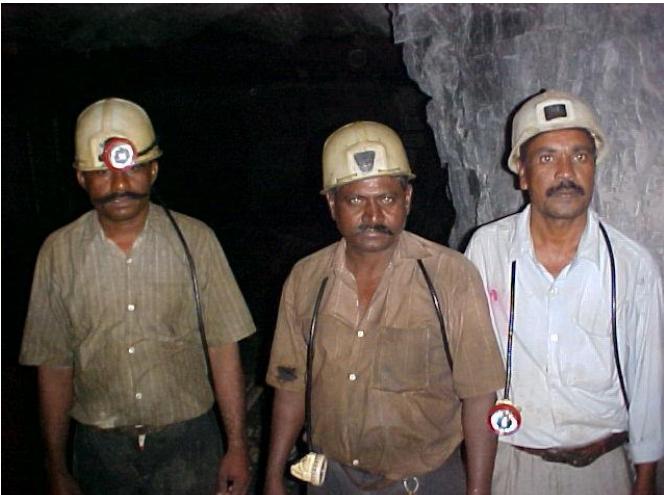
Central shaft



Drilling at depth



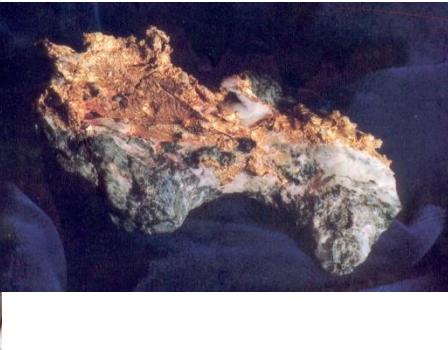
Friendly miners



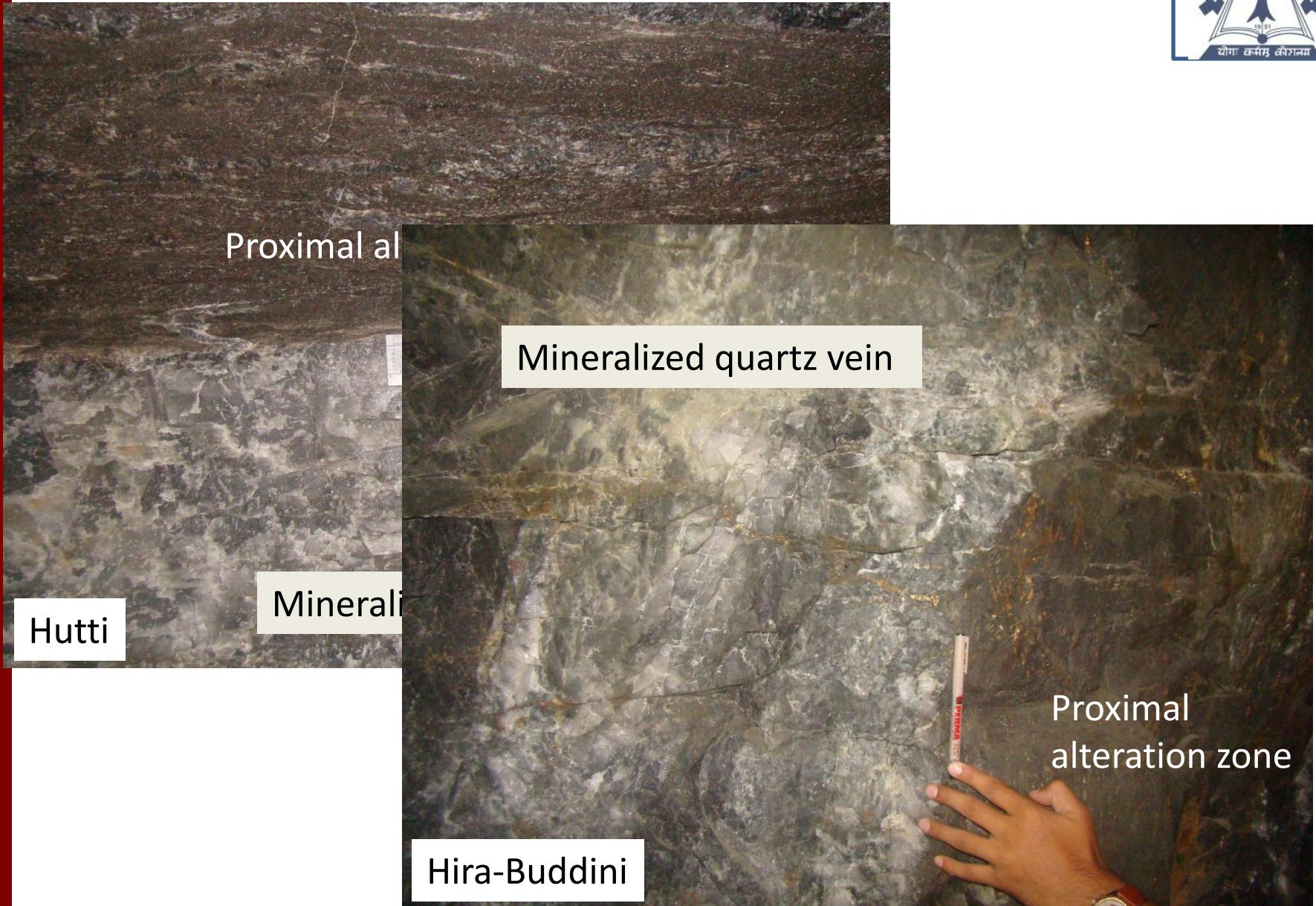
Security checking



Native gold



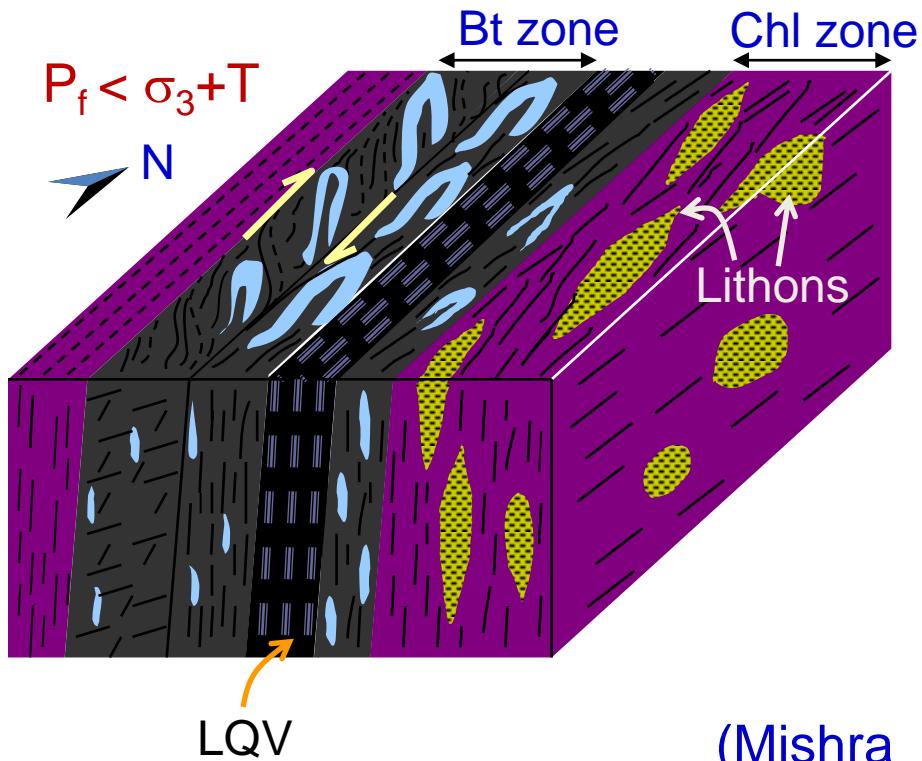
Hutti and Hira-Buddini:underground



Hutti

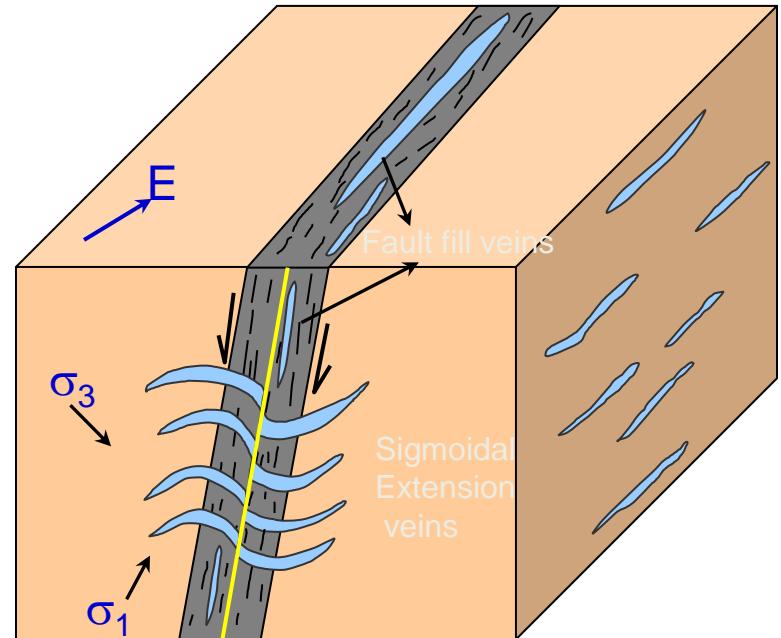
D_2 and D_3 veins (LQV) are mineralized (D_1-D_5)

Absence of extension veins and extensional shear veins →

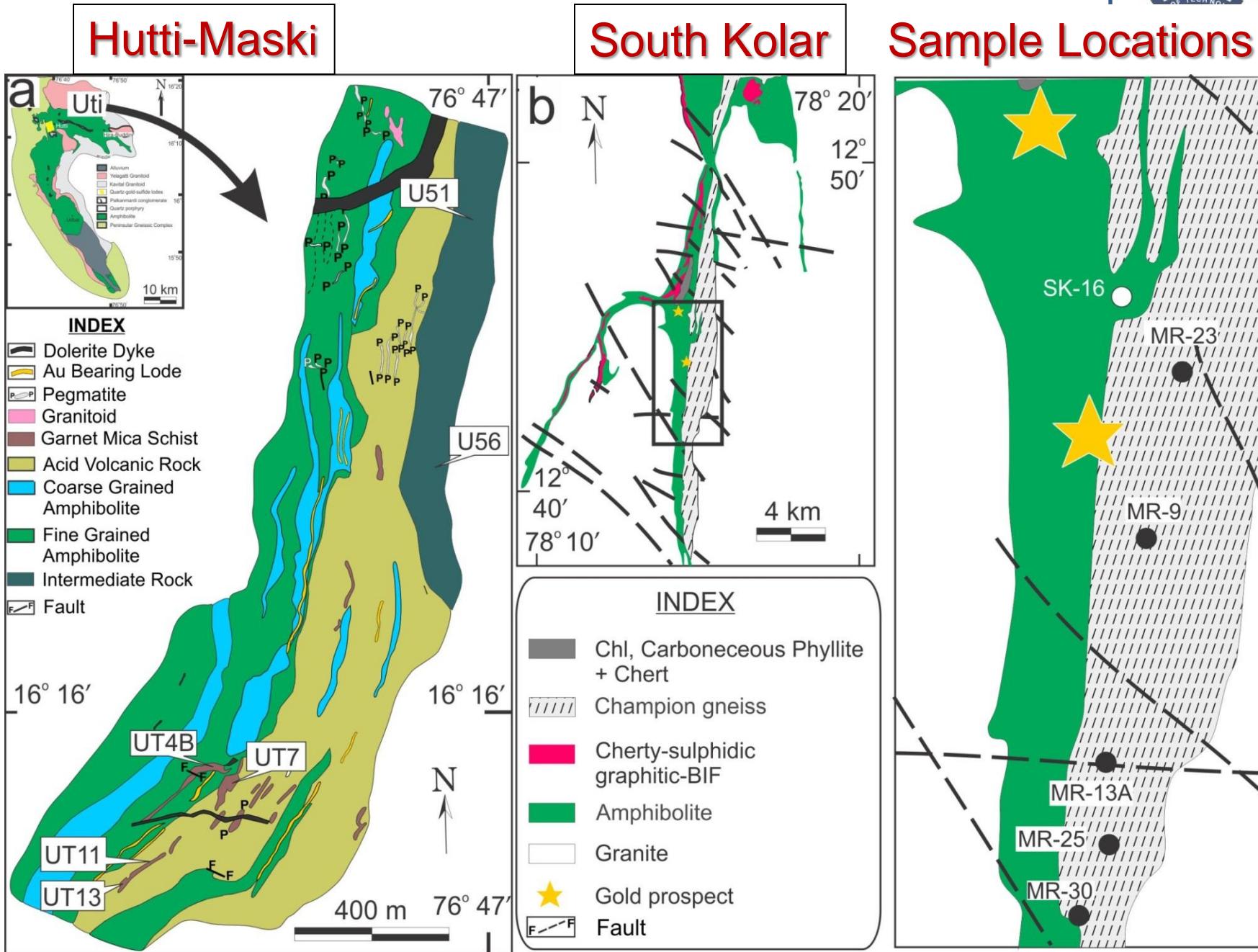


Hira-Buddini

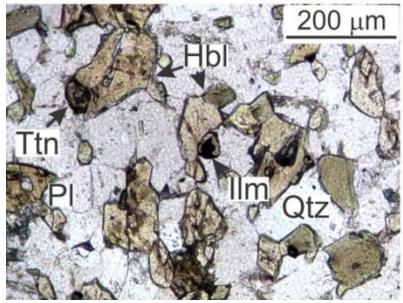
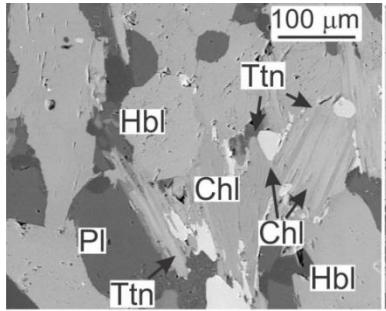
Presence of sigmoidal extension veins →
 $P_f \geq \sigma_3 + T$



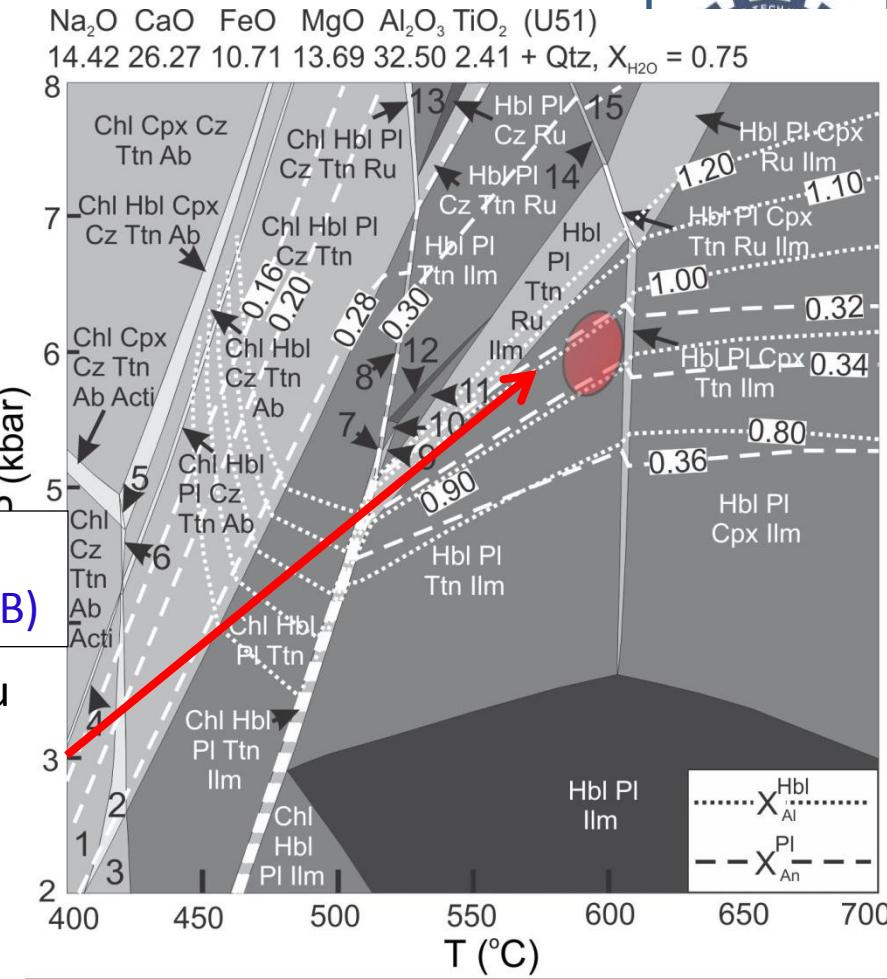
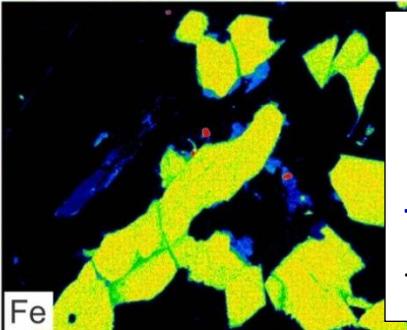
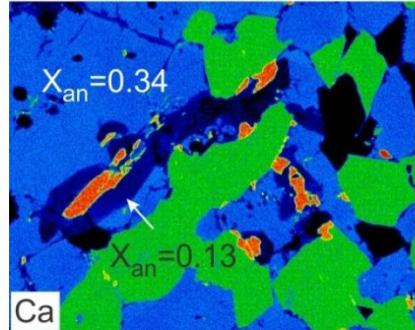
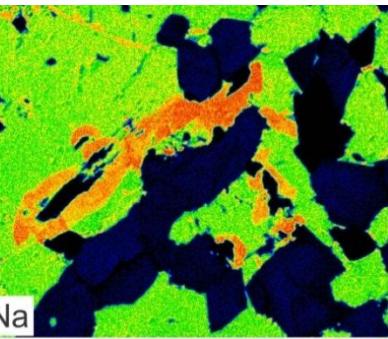
(Mishra & Pal, 2008, Econ Geol)



Uti (HMGB) metamorphism



Hbl composition
 $\text{Al(IV)} 0.60 \rightarrow 0.8$
 $\text{Na} 0.14 \rightarrow 0.24$,
 $X_{\text{Fe}} 0.46 \rightarrow 0.49$,
 Total Al 0.80 → 1.06 apfu

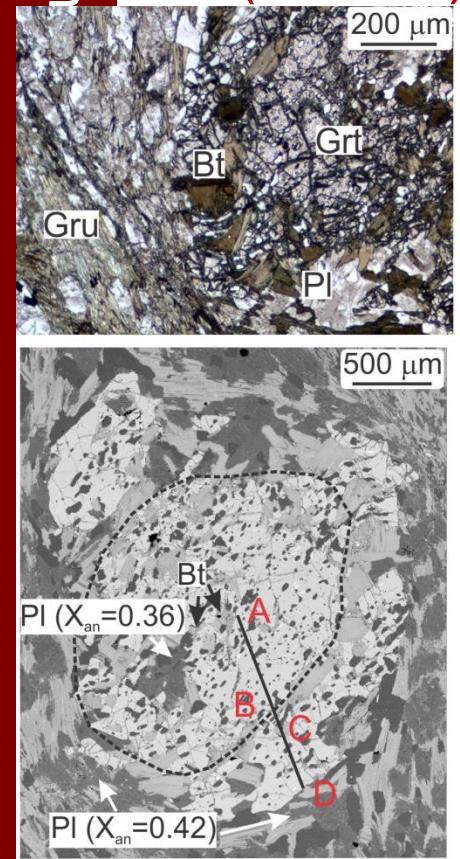


Mineral thermobarometry

P = 5.3–6.1 kbar (Hornblende-Plagioclase barometry, Bhadra and Bhattacharya, 2007, AM)

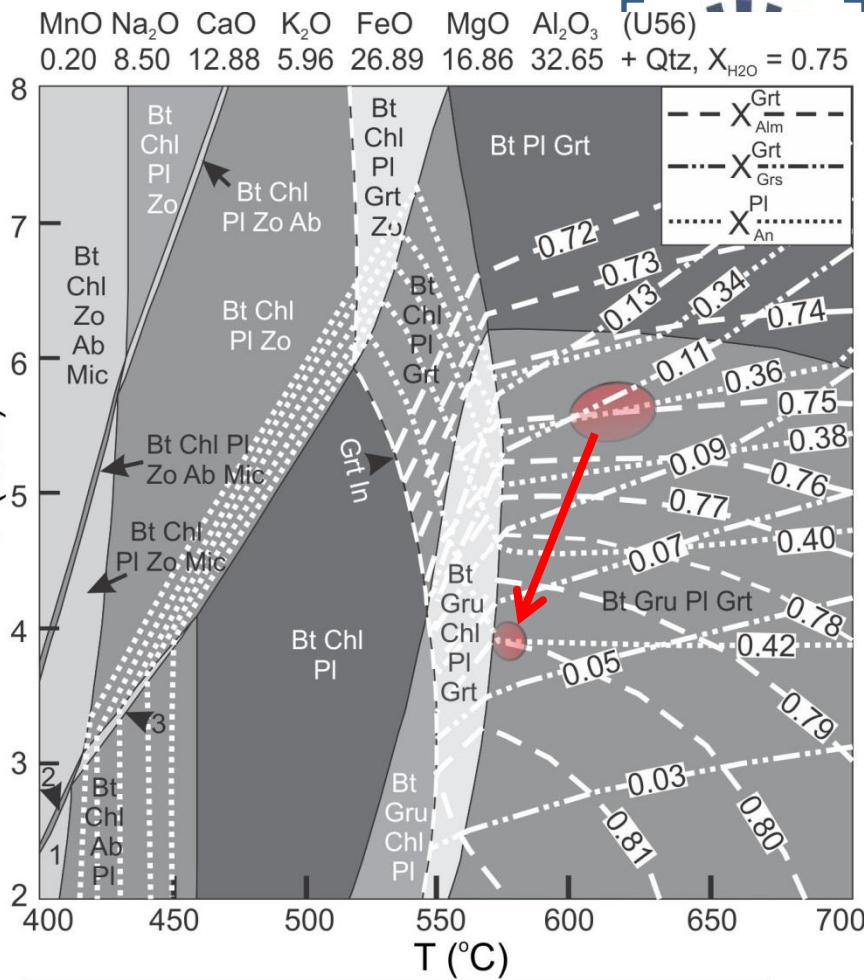
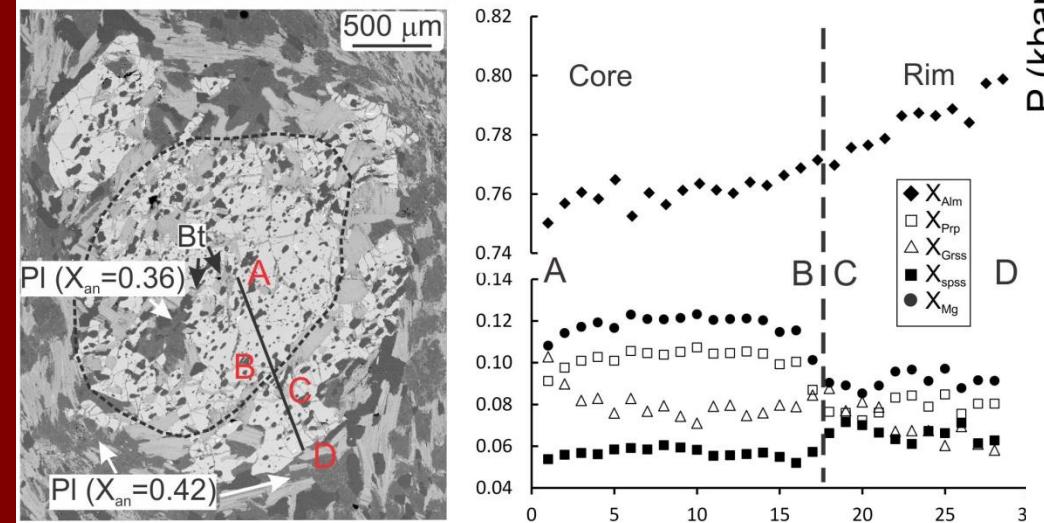
T = 586 – 617 °C (Hornblende-Plagioclase thermometry, Holland and Blundy, 1994, CMP)

Uti (HMGB) metamorphism



Sample U56
(HMGB)

Grt-Bt-Gru-Pl-Qtz



Mineral Thermobarometry

Garnet core-included Biotite-included Plagioclase

GBPQ barometry (Wu et al., 2004, *JP*)

$P = 5.1\text{--}5.7 \text{ kbar}$

Garnet-Biotite thermometry (Ferry and Spear, 1978, *CMP*; Bhattacharya et al., 1992, *CMP*)

$T = 620\text{--}645 \text{ }^{\circ}\text{C}$

Mineral thermobarometry

Garnet rim-near neighbor Biotite-near neighbor Plagioclase

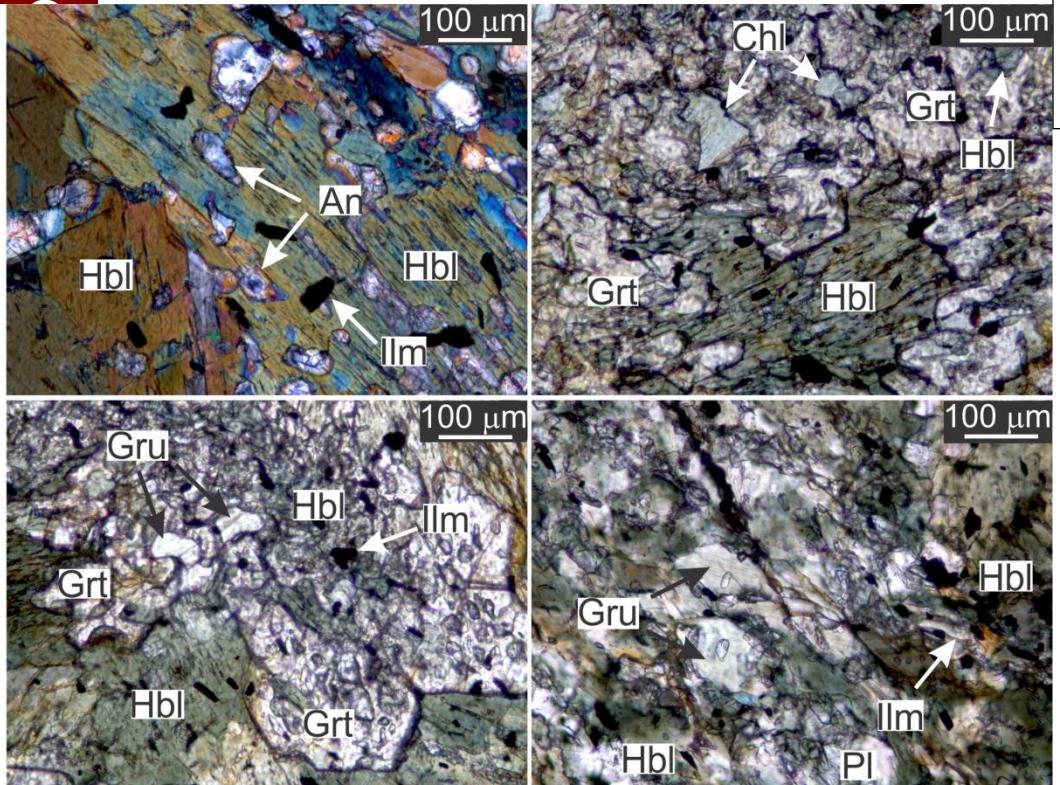
GBPQ barometry

$P = 3.6\text{--}4.2 \text{ kbar}$

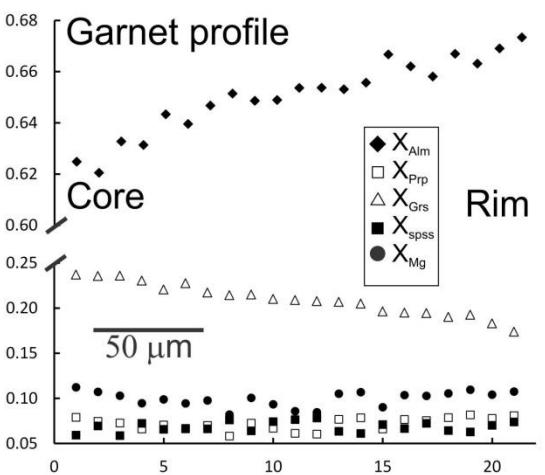
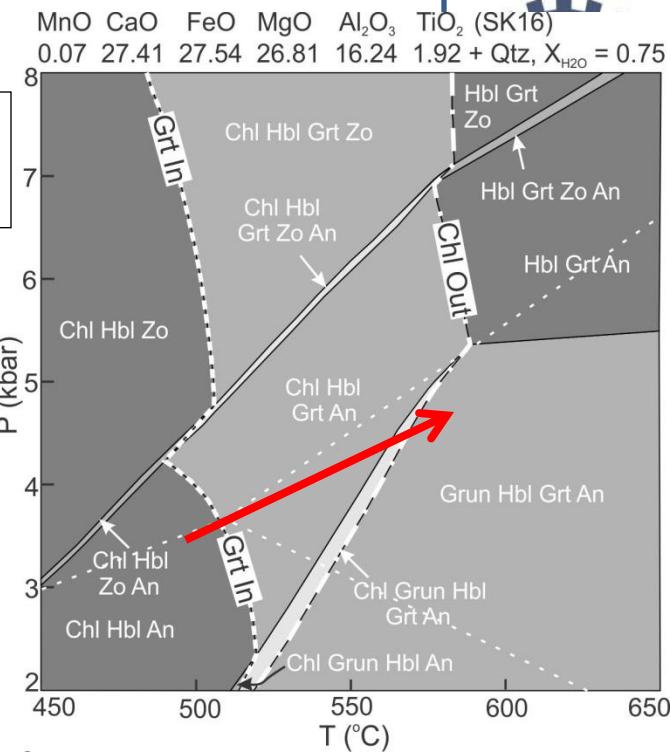
Garnet-Biotite thermometry

$T = 617\text{--}632 \text{ }^{\circ}\text{C}$

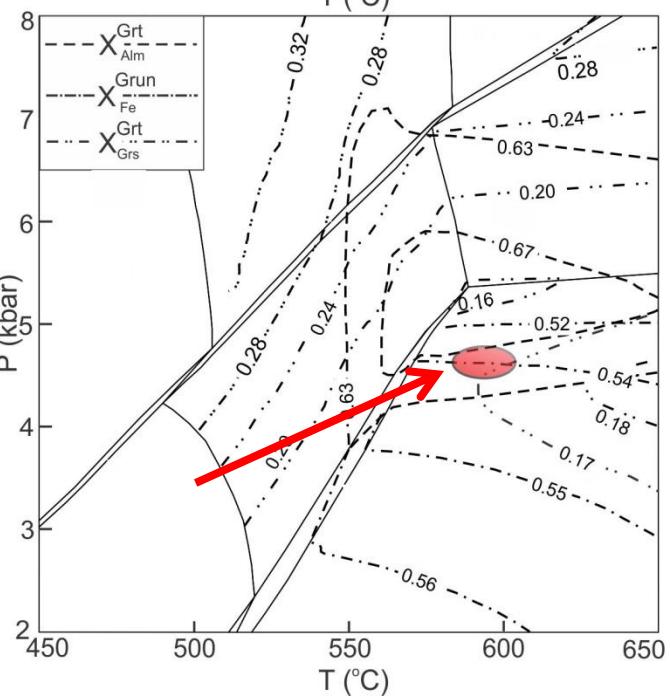
SKB metamorphism



SK16
(SKGB)



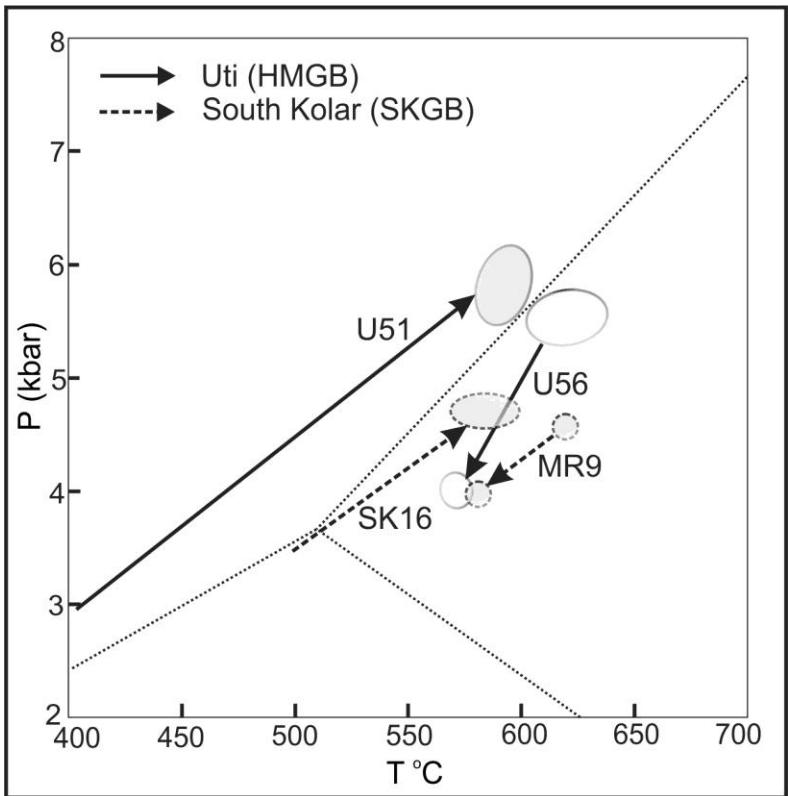
GPHQ barometry (Kohn and Spear, 1990, AM)
 $P = 3.99\text{--}4.21 \text{ kbar}$
Grt-Amph thermometry (Graham and Powell, 1984, JMG)
 $T = 580\text{--}590 \text{ }^{\circ}\text{C}$



Greenstone metamorphism: summary



- HMGB underwent middle amphibolite facies metamorphism with concomitant increase in P and T on prograde path. Garnet retrograde zoning indicate near isothermal decompression
- SKGB underwent lower amphibolite facies metamorphism with synchronous increase in P and T in the prograde path followed by decompressional cooling retrogression



Hazarika et al (2015) *Journal of Geology*

Rationale behind monazite chemical dating

Poor man's tool



$$\text{Pb}_m = \text{Th} (e^{\lambda 232t-1}) + 0.9928 \text{ U} (e^{\lambda 238t-1}) + 0.0072 \text{ U} (e^{\lambda 235t-1}) = \text{Pb}_e$$

where Pb_m and Pb_e are measured and estimated Pb contents

$$t \Rightarrow \text{Pb}_m - \text{Pb}_e \approx 10^{-4}$$

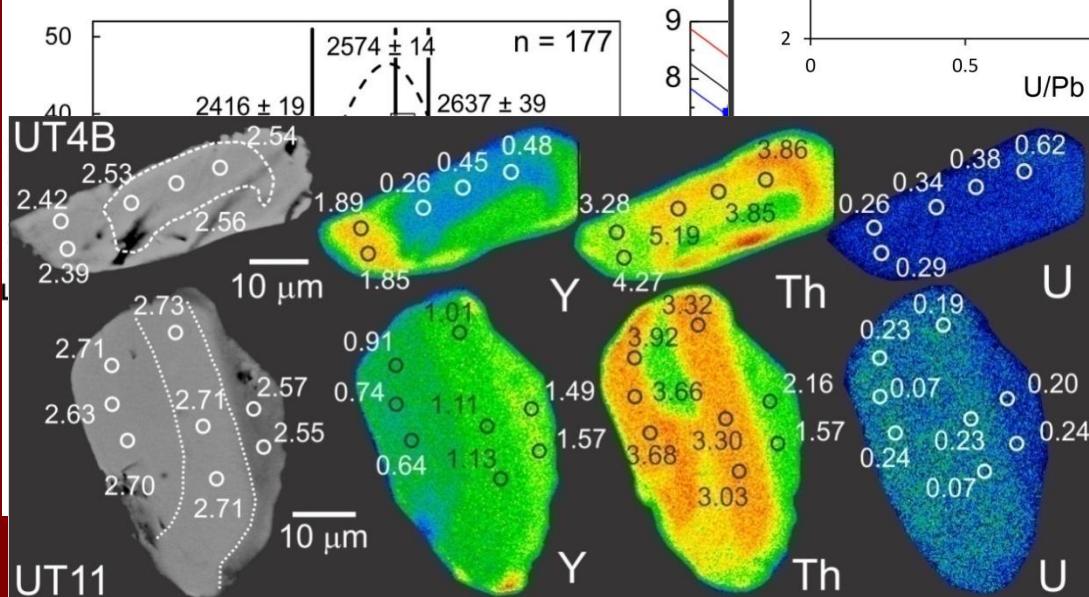
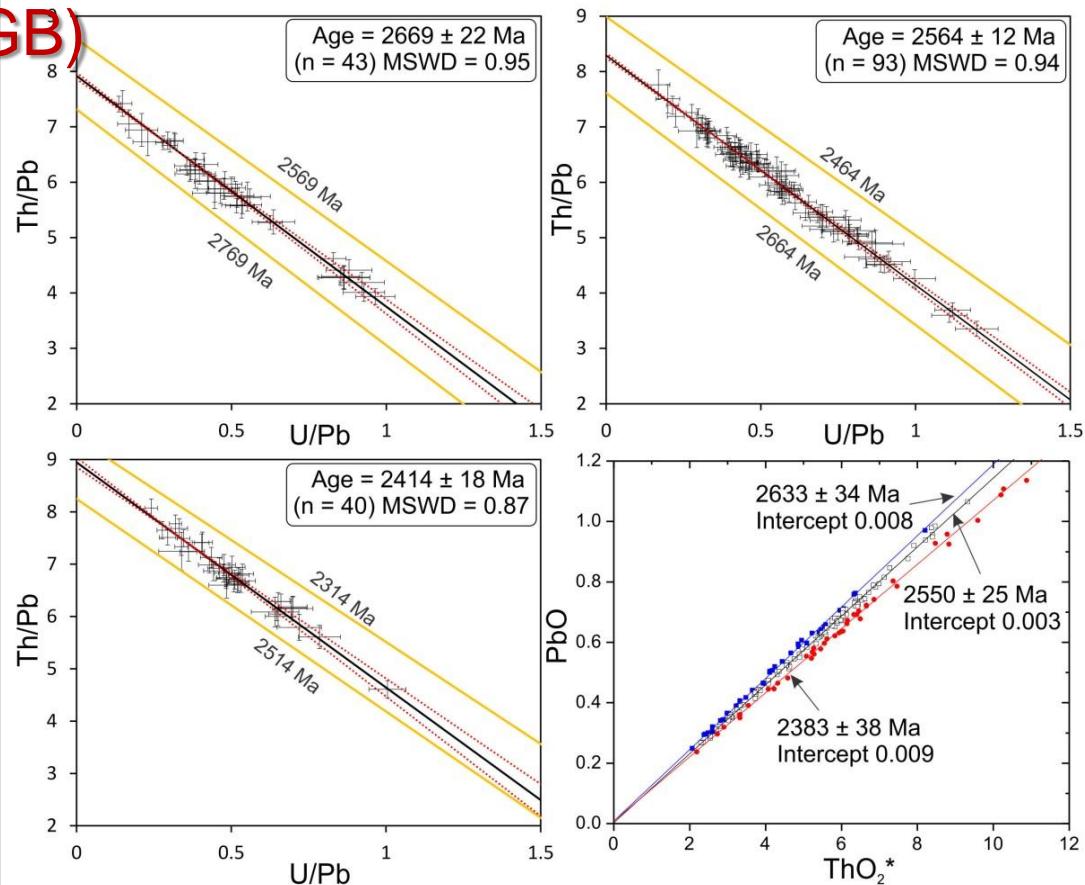
(Montel et al, 1996, GCA)

$$^{238}\text{U}/(^{238}\text{U} + ^{235}\text{U}) = 0.9928, \quad ^{235}\text{U}/(^{238}\text{U} + ^{235}\text{U}) = 0.0072$$

- Insignificant initial Pb
- High (>900 °C closure temperature)

Monazite dating (HMGB)

Age Domain A → within Grt
 2743–2624 Ma (n = 43)
 Age Domain B → Matrix
 2618–2503 Ma (n = 93)
 Age Domain C → Near Veins
 2481–2352 Ma (n = 40)
 Overlapping 2σ errors
 (64–225 Ma)



Cocherie and Albarede, 2001, GCA;
 Suzuki and Kato, 2008, GR

<u>Population</u>	<u>B</u>	<u>C</u>	
Garnet	UT7	0.37	2.08
Present	UT4B	0.52	1.92
Garnet	UT11	1.28	1.24
Absent	UT13	1.19	1.18

Monazite dating (SKB)

Pl

Bt

Age Domain A → within Grt and Cores of zoned Mnz

2736–2612 Ma (n = 26)

Age Domain B → Matrix

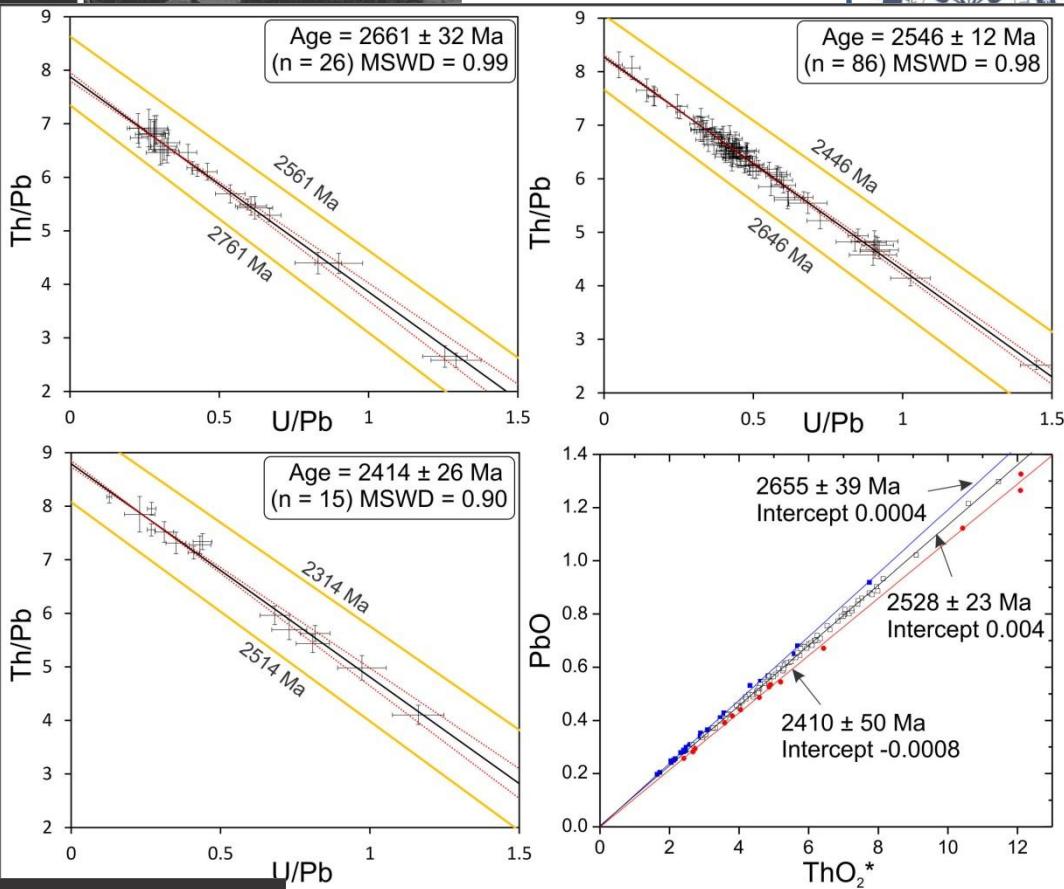
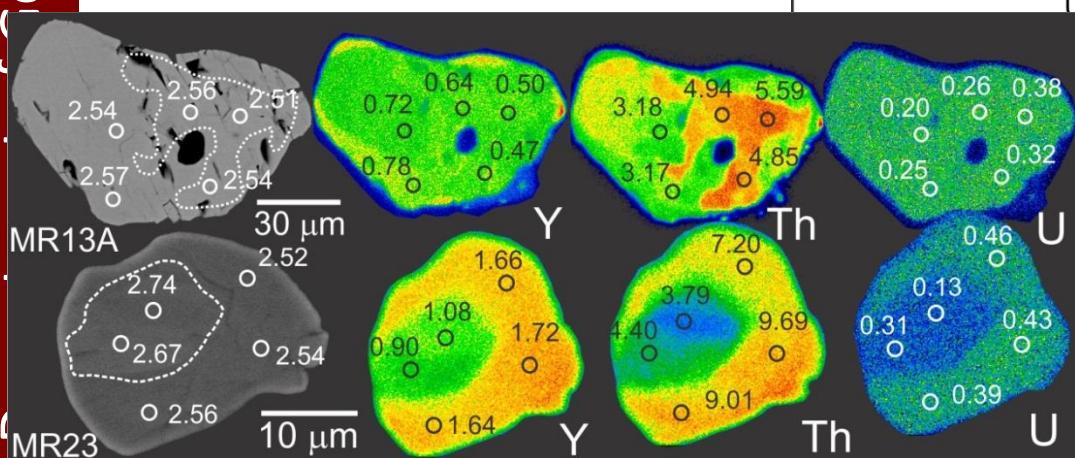
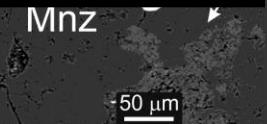
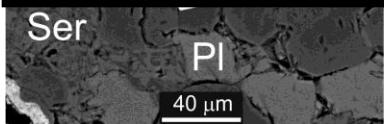
2604–2488 Ma (n = 93)

Age Domain C → Near sericitization

2459–2343 Ma (n = 15)

Overlapping 2 σ errors

(50–217 Ma)



Y-content (oxide wt.%) in monazite

Population

→ B

Garnet { MR13A 0.78

Present { MR25 0.32

Garnet { MR23 1.71

Absent { MR30 1.59

Summary: monazite ages



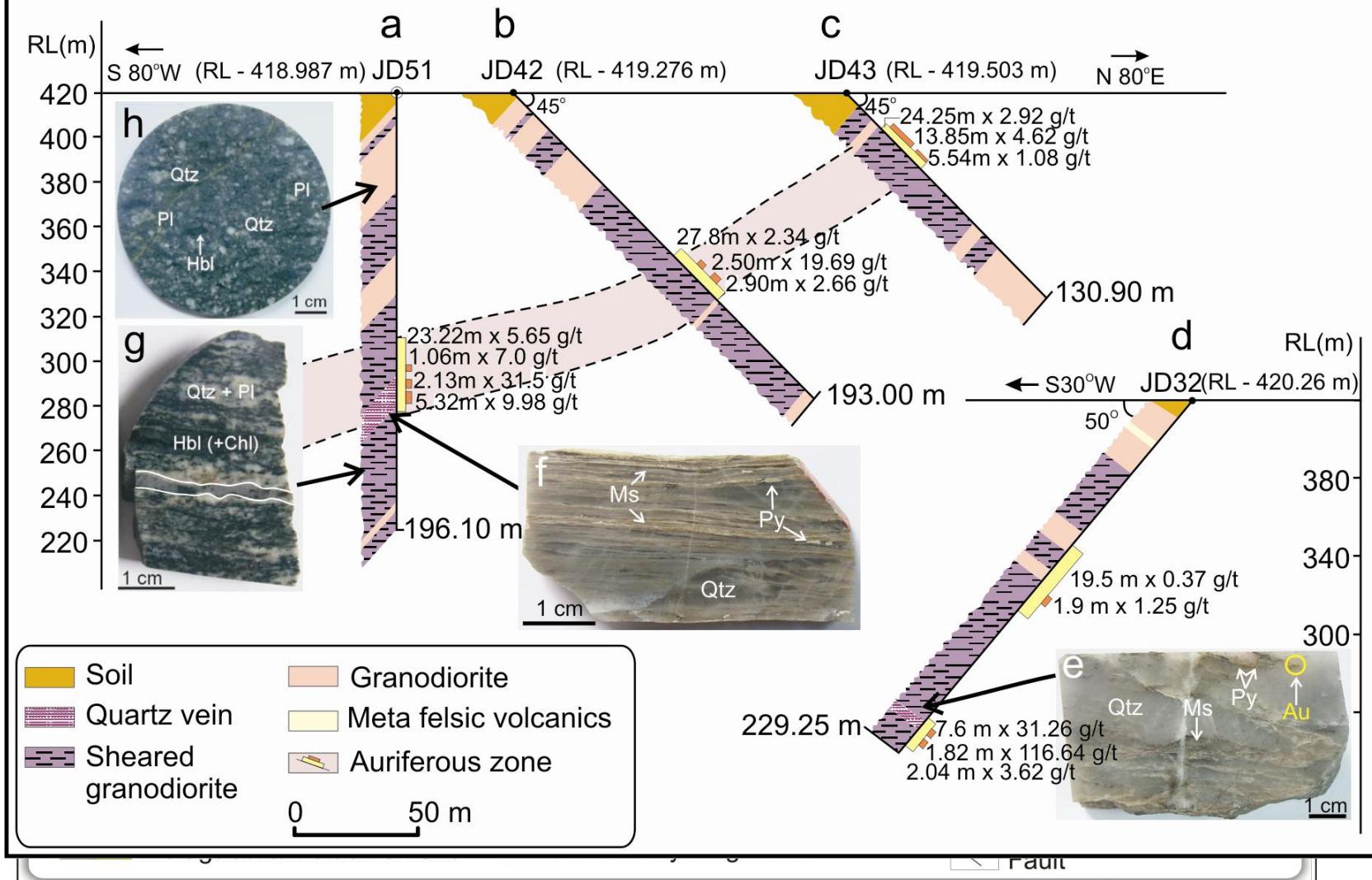
- Although ~400 km apart; HMGB and SKGB witnessed similar geochronological evolution
- Greenstone **volcanism** at ~2.67–2.66 Ga
- Greenstone **metamorphism** at ~2.56–2.55 Ga
- Hydrothermal alteration at ~2.41 Ga

Hazarika et al (2015) *Journal of Geology*

Jonnagiri granitoid-hosted Au deposit



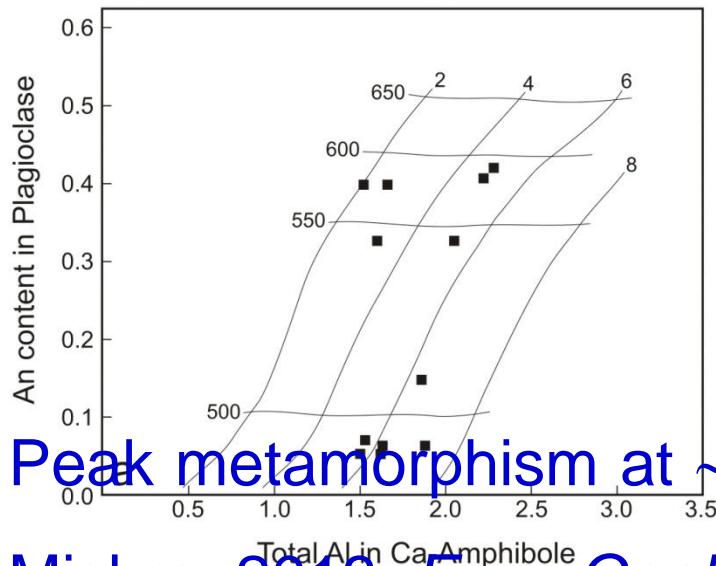
Bole hole logs (courtesy GSI)



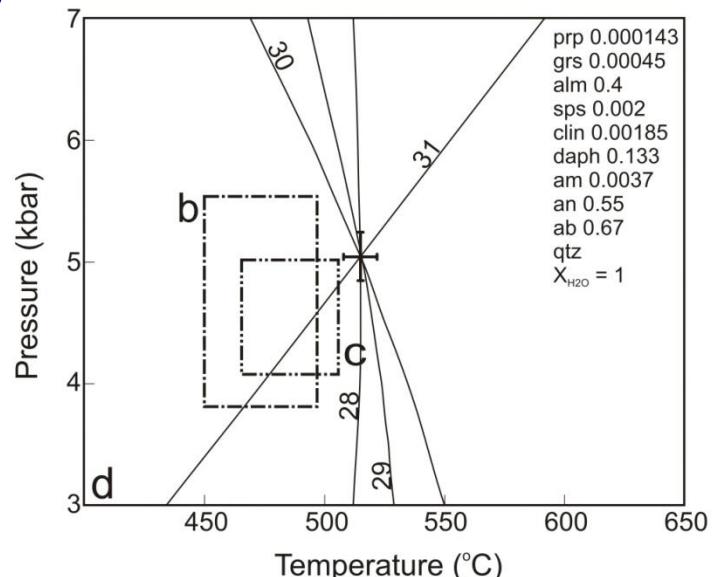
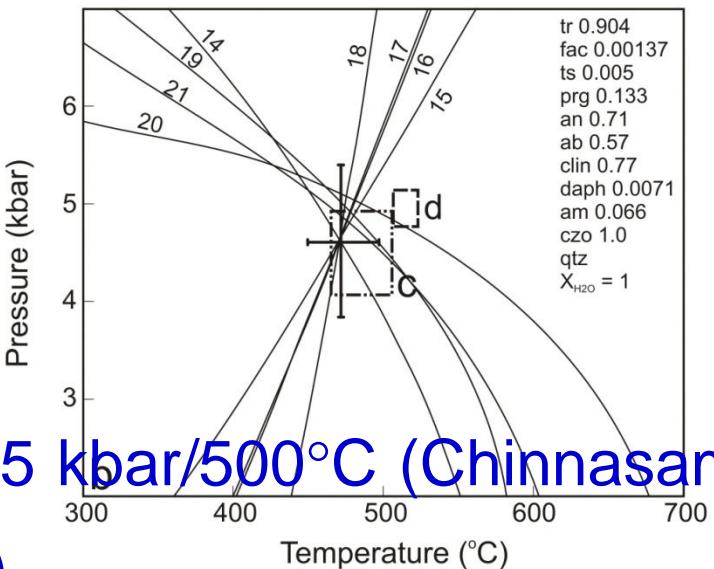
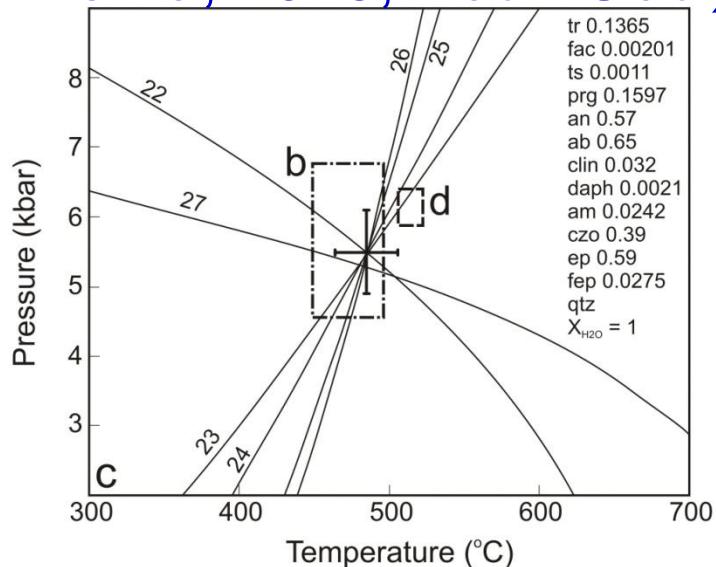
(Redrawn from Jairam et al, 2001)

CS-EB meeting, Bangalore, November 28-29, 2015

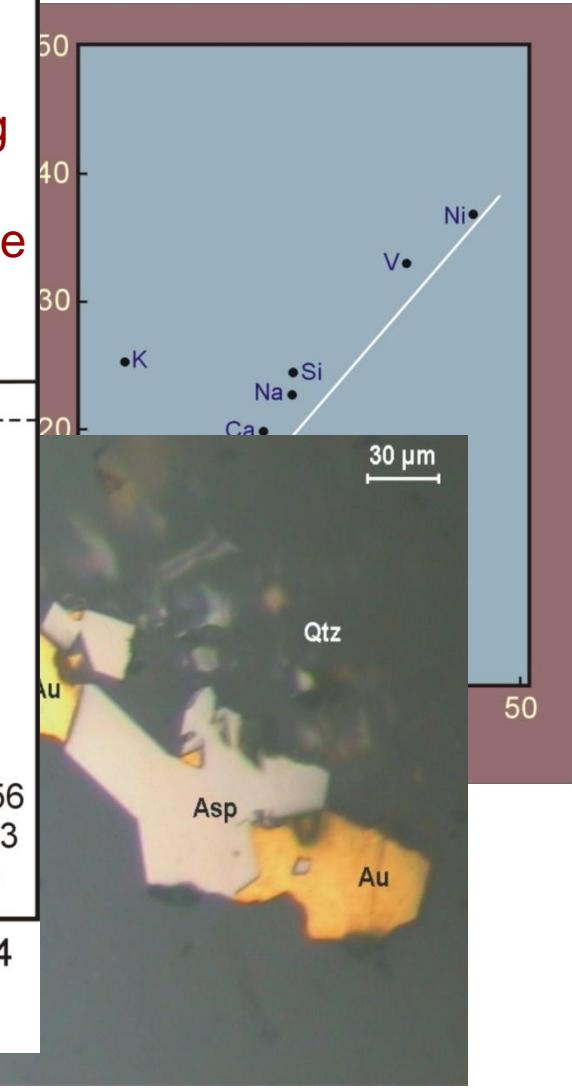
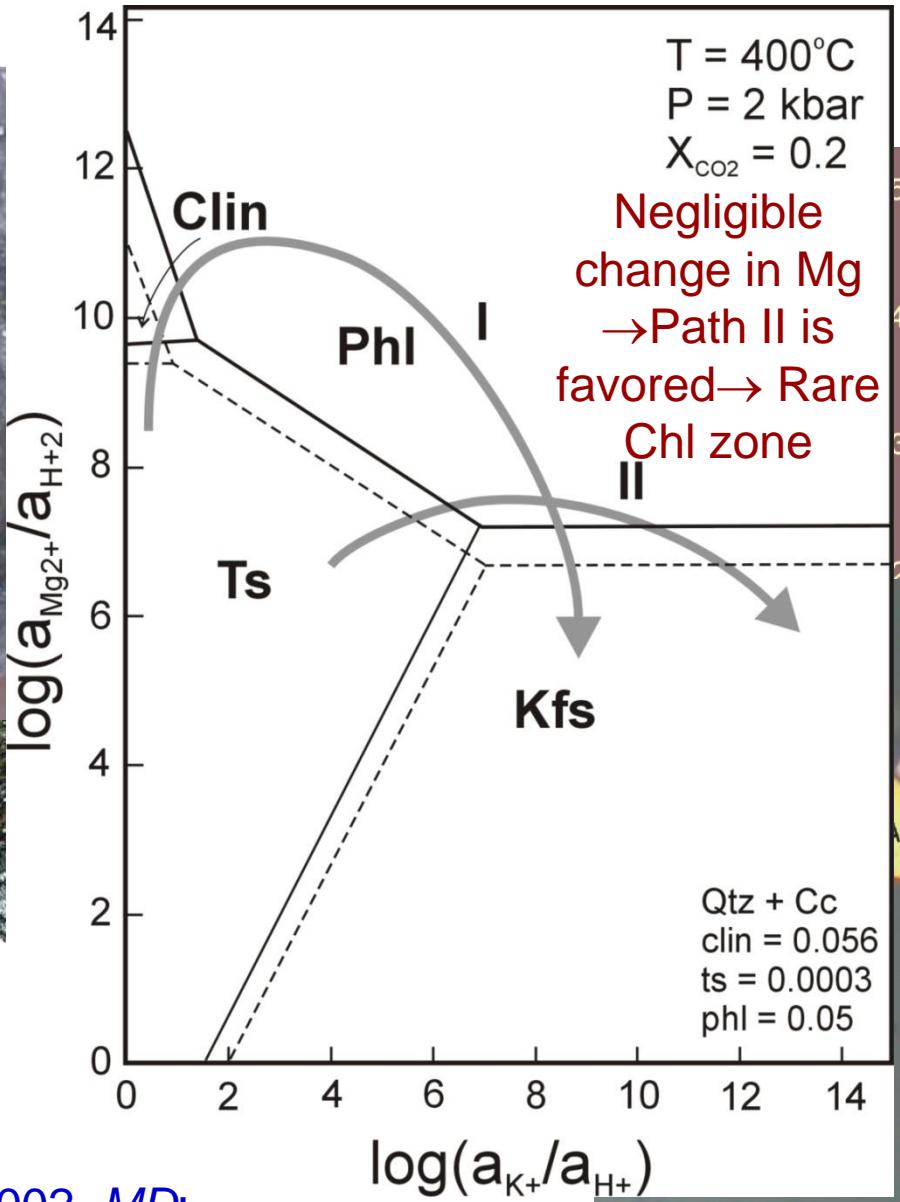
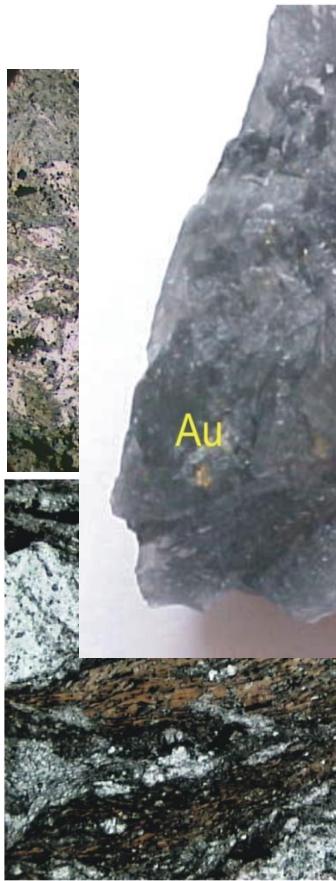
Thermobarometry of Jonnagiri greenstones



Peak metamorphism at ~5 kbar/500°C (Chinnasamy & Mishra, 2013, *Econ Geol*)

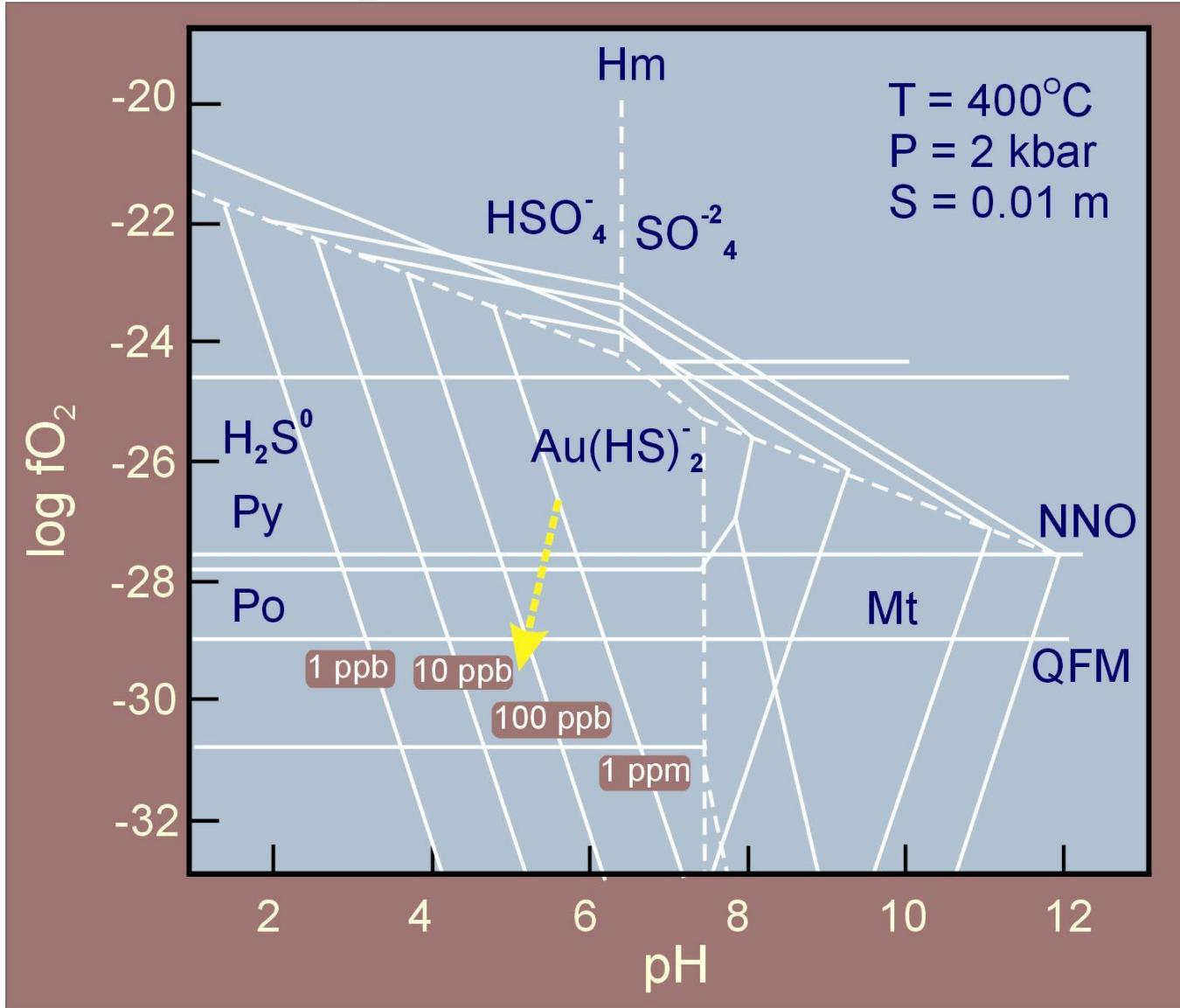


Hutti : alteration and mineralization



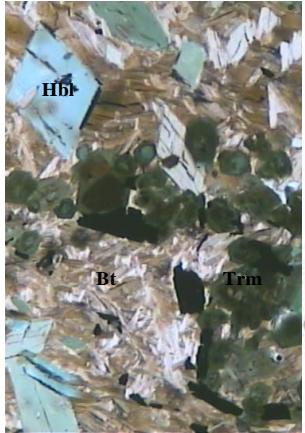
(Pal & Mishra, 2002, *MD*;
Mishra & Pal, 2008, *EG*)

Hutti: $f\text{O}_2$ -pH variation in D2 veins

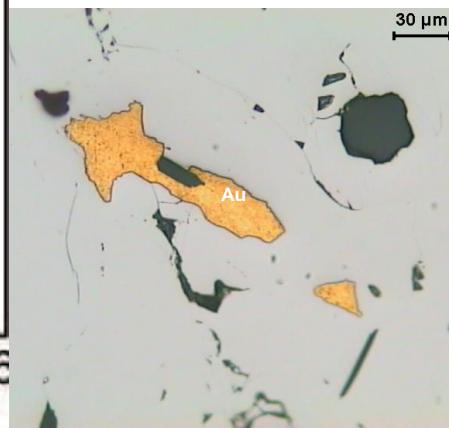
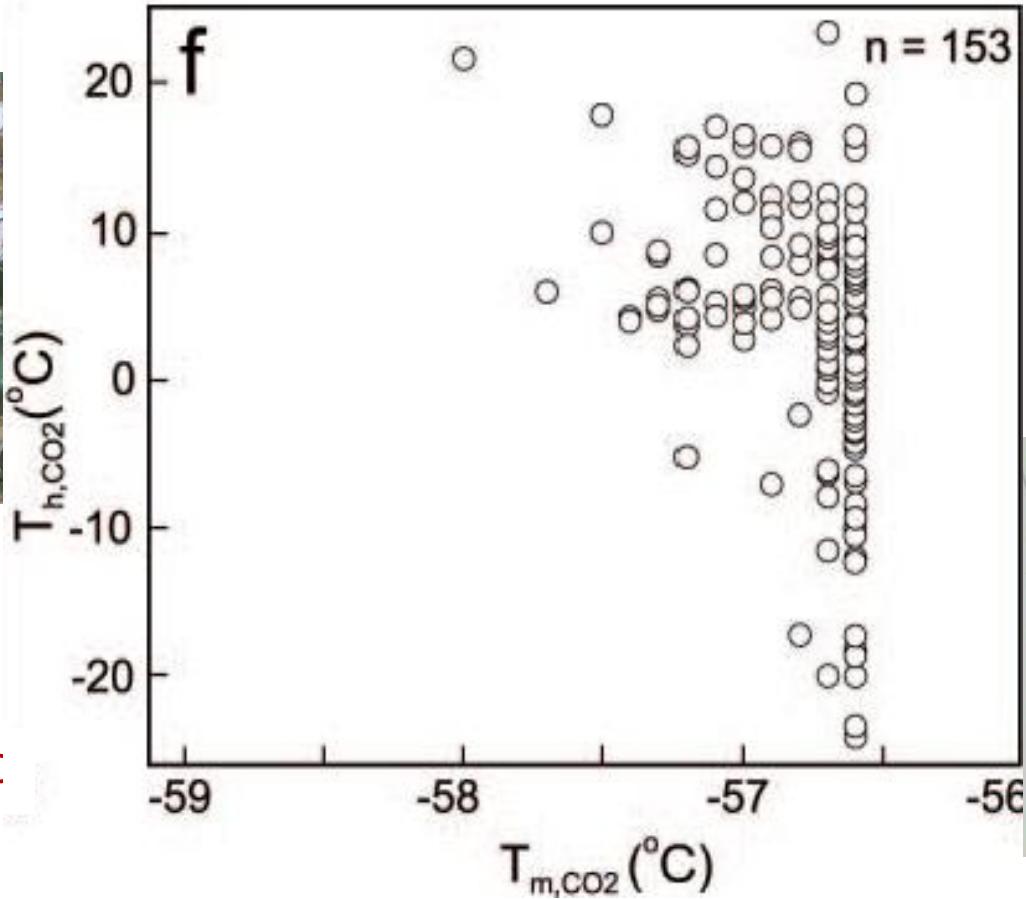


Solubility drop by decrease in $f\text{O}_2$ and marginal fall in pH

Hira-Buddini: Alteration and Au occurrence

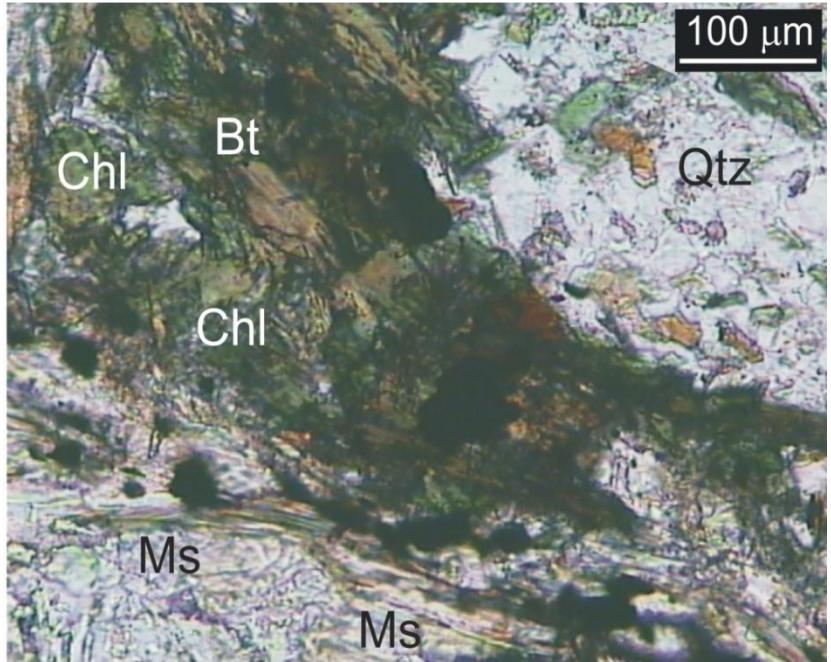
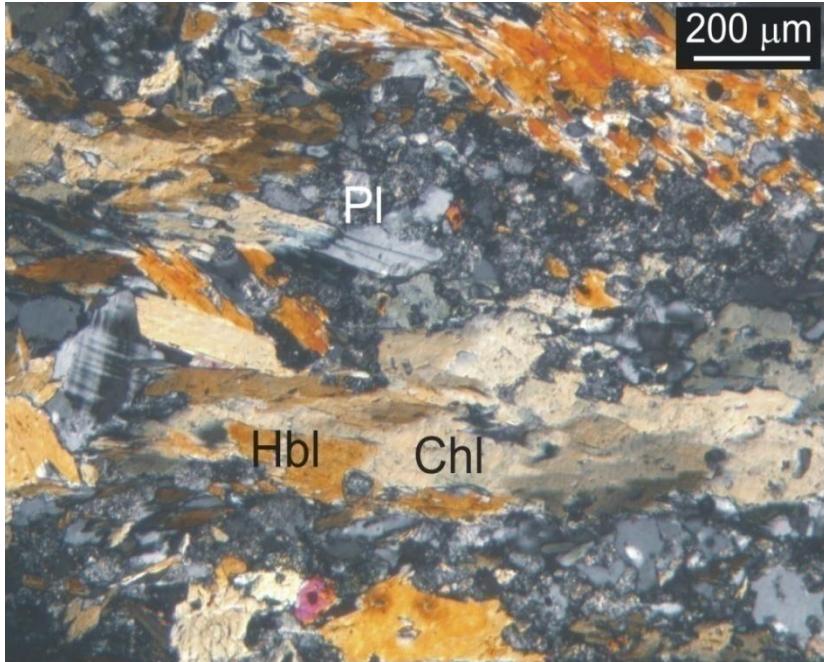


Distal: Chl +
and inner zc
2007, MD)



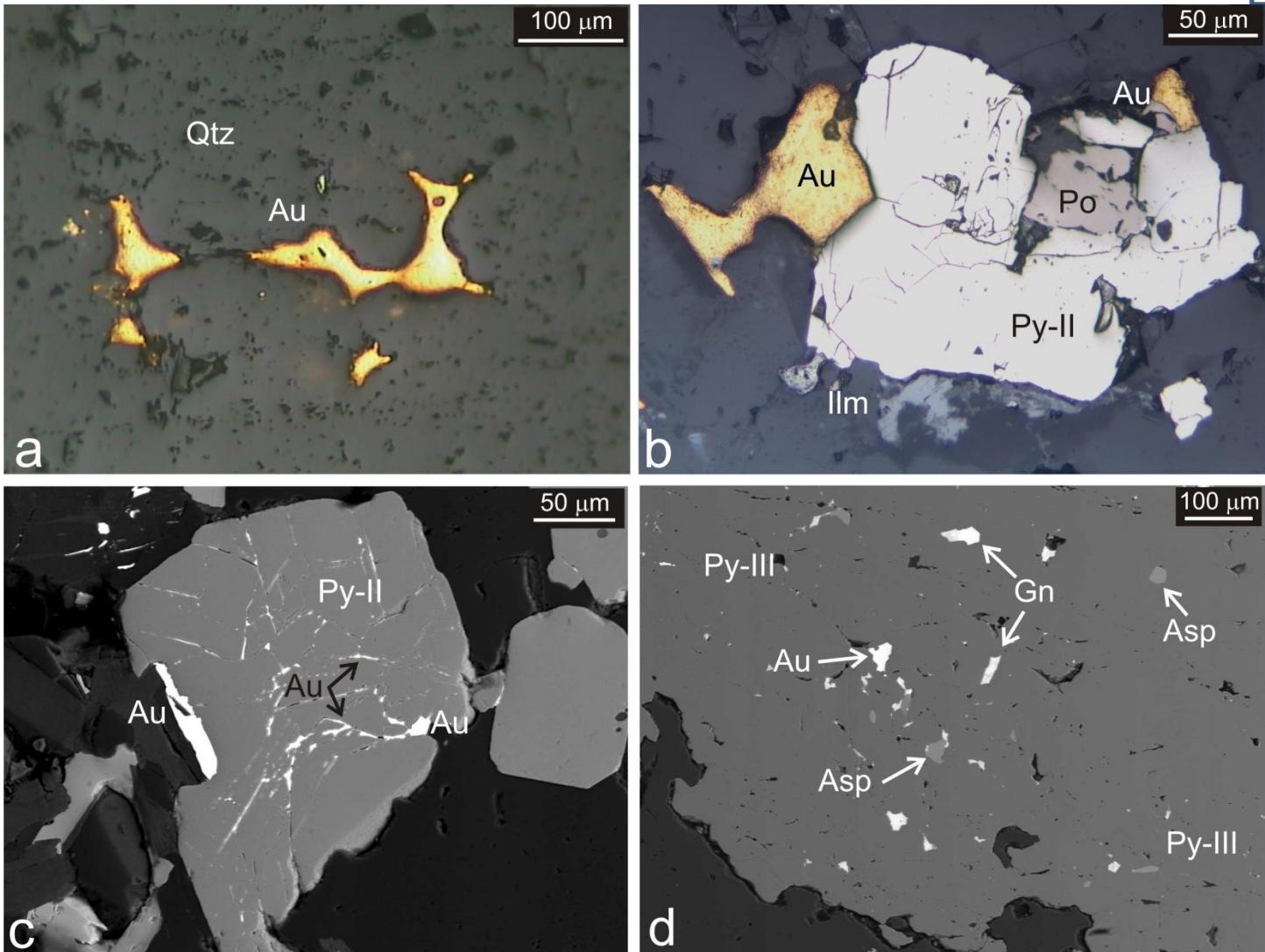
Pressure fluctuations (variation in T_{h,CO_2}) aided fluid immiscibility (Fault valve/Sibson Cycle) \rightarrow decrease in ΣS favoring gold precipitation

Jonnagiri: alteration

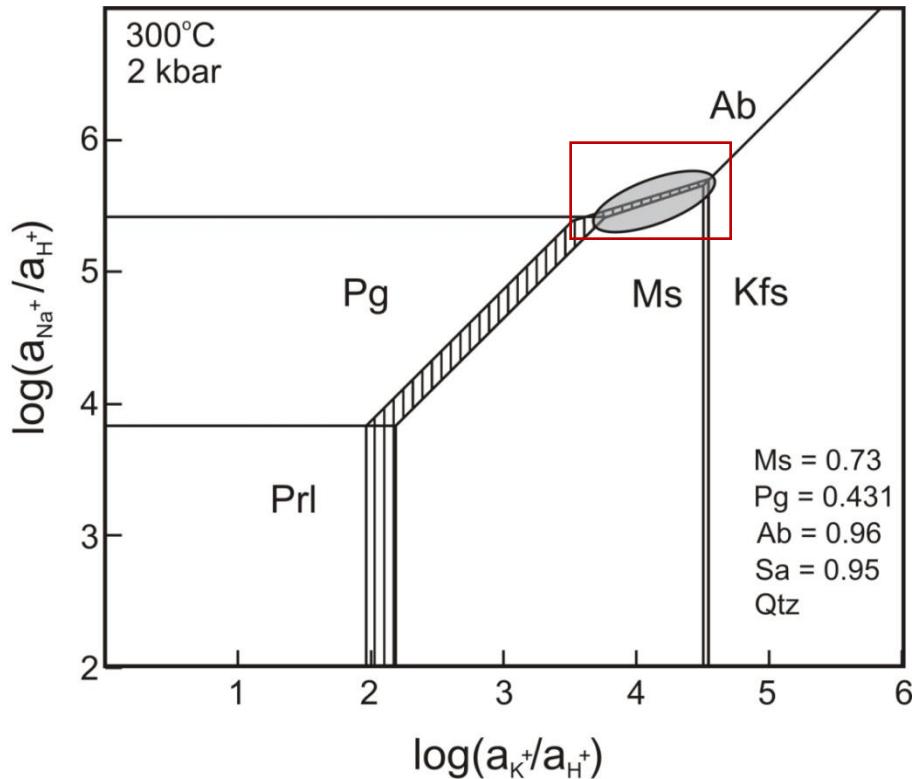
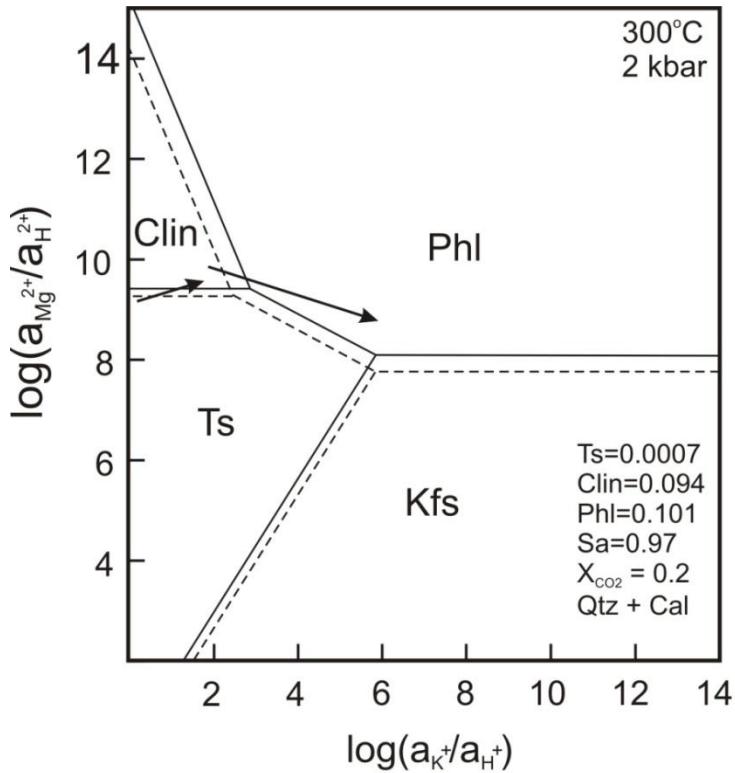


Proximal: Chl + Bt (+Ms)

Jonnagiri: Au-bearing inner zone



Jonnagiri: alteration



Proximal zone: Hbl → Chl → Bt

Inner zone: Ms + Kfs + Ab

Jonnagiri: alteration



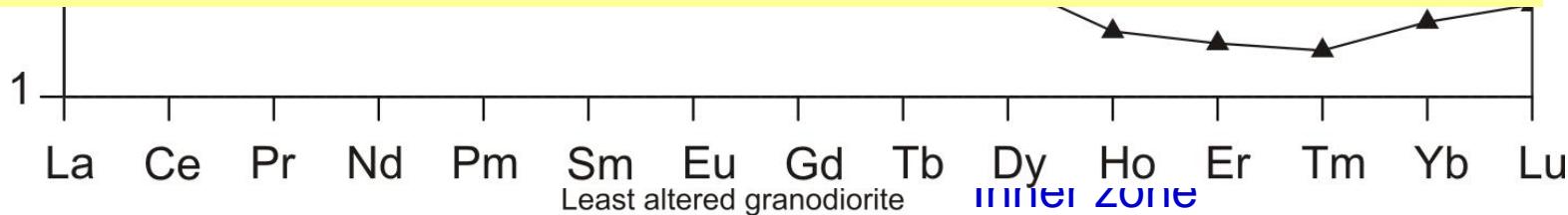
1000

Least altered granodiorite
Altered granodiorite

Proximal: gain in S, C, K, Si, and Rb and depletion of Ca, Mg, and Fe

Inner: addition of Si, K, S, C, and Au and depletion in REEs (high W/R)

(Chinnasamy & Mishra, 2013, *Econ Geol*)





Craton scale ore fluid uniformity

Composition: Initial low saline (4 to 10 wt. % equivalent) $\text{H}_2\text{O}-\text{CO}_2-\text{CH}_4-\text{NaCl}-\text{Au}(\text{HS})_2^-$ fluid ← phase separation by various means (**NO MAGMATIC SIGNATURE**) analogous to many in Yilgarn Craton/ Canadian Shield (Saravanan et al, 2009, OGR)

P-T of ore formation

Hutti: (i) D2: 2–3 kbar/ $\geq 400^\circ\text{C}$ (Mishra and Pal, 2008 EG; Rogers et al, 2013, MD), (ii) D3: 1.0 –1.7 kbar/ $280^\circ - 320^\circ\text{C}$ (Pal and Mishra, 2002 MD)

Kolar: 1.80 kbar/ 280°C (Mishra and Panigrahi, 1999 MD)

Ramagiri: 1.45 –1.7 kbar / $240^\circ - 270^\circ\text{C}$ (Mishra and Panigrahi, 1999 MD)

Ajjanahalli: 0.8 –1.15 kbar/ $265^\circ - 310^\circ\text{C}$ (Pal and Mishra, 2003 GR)

Jonnagiri: 1.54 – 2.50 kbar/ $290 - 320^\circ\text{C}$ (Saravanan et al, 2009 OGR)

Gadag: 1 kbar/ 216°C to 321°C (Swain et al., 2015 OGR)

Stable isotopes

Sulfur



$\delta^{34}\text{S}_{\text{Sulfide}}$ (+1.1 to +7.1 ‰) $\delta^{34}\text{S}_{\text{H}_2\text{S}} = +0.2$ to +5.8 ‰
(Hutti, Hira-Buddini, Uti, Chigargunta, Ajjanahalli, Jonnagiri)
→ magmatic or average crustal source of sulfur
(Saravanan & Mishra, 2009, MD)

Carbon and oxygen

Carbonate $\delta^{13}\text{C} = 5.5 \pm 1.3\text{\textperthousand}$, $\delta^{18}\text{O} = 4.1 \pm 2.7\text{\textperthousand}$ (Ajjanahalli)
 $6.2 \pm 1.9\text{\textperthousand}$ $14.1 \pm 0.5\text{\textperthousand}$ (GR Halli)
 $-2.2\text{\textperthousand}$ to $-9.7\text{\textperthousand}$ $12.0\text{\textperthousand}$ to $30.5\text{\textperthousand}$ (Gadag)

Average fluid $\delta^{13}\text{C} = 5.81 \pm 1.14\text{\textperthousand}$ $\delta^{18}\text{O} = 13.78 \pm 5.1\text{\textperthousand}$
 $4.64 \pm 0.7\text{\textperthousand}$, $6.50 \pm 0.6\text{\textperthousand}$
 $-2.1\text{\textperthousand}$ to $-9.6\text{\textperthousand}$ $6.8\text{\textperthousand}$ to $25.9\text{\textperthousand}$

(Sarangi et al., 2012, JAES; Swain et al., 2015, OGR)

Fl and stable isotope data: global synthesis



- H₂O–NaCl–CO₂±CH₄ fluid (Na >K > Mg and Ca)
- Salinity = 3 to 7 wt. % NaCl equivalent
- X(CO₂) = 0.1 to 0.25
- S = 57–1840 ppm (~0.006 to 0.18 mol%) mostly <200 ppm
- pH = ~5.5
- Au = 3–30 ppb Au (under saturated)
- P = 1 to 5 kbar
- T = 220 to 450 °C
- δ¹⁸O = +6 to +11‰ (Precambrian) = +7 to +13‰ (Phanerozoic)
- δD = -20 to -80‰ (insignificant lighter meteoric H₂O)
- δ¹³C_{fluid} = -11 and +2‰ (non-mantle source and insignificant fluid mixing)
- δ³⁴Sulphide = -20 to +25 ‰
(Ridley & Diamond, 2000, Rev. Econ Geol.; Goldfarb & Groves, 2015, *Lithos*)

Some exceptions: possible non-metamorphic fluids'

Late Buddi
2007,
Late from



at Hirai et al,
at Hutt MD)

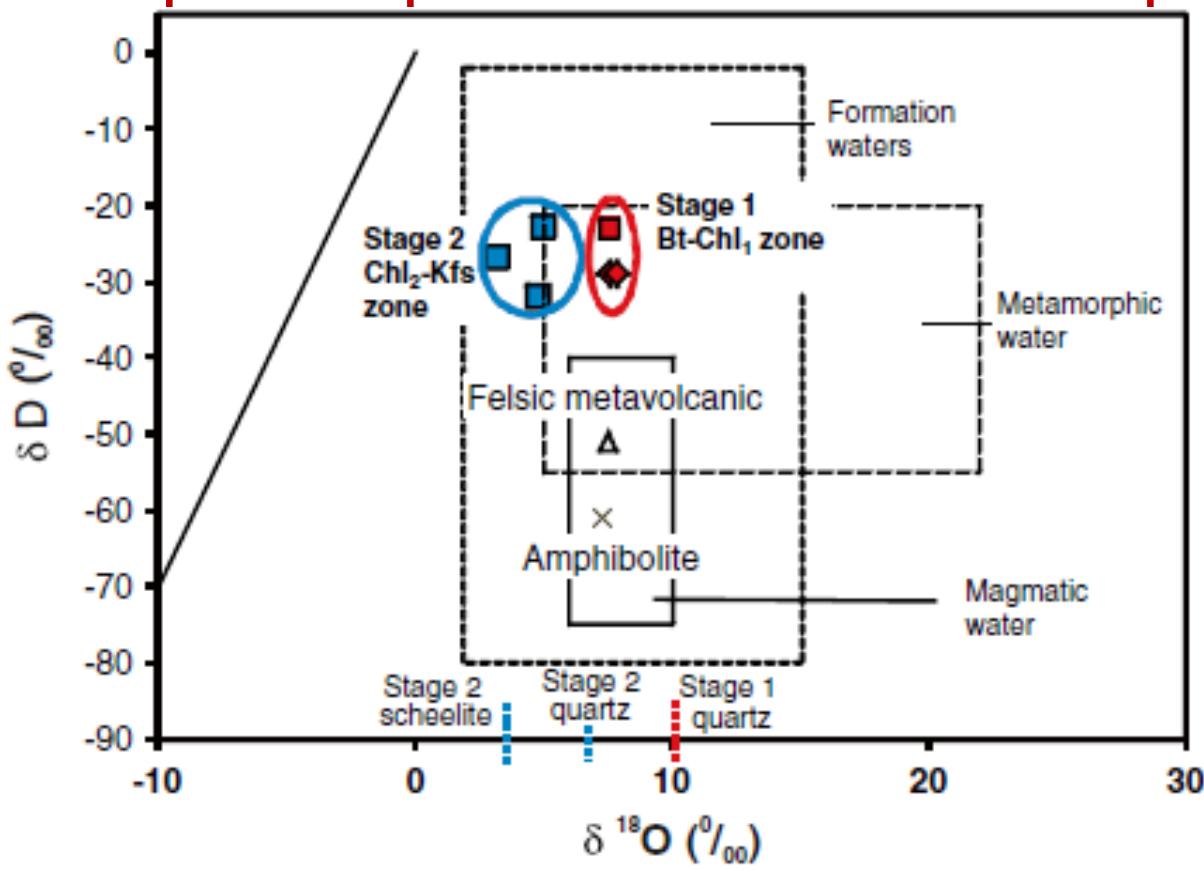


Fig.7 of Rogers et al (2013, MD): fields defined by data points for Hutt stage 1 and stage 2 mineralizations are much closer to each other, within fields of metamorphic and formation waters, away from the magmatic water field?

Geochemistry of hydrothermal minerals



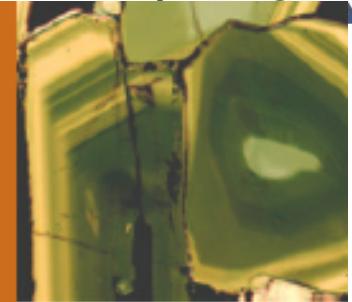
Tourmaline as a Recorder of Ore-Forming Processes

John F. Slack¹ and Robert B. Trumbull²

1811-5209/11/0007-0321\$2.50 DOI: 10.2113/gselements.7.5.321

Tourmaline occurs in diverse types of hydrothermal mineral deposits and can be used to constrain the nature and evolution of ore-forming fluids. Because of its broad range in composition and retention of chemical and isotopic signatures, tourmaline may be the only robust recorder of original mineralizing processes in some deposits. Microtextures and in situ analysis of compositional and isotopic variations in ore-related tourmaline provide valuable insights into hydrothermal systems in seafloor, sedimentary, magmatic, and metamorphic environments. Deciphering the hydrothermal record in tourmaline also holds promise for aiding exploration programs in the search for new ore deposits.

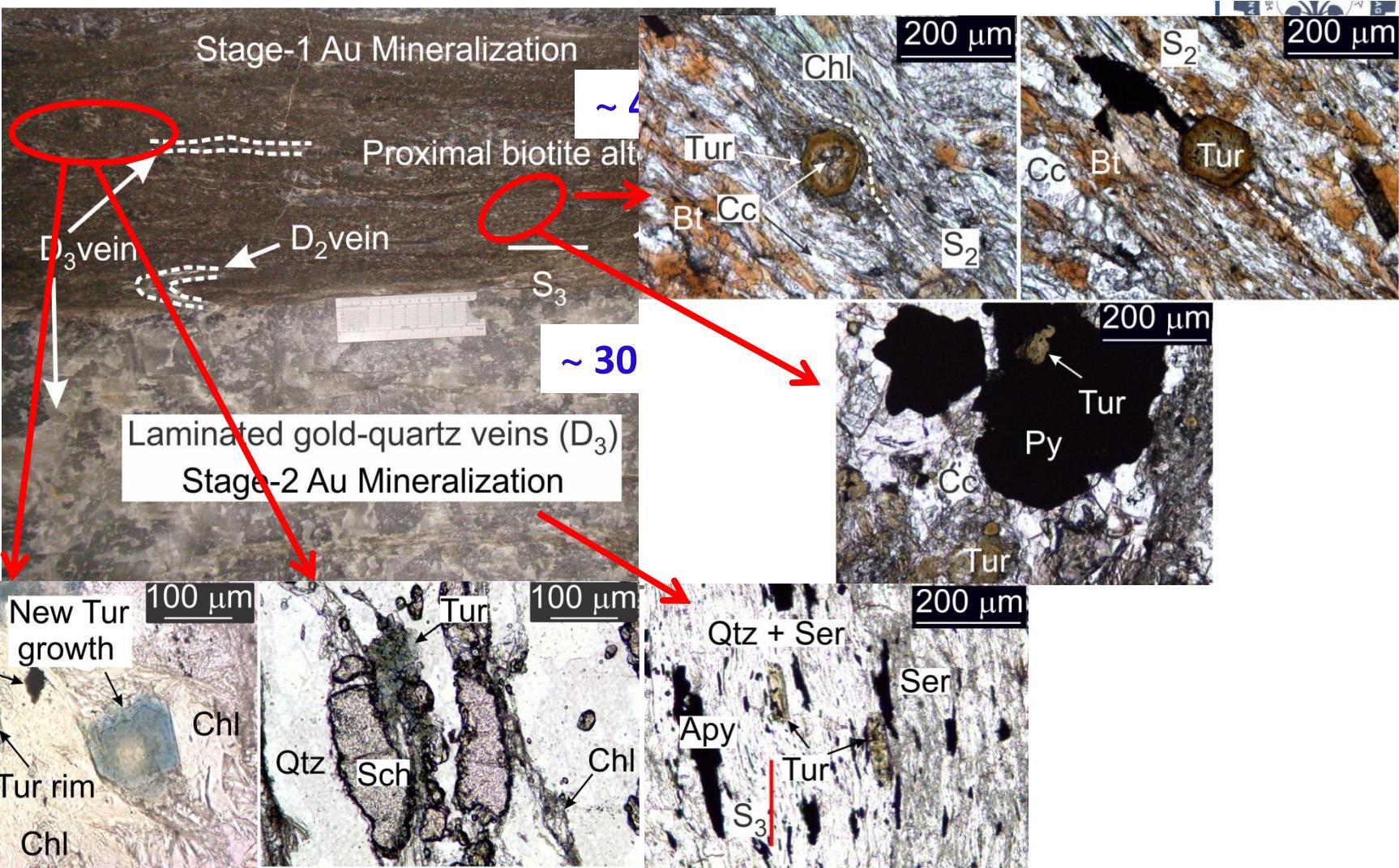
The formation of hydrothermal tourmaline requires the fortuitous convergence of complex processes: derivation of boron and other essential elements from one or more sources; transport in fluids of diverse provenance (magmatic-derived, metamorphic, basinal or evaporitic brines, heated meteoric water, evolved seawater); and finally precipitation in settings and conditions that may coincide with the deposition of economi-



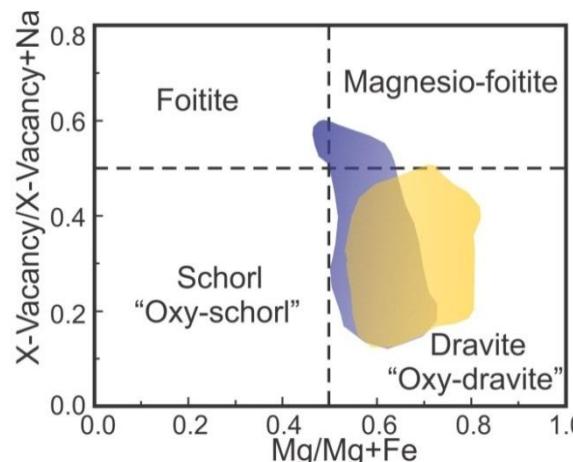
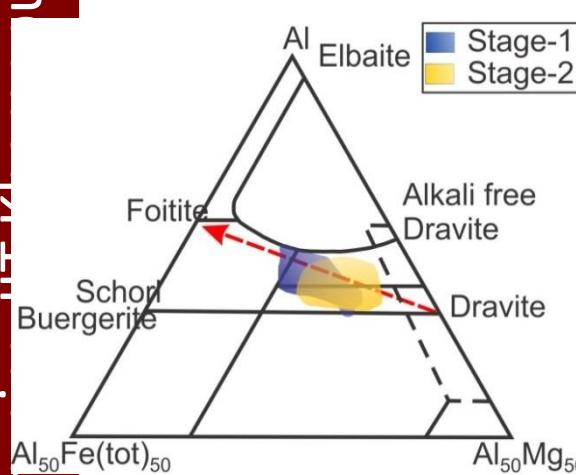
Zoned tourmaline from the Ore Knob VMS deposit, North Carolina, USA. The matrix (black) is chalcopyrite.

Slack and Trumbull (2011) *Elements*

Fluid proxy: Hydrothermal mineral geochemistry Tourmaline



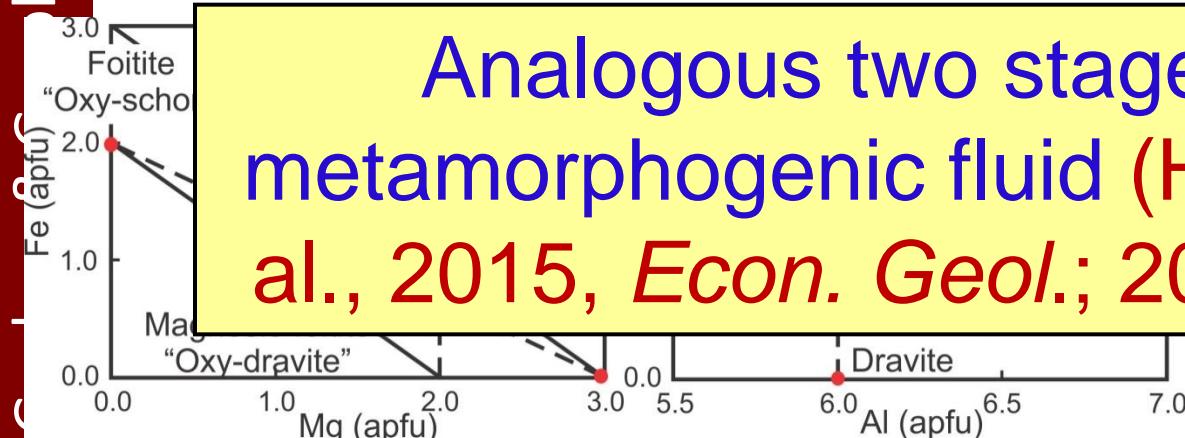
Fluid proxy: Hydrothermal mineral geochemistry Tourmaline



Characteristics

- Low Na contents (< 0.61 apfu) and high X-site vacancies (up to 0.53 pfu) → low salinity fluid

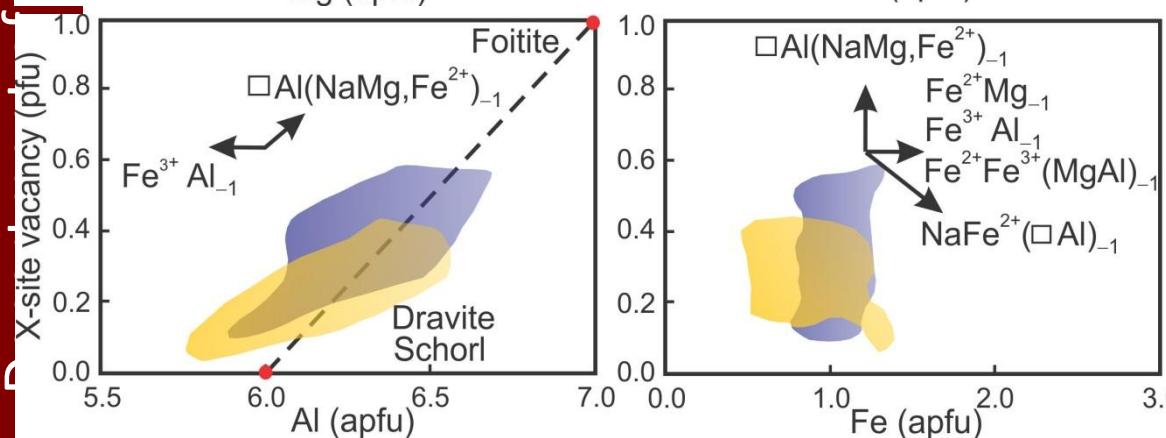
- Dravite-Foitite trend



Analogous two stage Hutton
metamorphogenic fluid (Hazarika et
al., 2015, *Econ. Geol.*; 2016, *OGR*)

Fe^{3+} in Z-site
Al Na substitutions
 $\text{Fe}^{3+}\text{Al}_{-1}$
and insignificant $\text{Fe}^{3+}\text{Al}_{-1}$
substitution → Prevalent Fe^{2+}

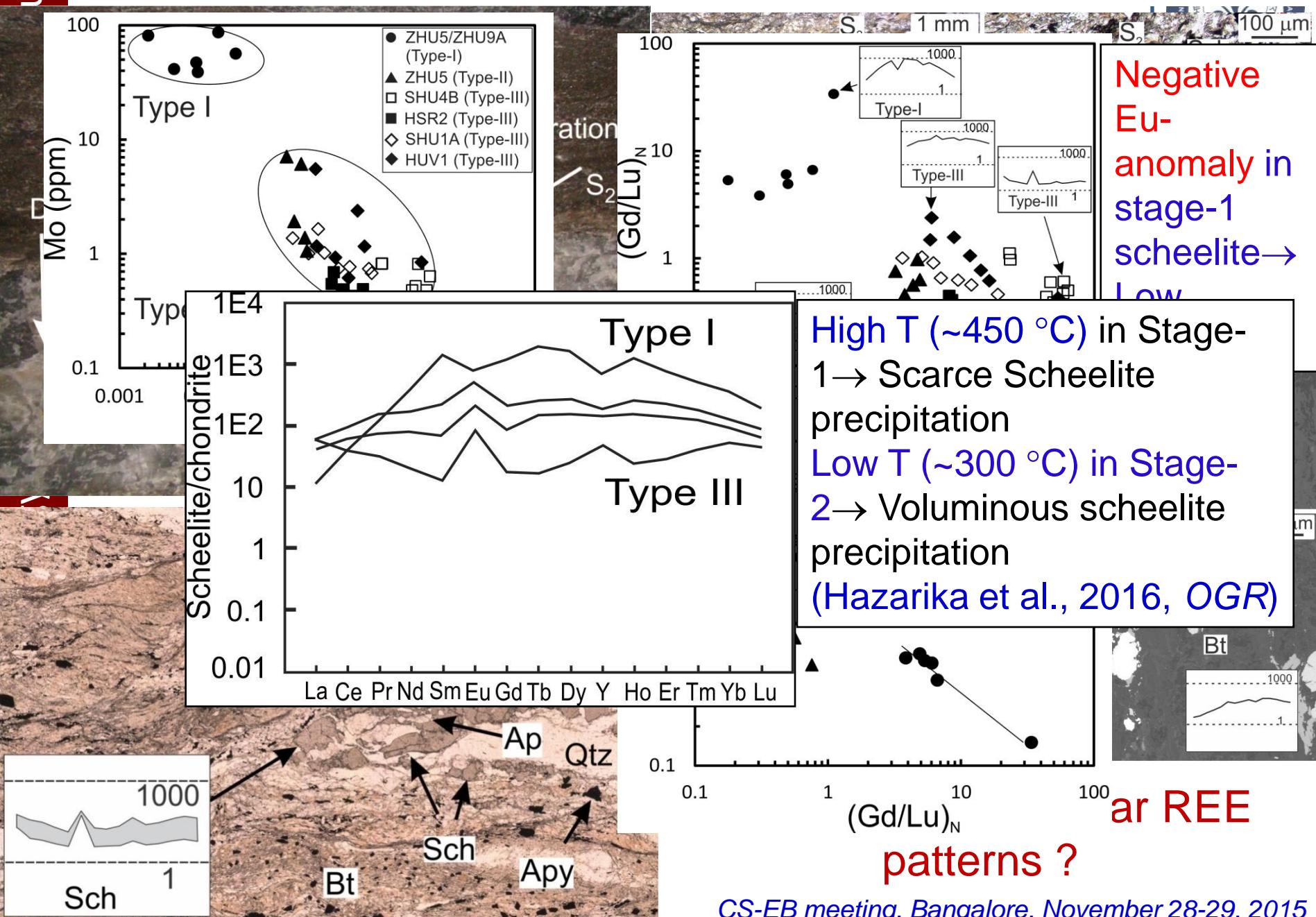
and insignificant $\text{Fe}^{3+}\text{Al}_{-1}$
substitution → Prevalent Fe^{2+}



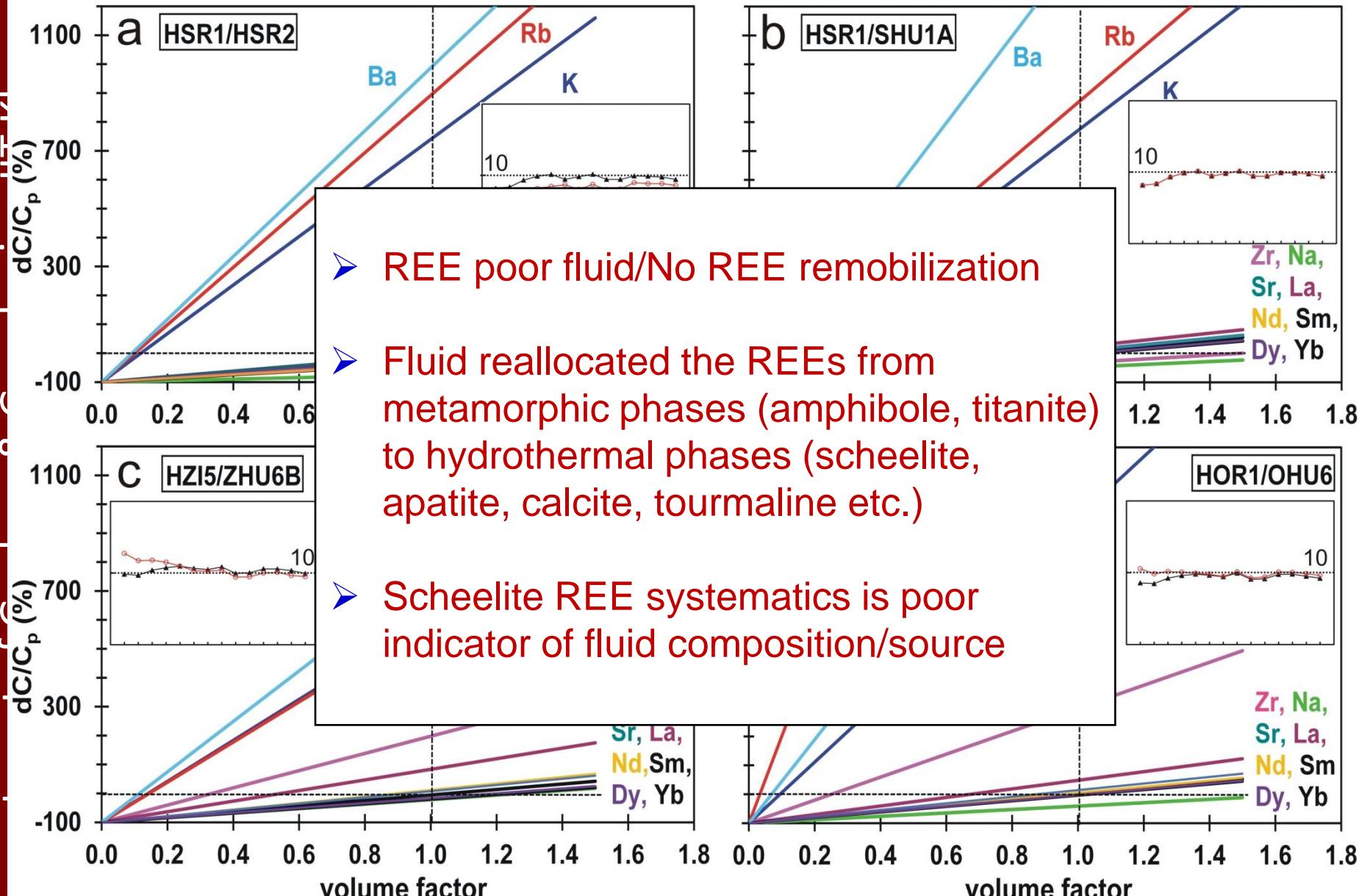
- Low $\text{Fe}^{3+}/\text{Fe}^{2+}$ → reducing fluid

- Low trace element contents (LILE and HFSE)

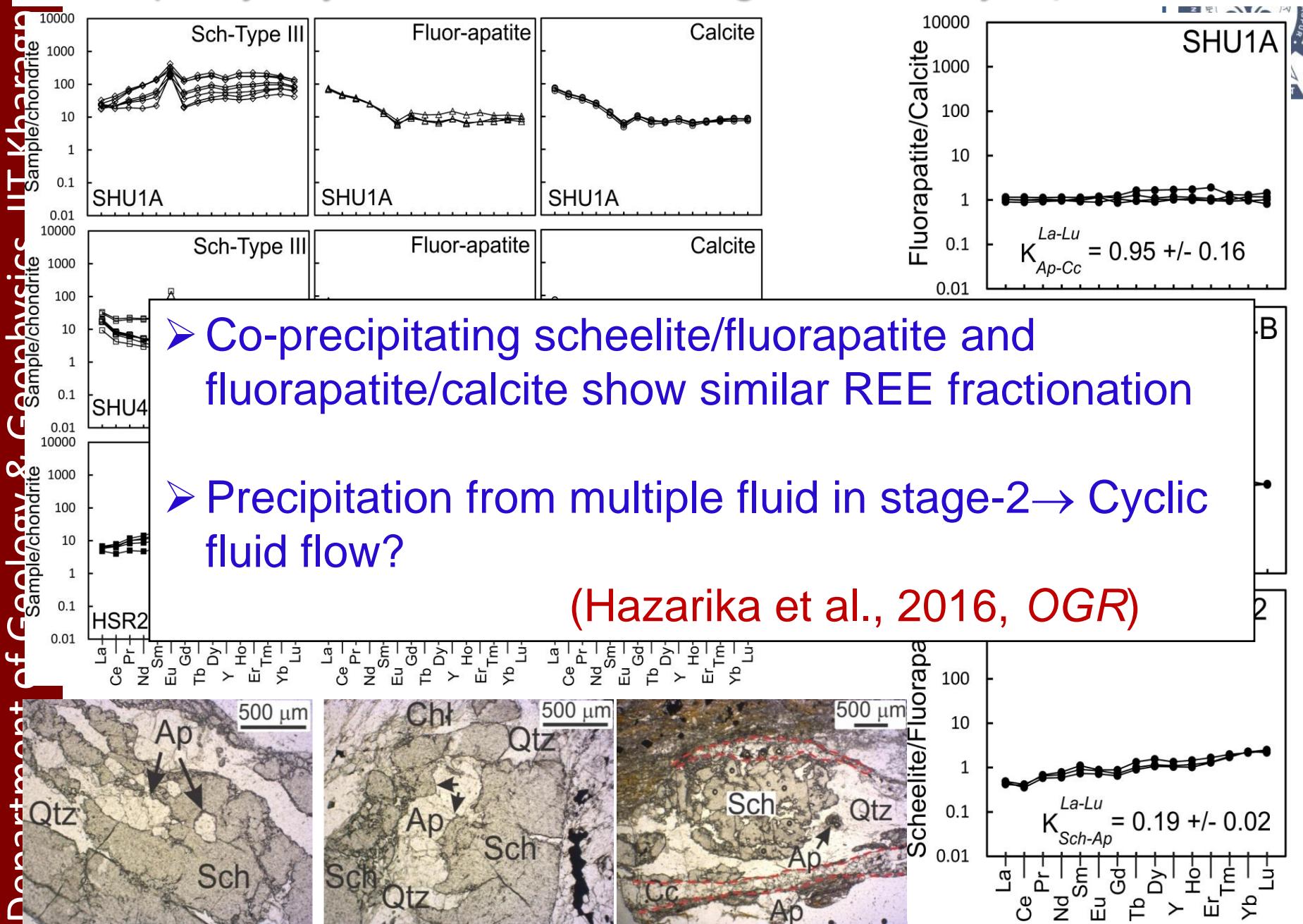
Fluid proxy: Hydrothermal mineral geochemistry Scheelite



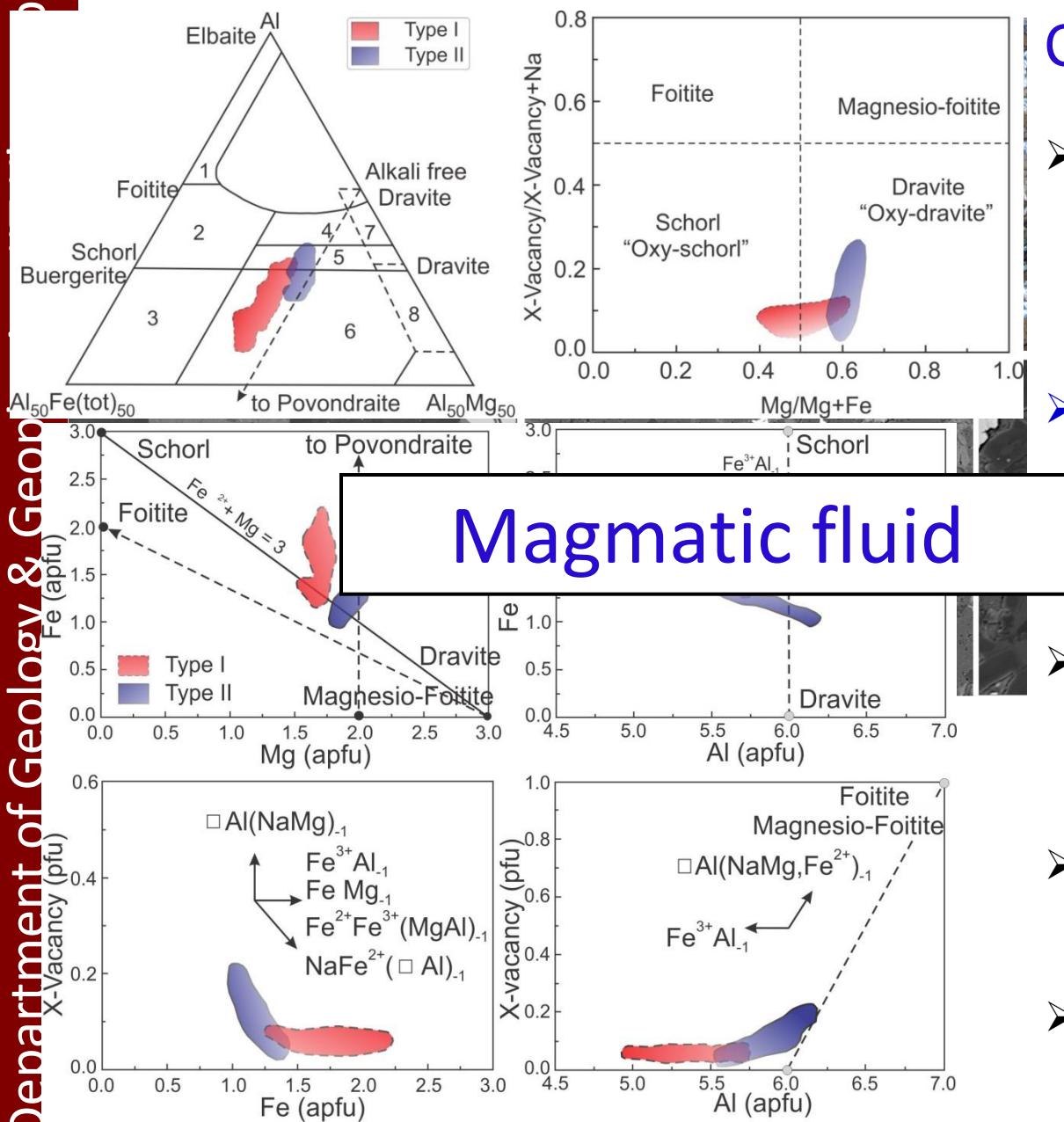
Hutti: Whole rock mass balance calculations



Fluid proxy: Hydrothermal mineral geochemistry Apatite/Calcite



Fluid proxy: Hydrothermal mineral geochemistry Hira-Buddini (Tur)

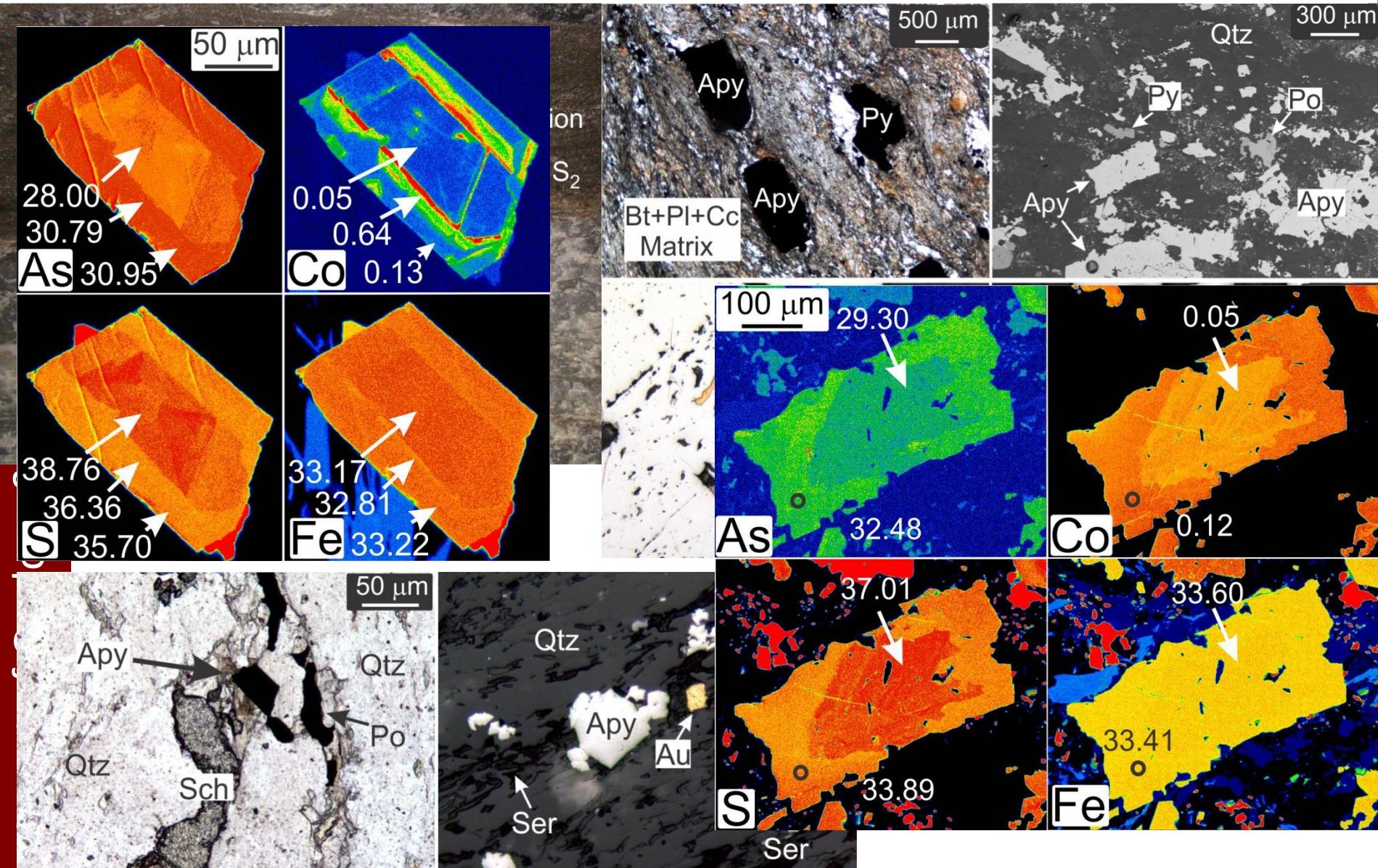


Characteristics (Type I)

- High Na contents (up to 0.84 apfu) and low X-site vacancies (< 0.14 pfu) → **High salinity of the fluid**
- Oxy-dravite-Povondraite trend
- $\text{Al} < 6 \text{ apfu} \rightarrow \text{Fe}^{3+} \text{ in Z-site}$
- Dominant $\text{Fe}^{3+}\text{Al}_{-1}$ substitution → Prevalent Fe^{3+}
- High $\text{Fe}^{3+}/\text{Fe}^{2+}$ → **oxidizing fluid**
- High contents of Pb, Sr and REE

Arsenopyrite/pyrite compositions

Hutti

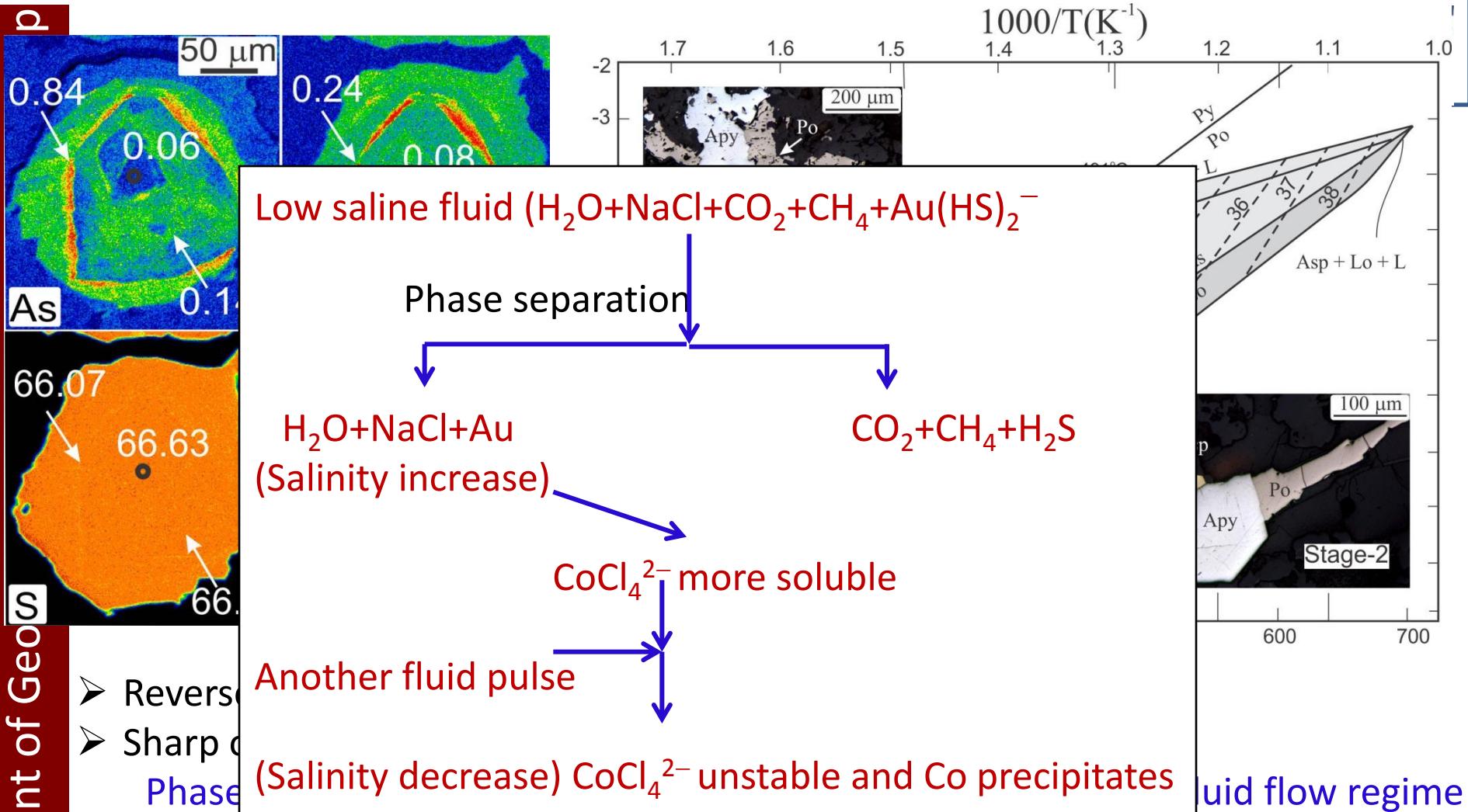


Stage-1: Py + Apy + Po; Stage-2:Apy + Po

g, Bangalore, November 28-29, 2015

Arsenopyrite/pyrite compositions

Hutti

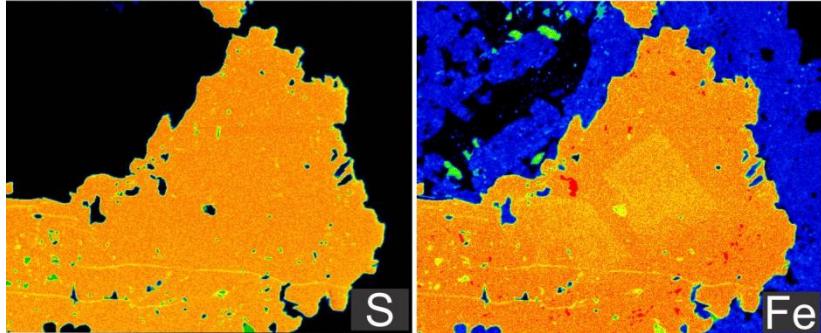
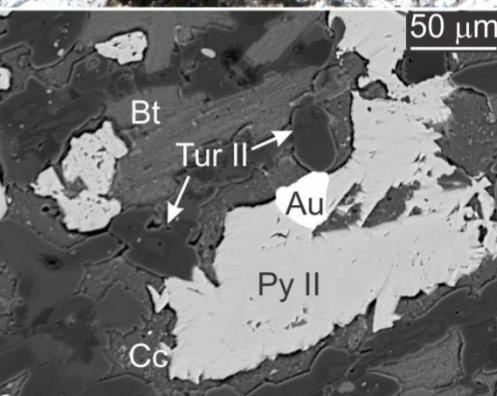
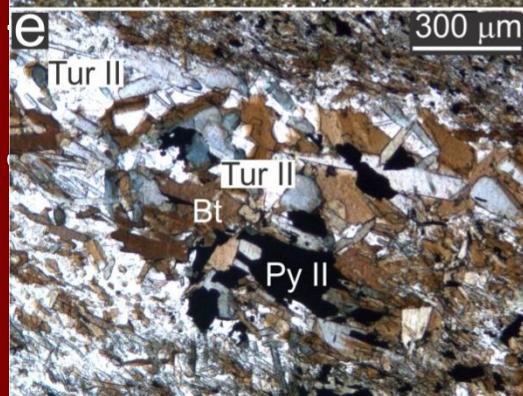
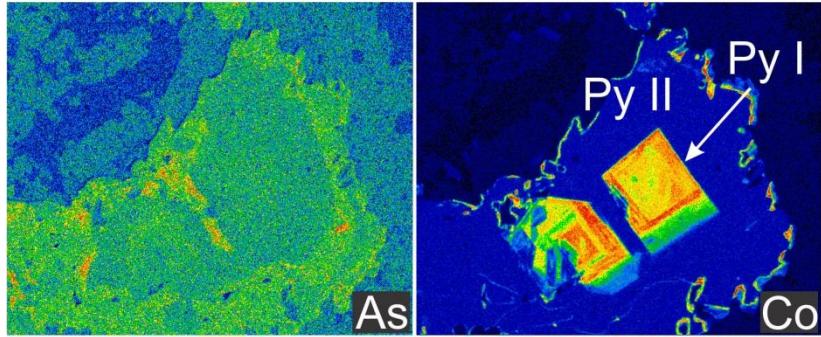
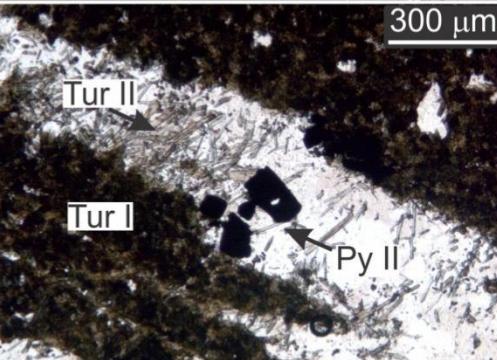
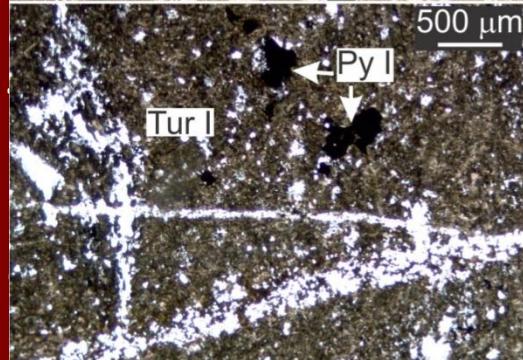
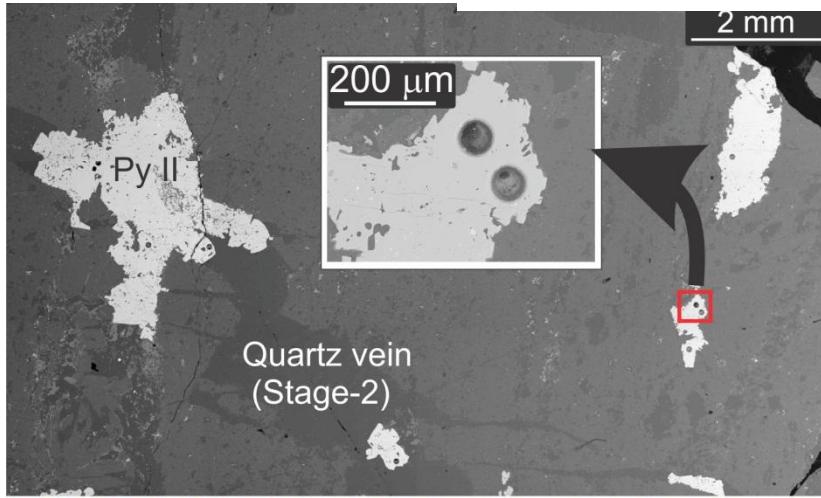
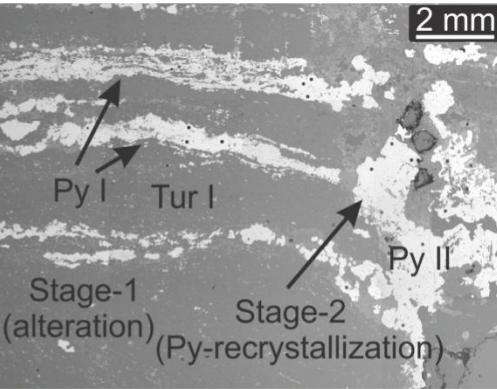
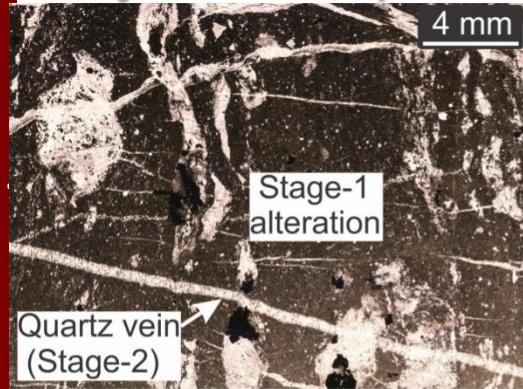


(Deditius et al., 2009, *Geology*) as a result of fault-valve actions

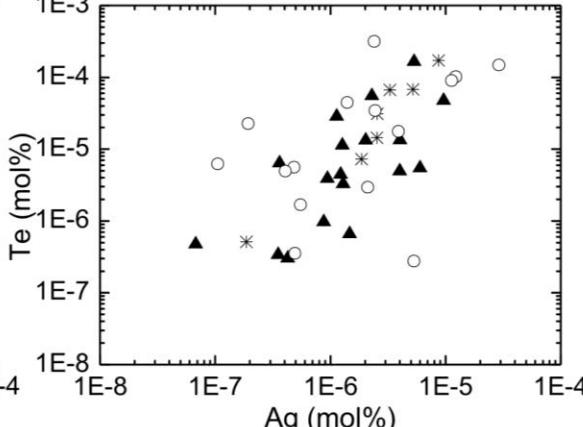
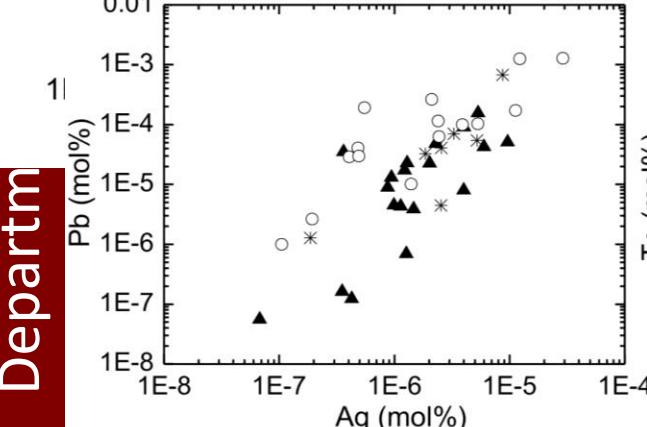
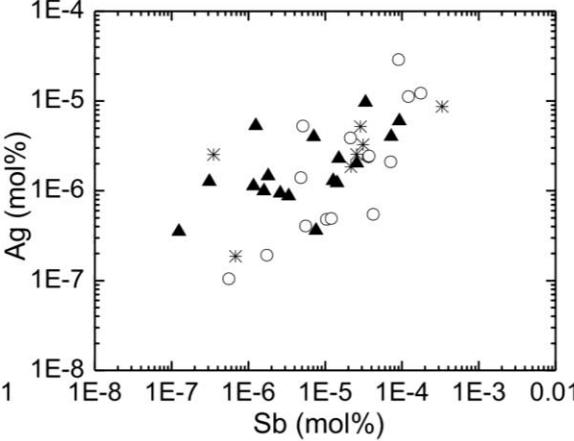
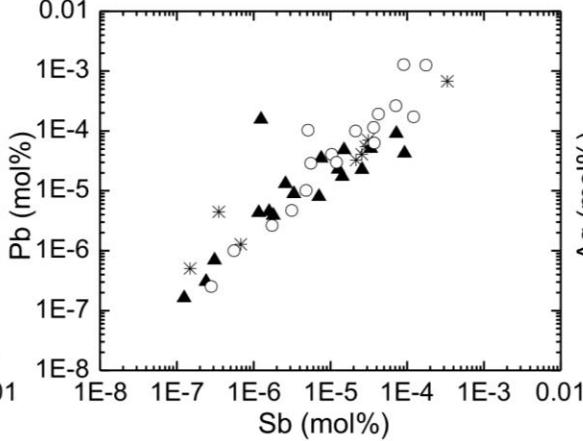
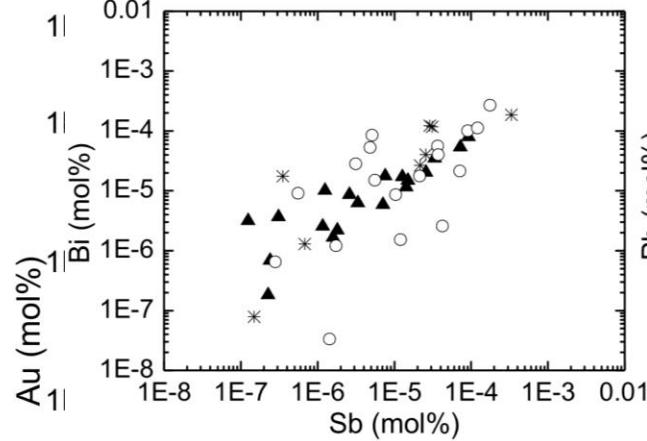
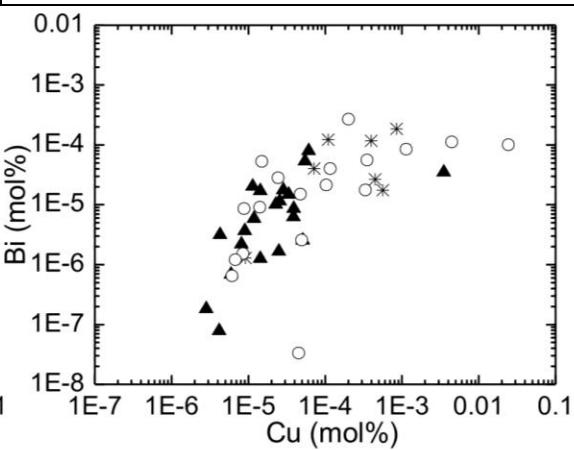
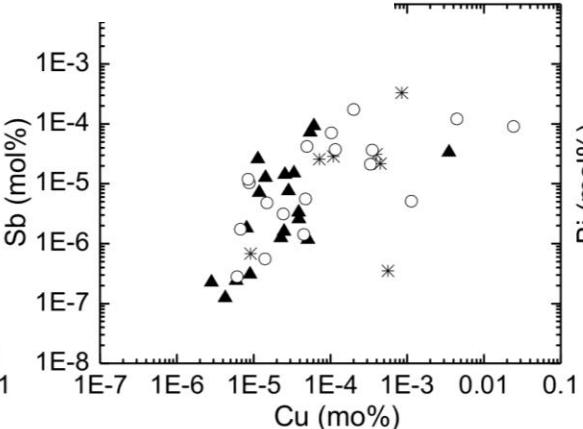
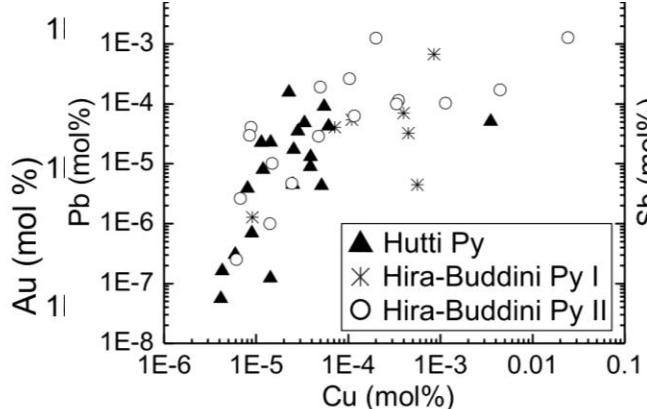
- Phase separation to $\text{NaCl}+\text{H}_2\text{O}$ and $\text{CO}_2+\text{CH}_4 \rightarrow$ fluctuation of fluid salinity \rightarrow Oscillatory cobalt zoning (Co transport salinity dependent, Liu et al., 2011, *GCA*)

Pyrite compositions

Hira-Buddini



Pyrite compositions (LA-ICP-MS)

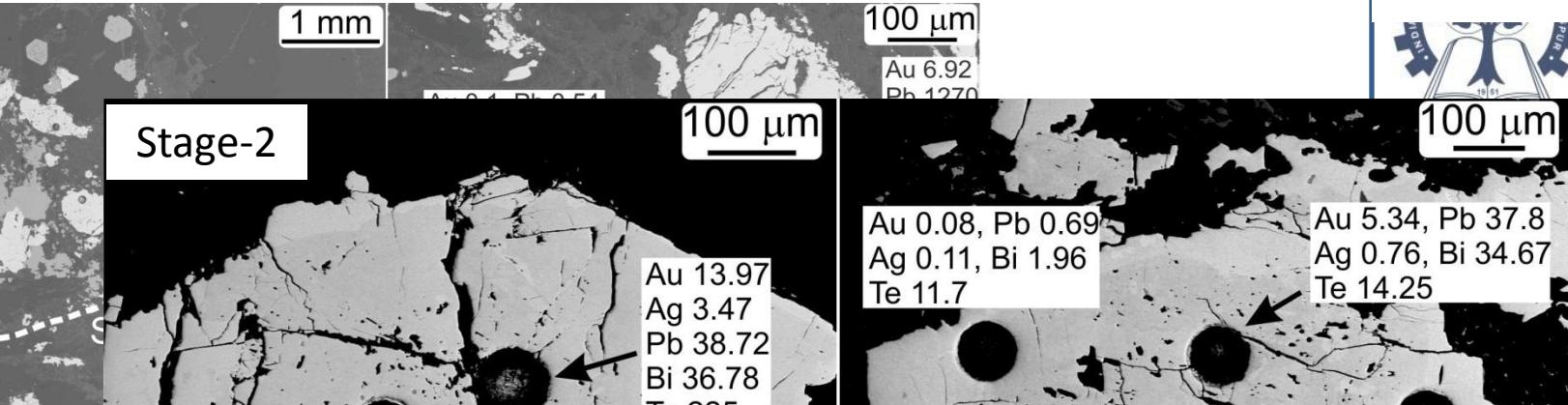


Chalcophile elements
show similar
geochemical behavior
in pyrite

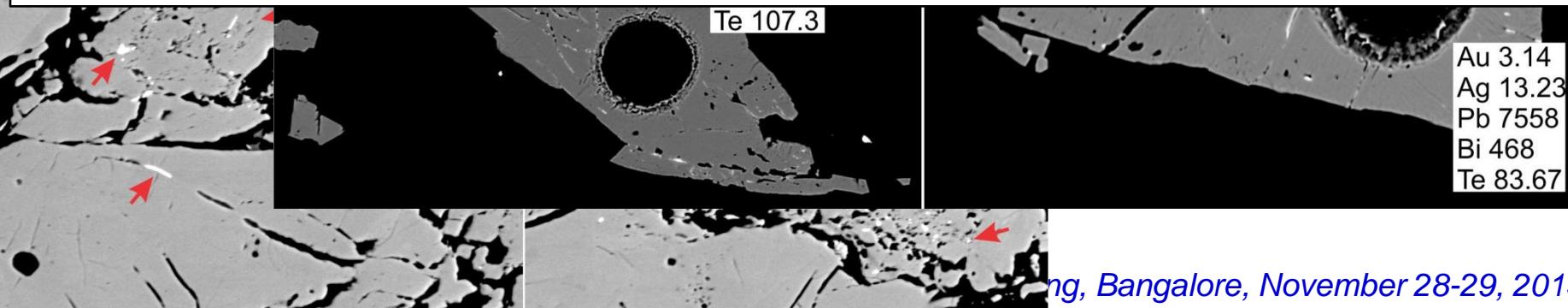
Arsenopyrite compositions (LA-ICP-MS)

Hutti

Stage-1



- High Au, Ag, Pb, Bi and Te contents along fractures
- No alteration in Apy adjacent to fractures
- 1. Precipitation of Au-Ag-Pb-Bi-Te micro-particles from ore fluid
- 2. Gold coupled with chalcophile elements in orogenic gold deposits
- 3. No role of As for Au incorporation into sulfides!



Synthesis

Magmatic fluid source model



- Granites are 50 Ma older than ore (Wallaby, Cleo-Sunrise, Kalgoorlie) and ~60 Ma younger than the ore (Ballarat and Bendigo) in Yilgarn Craton (WA)?
Ages of metamorphism of 2564 ± 12 Ma in HMGB and
- ~~Explanation of low salinity $H_2O-NaCl-CO_2-CH_4-Au(Ms)_2$ fluids~~ granodiorite; Rogers et al., 2007) are indistinguishable
- ~~Without errors making a geochronology-based resolution~~ of the existing debate on metamorphic vs. magmatic fluid
- ~~Source impossible?~~

Metamorphic fluid source model (Goldfarb & Groves, 2015, *Lithos*)

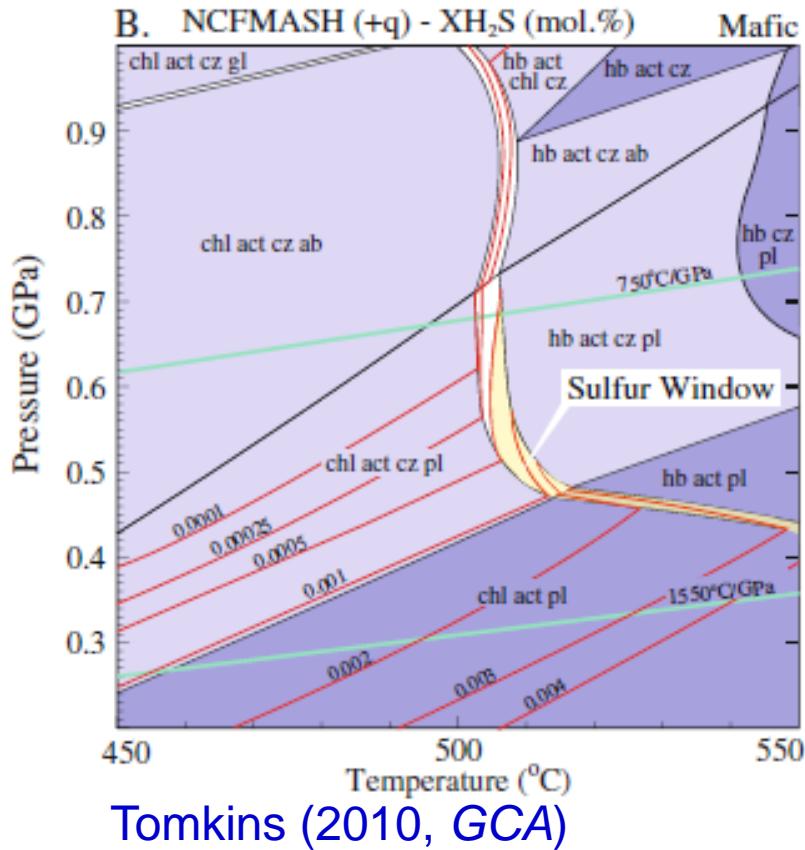
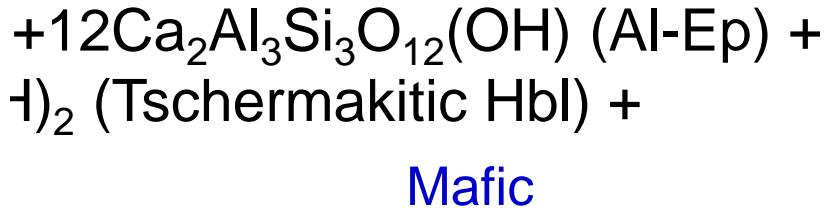
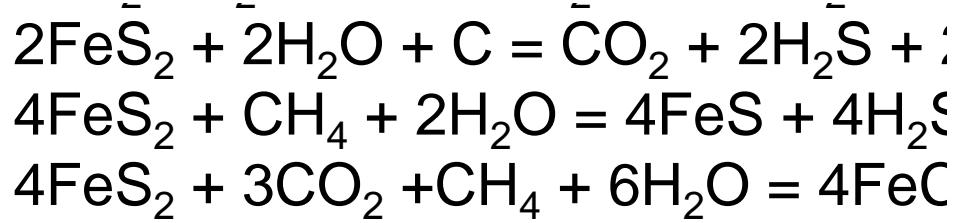
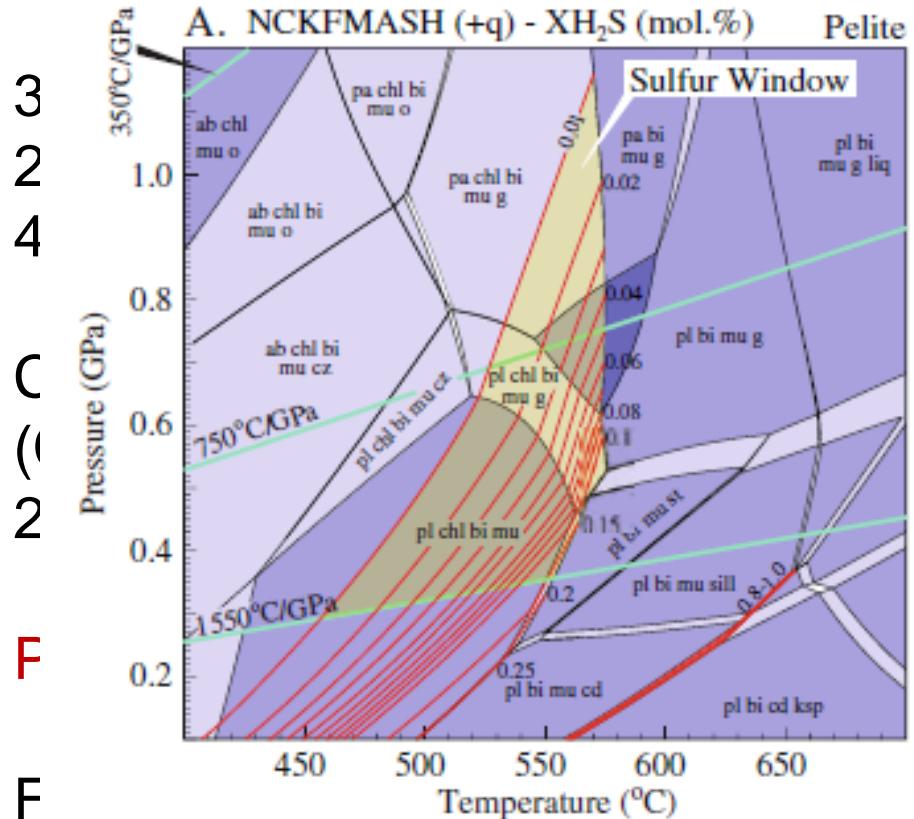
- Mostly in greenschist facies terrain (Yilgarn & Abitibi)
- Unlike amphibolite facies in Kolar, Hutti, Jonnagiri?

S-windows during metamorphism

Chl-breakdown reactions in mafic rocks



Pelitic



Accepted model with exploration implications



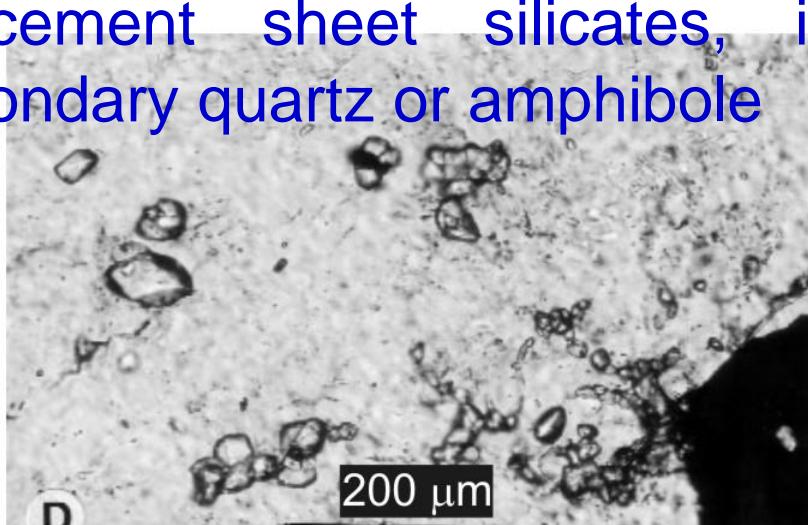
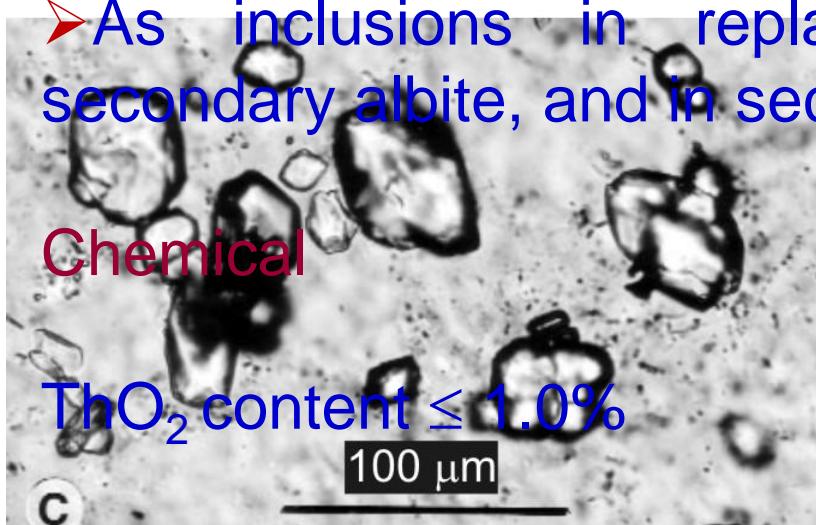
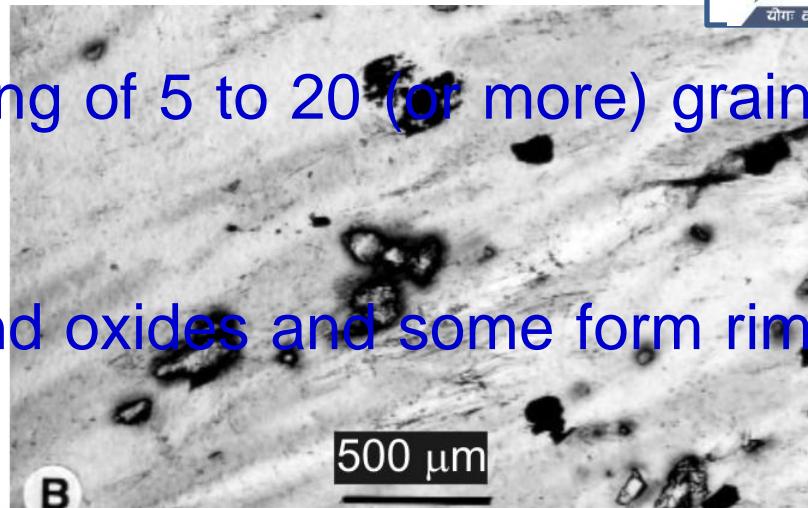
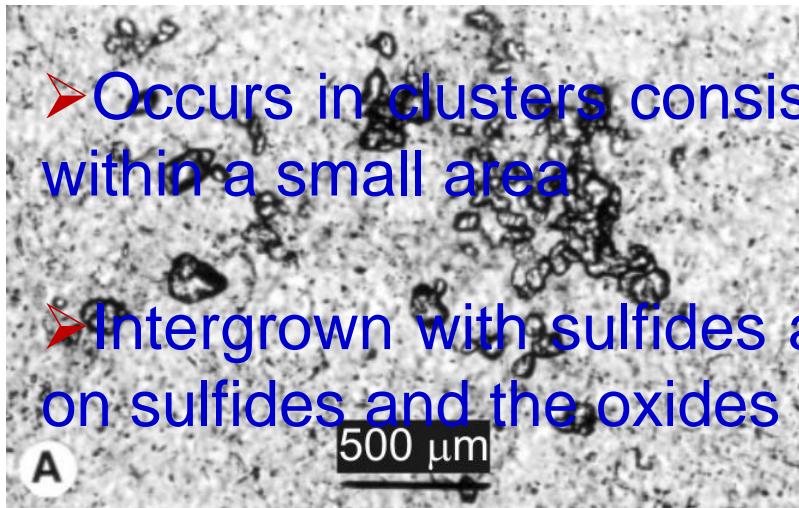
- Metamorphic devolatilization of greenstone pile
- Maximum H₂O released at greenschist-amphibolite facies boundary by terminal breakdown of chlorite at 550–600 °C/ 5 kbar)
- Explains observed low salinity H₂O-NaCl-CO₂±CH₄-Au(HS)₂⁻ fluid
- Coupled with pyrite desulfidation generating S-rich fluid ← Au(HS)₂⁻ complexing
- Pyrite acts as source of S and Au
- Gold-only deposits are expected to occur in the greenschist facies greenstones



Scopes for future research

- Timing of mineralization/hydrothermal alteration
- Rb-Sr in biotite and whole rock from hydrothermal zones (TIMS)
 - Sm-Nd in scheelite (TIMS/ HR-MC-ICP-MS)
 - Re-Os in sulfides (**complicated chemistry**)
 - Fluid source characterization by $^{87}\text{Sr}/^{86}\text{Sr}$ and $\delta^{11}\text{B}$ in tourmaline (HR-MC-ICP-MS)
 - Thermodynamic evaluation of fluid CO_2 by construction of isobaric $T\text{-}X_{\text{CO}_2}$ psedosections of the hydrothermally altered host rock (Elmer et al, 2007, *Econ Geol*)

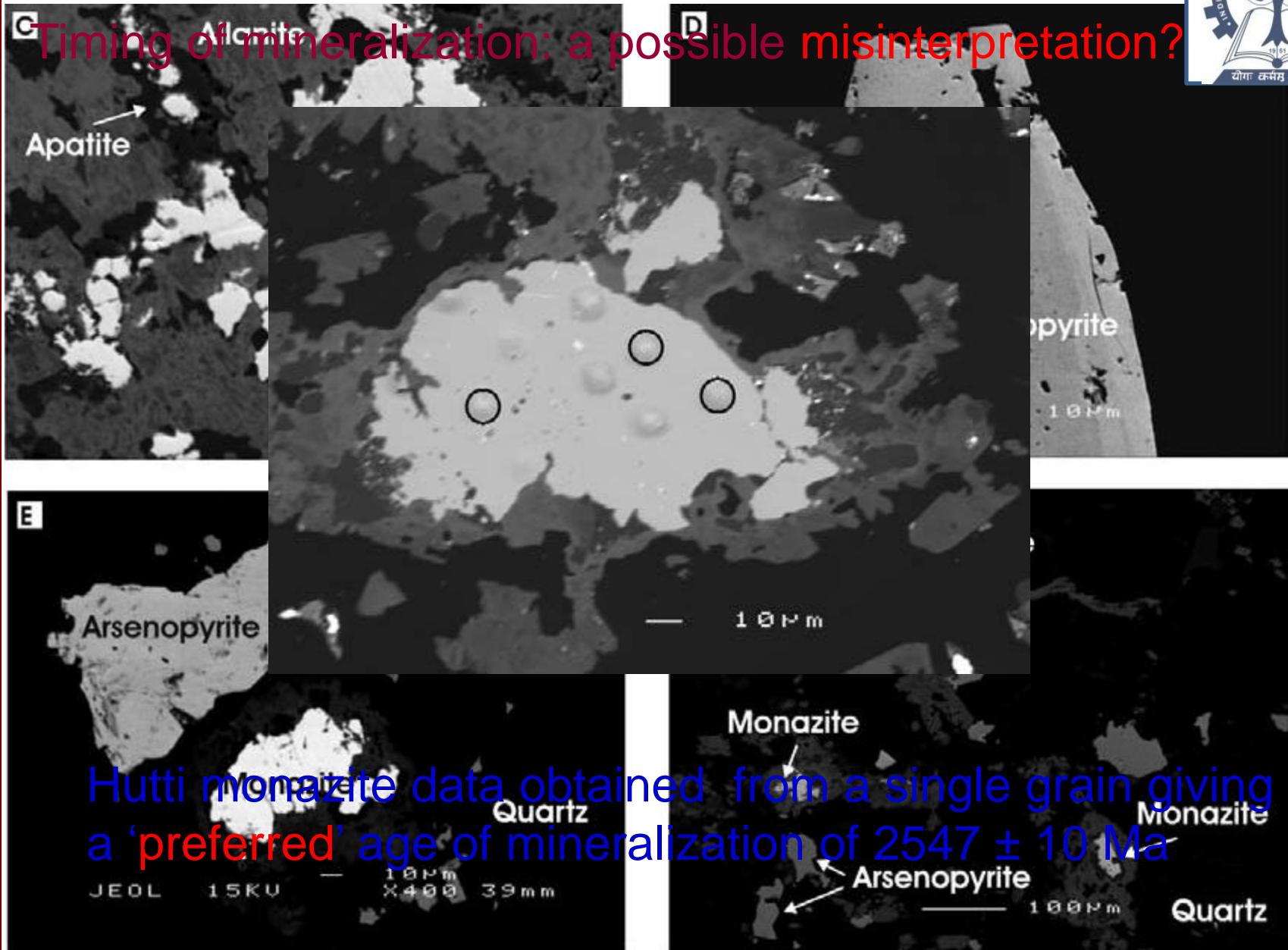
Hydrothermal monazites: textural and chemical criteria



(Schandl & Gorton, 2004, *Econ Geol*)

But there are monazite clusters as well!

Timing of mineralization: a possible misinterpretation?



Hutti monazite data obtained from a single grain giving a ‘preferred’ age of mineralization of 2547 ± 10 Ma

Acknowledgements

- DST (3 (2008) to JEOL JSM
 - IIT Kgp f
 - Rupam a
 - Saptarsh
 - D. Upad
 - My former students Nabarun (Pal), Ranjan (Roy), Amit (Basu Sarbadhikari), Sakthi (Salavanan Chinnasamy), and presently Pranjit (Hazarika) who taught me to think more about the gold ores
- Its only words
and words all that I have*

