

ASSIGNMENT-1

PHYSICS-II

SHUBHAM GARG

9919103057

BATCH: R2

Q.1: Determine E_{avg} , E_{rms} and E_{mp} by using molecular energy distribution with energies between E and $(E+dE)$ in a sample of an ideal gas that contains N molecules and whose absolute temperature is T .

Solution: we know that -

$$n(E)dE = g(E)dE f(E) \\ = g(E) A e^{-E/kT} dE$$

$$\therefore \boxed{n(E)dE = \frac{2\pi N}{(\pi kT)^{3/2}} \sqrt{E} e^{-E/kT} dE}$$

Average Energy \rightarrow

$$\begin{aligned} \langle E \rangle &= \frac{\text{Total Energy}}{\text{Total no. of Particles}} \\ &= \frac{\int_0^{\infty} E n(E) dE}{N} \\ &= \frac{\int_0^{\infty} \frac{2\pi N \sqrt{E} e^{-E/kT}}{(\pi kT)^{3/2}} dE}{N} \\ &= \frac{2\pi}{(\pi kT)^{3/2}} \int_0^{\infty} E^{3/2} e^{-E/kT} dE \end{aligned}$$

By using Gamma's function \rightarrow

$$\boxed{\int_0^{\infty} x^n e^{-\alpha x} dx = \frac{n!}{\alpha^{n+1}}}$$

$$\therefore \int_0^{\infty} n^{3/2} e^{-\alpha n} dn = \frac{3}{4\alpha^2} \sqrt{\pi/\alpha}$$

$$\Rightarrow \bar{\epsilon} = \frac{2\pi}{(\pi kT)^{3/2}} \times \frac{3}{4\left(\frac{1}{kT}\right)^2} \sqrt{\pi kT}$$

$$* \boxed{(\bar{\epsilon}) = \frac{3}{2} kT}$$

Rms value of Energy -

$$\begin{aligned} \sqrt{(\bar{\epsilon})^2} &= \sqrt{\frac{\int_0^{\infty} \epsilon^2 n(\epsilon) d\epsilon}{N}} \\ &= \sqrt{\frac{\int_0^{\infty} \frac{2\pi \epsilon^2 \sqrt{\epsilon} \cdot e^{-\epsilon/kT}}{(\pi kT)^{3/2}} d\epsilon}{N}} \\ &= \sqrt{\frac{\int_0^{\infty} \frac{2\pi \epsilon^{5/2} e^{-\epsilon/kT}}{(\pi kT)^{3/2}} d\epsilon}{N}} \end{aligned}$$

Now using Gamma's formula.

$$\int_0^{\infty} n^{\alpha} e^{-\alpha n} dn = \frac{n!}{\alpha^{n+1}}$$

$$\sqrt{(\bar{\epsilon})^2} = \sqrt{\frac{15(kT)^2}{4}}$$

$$* \boxed{\sqrt{(\bar{\epsilon})^2} = \sqrt{15} \frac{kT}{2}}$$

Most Probable Energy \rightarrow

for obtaining the expression for most probable energy

$$\text{put } \frac{dn(\epsilon)}{d\epsilon} = 0$$

Hence, we get \rightarrow

$$\frac{2\pi N}{(\pi kT)^{3/2}} \left[\frac{1}{2} \epsilon^{-1/2} e^{-\epsilon/kT} + \epsilon^{1/2} \left(\frac{-1}{kT} \right) e^{-\epsilon/kT} \right] = 0.$$

$$\Rightarrow \epsilon^{1/2} e^{-\epsilon/kT} \left[\frac{1}{2\epsilon} - \frac{1}{kT} \right] = 0.$$

$$\therefore \frac{\epsilon}{kT} = \frac{1}{2}$$

$$\Rightarrow * \boxed{\epsilon = \frac{1}{2} kT}$$

Q.2: Write down the number of particles with velocities v and $v+dv$ from molecular energy distribution and calculate ratio between E_{avg} , E_{rms} , E_{mp} .

Solution- We know that $n(v)dv = 4\pi N \left(\frac{m}{2\pi kT} \right)^{3/2} v^2 e^{-\frac{mv^2}{2kT}} dv$

$$\text{Since } E_{avg} = \frac{3}{2} kT, E_{rms} = \frac{\sqrt{15}}{2} kT, E_{mp} = \frac{1}{2} kT$$

$$\therefore E_{avg} : E_{rms} : E_{mp} = \frac{3}{2} kT : \frac{\sqrt{15}}{2} kT : \frac{1}{2} kT$$
$$= \boxed{3 : \sqrt{15} : 1} *$$

$$\text{Also, } n(\epsilon)d\epsilon = \frac{2\pi N}{(\pi kT)^{3/2}} \sqrt{\epsilon} e^{-\epsilon/kT} d\epsilon$$

$$\text{Now } \epsilon = \frac{1}{2} mv^2$$

$$\therefore dE = mv dv$$

$$\therefore n(v) dv = \frac{21TN}{(1\pi kT)^{3/2}} \sqrt{\frac{1}{2}mv^2} e^{-\frac{1}{2} \frac{mv^2}{kT}} mv dv$$

$$* \boxed{n(v) dv = \frac{41TN}{(2\pi kT)^{3/2}} m^{3/2} v^2 e^{-\frac{1}{2} \frac{mv^2}{kT}} dv}$$

Q.3: Prove average energy of a free electron gas at $T=0$ is $3/5$ of Fermi energy.

Solution-

we know, in Fermi Dirac Energy Distribution \rightarrow

$$n(E) dE = g(E) f_{FD}(E) dE$$

$$\text{Also, } g(E) = \frac{3N}{2} E_F^{-3/2} E^{1/2}$$

$$n(E) dE = \frac{3N}{2} \frac{E_F^{-3/2} E^{1/2}}{e^{(E-E_F)/kT} + 1} dE$$

$$\text{Average Energy, } \bar{E} = \frac{\int_0^\infty E n(E) dE}{\int_0^\infty n(E) dE}$$

$$\bar{E}_0 = 3 E_F^{-3/2} \int_0^{E_F} \frac{E^{3/2} N}{e^{(E-E_F)/kT} + 1} dE$$

$$\text{at } T=0 : e^{(E-E_F)/kT} = e^{-\infty} = 0$$

$$\text{Hence } * \boxed{\bar{E}_0 = \frac{3}{5} E_F}$$

Thus, average energy of a free electron gas at $T=0$ is $3/5$ of Fermi Energy.

NOTE: * marked are the results.